

## Table 6.1 – Odor Threshold Values

The table contains the following information:

Chemical Name, CAS Number, Chemical Formula, Chemical Molecular weight”

- Range of Referenced Values
- Odor Character Description(s)
- ACGIH® Threshold Limit Value (TLV)®
- OSHA Permissible Exposure Limit (PEL)
- AIHA® WEEL® Value

Abbreviations/Definitions used in table:

- Alliacious – Resembling garlic or onion in smell or taste
- BEI – Biological Exposure Indices
- DSEN – May cause dermal sensitization
- Empyreumatic – Being or having an odor of burnt organic material as a result of decomposition
- Etherous / Ethereal – Resembling or pertaining to ether
- Fusel – Hot acrid oily liquid
- H – Aerosol only
- IFV – Measured as Inhalable fraction and vapor
- L – Exposure to carcinogens should be kept to a minimum
- Q – Absorbed rapidly through the skin in molten/heated liquid form in amounts that have caused rapid death in humans
- SEN – Sensitization
- Skin – Potential exposure by the cutaneous route
- (W) – Worker exposure by all routes should be minimized to the fullest extent possible

Table 6.1 – Odor Threshold Values

#	Compound Name CAS Number Formula Molecular Weight	Range of Odor Values (ppm)	Odor Character	ACGIH TLV® (ppm)	OSHA PEL (ppm)	AIHA WEEL® (ppm)
1	<b>Acetaldehyde</b> 75-07-0 C <sub>2</sub> H <sub>4</sub> O 44.05	0.0015 – 1,000	pungent, fruity, suffocating, fresh, green	C = 25	TWA = 200	–
2	<b>Acetic Acid</b> 64-19-7 C <sub>2</sub> H <sub>4</sub> O <sub>2</sub> 60.05	0.0004 – 204	pungent, vinegar	STEL = 15 TWA = 10	TWA = 10	–
3	<b>Acetic Anhydride</b> 108-24-7 C <sub>4</sub> H <sub>6</sub> O <sub>3</sub> 102.09	0.12 – 0.36	sour, acid	TWA = 1 C = 3	TWA = 5	–
4	<b>Acetone</b> 67-64-1 C <sub>3</sub> H <sub>6</sub> O 58.08	0.40 – 11,745	sweet, fruity, ethereal	TWA = 500 STEL = 750 BEI	TWA = 1,000	–
5	<b>Acetonitrile</b> 75-05-8 C <sub>2</sub> H <sub>3</sub> N 41.05	13 – 1,161	etherish	TWA = 20 Skin	TWA = 40	–
6	<b>Acetophenone</b> 98-86-2 C <sub>8</sub> H <sub>8</sub> O 120.15	0.00024 – 0.59	sweet, almond, pungent, oranges, river water	TWA = 10	–	TWA = 10
7	<b>Acetylene</b> 74-86-2 C <sub>2</sub> H <sub>2</sub> 26.02	226 – 2584	gassy, garlic	Simple Asphyxiant	–	–
8	<b>Acrolein</b> 107-02-8 C <sub>3</sub> H <sub>4</sub> O 56.06	0.0036 – 1.8	pungent	C = 0.1 Skin	TWA = 0.1	–
9	<b>Acrylic Acid</b> 79-10-7 C <sub>3</sub> H <sub>4</sub> O <sub>2</sub> 72.06	0.092 – 1.0	rancid, plastic, sweet	TWA = 2 Skin	–	–
10	<b>Acrylonitrile</b> 107-13-1 C <sub>3</sub> H <sub>3.5</sub> N 53.06	1.6 – 22	onion, garlic	TWA = 2 Skin	TWA = 2 Skin	–
11	<b>Allyl Alcohol</b> 107-18-6 C <sub>3</sub> H <sub>6</sub> O 58.08	0.51 – 35	mustard	TWA = 0.5 Skin	TWA = 2 Skin	–
12	<b>Allyl Chloride</b> 107-05-1 C <sub>3</sub> H <sub>5</sub> Cl 76.53	0.48 – 5.9	pungent	TWA = 1 STEL = 2 Skin	TWA = 1	–
13	<b>Allyl Isothiocyanate</b> 57-06-7 C <sub>4</sub> H <sub>7</sub> NS 99.15	0.0091 – 1.97	irritating	–	–	STEL = 1 Skin DSEN

Table 6.1 – Odor Threshold Values, cont.

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14	<b>Ammonia</b> 7664-41-7 NH <sub>3</sub> 17.03	0.043 – 60.3	pungent, irritating	TWA = 25 STEL = 35	TWA = 50	–
15	<b>n-Amyl Acetate</b> 628-63-7 C <sub>7</sub> H <sub>14</sub> O <sub>2</sub> 130.18	0.007 – 43	banana, ethereal	TWA = 50 STEL = 100	TWA = 100	–
16	<b>Aniline</b> 62-53-3 C <sub>6</sub> H <sub>5</sub> N 93.12	0.012 – 10	pungent, oily, empyreumatic	TWA = 2 Skin BEI	TWA = 5 Skin	–
17	<b>Arsine</b> 7784-42-1 AsH <sub>3</sub> 77.93	<1.0	garlic	TWA = 0.005	TWA = 0.05	–
18	<b>Benzaldehyde</b> 100-52-7 C <sub>7</sub> H <sub>6</sub> O 106.12	0.0015 – 783	bitter almond, fruit, vanilla	–	–	TWA = 2 DSEN
19	<b>Benzene</b> 71-43-2 C <sub>6</sub> H <sub>6</sub> 78.11	0.47 – 313	aromatic, sweet, solvent, empyreumatic	TWA = 0.5 STEL = 2.5 Skin, BEI	TWA = 1	–
20	<b>Benzoyl Chloride</b> 98-88-4 C <sub>7</sub> H <sub>5</sub> ClO 140.56	0.0021 – 0.0063	pungent	C = 0.5	–	C = 5 Skin, DSEN
21	<b>Benzyl Acetate</b> 140-11-4 C <sub>9</sub> H <sub>10</sub> O <sub>2</sub> 150.17	0.00016 – 22	pears, plastic, ethereal, anise	TWA = 10	–	–
22	<b>Benzyl Chloride</b> 100-44-7 C <sub>7</sub> H <sub>7</sub> Cl 126.58	0.041 – 0.046	pungent	TWA = 1	TWA = 1	–
23	<b>Biphenyl</b> 92-52-4 C <sub>12</sub> H <sub>10</sub> 154.2	0.00052 – 0.0095	pleasant, butter-like	TWA = 0.2	TWA = 0.2	–
24	<b>Boron Trifluoride</b> 7637-07-2 BF <sub>3</sub> 67.82	1.5	pungent	C = 1	C = 1	–
25	<b>Bromine</b> 7726-95-6 Br <sub>2</sub> 159.83	<0.0099 – 0.99	alliaceous, sharp, irritating	TWA = 0.1 STEL = 0.2	TWA = 0.1	–
26	<b>Bromoform</b> 75-25-2 CHBr <sub>3</sub> 252.77	0.19 – 15	chloroform, sweet, suffocating	TWA = 0.5	TWA = 0.5 Skin	–

**Table 6.1 – Odor Threshold Values, cont.**

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27	<b>1,3-Butadiene</b> 106-99-0 C <sub>4</sub> H <sub>6</sub> 54.09	0.099 – 76	aromatic, rubber	TWA = 2	TWA = 1 STEL = 5	–
28	<b>Butane, all isomers</b> 106-97-8, 75-28-5 C <sub>4</sub> H <sub>10</sub> 58.12	0.421 – 5,048	natural gas	STEL = 1,000	–	–
29	<b>Butenes, all isomers</b> 106-98-9, 107-01-7, 590-18-1 624-64-6, 25167-67-3, 115-11-7 C <sub>4</sub> H <sub>8</sub> 56.11	0.362 – 2,126	petroleum	TWA = 250	–	–
30	<b>2-Butoxyethanol</b> 111-76-2 C <sub>8</sub> H <sub>18</sub> O <sub>2</sub> 118.17	0.08 – 0.35	sweet, ester, musty	TWA = 20 BEI	TWA = 50 Skin	–
31	<b>2-Butoxyethyl Acetate</b> 112-07-2 C <sub>10</sub> H <sub>18</sub> O <sub>3</sub> 160.21	0.107 – 0.99	fruity	TWA = 20	–	–
32	<b>n-Butyl Acetate</b> 123-86-4 C <sub>8</sub> H <sub>16</sub> O <sub>2</sub> 116.16	0.00013 – 368	sweet, banana	TWA = 150 STEL = 200	TWA = 150	–
33	<b>sec-Butyl Acetate</b> 105-46-4 C <sub>8</sub> H <sub>16</sub> O <sub>2</sub> 116.16	0.0025 – 4.76	fruity	TWA = 200	TWA = 200	–
34	<b>tert-Butyl Acetate</b> 540-88-5 C <sub>8</sub> H <sub>16</sub> O <sub>2</sub> 116.16	0.008 – 1.31	mild	TWA = 200	TWA = 200	–
35	<b>n-Butyl Acrylate</b> 141-32-2 C <sub>8</sub> H <sub>14</sub> O <sub>2</sub> 128.17	0.00029 – 0.101	sweet, rancid, plastic	TWA = 2 SEN	–	–
36	<b>n-Butyl Alcohol</b> 71-36-3 C <sub>4</sub> H <sub>10</sub> O 74.12	0.0033 – 990	sweet, malty, alcohol, medicinal	TWA = 20	TWA = 100	–
37	<b>sec-Butyl Alcohol</b> 78-92-2 C <sub>4</sub> H <sub>10</sub> O 74.12	0.043 – 94	sweet, malty alcohol	TWA = 100	TWA = 150	–
38	<b>tert-Butyl Alcohol</b> 75-65-0 C <sub>4</sub> H <sub>10</sub> O 74.12	3.3 – 957	sweet alcohol	TWA = 100	TWA = 100	–

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39	<b>n-Butylamine</b> 109-73-9 C <sub>4</sub> H <sub>11</sub> N 73.14	0.08 – 13.9	sour ammoniacal	C = 5 Skin	C = 5 Skin	–
40	<b>n-Butyl Lactate</b> 138-22-7 C <sub>8</sub> H <sub>16</sub> O <sub>2</sub> 146.21	0.000000049	mild	TWA = 5	–	–
41	<b>Butyl Mercaptan</b> 109-79-5 C <sub>4</sub> H <sub>10</sub> S 90.19	0.0000027 – 4.9	skunk	TWA = 0.5	TWA = 10	–
42	<b>p-tert-Butyl Toluene</b> 98-51-1 C <sub>11</sub> H <sub>16</sub> 148.24	<5.031	gasoline	TWA = 1	TWA = 10	–
43	<b>Butyraldehyde</b> 123-72-8 C <sub>4</sub> H <sub>8</sub> O 72.11	0.0003 – 5.09	pungent	–	–	TWA = 25
44	<b>Camphor, synthetic</b> 76-22-2 C <sub>10</sub> H <sub>16</sub> O 152.23	0.0026 – 7.2	camphorous	TWA = 2 STEL = 3	TWA = 0.321	–
45	<b>Caprolactam</b> 105-60-2 C <sub>7</sub> H <sub>13</sub> NO 113.16	0.065	mild	TWA 1.08 IFV	–	–
46	<b>Carbon Dioxide</b> 124-38-9 CO <sub>2</sub> 44.01	39,000 – 600,136	–	TWA = 5000 STEL = 30000	TWA = 5000	–
47	<b>Carbon Disulfide</b> 75-15-0 CS <sub>2</sub> 76.14	0.016 – 32	vegetable, sulfide, medicinal	TWA = 1 Skin BEI	TWA = 20 C = 30	–
48	<b>Carbon Tetrachloride</b> 56-23-5 CCl <sub>4</sub> 153.84	1.68 – 720	sweet, ethereal, dry cleaner, aromatic	TWA = 5 STEL = 10 Skin	TWA = 10 C = 25	–
49	<b>Carbonyl Sulfide</b> 463-58-1 COS 60.08	0.057 – 0.102	unpleasant	TWA = 5	–	–
50	<b>Chlorine</b> 7782-50-5 Cl <sub>2</sub> 70.91	0.021 – 4.9	suffocating, sharp, bleach	TWA = 0.5 STEL = 1	C = 1	–
51	<b>Chlorine Dioxide</b> 10049-04-4 ClO <sub>2</sub> 67.46	15	chlorine	TWA = 0.1 STEL = 0.3	TWA = 0.1	–

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52	<b>2-Chloroacetophenone</b> 532-27-4 C <sub>8</sub> H <sub>7</sub> ClO 154.59	0.016 - 0.111	fruity	TWA = 0.05	TWA = 0.05	–
53	<b>Chlorobenzene</b> 108-90-7 C <sub>6</sub> H <sub>5</sub> Cl 112.56	0.087 - 13	almond-like, shoe polish	TWA = 10 BEI	TWA = 75	–
54	<b>Chlorodifluoromethane</b> 75-45-6 CHClF <sub>2</sub> 86.47	200,192	ethereal	TWA = 1,000	–	–
55	<b>Chloroform</b> 67-66-3 CHCl <sub>3</sub> 119.38	0.102 – 1,413	sweet, ethereal, suffocating	TWA = 10	C = 50	–
56	<b>Chloropicrin</b> 76-06-2 CCl <sub>3</sub> NO <sub>2</sub> 164.38	1.09	chlorine	TWA = 0.1	TWA = 0.1	–
57	<b>b-Chloroprene</b> 126-99-8 C <sub>5</sub> H <sub>7</sub> Cl 88.54	0.11 – 276	rubber	TWA = 10 Skin	TWA = 25 Skin	–
58	<b>Chlorotoluene, o-isomer</b> 95-49-8 C <sub>7</sub> H <sub>7</sub> Cl 126.58	0.18 – 0.270	aromatic	TWA = 50	–	–
59	<b>Citral</b> 5392-40-5 C <sub>10</sub> H <sub>16</sub> O 152.23	0.000024 – 0.032	lemon, flowery, citrous	TWA = 5 IFV, Skin SEN	–	–
60	<b>Cresol, all isomers</b> 1319-77-3, 95-48-7 108-39-4, 106-44-5 C <sub>7</sub> H <sub>8</sub> O 108.13	0.00005 – 0.0090	creosote, phenol, irritating, smoky, empyreumatic, burnt plastic	TWA = 4.5 IFV, Skin	TWA = 5 Skin	–
61	<b>Crotonaldehyde</b> 4170-30-3, 123-73-9 C <sub>5</sub> H <sub>8</sub> O 70.09	0.02 – 0.59	pungent	C = 0.3 Skin	TWA = 2	–
62	<b>Cumene</b> 98-82-8 C <sub>9</sub> H <sub>12</sub> 120.19	0.008 – 1.3	sharp	TWA = 50	TWA = 50 Skin	–
63	<b>Cumene Hydroperoxide</b> 80-15-9 C <sub>9</sub> H <sub>12</sub> O <sub>2</sub> 152.19	0.0048	sharp, irritating	–	–	TWA = 1 Skin
64	<b>Cyanogen</b> 460-19-5 CN 26.02	>500	almonds	TWA = 10	–	–

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65	<b>Cyanogen Chloride</b> 506-77-4 CClN 61.47	0.994	acid	C = 0.3	–	–
66	<b>Cyclohexane</b> 110-82-7 C <sub>6</sub> H <sub>12</sub> 84.16	0.52 – 784	pungent	TWA = 100	TWA = 300	–
67	<b>Cyclohexanol</b> 108-93-0 C <sub>6</sub> H <sub>12</sub> O 100.16	0.058 – 0.491	camphorous	TWA = 50 Skin	TWA = 50	–
68	<b>Cyclohexanone</b> 108-94-1 C <sub>6</sub> H <sub>10</sub> O 98.14	0.052 – 219	sweet, sharp	TWA = 20 Skin	TWA = 50	–
69	<b>Cyclohexene</b> 110-83-8 C <sub>6</sub> H <sub>10</sub> 82.14	0.18	sweet	TWA = 300	TWA = 300	–
70	<b>Cyclohexylamine</b> 108-91-8 C <sub>6</sub> H <sub>13</sub> N 99.17	2.42	ammonia	TWA = 10	–	–
71	<b>Cyclopentadiene</b> 542-92-7 C <sub>5</sub> H <sub>6</sub> 66.1	1.8	terpene-like, pine, fruit	TWA = 75	TWA = 75	–
72	<b>Decaborane</b> 17702-41-9 B <sub>10</sub> H <sub>14</sub> 122.31	0.06	pungent	TWA = 0.05 STEL = 0.15 Skin	TWA = 0.05 Skin	–
73	<b>1-Decene</b> 872-05-9 C <sub>10</sub> H <sub>20</sub> 140.27	6.45	pleasant	–	–	TWA = 100
74	<b>Diacetone Alcohol</b> 123-42-2 C <sub>6</sub> H <sub>12</sub> O <sub>2</sub> 116.16	0.27 – 13	sweet	TWA = 50	TWA = 50	–
75	<b>Diacetyl</b> 431-03-8 C <sub>4</sub> H <sub>6</sub> O <sub>2</sub> 86.09	0.000002 – 2.9	pleasant, buttery	TWA = 0.01 STEL = 0.02	–	–
76	<b>Diallylamine</b> 124-02-7 C <sub>6</sub> H <sub>11</sub> N 97.16	2	disagreeable	–	–	TWA = 1 Skin
77	<b>Diborane</b> 19287-45-7 H <sub>2</sub> B <sub>2</sub> 27.69	1.8 – 3.5	repulsive	TWA = 0.1	TWA = 0.1	–

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78	<b>2,3-Dibromo-1-Chloropropane</b> 96-12-8 C <sub>3</sub> H <sub>5</sub> BrCl 236.33	0.01 – 0.031	irritating	–	TWA = 0.001	–
79	<b>Dibutylamine</b> 111-92-2 C <sub>8</sub> H <sub>17</sub> N 129.24	0.079 – 0.770	amine	–	–	C = 5 Skin
80	<b>Dibutyl Phthalate</b> 84-74-2 C <sub>18</sub> H <sub>22</sub> O <sub>4</sub> 278.34	0.023	–	TWA = 0.44	TWA = 0.44	–
81	<b>Dichloroacetic Acid</b> 79-43-6 C <sub>2</sub> H <sub>2</sub> Cl <sub>2</sub> O <sub>2</sub> 128.94	0.044	–	TWA = 0.5 Skin	–	–
82	<b>Dichlorobenzene, o- isomer</b> 95-50-1 C <sub>6</sub> H <sub>4</sub> Cl <sub>2</sub> 147.01	0.02 – 50	camphor	TWA = 25 STEL = 50	C = 50	–
83	<b>Dichlorobenzene, p- isomer</b> 106-46-7 C <sub>6</sub> H <sub>4</sub> Cl <sub>2</sub> 147.01	0.121 – 15	camphor, mothballs	TWA = 10	TWA = 75	–
84	<b>Dichlorodifluoromethane</b> 75-71-8 CCl <sub>2</sub> F <sub>2</sub> 120.91	199,790	ethereal	TWA = 1,000	TWA = 1,000	–
85	<b>1,1-Dichloroethane</b> 75-34-3 C <sub>2</sub> H <sub>4</sub> Cl <sub>2</sub> 98.97	49 – 1,359	chloroform, aromatic	TWA = 100	TWA = 100	–
86	<b>1,2-Dichloroethylene, all isomers</b> 156-60-5, 156-59-2, 540-59-0 C <sub>2</sub> H <sub>2</sub> Cl <sub>2</sub> 96.94	277	pleasant	TWA = 200	TWA = 200	–
87	<b>2,4-Dichlorophenol</b> 120-83-2 C <sub>6</sub> H <sub>4</sub> Cl <sub>2</sub> O 163	0.000041	medicinal, phenolic, leather-like, fish sauce	–	–	TWA = 1 Skin, Q
88	<b>1,3-Dichloropropene</b> 542-75-6 C <sub>3</sub> H <sub>3</sub> Cl <sub>2</sub> 110.97	<0.99	sweet, pungent	TWA = 1 Skin	–	–
89	<b>Dicyclopentadiene</b> 77-73-6 C <sub>10</sub> H <sub>12</sub> 132.21	0.00019 – 0.02	sweet, sharp	TWA = 5	–	–



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90	<b>Diethanolamine</b> 111-42-2 $C_4H_{11}NO_2$ 105.14	0.279	ammonia, amine, rotten fish	TWA = 0.2 IFV Skin	–	–
91	<b>Diethylamine</b> 109-89-7 $C_4H_{11}N$ 73.14	0.0033 – 14.3	musty, fishy, amine	TWA = 5 STEL = 15	TWA = 25	–
92	<b>2-Diethylaminoethanol</b> 100-37-8 $C_8H_{17}ON$ 117.19	0.01 – 0.25	amine, sharp, ammoniacal	TWA = 2 Skin	TWA = 10 Skin	–
93	<b>Diethylbenzenes, mixed isomers</b> 25340-17-4, 135-01-3 105-05-5, 141-93-5 $C_{10}H_{14}$ 134.22	0.00038 – 0.071	–	–	–	TWA = 5
94	<b>Diethyl Ketone</b> 96-22-0 $C_8H_{16}O$ 86.13	0.85 – 14	acetone, fingernail polish remover	TWA = 200 STEL = 300	–	–
95	<b>Diethyl Phthalate</b> 84-66-2 $C_{12}H_{14}O_4$ 222.24	0.036 – 0.363	–	TWA = 0.55	–	–
96	<b>Diisobutyl Ketone</b> 108-83-8 $C_8H_{16}O$ 142.24	<0.103 – 1.6	peppermint	TWA = 25	TWA = 50	–
97	<b>Diisopropylamine</b> 108-18-9 $C_6H_{13}N$ 101.19	0.014 – 4.2	amine, fishy	TWA = 5 Skin	TWA = 5 Skin	–
98	<b>N,N-Dimethylacetamide</b> 127-19-5 $C_4H_9NO$ 87.12	48	faint, ammonia	TWA = 10 Skin BEI	TWA = 10 Skin	–
99	<b>Dimethylamine</b> 124-40-3 $C_2H_7N$ 45.08	0.00076 – 4.2	ammoniacal, rotten fish	TWA = 5 STEL = 15	TWA = 10	TWA = 1
100	<b>Dimethylaniline</b> 121-69-7 $C_8H_9N$ 121.18	0.001 – 0.2	oily	TWA = 5 STEL = 10 Skin BEI	TWA = 5 Skin	–
101	<b>Dimethyl Disulfide</b> 624-92-0 $C_2H_6S_2$ 94.2	0.00029 – 1.45	garlic, putrid, asparagus	TWA = 0.5 Skin	–	–
102	<b>Dimethyl Ether</b> 115-10-6 $C_2H_6O$ 46.07	161 – 228	ethereal	–	–	TWA = 1,000

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103	<b>Dimethyl Formamide</b> 68-12-2 $C_2H_5ON$ 73.09	0.047 – 100	fishy	TWA = 10 Skin BEI	TWA = 10 Skin	–
104	<b>1,1-Dimethylhydrazine</b> 57-14-7 $C_2H_6N_2$ 60.1	<0.31 – 14	fishy	TWA = 0.01 Skin	TWA = 0.5 Skin	–
105	<b>Dimethyl Sulfide</b> 75-18-3 $C_2H_6S$ 62.13	0.00012 – 8.11	disagreeable, asparagus, putrid	TWA = 10	–	–
106	<b>4,6-Dinitro-o-Cresol</b> 534-52-1 $C_7H_5N_2O_5$ 198.13	0.00049 - 0.00259	–	TWA = 0.025 Skin	TWA = 0.025 Skin	–
107	<b>1,4-Dioxane</b> 123-91-1 $C_4H_8O_2$ 88.1	0.8 – 2609	sweet, alcohol	TWA = 20 Skin	TWA = 100 Skin	–
108	<b>1,3-Dioxolane</b> 646-06-0 $C_3H_4O_2$ 74.08	16.8 – 63.4	–	TWA = 20	–	–
109	<b>Diphenylamine</b> 122-39-4 $C_{12}H_{11}N$ 169.22	0.022 – 0.188	floral	TWA = 1.44	–	–
110	<b>Dodecyl Mercaptan</b> 112-55-0 $C_{12}H_{26}S$ 202.4	0.00000011 – 0.000097	skunk	TWA = 0.1 SEN	–	–
111	<b>Epichlorohydrin</b> 106-89-8 $C_2H_5ClO$ 92.53	0.08 – 12	chloroform	TWA = 0.5 Skin	TWA = 5 Skin	–
112	<b>Ethane</b> 74-84-0 $C_2H_6$ 30.07	20,328 – 730,973	–	TWA = 1000	–	–
113	<b>Ethanolamine</b> 141-43-5 $C_2H_7NO$ 61.08	2.6 – 24	ammonia	TWA = 3 STEL = 6	TWA = 3	–
114	<b>2-Ethoxyethanol</b> 110-80-5 $C_4H_{10}O_2$ 90.12	0.3 – 49	sweet, musty	TWA = 5 Skin BEI	TWA = 200 Skin	–
115	<b>2-(2-Ethoxyethoxy) ethanol</b> 111-90-0 $C_6H_{14}O_3$ 134.17	<0.219 – 1.09	mild, pleasant	–	–	TWA = 25

Table 6.1 – Odor Threshold Values, cont.

#	Compound Name CAS Number Formula Molecular Weight	Range of Odor Values (ppm)	Odor Character	ACGIH TLV® (ppm)	OSHA PEL (ppm)	AIHA WEEL® (ppm)
116	<b>2-Ethoxyethyl Acetate</b> 111-15-9 C <sub>8</sub> H <sub>14</sub> O <sub>2</sub> 132.16	0.048 – 0.13	sweet, ester	TWA = 5 Skin BEI	TWA = 100 Skin	–
117	<b>Ethyl Acetate</b> 141-78-6 C <sub>4</sub> H <sub>8</sub> O <sub>2</sub> 88.1	0.09 – 190	fruity, sweet, fingernail polish, ethereal	TWA = 400	TWA = 400	–
118	<b>Ethyl Acrylate</b> 140-88-5 C <sub>7</sub> H <sub>10</sub> O <sub>2</sub> 100.11	0.0000066 – 0.0032	sweet, ester, plastic, alcohol, sharp, ammoniacal	TWA = 5 STEL = 15	TWA = 25 Skin	–
119	<b>Ethyl Alcohol</b> 64-17-5 C <sub>2</sub> H <sub>6</sub> O 46.07	0.09 – 40334	vinous, alcohol	STEL = 1000	TWA = 1000	–
120	<b>Ethylamine</b> 75-04-7 C <sub>2</sub> H <sub>7</sub> N 45.08	0.027 – 3.5	ammonia	TWA = 5 STEL = 15 Skin	TWA = 10	–
121	<b>Ethyl Amyl Ketone</b> 541-85-5 C <sub>11</sub> H <sub>20</sub> O 128.21	5.9	solvent, sharp	TWA = 10	TWA = 25	–
122	<b>Ethyl Benzene</b> 100-41-4 C <sub>8</sub> H <sub>10</sub> 106.16	<0.002 – 18	oily, solvent	TWA = 20 STEL = 125 BEI	TWA = 100	–
123	<b>Ethyl Bromide</b> 74-96-4 C <sub>2</sub> H <sub>5</sub> Br 108.97	2.7 – 3.6	ethereal	TWA = 5 Skin	TWA = 200	–
124	<b>Ethyl Chloride</b> 75-00-3 C <sub>2</sub> H <sub>5</sub> Cl 64.51	3.8 – 379	pungent	TWA = 100 Skin	TWA = 1000	–
125	<b>Ethylene</b> 74-85-1 C <sub>2</sub> H <sub>4</sub> 28.05	17 – 1029	grassy	TWA = 200	–	–
126	<b>Ethylene Chlorohydrin</b> 107-07-3 C <sub>2</sub> H <sub>4</sub> ClO 80.51	0.36	ethereal	C = 1 Skin	TWA = 5 Skin	–
127	<b>Ethylenediamine</b> 107-15-3 C <sub>2</sub> H <sub>6</sub> N <sub>2</sub> 60.1	1.3 – 4.5	ammonia	TWA = 10 Skin	TWA = 10	–
128	<b>Ethylene Dibromide</b> 106-93-4 C <sub>2</sub> H <sub>2</sub> Br <sub>2</sub> 187.86	<10	sweet	Skin	TWA = 20 C = 30	–

Table 6.1 – Odor Threshold Values, cont.

#	Compound Name CAS Number Formula Molecular Weight	Range of Odor Values (ppm)	Odor Character	ACGIH TLV® (ppm)	OSHA PEL (ppm)	AIHA WEEL® (ppm)
129	<b>Ethylene Dichloride</b> 107-06-2 $C_2H_2Cl_2$ 98.96	4.3 – 988	sweet	TWA = 10	TWA = 50 C = 100	–
130	<b>Ethylene Glycol</b> 107-21-1 $C_2H_6O_2$ 62.07	5.12	–	C = 39 H	–	–
131	<b>Ethyleneimine</b> 151-56-4 $C_2H_5N$ 43.07	0.71 – 2	ammonia	TWA = 0.05 STEL = 0.1 Skin	1910.1003 carcinogen	–
132	<b>Ethylene Oxide</b> 75-21-8 $C_2H_4O$ 44.65	0.82 – 690	sweet, olefinic	TWA = 1	TWA = 1 STEL = 5	–
133	<b>Ethyl Ether</b> 60-29-7 $C_4H_{10}O$ 74.12	0.165 – 1,924	anesthetic, ethereal	TWA = 400 STEL = 500	TWA = 400	–
134	<b>Ethyl Formate</b> 109-94-4 $C_4H_8O_2$ 74.08	2.7 – 30	aromatic	STEL = 100	TWA = 100	–
135	<b>Ethylidene Norbornene</b> 16219-75-3 $C_{12}H_{12}$ 120.19	0.007 – 0.08	turpentine	C = 5	–	–
136	<b>Ethyl Mercaptan</b> 75-08-1 $C_2H_5S$ 62.13	0.0000087 – 18	rotten cabbage	TWA = 0.5	C = 10	–
137	<b>N-Ethylmorpholine</b> 100-74-3 $C_6H_{13}NO$ 115.18	0.085 – 0.25	ammonia	TWA = 5 Skin	TWA = 20 Skin	–
138	<b>Ethyl Silicate</b> 78-10-4 $C_2H_6SiO_4$ 208.3	3.6 – 85	sweet, alcohol	TWA = 10	TWA = 100	–
139	<b>Fluorine</b> 7782-41-4 $F_2$ 37.997	0.097 – 0.19	pungent	TWA = 1 STEL = 2	TWA = 0.1	–
140	<b>Formaldehyde</b> 50-00-0 $CH_2O$ 30.03	0.027 – 9,770	pungent	C = 0.3 SEN	TWA = 0.75 STEL = 2	–
141	<b>Formic Acid</b> 64-18-6 $CH_2O_2$ 46.02	0.52 – 340	sharp	TWA = 5 STEL = 10	TWA = 5	–

Table 6.1 – Odor Threshold Values, cont.

#	Compound Name CAS Number Formula Molecular Weight	Range of Odor Values (ppm)	Odor Character	ACGIH TLV® (ppm)	OSHA PEL (ppm)	AIHA WEEL® (ppm)
142	<b>Furan</b> 110-00-9 C <sub>4</sub> H <sub>4</sub> O 68.07	10.06	strong	–	–	(W)
143	<b>Furfural</b> 98-01-1 C <sub>6</sub> H <sub>4</sub> O <sub>2</sub> 96.08	0.002 – 0.713	bread, almond	TWA = 2 Skin BEI	TWA = 5 Skin	–
144	<b>Furfuryl Alcohol</b> 98-00-0 C <sub>7</sub> H <sub>6</sub> O <sub>2</sub> 98.1	8	sweet, ether, alcohol	TWA = 10 STEL = 15 Skin	TWA = 50	–
145	<b>Glutaraldehyde</b> 111-30-8 C <sub>5</sub> H <sub>8</sub> O <sub>2</sub> 100.12	0.00037 – 0.039	–	C = 0.05 SEN	–	–
146	<b>Halothane</b> 151-67-7 C <sub>2</sub> HBrClF <sub>3</sub> 197.4	33	chloroform	TWA = 50	–	–
147	<b>Heptane, all isomers</b> 142-82-5, 590-35-2, 565-59-3 108-08-7, 591-76-4, 589-34-4 C <sub>7</sub> H <sub>16</sub> 100.2	0.41 – 732	gasoline	TWA = 400 STEL = 500	TWA = 500	–
148	<b>Hexachlorocyclopentadiene</b> 77-47-4 C <sub>5</sub> Cl <sub>6</sub> 272.77	0.15	pungent	TWA = 0.01	–	–
149	<b>1,6-Hexamethylene Diisocyanate</b> 822-06-0 C <sub>6</sub> H <sub>10</sub> N <sub>2</sub> O <sub>2</sub> 168.19	0.005 – 0.01	–	TWA = 0.005	–	–
150	<b>n-Hexane</b> 110-54-3 C <sub>6</sub> H <sub>14</sub> 86.17	1.50 – 248	gasoline	TWA = 50 Skin BEI	TWA = 500	–
151	<b>Hexane, isomers except n-hexane</b> 107-83-5, 96-14-0, 75-83-2, 79-29-8 C <sub>6</sub> H <sub>14</sub> 86.17	0.426 – 20	gasoline	TWA = 500 STEL = 1,000	–	–
152	<b>1,6-Hexanediamine</b> 124-09-4 C <sub>6</sub> H <sub>16</sub> N <sub>2</sub> 116.2	0.00067	–	TWA = 0.5	–	TWA = 1
153	<b>1-Hexene</b> 592-41-6 C <sub>6</sub> H <sub>12</sub> 84.16	0.139	petroleum	TWA = 50	–	–

**Table 6.1 – Odor Threshold Values, cont.**

#	Compound Name CAS Number Formula Molecular Weight	Range of Odor Values (ppm)	Odor Character	ACGIH TLV® (ppm)	OSHA PEL (ppm)	AIHA WEEL® (ppm)
154	<b>sec-Hexyl Acetate</b> 108-84-9 C <sub>8</sub> H <sub>16</sub> O <sub>2</sub> 144.21	<0.068 – 0.39	banana, pear, fruity	TWA = 50	TWA = 50	–
155	<b>n-Hexyl Alcohol</b> 111-27-3 C <sub>6</sub> H <sub>14</sub> O 102.18	0.0024 – 16	green grass, plastic	–	–	TWA = 40 Eye irritation
156	<b>Hexylene glycol</b> 107-41-5 C <sub>6</sub> H <sub>14</sub> O <sub>2</sub> 118.18	3.93	mild, sweet	C = 25	–	–
157	<b>Hydrazine</b> 302-01-2 N <sub>2</sub> H <sub>4</sub> 32.05	3.0 – 4.0	ammonia	TWA = 0.01 Skin	TWA = 1 Skin	–
158	<b>Hydrogen Chloride</b> 7647-01-0 HCl 36.47	0.06 – 10	sharp, irritating	C = 2	C = 5	–
159	<b>Hydrogen Cyanide</b> 74-90-8 CHN 27.03	0.009 – 5.43	almonds	C = 4.7 Skin	TWA = 10 Skin	–
160	<b>Hydrogen Fluoride</b> 7664-39-3 HF 20.01	0.04	highly corrosive, irritating	TWA = 0.5 C = 2 Skin, BEI	TWA = 3	–
161	<b>Hydrogen Selenide</b> 7783-07-5 H <sub>2</sub> Se 80.98	<0.3	garlic	TWA = 0.05	TWA = 0.05	–
162	<b>Hydrogen Sulfide</b> 7783-06-4 H <sub>2</sub> S 34.08	0.00004 – 1.4	rotten eggs	TWA = 1 STEL = 5	C = 20	–
163	<b>Indene</b> 95-13-6 C <sub>9</sub> H <sub>8</sub> 116.15	0.0027 – 0.0042	–	TWA = 5	–	–
164	<b>Iodine</b> 7553-56-2 I <sub>2</sub> 253.81	0.973	sharp, alliaceous	TWA = 0.01 STEL = 0.1 IFV	C = 0.1	–
165	<b>Iodoform</b> 75-47-8 CHI <sub>3</sub> 393.78	0.000019 – 1.12	chemical, etherish	TWA = 0.6	–	–
166	<b>Isoamyl Acetate</b> 123-92-2 C <sub>8</sub> H <sub>16</sub> O <sub>2</sub> 130.18	0.00075 – 366	banana, fresh	TWA = 50 STEL = 100	TWA = 100	–

**Table 6.1 – Odor Threshold Values, cont.**

#	Compound Name CAS Number Formula Molecular Weight	Range of Odor Values (ppm)	Odor Character	ACGIH TLV® (ppm)	OSHA PEL (ppm)	AIHA WEEL® (ppm)
167	<b>Isomyl Alcohol</b> 123-51-3 C <sub>7</sub> H <sub>14</sub> O 88.15 bitter	0.00169 – 1.75	sweet, malty, rancid, rubber,	TWA = 100 STEL = 125	TWA = 100	–
168	<b>Isobutyl Acetate</b> 110-19-0 C <sub>8</sub> H <sub>16</sub> O <sub>2</sub> 116.16	0.008 – 129	sweet, ester, medicinal	TWA = 150	TWA = 150	–
169	<b>Isobutyl Alcohol</b> 78-83-1 C <sub>7</sub> H <sub>16</sub> O 74.12	0.01 – 165	sweet, fusel, musty, alcohol, rubber, latex	TWA = 50	TWA = 100	–
170	<b>Isobutyraldehyde</b> 78-84-2 C <sub>4</sub> H <sub>8</sub> O 72.11	0.00034 – 0.139	pungent	–	–	TWA = 25
171	<b>Isooctyl Alcohol</b> 26952-21-6, 60435-70-3 C <sub>8</sub> H <sub>18</sub> O 130.23	0.0092 – 0.049	faint, pleasant	TWA = 50 Skin	–	–
172	<b>Isophorone</b> 78-59-1 C <sub>11</sub> H <sub>20</sub> O 138.2	0.0003 – 0.19	sharp	C = 5	TWA = 25	–
173	<b>Isoprene</b> 78-79-5 C <sub>5</sub> H <sub>8</sub> 68.12	0.047 – 3.59	aromatic	–	–	TWA = 2
174	<b>Isopropyl Acetate</b> 108-21-4 C <sub>7</sub> H <sub>14</sub> O <sub>2</sub> 102.13	0.160 – 41	fruity	TWA = 100 STEL = 200	TWA = 250	–
175	<b>Isopropyl Alcohol</b> 67-63-0 C <sub>3</sub> H <sub>8</sub> O 60.09	1.0 – 2,197	sharp, rubbing alcohol	TWA = 100	TWA = 200	–
176	<b>Isopropylamine</b> 75-31-0 C <sub>3</sub> H <sub>9</sub> N 59.08	0.025 – 0.70	ammoniacal, amine	TWA = 5 STEL = 10	TWA = 5	–
177	<b>Isopropyl Ether</b> 108-20-3 C <sub>6</sub> H <sub>14</sub> O 102.17	0.017 – 0.053	sweet	TWA = 250 STEL = 310	TWA = 500	–
178	<b>d-Limonene</b> 138-86-3 C <sub>10</sub> H <sub>16</sub> 136.23	0.0018 – 0.31	lemon, plastic, citrus, rubber, terpeny	–	–	TWA = 30
179	<b>Maleic Anhydride</b> 108-31-6 C <sub>4</sub> H <sub>2</sub> O <sub>3</sub> 98.06	0.25 – 0.32	acid	TWA = 0.0025 IFV, SEN	TWA = 0.25	–

Table 6.1 – Odor Threshold Values, cont.

#	Compound Name CAS Number Formula Molecular Weight	Range of Odor Values (ppm)	Odor Character	ACGIH TLV® (ppm)	OSHA PEL (ppm)	AIHA WEEL® (ppm)
180	<b>Mercaptoethanol</b> 60-24-2 C <sub>2</sub> H <sub>6</sub> O <sub>2</sub> 78.13	0.075	–	–	–	TWA = 0.2 Skin
181	<b>Mesityl Oxide</b> 141-79-7 C <sub>9</sub> H <sub>8</sub> O 98.14	0.017 – 12	sweet	TWA = 15 STEL = 25	TWA = 25	–
182	<b>Methacrylic acid</b> 79-41-4 C <sub>5</sub> H <sub>8</sub> O <sub>2</sub> 86.09	0.54 – 2.84	pungent	TWA = 20	–	–
183	<b>Methacrylonitrile</b> 126-98-7 C <sub>5</sub> H <sub>7</sub> N 67.09	2.95 – 6.9	–	TWA = 1 Skin	–	–
184	<b>Methane</b> 74-82-8 CH <sub>4</sub> 16.04	2,896,197	–	TWA = 1,000	–	–
185	<b>2-Methoxyethanol</b> 109-86-4 C <sub>3</sub> H <sub>8</sub> O <sub>2</sub> 76.09 alcohol	<0.096 – 90	sweet,	TWA = 0.1 Skin BEI	TWA = 25	–
186	<b>2-Methoxyethyl Acetate</b> 110-49-6 C <sub>5</sub> H <sub>10</sub> O <sub>3</sub> 118.13	0.33 – 0.64	sweet, ester	TWA = 0.1 Skin BEI	TWA = 25	–
187	<b>1-Methoxy-2-Propanol</b> 107-98-2 C <sub>4</sub> H <sub>10</sub> O <sub>2</sub> 90.12	8.39 – 33	etherish, ammonia	TWA = 100 STEL = 150	–	–
188	<b>1-Methoxy-2-Propanol Acetate</b> 108-65-6 C <sub>6</sub> H <sub>12</sub> O <sub>3</sub> 132.16	0.0029 – 0.13	–	–	–	TWA = 50
189	<b>Methyl Acetate</b> 79-20-9 C <sub>3</sub> H <sub>6</sub> O <sub>2</sub> 74.08	0.17 – 2,848	fruity	TWA = 200 STEL = 250	TWA = 200	–
190	<b>Methyl Acrylate</b> 96-33-3 C <sub>5</sub> H <sub>8</sub> O <sub>2</sub> 86.09	0.003 – 0.025	plastic, sharp, airplane glue	TWA = 2 Skin SEN	TWA = 10 Skin	–
191	<b>Methyl Alcohol</b> 67-56-1 CH <sub>3</sub> O 32.04	3.05 – 198,686	sour, sweet, alcohol	TWA = 200 STEL = 250 Skin BEI	TWA = 200	–
192	<b>Methylamine</b> 74-89-5 CH <sub>3</sub> N 31.06	0.00075 - 4.8	fishy	TWA = 5 STEL = 15	TWA = 10	–



Table 6.1 – Odor Threshold Values, cont.

#	Compound Name CAS Number Formula Molecular Weight	Range of Odor Values (ppm)	Odor Character	ACGIH TLV® (ppm)	OSHA PEL (ppm)	AIHA WEEL® (ppm)
193	<b>Methyl n-amyl Ketone</b> 110-43-0 C <sub>7</sub> H <sub>14</sub> O 114.18	0.00075 – 0.71	sweet, mushroom	TWA = 50	TWA = 100	–
194	<b>N-Methyl Aniline</b> 100-61-8 C <sub>7</sub> H <sub>9</sub> N 107.15	1.6 – 2.0	–	TWA = 0.5 Skin BEI	TWA = 2 Skin	–
195	<b>2-Methyl Butyl Acetate</b> 624-41-9 C <sub>8</sub> H <sub>16</sub> O <sub>2</sub> 130.18	0.026 – 0.039	–	TWA = 50 STEL = 100	–	–
196	<b>Methyl tert-Butyl Ether</b> 1634-04-4 C <sub>5</sub> H <sub>12</sub> O 88.15	0.03 – 0.17	anesthetic	TWA = 50	–	–
197	<b>Methyl n-Butyl Ketone</b> 591-78-6 C <sub>8</sub> H <sub>16</sub> O 100.16	0.024 – 1.15	sweet, paint	TWA = 5 STEL = 10 Skin BEI	TWA = 100	–
198	<b>Methyl Chloride</b> 74-87-3 CH <sub>2</sub> Cl 50.49	>10	sweet, etherish	TWA = 50 STEL = 100	TWA = 100 Skin	–
199	<b>Methyl Chloroform</b> 71-55-6 C <sub>2</sub> H <sub>3</sub> Cl <sub>3</sub> 133.42	0.97 – 715	sweet, etherish	TWA = 350 STEL = 450 BEI	TWA = 350	–
200	<b>Methyl 2-Cyanoacrylate</b> 137-05-3 C <sub>5</sub> H <sub>5</sub> NO <sub>2</sub> 111.1	0.99 – 2.97	–	TWA = 0.2	–	–
201	<b>Methylcyclohexane</b> 108-87-2 C <sub>7</sub> H <sub>14</sub> 98.19	0.149	petroleum	TWA = 400	TWA = 500	–
202	<b>2-Methylcyclohexanone</b> 583-60-8 C <sub>7</sub> H <sub>12</sub> O 112.17	0.181	acetone	TWA = 50 STEL = 75 Skin	TWA = 100 Skin	–
203	<b>Methylene Bisphenyl Isocyanate</b> 101-68-8 C <sub>15</sub> H <sub>16</sub> O <sub>2</sub> N <sub>2</sub> 250	0.39	–	TWA = 0.005	C = 0.02	–
204	<b>Methylene Chloride</b> 75-09-2 CH <sub>2</sub> Cl <sub>2</sub> 84.94	1.2 – 440	sweet	TWA = 50 BEI	TWA = 25	–
205	<b>Methyl Ethyl Ketone</b> 78-93-3 C <sub>5</sub> H <sub>10</sub> O 72.1	0.07 – 339	sweet, sharp	TWA = 200 STEL = 300 BEI	TWA = 200	–

**Table 6.1 – Odor Threshold Values, cont.**

#	Compound Name CAS Number Formula Molecular Weight	Range of Odor Values (ppm)	Odor Character	ACGIH TLV® (ppm)	OSHA PEL (ppm)	AIHA WEEL® (ppm)
206	<b>Methyl Formate</b> 107-31-3 $C_1H_2O_2$ 60.06	67 – 2,809	ethereal	TWA = 100 STEL = 150	TWA = 100	–
207	<b>Methyl Hydrazine</b> 60-34-4 $CH_2N_2$ 46.07	1 – 3	ammonia	TWA = 0.01 Skin	C = 0.2 Skin	–
208	<b>Methyl Isoamyl Ketone</b> 110-12-3 $C_7H_{14}O$ 114.2	0.0021 – 0.135	sweet, sharp	TWA = 50	TWA = 100	–
209	<b>Methyl Isobutyl Ketone</b> 108-101 $C_6H_{12}O$ 100.16	0.03 – 16	sweet, sharp	TWA = 20 STEL = 75 BEI	TWA = 100	–
210	<b>Methyl Isocyanate</b> 624-83-9 $C_1H_3NO$ 57.05	2.14	–	TWA = 0.02 Skin	TWA = 0.02 Skin	–
211	<b>Methyl Isopropyl Ketone</b> 563-80-4 $C_6H_{12}O$ 86.14	0.51 – 4.8	sweet, sharp	TWA = 20	–	–
212	<b>Methyl Mercaptan</b> 74-93-1 $CH_3S$ 48.11	0.000000000000051 – 0.56	rotten cabbage, garlic	TWA = 0.5	C = 10	–
213	<b>Methyl Methacrylate</b> 80-62-6 $C_5H_8O_2$ 100.13	0.014 – 0.66	plastic, sharp	TWA = 50 STEL = 100 SEN	TWA = 100	–
214	<b>2-Methylnaphthalene</b> 91-57-6 $C_{11}H_{10}$ 142.2	0.00069	–	TWA = 0.5 Skin	–	–
215	<b>Methyl Parathion</b> 298-00-0 $C_8H_{11}NO_3PS$ 263.23	0.0012	pungent	TWA = 0.002 IFV Skin	–	–
216	<b>4-Methyl-2-Propanol</b> 108-11-2 $C_4H_{10}O$ 102.17	0.335 – 0.526	–	TWA = 25 STEL = 40 Skin	TWA = 25 Skin	–
217	<b>Methyl Propyl Ketone</b> 107-87-9 $C_6H_{12}O$ 86.17	0.028 – 65	ingernail polish	STEL = 150	TWA = 200	–
218	<b>n-Methyl-2-Pyrrolidone</b> 872-50-4 $C_5H_9NO$ 99.13	4.2 – 10	amine	–	–	TWA = 10 Skin

Table 6.1 – Odor Threshold Values, cont.

#	Compound Name CAS Number Formula Molecular Weight	Range of Odor Values (ppm)	Odor Character	ACGIH TLV® (ppm)	OSHA PEL (ppm)	AIHA WEEL® (ppm)
219	<b>Methyl Styrene</b> 98-83-9 C <sub>9</sub> H <sub>10</sub> 118.18	0.02 – 49.7	–	TWA = 10	C = 100	–
220	<b>Methyl Vinyl Ketone</b> 78-94-4 C <sub>6</sub> H <sub>8</sub> O 70.09	0.174	pungent	C = 0.2 Skin SEN	–	–
221	<b>Monochloroacetic Acid</b> 79-11-8 C <sub>2</sub> H <sub>3</sub> ClO <sub>2</sub> 94.5	0.013 – 0.155	–	TWA = 0.5 IFV Skin	–	TWA = 0.5 Skin
222	<b>Morpholine</b> 110-91-8 C <sub>4</sub> H <sub>9</sub> NO 87.12	0.011 – 0.070	fishy, amine	TWA = 20 Skin	TWA = 20 Skin	–
223	<b>Naphthalene</b> 91-20-3 C <sub>10</sub> H <sub>8</sub> 128.16	0.0019 – 1.02	tar, creosote, mothballs, emphyreumatic	TWA = 10 STEL = 15 Skin	TWA = 10	–
224	<b>1-Naphthylamine</b> 134-32-7 C <sub>10</sub> H <sub>9</sub> N 143.19	0.024 – 0.050	–	–	1910.1003 carcinogen	–
225	<b>2-Naphthylamine</b> 91-59-8 C <sub>10</sub> H <sub>9</sub> N 143.19	0.24 – 0.32	–	–	1910.1003 carcinogen	–
226	<b>Nickel Carbonyl</b> 13463-39-3 Ni(CO) <sub>4</sub> 170.73	0.5 – 3	sooty	TWA = 0.05	TWA = 0.001	–
227	<b>Nicotine</b> 54-11-5 C <sub>10</sub> H <sub>14</sub> N <sub>2</sub> 162.23	0.0099	–	TWA = 0.075 Skin	TWA = 0.075 Skin	–
228	<b>Nitric Acid</b> 7697-37-2 HNO <sub>3</sub> 63.02	0.27	suffocating	TWA = 2 STEL = 4	TWA = 2	–
229	<b>Nitrobenzene</b> 98-95-3 C <sub>6</sub> H <sub>5</sub> NO <sub>2</sub> 123.11	0.0004 – 29	almonds, shoe polish, pungent	TWA = 1 Skin BEI	TWA = 1 Skin	–
230	<b>Nitrogen Dioxide</b> 10102-44-0 NO <sub>2</sub> 46.01	0.058 – 0.5	bleach	TWA = 0.2	C = 5	–
231	<b>Nitromethane</b> 75-52-5 CH <sub>3</sub> NO <sub>2</sub> 61.04	50	–	TWA = 20	TWA = 100	–

Table 6.1 – Odor Threshold Values, cont.

#	Compound Name CAS Number Formula Molecular Weight	Range of Odor Values (ppm)	Odor Character	ACGIH TLV® (ppm)	OSHA PEL (ppm)	AIHA WEEL® (ppm)
232	<b>1-Nitropropane</b> 108-03-2 C <sub>3</sub> H <sub>7</sub> NO <sub>2</sub> 89.09	7.7 – 140	–	TWA = 25	TWA = 25	–
233	<b>2-Nitropropane</b> 79-46-9 C <sub>3</sub> H <sub>7</sub> NO <sub>2</sub> 89.09	4.94 – 288	fruity	TWA = 10	TWA = 25	–
234	<b>N-Nitrosodimethylamine</b> 62-75-9 C <sub>2</sub> H <sub>5</sub> N <sub>2</sub> O 74.08	0.0079 – 0.013	faint	Skin, L	1910.1003 carcinogen	–
235	<b>Nonane</b> 111-84-2 C <sub>9</sub> H <sub>20</sub> 128.26	2.3 – 21	gasoline	TWA = 200	–	–
236	<b>Octane, all isomers</b> 111-65-9, 540-84-1, 86290-81-5 C <sub>8</sub> H <sub>18</sub> 114.22	0.66 – 235	gasoline, oil	TWA = 300	TWA = 500 n-Octane only	–
237	<b>1-Octanol</b> 111-87-5 C <sub>8</sub> H <sub>18</sub> O 130.23	0.0009 – 1.69	penetrating	–	–	TWA = 50
238	<b>1-Octene</b> 111-66-0 C <sub>8</sub> H <sub>16</sub> 112.21	0.001 – 206	–	–	–	TWA = 75
239	<b>Oxygen Difluoride</b> 7783-41-7 OF <sub>2</sub> 54	0.0996	strong, peculiar	C = 0.05	TWA = 0.05	–
240	<b>Ozone</b> 10028-15-6 O <sub>3</sub> 48	0.0031 – 0.25	pungent, thunder storm	TWA = 0.05	TWA = 0.1	–
241	<b>Pentaborane</b> 19624-22-7 B <sub>5</sub> H <sub>9</sub> 63.17	0.97	pungent	TWA = 0.005 STEL = 0.015	TWA = 0.005	–
242	<b>Pentane, all isomers</b> 78-78-4, 109-66-0, 463-82-1 C <sub>5</sub> H <sub>12</sub> 72.15	1.29 – 1147	sweet	TWA = 600	TWA = 1,000	–
243	<b>2,4-Pentanedione</b> 123-54-6 C <sub>5</sub> H <sub>8</sub> O <sub>2</sub> 100.12	0.0098 – 0.0195	–	TWA = 25 Skin	–	–

Table 6.1 – Odor Threshold Values, cont.

#	Compound Name CAS Number Formula Molecular Weight	Range of Odor Values (ppm)	Odor Character	ACGIH TLV® (ppm)	OSHA PEL (ppm)	AIHA WEEL® (ppm)
244	<b>Pentanol, all isomers</b> 71-41-0, 75-85-4, 75-84-3, 123-51-3, 137-32-6, 584-02-1, 598-75-4, 6032-29-7, 30899-19-5, 94624-12-1 $C_5H_{12}O$ 88.15	0.0055 – 305	–	–	–	TWA = 100 71-41-0 only
245	<b>Perchloroethylene</b> 127-18-4 $C_2Cl_4$ 165.84	0.767 – 71	etherish	TWA = 25 STEL = 100 BEI	TWA = 100 C = 200	–
246	<b>Perchloryl Fluoride</b> 7616-94-6 $ClFO$ 70.45	14.58	sweet	TWA = 3 STEL = 6	TWA = 3	–
247	<b>Phenol</b> 108-95-2 $C_6H_5OH$ 94.11	0.0045 – 1.95	medicinal, acid, ink, creosote, emphyreumatic	TWA = 5 Skin BEI	TWA = 5 Skin	–
248	<b>Phenyl Mercaptan</b> 108-98-5 $C_6H_5S$ 110.17	0.00003 – 0.00027	putrid	TWA = 0.1 Skin	–	–
249	<b>Phosgene</b> 75-44-5 $COCl_2$ 98.92	0.12 – 5.7	hay like	TWA = 0.1	TWA = 0.1	–
250	<b>Phosphine</b> 7803-51-2 $PH_3$ 34	0.01 – 5	garlic	TWA = 0.3 STEL = 1	TWA = 0.3	–
251	<b>Phthalic Anhydride</b> 85-44-9 $C_8H_4O_3$ 148.1	0.053	choking	TWA = 1 SEN	TWA = 2	–
252	<b>Picolines</b> 109-06-8, 108-99-6, 108-89-4 $C_7H_7N$ 93.13	0.0026 – 0.0236	strong, unpleasant	–	–	TWA = 2 STEL = 5 Skin
253	<b>Piperdine</b> 110-89-4 $C_8H_{11}N$ 85.15	0.14 – <2	pepper	–	–	TWA = 1 Skin
254	<b>Propane</b> 74-98-6 $C_3H_8$ 44.09	1497 – 19964	natural gas	TWA = 1000	TWA = 1000	–
255	<b>Propionaldehyde</b> 123-38-6 $C_3H_6O$ 58.08	0.001 – 101	fruity	TWA = 20	–	TWA = 20

**Table 6.1 – Odor Threshold Values, cont.**

#	Compound Name CAS Number Formula Molecular Weight	Range of Odor Values (ppm)	Odor Character	ACGIH TLV® (ppm)	OSHA PEL (ppm)	AIHA WEEL® (ppm)
256	<b>Propionic Acid</b> 79-09-4 C <sub>3</sub> H <sub>6</sub> O <sub>2</sub> 74.08	0.00099 – 4.65	sour	TWA = 10	–	–
257	<b>n-Propyl Acetate</b> 109-60-4 C <sub>6</sub> H <sub>10</sub> O <sub>2</sub> 102.13	0.048 – 87	sweet, ester	TWA = 200 STEL = 250	TWA = 200	–
258	<b>Propyl Alcohol</b> 71-23-8 C <sub>3</sub> H <sub>8</sub> O 60.09	<0.031 – 10172	sweet, alcohol	TWA = 100	TWA = 200	–
259	<b>Propylene</b> 115-07-1 C <sub>3</sub> H <sub>6</sub> 42.08	10.1 – 99	gassy, aromatic	TWA = 500	–	–
260	<b>Propylene Dichloride</b> 78-87-5 C <sub>3</sub> H <sub>4</sub> Cl <sub>2</sub> 112.99	0.26 – 8.66	sweet	TWA = 10 SEN	TWA = 75	–
261	<b>Propylene Glycol</b> 57-55-6 C <sub>3</sub> H <sub>8</sub> O <sub>2</sub> 76.09	5.14	–	–	–	TWA = 3.2
262	<b>Propylene Glycol Dinitrate</b> 6423-43-4 C <sub>3</sub> H <sub>4</sub> N <sub>2</sub> O <sub>6</sub> 166.09	0.236	disagreeable	TWA = 0.05 Skin BEI	–	–
263	<b>Propylene Oxide</b> 75-56-9 C <sub>3</sub> H <sub>6</sub> O 58.08	10 – 199	sweet	TWA = 2 SEN	TWA = 100	–
264	<b>Pyridine</b> 110-86-1 C <sub>5</sub> H <sub>5</sub> N 79.1	0.01 – 12	burnt, pungent, nauseating	TWA = 1	TWA = 5	–
265	<b>Quinoline</b> 91-22-5 C <sub>8</sub> H <sub>7</sub> N 129.16	0.0057 – 5.3	peculiar	–	–	TWA = 0.001 Skin
266	<b>Quinone</b> 106-51-4 C <sub>6</sub> H <sub>4</sub> O <sub>2</sub> 108.09	0.011 – 0.10	pungent	TWA = 0.1	TWA = 0.1	–
267	<b>Styrene, monomer</b> 100-42-5 C <sub>8</sub> H <sub>8</sub> 104.14	0.0028 – 61	sharp, sweet	TWA = 20 STEL = 40 BEI	TWA = 100 C = 200	–

**Table 6.1 – Odor Threshold Values, cont.**

#	Compound Name CAS Number Formula Molecular Weight	Range of Odor Values (ppm)	Odor Character	ACGIH TLV® (ppm)	OSHA PEL (ppm)	AIHA WEEL® (ppm)
268	<b>Sulfur Dioxide</b> 7446-09-5 SO <sub>2</sub> 64.07	0.33 - 8	metallic	STEL = 0.25	TWA = 5	-
269	<b>Sulfur Hexafluoride</b> 2551-62-4 F <sub>6</sub> S 146.06	4,017,527	-	TWA = 1,000	TWA = 1,000	-
270	<b>Sulfuric Acid</b> 7664-93-9, 8014-95-7 H <sub>2</sub> SO <sub>4</sub> 98.08	0.15	-	TWA = 0.05	TWA = 0.25	
271	<b>1,1,2,2-Tetrabromoethane</b> 79-27-6 C <sub>2</sub> H <sub>2</sub> Br <sub>4</sub> 345.65	<0.99	camphor, pungent	TWA = 0.1 IFV	TWA = 1	-
272	<b>1,1,2,2-Tetrachloroethane</b> 79-34-5 C <sub>2</sub> H <sub>2</sub> Cl <sub>4</sub> 167.9	0.233 – 7.3	solvent	TWA = 1 Skin	TWA = 5 Skin	-
273	<b>Tetrahydrofuran</b> 109-99-9 C <sub>4</sub> H <sub>8</sub> O 72.1	0.092 – 61	ether	TWA = 50 STEL = 100 Skin	TWA = 200	-
274	<b>Thioglycolic Acid</b> 68-11-1 C <sub>2</sub> H <sub>3</sub> O <sub>2</sub> S 92.12	0.00021	unpleasant	TWA = 1 Skin	-	-
275	<b>Toluene</b> 108-88-3 C <sub>7</sub> H <sub>8</sub> 92.13	0.021 – 157	sour, burnt	TWA = 20 BEI	TWA = 200 C = 300	-
276	<b>Toluene 2,4- or 2,6-Diisocyanate</b> 584-84-9, 91-08-7 C <sub>7</sub> H <sub>6</sub> N <sub>2</sub> O <sub>2</sub> 174.06	0.02 – 2	-	TWA = 0.005 STEL = 0.02 SEN	C = 0.02 584-84-9 only	-
277	<b>o-Toluidine</b> 95-53-4 C <sub>7</sub> H <sub>7</sub> N 107.15	0.025 – 6.6	aromatic, amine, empyreumatic	TWA = 2 Skin BEI	TWA = 5 Skin	-
278	<b>m-Toluidine</b> 108-44-1 C <sub>7</sub> H <sub>7</sub> N 107.15	0.46 – 5.9	empyreumatic	TWA = 2 Skin BEI	-	-
279	<b>p-Toluidine</b> 106-49-0 C <sub>7</sub> H <sub>7</sub> N 107.15	0.027 – 3.2	amine, empyreumatic	TWA = 2 Skin BEI	-	-

Table 6.1 – Odor Threshold Values, cont.

#	Compound Name CAS Number Formula Molecular Weight	Range of Odor Values (ppm)	Odor Character	ACGIH TLV® (ppm)	OSHA PEL (ppm)	AIHA WEEL® (ppm)
280	<b>Trichloroacetic acid</b> 76-03-9 C <sub>2</sub> HCl <sub>3</sub> O <sub>2</sub> 163.39	0.24 – 0.37	–	TWA = 1	–	–
281	<b>1,2,4-Trichlorobenzene</b> 120-82-1 C <sub>6</sub> H <sub>3</sub> Cl <sub>3</sub> 181.46	2.96	aromatic	C = 5	–	–
282	<b>Trichloroethylene</b> 79-01-6 C <sub>2</sub> HCl <sub>3</sub> 131.4	0.5 – 167	ether, solvent	TWA = 10 STEL = 25 BEI	TWA = 100 C = 200	–
283	<b>Trichlorofluoromethane</b> 75-69-4 CCl <sub>3</sub> F 137.37	5 – 200,057	–	C = 1000	TWA = 1000	–
284	<b>Triethanolamine</b> 102-71-6 C <sub>6</sub> H <sub>15</sub> NO <sub>3</sub> 149.19	>10	mild, ammonia	TWA = 0.82	–	–
285	<b>Triethylamine</b> 121-44-8 C <sub>6</sub> H <sub>15</sub> N 101.19	0.005 – 2.9	fishy, amine	TWA = 1 STEL = 3 Skin	TWA = 25	–
286	<b>Trimethylamine</b> 75-50-3 C <sub>3</sub> H <sub>9</sub> N 59.11	0.00002 – 1.82	fishy, pungent	TWA = 5 STEL = 15	–	TWA = 1
287	<b>Trimethyl Benzene, all isomers</b> 95-63-6, 108-67-8, 526-73-8, 25551-13-7 C <sub>9</sub> H <sub>12</sub> 120.19	0.006 – 2.4	aromatic	TWA = 25	–	–
288	<b>Trimethyl Phosphite</b> 121-45-9 C <sub>3</sub> H <sub>9</sub> O <sub>3</sub> P 124.08	0.000099	pungent	TWA = 2	–	–
289	<b>Turpentine &amp; monoterpenes</b> 80-56-8, 127-91-3, 13466-78-9, 8006-64-2 C <sub>10</sub> H <sub>16</sub> 136.23	0.00006 – 19	turpentine, rosiny, pine tree, camphorous, fir needles	TWA = 20 SEN	TWA = 100 80006-64-2 only	–
290	<b>n-Valeraldehyde</b> 110-62-3 C <sub>5</sub> H <sub>10</sub> O 86.13	0.0004 – 4.97	sickening, rancid, decayed	TWA = 50	–	–



**Table 6.1 – Odor Threshold Values, cont.**

#	Compound Name CAS Number Formula Molecular Weight	Range of Odor Values (ppm)	Odor Character	ACGIH TLV® (ppm)	OSHA PEL (ppm)	AIHA WEEL® (ppm)
291	<b>Vanillin</b> 121-33-5 C <sub>8</sub> H <sub>8</sub> O <sub>3</sub> 152.15	0.00000016 – 0.0929	vanilla, caramel, sweet	–	–	TWA = 1.6
292	<b>Vinyl Acetate</b> 108-05-4 C <sub>4</sub> H <sub>6</sub> O <sub>2</sub> 86.09	0.12 – 0.4	sour, sharp	TWA = 10 STEL = 15	–	–
293	<b>Vinyl Chloride</b> 75-01-4 C <sub>2</sub> H <sub>3</sub> Cl 62.5	203 – 356	sweet	TWA = 1	TWA = 1 STEL = 5	–
294	<b>Vinylidene chloride</b> 75-35-4 C <sub>2</sub> H <sub>2</sub> Cl <sub>2</sub> 96.94	50 – 1387	chloroform	TWA = 5	–	–
295	<b>Xylene (o-, m-, p-, isomers)</b> 1330-20-7, 95-47-6 108-38-3, 106-42-3 C <sub>8</sub> H <sub>10</sub> 106.16	0.012 – 316	sweet, empyreumatic	TWA = 100 STEL = 150 BEI	TWA = 100	

## **Table 6.2 – Methods Summary of Reviewed Articles**

Threshold methodologies were reviewed according to the criteria discussed in Section 4.4.

The table contains the following information:

- Source (Last name of first author) and publication date
- Panel size
- Panel selection criteria (i.e., trained, screened, etc.)
- Panel calibration
- Vapor modality (usually air; however, in a few cases water vapor or water)
- Diluent (unless specified otherwise in the paper, it was assumed to be air)
- Presentation mode (type of instrument at interface)
- Analytic measure
- Flow rate
- Threshold type
- Concentration series
- Trials (greater than one trial)
- Forced choice
- Concentration interval (less than or equal to a three-fold step size)

### Abbreviations Used in Table

- |                            |   |
|----------------------------|---|
| • R = recognition          | • R = random  |
| • D = detection            | • V = variable  |
| • MP = minimum perceptible | • U-D = up-down series  |
| • I = intensity            | • ng – information not given in article                           |
| • A = ascending            | • nd – data not determined, usually in a foreign language article |
| • D = descending           | • VDD8 = Vapor Delivery Device                                    |

Table 6.2 – Methods Summary of Reviewed Articles

Source (Year)	Panel			Presentation Apparatus					Presentation Method					Note #
	Size	Selection Criteria	Calibration	Vapor Modality	Diluent	Presentation Mode	Analytic Measure	Flow Rate	Threshold Type	Conc. Series	Trials	Forced - Choice	Conc. Interval	
Adams (1968)	114 - 789	no	no	air	pure air	odor hood	yes	2-5 lpm	D	A + D + R	yes	no	yes	
Ahlstrom (1986a)	40	yes	yes	air	air	odor hood	yes	100 lpm	D	A + D + R	yes	yes	yes	
Ahlstrom (1968b)	64	yes	no	air	air	mobile olfactometer	yes	100 lpm	D	A	yes	yes	yes	
Akhemedov (1968)	4	yes	no	air	carbon filtered	cylinder	yes	15 lpm	MP	ng	yes	yes	yes	36
Alibaev (1970)	25	yes	no	air	carbon filtered	cylinder	yes	15 lpm	MP	ng	yes	yes	yes	36
Allison (1919)	ng	ng	ng	air	pure air	glass	no	≥8 lpm	D	ng	ng	no	ng	1
Amdur (1953)	14	ng	ng	air	air	face mask	yes	ng	R	ng	ng	no	yes	1
Amoore (1978)	>10	ng	ng	water	water or buffered water	flask	no	static	D	D	ng	yes	yes	56
Amoore (1977)	>10	ng	ng	water	water or buffered water	flask	no	static	D	D	ng	yes	yes	20, 56
Andreescheva (1964)	29	yes	yes	air	carbon filtered	cylinder	yes	15 lpm	MP	ng	yes	yes	yes	36
Andreescheva (1968)	26	yes	no	air	carbon filtered	cylinder	yes	15 lpm	MP	ng	yes	yes	yes	36
Appell (1969)	ng	ng	ng	water	water	bottle	ng	static	MP	ng	ng	ng	yes	17
Babin (1965)	ng	ng	ng	air	ng	ng	ng	ng	ng	ng	ng	ng	ng	2

Table 6.2 – Methods Summary of Reviewed Articles, cont.

Source (Year)	Panel			Presentation Apparatus					Presentation Method					Note #
	Size	Selection Criteria	Calibration	Vapor Modality	Diluent	Presentation Mode	Analytic Measure	Flow Rate	Threshold Type	Conc. Series	Trials	Forced - Choice	Conc. Interval	
Baikov (1963)	nd	nd	nd	air	nd	nd	nd	nd	MP	nd	nd	nd	nd	46
Baikov (1973)	28	nd	nd	air	nd	nd	nd	nd	MP	nd	nd	nd	nd	46
Basmadzhieva (1968)	13	ng	ng	air	ng	ng	ng	0.2-0.6 lpm	MP	ng	yes	yes	yes	36
Baydar (1993)	79	yes	no	air	air	olfactometer	yes	ng	D + R	A	yes	yes	yes	12
Belkov (1969)	ng	ng	ng	air	ng	ng	ng	ng	ng	ng	ng	ng	ng	2
Berglund (1992)	44	yes	no	air	sodium hydroxide	dynamic olfactometer	yes	100 lpm	D	A	yes	yes	yes	52
Berzins (1967)	18	ng	ng	air	ng	ng	ng	ng	ng	ng	ng	ng	ng	2
Bezpalikova (1967a)	23	nd	nd	air	nd	nd	nd	nd	MP	nd	nd	nd	yes	45
Blank (1993)	7	yes	ng	air	nitrogen	GC-olfactometry	yes	0.01 lpm	D	ng	yes	ng	ng	50
Blank (1989)	ng	ng	ng	ng	ng	GC-olfactometry	ng	ng	D	ng	ng	ng	ng	
Blinova (1965)	9 - 10	ng	ng	air	ng	gas mask	ng	ng	MP	ng	yes	ng	ng	46
Bocca (1964)	3	ng	ng	air	ng	blast	ng	167 lpm	D	D	yes	ng	ng	20, 62
Bokowa (2012)	3	yes	ng	ng	ng	dynamic olfactometer	ng	ng	D	ng	ng	ng	ng	

Table 6.2 – Methods Summary of Reviewed Articles, cont.

Source (Year)	Panel			Presentation Apparatus					Presentation Method					Note #
	Size	Selection Criteria	Calibration	Vapor Modality	Diluent	Presentation Mode	Analytic Measure	Flow Rate	Threshold Type	Conc. Series	Trials	Forced - Choice	Conc. Interval	
Boriskova (1957)	12	nd	nd	air	nd	nd	nd	nd	MP	nd	nd	nd	nd	46
Brunekreef (1980)	4	ng	ng	air	air	ng	ng	ng	ng	ng	ng	yes	ng	59
Buettner (2001a)	10	yes	yes	air	water	bottle	ng	static	D	ng	yes	ng	ng	
Buettner (2001b)	ng	ng	ng	ng	ng	ng	ng	ng	D	ng	ng	ng	ng	
Bushtueva (1962)	ng	yes	ng	air	Clean air	cylinder	ng	ng	ng	ng	ng	no	ng	1
Buttery (1969)	>10	ng	ng	water	water	bottles	no	static	D + R	D	ng	yes	yes	20, 56
Cain (1987)	57, 72	yes	yes	air	propane, argon	olfactometer, bottles	yes	static, 180 lpm, 85 lpm	D + R	A	yes	yes	yes	51
Cain (2005)	33	yes	ng	air	silicon oil, water	glass vessel	yes	static, 3 lpm	D, S, I (eye)	A	yes	yes	yes	49
Cain (2007a)	10	yes	ng	air	air	VDD8	yes	40 lpm	D	yes	yes	yes	yes	
Cain (2008)	48	yes	ng	air	mineral oil, water	squeeze bottles	yes	static	D	A	yes	yes	yes	
Cain (2009)	29, 26	yes	no	air	nitrogen, air	VDD8	yes	78 lpm, 10 lpm	D + I (eye)	A	yes	yes	yes	49
Cain (1969)	12	ng	ng	vapor	Diethylphthalate	test tubes	no	static	R	A + D	yes	no	yes	19
Cain (1977)	2	ng	no	air	water	glass vessel	yes	static	D	A + D	yes	yes	ng	25

Table 6.2 – Methods Summary of Reviewed Articles, cont.

Source (Year)	Panel			Presentation Apparatus					Presentation Method					Note #
	Size	Selection Criteria	Calibration	Vapor Modality	Diluent	Presentation Mode	Analytic Measure	Flow Rate	Threshold Type	Conc. Series	Trials	Forced - Choice	Conc. Interval	
Gain (2007b)	50	yes	ng	air	air	VDD8	yes	40 lpm	D	yes	yes	yes	yes	
Gain (2010)	70, 17	yes	yes	air	nitrogen	VDD8	yes	1 lpm	D + eye irritation	ng	ng	ng	ng	49
Cancho (2001)	5 - 6	yes	ng	air	methanol or MTBE	Flavor Profile or GC-O	yes	ng	D	ng	ng	ng	ng	55
Catana (2012)	248	no	no	ng	ng	sniffin sticks	ng	ng	R	ng	ng	ng	ng	
Cederlof (1966)	30	ng	ng	air	air	hood	ng	100 lpm	D	A	ng	yes	yes	
Cerny (1994)	3	yes	ng	ng	ng	GC-olfactometry	ng	ng	D	ng	ng	ng	ng	
Chao-Chen-Tzi (1959)	13	nd	nd	air	nd	nd	nd	nd	MP	nd	nd	nd	nd	46
Cheesman (1959)	10-20	ng	ng	air	air	tube	no	yes	D	V	5	no	yes	29, 57
Clausen (1955)	ng	ng	ng	air	pure air	tube	ng	stream	D	D	ng	yes	ng	1, 20
Cometto-Muniz (1990)	7	yes	ng	air	mineral oil	squeeze bottles	yes	static	D	A	yes	yes	yes	
Cometto-Muniz (1991)	4	yes	no	air	mineral oil	squeeze bottles	yes	static	D, I	A	yes	yes	yes	
Cometto-Muniz (1993)	8	yes	ng	air	water, mineral oil	squeeze bottles	yes	static	D	A	yes	yes	yes	49
Cometto-Muniz (2002)	8	yes	ng	air	mineral oil	squeeze bottles	yes	static	D	A	yes	yes	yes	

**Table 6.2 – Methods Summary of Reviewed Articles, cont.**

Source (Year)	Panel			Presentation Apparatus					Presentation Method					Note #
	Size	Selection Criteria	Calibration	Vapor Modality	Diluent	Presentation Mode	Analytic Measure	Flow Rate	Threshold Type	Conc. Series	Trials	Forced - Choice	Conc. Interval	
Cometto-Muniz (2008)	34	yes	yes	air	air	VDD8	yes	40 lpm	D	A	yes	yes	yes	44
Cometto-Muniz (2008)	36	yes	yes	air	air	VDD8	yes	40 lpm	D	A	yes	yes	yes	44
Cometto-Muniz (2009a)	39	yes	yes	air	air	VDD8	yes	40 lpm	D	A	yes	yes	yes	44
Cometto-Muniz (2009b)	36	yes	yes	air	air	VDD8	yes	40 lpm	D	A	yes	yes	yes	44
Cometto-Muniz (2010a)	16	yes	yes	air	air	VDD8	yes	40 lpm	D	A	yes	yes	yes	41
Cometto-Muniz (2010b)	14	yes	yes	air	air	VDD8	yes	40 lpm	D	A	yes	yes	yes	41
Cometto-Muniz (1998a)	11	yes	ng	air	mineral oil	squeeze bottles	yes	static	D	A	yes	yes	yes	
Cometto-Muniz (1998b)	4	yes	yes	air	mineral oil	squeeze bottles	yes	static	D	A	yes	yes	yes	49
Cometto-Muniz (2003)	varied	yes	ng	air	ng	glass bottles	yes	static	D	A	yes	yes	yes	
Cometto-Muniz (2004)	10	yes	ng	?	mineral oil	bottle	yes	static	D, S, I (eye)	yes	yes	yes	yes	
Cometto-Muniz (2005)	varied	yes	ng	air	mineral oil	glass bottles	yes	static	D	A	yes	yes	yes	
Cometto-Muniz (1999)	4, 14	yes	yes	air	mineral oil	squeeze bottles	yes	static	D	A	yes	yes	yes	49
Corbit (1971)	3	yes	no	air	air	nose port	no	2 lpm	D	A	5	yes	yes	37

Table 6.2 – Methods Summary of Reviewed Articles, cont.

	Panel			Presentation Apparatus					Presentation Method					
Source (Year)	Size	Selection Criteria	Calibration	Vapor Modality	Diluent	Presentation Mode	Analytic Measure	Flow Rate	Threshold Type	Conc. Series	Trials	Forced - Choice	Conc. Interval	Note #
Cormack (1974)	4-6	yes	ng	air	air	Room	yes	static	D	ng	ng	no	yes	
Crawford (1984)	4	ng	ng	ng	ng	Triangle Olfactometer	ng	ng	D	A	yes	yes	ng	
Czerny (2008)	ng	ng	ng	ng	water	ng	ng	ng	D	ng	ng	ng	ng	
Czerny (2011)	13	yes	no	air	water	GC-olfactometry	yes	ng	D	D	yes	no	no	
Dalton (1997b)	90	yes	no	air	propylene glycol, mineral oil	squeeze bottles	yes	static	D + I	A	yes	yes	yes	
Dalton (2000)	40	yes	yes	air	air	bottle	yes	static	D, I	U-D	yes	yes	yes	
Dalton (2007)	15	yes	yes	air	air	VDD8	yes	40 lpm	D	A & D	no	yes	yes	47
Davis (1973)	3	ng	ng	air	Clean air	funnel	ng	20 lpm	D	D	ng	no	yes	37
Dixon (1977)	II	yes	ng	air	Oxygen	tube	yes	7-8 lpm	D	U-D	ng	yes	yes	38
Dobrinskii (1964)	ng	ng	ng	air	ng	ng	yes	ng	MP	ng	ng	ng	ng	46
Doty (1984)	1955	ng	ng	ng	ng	ng	ng	ng	D	ng	ng	ng	ng	51
Doty (1988)	36	yes	ng	air	Clean air	bottles	yes	static	D	A + D	ng	yes	yes	
Dravnieks (1971)	5-7	ng	ng	air	ng	ng	yes	ng	D	ng	ng	ng	ng	



Table 6.2 – Methods Summary of Reviewed Articles, cont.

Source (Year)	Panel			Presentation Apparatus					Presentation Method					Note #
	Size	Selection Criteria	Calibration	Vapor Modality	Diluent	Presentation Mode	Analytic Measure	Flow Rate	Threshold Type	Conc. Series	Trials	Forced - Choice	Conc. Interval	
Dravnieks (1972)	nd	nd	nd	water	nd	nd	nd	static	D	nd	nd	nd	nd	
Dravnieks (1968)	ng	ng	ng	air	ng	ng	yes	ng	ng	ng	ng	ng	ng	
Dravnieks (1973)	9	Yes	ng	Air	Pure air	glass port	No	0.6 Lpm	D	A	ng	Yes	Yes	
Dravnieks (1974)	9	ng	ng	Air	Pure air	glass port	No	0.15 Lpm	D	A	ng	Yes	Yes	
Duan-Fen-Djuj (1959)	12	nd	nd	air	nd	nd	nd	nd	MP	nd	nd	nd	nd	46
Dubrovskaya (1961)	12	ng	ng	air	ng	ng	ng	ng	MP	ng	yes	ng	yes	46
Dubrovskaya (1973)	18	yes	ng	air	carbon filtered	cylinder	yes	15 lpm	MP	ng	yes	yes	yes	36
Dubrovskaya (1957)	12	ng	ng	air	ng	ng	ng	ng	MP	ng	yes	ng	yes	36
Dubrovskaya (1969)	22	yes	ng	air	carbon filtered	cylinder	yes	15 lpm	MP	ng	yes	yes	yes	36
Dumas (1974)	ng	ng	ng	ng	ng	ng	ng	ng	ng	ng	ng	ng	ng	
Eglite (1968)	20	yes	ng	air	carbon filtered	cylinder	yes	15 lpm	MP	ng	yes	yes	yes	36
Elfimova (1967)	18	ng	ng	air	ng	ng	ng	ng	ng	ng	yes	ng	ng	2
Feddes (2001)	24	yes	yes	air	air	olfactometer	Indirectly	Up to 20 lpm	D	A	yes	yes	ng	8

Table 6.2 – Methods Summary of Reviewed Articles, cont.

Source (Year)	Panel			Presentation Apparatus					Presentation Method					Note #
	Size	Selection Criteria	Calibration	Vapor Modality	Diluent	Presentation Mode	Analytic Measure	Flow Rate	Threshold Type	Conc. Series	Trials	Forced - Choice	Conc. Interval	
Feldman (1960)	nd	nd	nd	air	nd	nd	nd	nd	MP	nd	nd	nd	nd	45
Feldman (1971)	15	yes	ng	air	ng	ng	yes	ng	MP	ng	ng	ng	yes	46
Feldman (1967)	20	yes	ng	air	carbon filtered	cylinder	yes	15 lpm	MP	ng	yes	yes	yes	36
Ferreira (1998)	4	yes	ng	air	helium	GC- olfactometry	yes	4 ml/min	D	ng	ng	ng	ng	33
Filatova (1962)	14	nd	nd	air	nd	nd	nd	nd	MP	nd	nd	nd	nd	46
Fischer (2008)	2	no	yes	air	Dichloromethane	nasal cone	yes	ng	D	D	no	no	ng	13
Flemming (1977)	18 - 20	yes	ng	air	Compressed air	tube	yes	10 lpm	R	A	no	no	yes	
Fluck (1976)	10	ng	ng	air	Room air	test room	yes	static	R	R	ng	no	yes	4
Fomin (1966)	18	ng	ng	air	ng	ng	ng	ng	MP	ng	ng	ng	ng	
Gijs (2000)	3	yes	ng	air	air	GC- olfactometry	yes	20 ml/min	D	yes	no	no	no	
Glindemann (2006)	4	yes	ng	air	ng	dilution olfactometer	yes	ng	D	D	ng	ng	ng	
Gofmekler (1967)	nd	nd	nd	air	nd	nd	nd	nd	MP	nd	nd	nd	nd	
Gofmekler (1960)	nd	nd	nd	air	nd	nd	nd	nd	MP	nd	nd	nd	nd	46

**Table 6.2 – Methods Summary of Reviewed Articles, cont.**

Source (Year)	Panel			Presentation Apparatus					Presentation Method					Note #
	Size	Selection Criteria	Calibration	Vapor Modality	Diluent	Presentation Mode	Analytic Measure	Flow Rate	Threshold Type	Conc. Series	Trials	Forced - Choice	Conc. Interval	
Gorlova (1970)	24	ng	ng	air	ng	ng	yes	ng	MP	ng	yes	ng	ng	
Greenman (2004)	7	yes	yes	air	several	bottles	no	0.2 lpm	D	ng	yes	ng	yes	14
Grigorieva (1964)	12	yes	ng	air	carbon filtered	cylinder	yes	15 lpm	MP	ng	yes	yes	yes	36
Gundlach (1939)	16-60	ng	ng	air	pure air	nose piece	no	0.041-0.35 lpm	D	ng	ng	yes	yes	24
Gusev (1965)	18-30	yes	ng	air	carbon filtered	cylinder	yes	15 lpm	MP	ng	yes	yes	yes	36
Guth (2001)	ng	ng	ng	ng	ng	ng	ng	ng	ng	ng	ng	ng	ng	13
Hartung (1971)	ng	ng	ng	air	ng	ng	yes	ng	ng	ng	ng	ng	ng	5
Hellman (1974)	5	yes	no	air	air	air stream	yes	20-80 lpm	D+R	A	yes	yes	yes	
Hellman (1973a,b)	5	yes	no	air	air	air stream	yes	20-80 lpm	D+R	A	yes	yes	yes	
Hesse (1926)	ng	ng	ng	air	ng	ng	ng	ng	ng	ng	ng	ng	ng	1
Hesse (1928)	ng	ng	ng	air	ng	ng	yes	ng	ng	ng	ng	ng	ng	1
Higuchi (2004)	ng	ng	ng	ng	ng	ng	ng	ng	ng	ng	ng	ng	ng	
Hildenskiold (1959)	nd	nd	nd	air	nd	nd	nd	nd	MP	nd	nd	nd	nd	45

Table 6.2 – Methods Summary of Reviewed Articles, cont.

	Panel			Presentation Apparatus					Presentation Method					
Source (Year)	Size	Selection Criteria	Calibration	Vapor Modality	Diluent	Presentation Mode	Analytic Measure	Flow Rate	Threshold Type	Conc. Series	Trials	Forced - Choice	Conc. Interval	Note #
Hollingsworth (1963)	ng	ng	ng	ng	ng	ng	ng	ng	ng	ng	ng	ng	ng	
Holmes (1915)	60	no	ng	air	air	booth	ng	High Velocity	D+R	A	yes	no	yes	
Homans (1978)	16	nd	nd	air	nd	nd	yes	yes	D	A	nd	nd	yes	29
Hori (1972)	5 - 10	no	ng	air	air	syringe	yes	static	ng	A	ng	no	ng	1
Hoshika (1997)	7	ng	ng	air	air	ng	yes	static	R	two	ng	ng	ng	53
Ifeadi (1972)	1	no	ng	air	Charcoal filtered	hood	yes	0.4 lpm	D	A	yes	no	yes	37
Imasheva (1963)	18	nd	nd	air	nd	nd	nd	nd	MP	nd	nd	nd	nd	46
Itskovich (1962)	ng	ng	ng	air	ng	ng	ng	ng	ng	ng	ng	ng	ng	2
Ivanov (1964)	11	nd	nd	air	nd	nd	nd	nd	MP	nd	nd	nd	nd	46
Jacobson (1955)	15 - 22	ng	ng	air	Room air	nostril piece	yes	static	D	A	ng	no	yes	
Jacobson (1956)	14 - 16	ng	ng	air	Room air	nostril piece	yes	static	D	A	ng	no	yes	
Jacobson (1958)	13	ng	ng	air	Room air	nostril piece	yes	static	D	A	ng	no	yes	
Jones (1954)	4	no	no	air	pure air	nostril piece	no	3 lpm	R	A	yes	no	ng	1, 46

**Table 6.2 – Methods Summary of Reviewed Articles, cont.**

Source (Year)	Panel			Presentation Apparatus					Presentation Method					Note #
	Size	Selection Criteria	Calibration	Vapor Modality	Diluent	Presentation Mode	Analytic Measure	Flow Rate	Threshold Type	Conc. Series	Trials	Forced - Choice	Conc. Interval	
Jones (1955a)	24	no	no	water	Mineral Spirits	flask	no	static	ng	A	yes	no	ng	1
Jones (1955b)	45	no	no	air	pure air	nostril piece	no	3 lpm	R	A	no	no	ng	1
Jones (1955c)	84	no	no	air	pure air	nostril piece	no	3 lpm	D	A	no	no	ng	1
Kaloyanova (1967)	10	yes	no	air	ng	ng	ng	ng	MP	ng	ng	ng	yes	
Kaloyanova (1968)	12	yes	ng	air	carbon filtered	cylinder	yes	15 lpm	MP	ng	yes	yes	yes	36
Katz (1930)	7 - 14	no	no	air	pure air	funnel	no	≥8 lpm	D	A	no	no	yes	
Kerka (1956)	6	yes	ng	air	nitrogen	test room	ng	static	D + R	ng	yes	no	ng	21
Khachatryan (1968)	3	ng	ng	air	ng	ng	ng	ng	ng	D	yes	ng	yes	38
Khachatryan (1969)	3	ng	ng	air	ng	ng	ng	ng	ng	ng	yes	ng	ng	36
Khiari (1992)	ng	yes	ng	ng	ng	GC- olfactometry	yes	ng	D	ng	ng	ng	ng	
Khikmatullaeva (1967)	21	yes	ng	air	ng	ng	ng	ng	MP	ng	ng	ng	ng	
Kinkead (1971a)	6	no	no	air	Test room	no	no	static	D	R	yes	no	yes	3
Kinkead (1971b)	3	no	no	air	Test room	no	no	static	D	ng	yes	no	2-5	37

Table 6.2 – Methods Summary of Reviewed Articles, cont.

Source (Year)	Panel			Presentation Apparatus					Presentation Method					Note #
	Size	Selection Criteria	Calibration	Vapor Modality	Diluent	Presentation Mode	Analytic Measure	Flow Rate	Threshold Type	Conc. Series	Trials	Forced - Choice	Conc. Interval	
Kleinbeck (2011)	44	yes	no	air	nitrogen	flow olfactometer	yes	ng	D + I	A	yes	yes	yes	
Kniebes (1969)	13 - 33	no	ng	air	nitrogen	test room		static	D	R	yes	no	yes	20
Komthong (2006)	3	yes	ng	air	air	GC-olfactometry	yes	ng	D	D	only two trials	no	ng	34
Korneev (1965)	22	yes	ng	air	carbon	cylinder	yes	15 lpm	MP	ng	yes	yes	yes	36
Kosiborod (1968)	22	yes	ng	air	carbon filtered	cylinder	yes	15 lpm	MP	ng	yes	yes	yes	36
Krackow (1953)	ng	ng	ng	air	ng	ng	ng	ng	D	ng	ng	ng	ng	
Krasovitskaya (1968)	11	yes	ng	air	Clean air	cylinder	yes	15 lpm	MP	ng	yes	yes	yes	36
Krichevskaya (1968)	21	yes	ng	air	Clean air	cylinder	yes	15 lpm	MP	ng	yes	yes	yes	36
Kristesashvili (1965)	12	ng	ng	air	ng	ng	ng	ng	MP	ng	ng	ng	ng	
Kulakov (1964)	19	yes	ng	air	Clean air	cylinder	yes	15 lpm	MP	ng	yes	yes	yes	36
Laffort (1987)	9	yes	ng	ng	ng	tedlar bags	ng	ng	D	ng	ng	ng	ng	32
Laffort (1973)	4	no	no	air	air	Mono rhinal valve	yes	0.4 lpm	D	R	yes	ng	ng	1, 20, 37
Laing (1975)	6	no	no	air	nitrogen	sniff port	yes	0.04 lpm	D	R + D	yes	no	yes	38

**Table 6.2 – Methods Summary of Reviewed Articles, cont.**

	Panel			Presentation Apparatus					Presentation Method					
Source (Year)	Size	Selection Criteria	Calibration	Vapor Modality	Diluent	Presentation Mode	Analytic Measure	Flow Rate	Threshold Type	Conc. Series	Trials	Forced - Choice	Conc. Interval	Note #
ding (1978)	16	yes	no	air	nitrogen	nose & mouth port	yes	10 lpm	R	R	yes	no	yes	42
ding (1982)	23	yes	yes	air	air	chamber & olfactometer	ng	10 lpm	D	A- $\Phi$	yes	yes	yes	
aska (2010)	20	yes	no	air	several	squeeze bottles	no	static	D	A	yes	yes	yes	10
aska (1991)	44	yes	ng	air	n-butanol	sniff bottles	estimated	static	D	A	yes	yes	ng	31
donardos (1969)	4	yes	no	air	Purified air	test room	no	static	100%	R	no	no	no	
L-Shen (1961)	15	nd	nd	air	nd	nd	nd	nd	MP	nd	nd	nd	nd	46
oginova (1957)	11	ng	ng	air	air	bifarate tube		ng	MP	ng	yes	no	yes	36
otsch (1997)	5	yes	yes	air	air	dynamic olfactometer	yes	ng	D, I	A- $\Phi$	yes	yes	yes	
Makhinya (1966)	19	yes	ng	air	carbon filtered	cylinder	yes	15 lpm	MP	ng	yes	yes	yes	36
Malyarova (1967)	ng	ng	ng	air	ng	ng	ng	ng	ng	ng	ng	ng	ng	2
Marin (1988)	8	yes	yes	air	air	G- olfactometry	yes	ng	D	D	yes	no	ng	54
Martirosyan (1970)	ng	ng	ng	air	ng	ng	ng	ng	ng	ng	ng	ng	ng	2
Mateson (1955)	ng	yes	yes	air	ng	glass funnel	ng	yes	ng	ng	ng	ng	ng	8, 29

Table 6.2 – Methods Summary of Reviewed Articles, cont.

Source (Year)	Panel			Presentation Apparatus					Presentation Method					Note #
	Size	Selection Criteria	Calibration	Vapor Modality	Diluent	Presentation Mode	Analytic Measure	Flow Rate	Threshold Type	Conc. Series	Trials	Forced - Choice	Conc. Interval	
May (1966)	16	yes	ng	air	air	flask	yes	static	D+R	U-D	yes	no	yes	60
McGinley (2003)	5	yes	no	air	air	Scentometer, Nasal Ranger	yes	16 - 20 lpm	D	A	yes	yes	yes	8
Melekhina (1958)	nd	nd	nd	air	nd	nd	nd	nd	MP	nd	nd	nd	nd	46
Melekhina (1968)	16	yes	ng	air	carbon filtered	cylinder	yes	15 lpm	MP	ng	yes	yes	yes	36
Melzner (2011)	25	yes	yes	air	air	olfactometer	yes	8 lpm	D	A	yes	yes	yes	9
Minaev (1966)	19	yes	ng	air	carbon filtered	cylinder	yes	15 lpm	MP	ng	yes	yes	yes	36
Miryakubova (1970)	ng	ng	ng	air	ng	ng	ng	ng	MP	ng	ng	ng	ng	2
Miyazawa (2009a)	12	yes	ng	air	water	GC-olfactometry	yes	30 lpm	D	A	yes	yes	yes	31
Mnatsakanyan (1962)	11	yes	ng	air	ng	ng	yes	ng	R	ng	yes	ng	yes	1
Molhave (2000)	12	yes	yes	Aor	air	olfactometer	ng	ng	D	A	no	yes	yes	18
Moskowitz (1974)	33	no	ng	air	air	nose port	yes	0.12 lpm	D	ng	ng	no	yes	40
Muhlen (1968)	4	ng	ng	air	air	hood	yes	static	R	U-D	ng	no	yes	37
Mukhamedova (1968)	22	yes	ng	air	carbon filtered	cylinder	yes	15 lpm	MP	ng	yes	yes	yes	36



Table 6.2 – Methods Summary of Reviewed Articles, cont.

Source (Year)	Panel			Presentation Apparatus					Presentation Method					Note #
	Size	Selection Criteria	Calibration	Vapor Modality	Diluent	Presentation Mode	Analytic Measure	Flow Rate	Threshold Type	Conc. Series	Trials	Forced - Choice	Conc. Interval	
Mukhitov (1971)	ng	ng	ng	air	ng	ng	ng	ng	ng	ng	ng	ng	ng	2
Mukhitov (1962)	14	nd	nd	air	nd	nd	nd	nd	MP	nd	nd	nd	nd	46
Mullins (1955)	9 - 23	no	no	air	air	nose port	no	4 lpm	R	ng	yes	no	ng	37
Murphy (1985)	9 - 20	yes	ng	air	air	dilution olfactometer	no	no	D	A + D	ng	yes	yes	63
Nader (1958)	10	no	no	air	pure air	mask	no	15-20 lpm	D	A	yes	no	yes	
Nagata (2003)	6	yes	yes	air	nitrogen	tedlar bag	yes	static	D	ng	ng	yes	ng	
Neuhaus (1957)	nd	nd	nd	air	nd	nd	nd	nd	D + R	nd	nd	nd	no	28
Nevers (1965)	6+	ng	ng	air	pure air	funnel	ng	1.25 lpm	I	R	ng	yes	yes	58
Nikiforov (1970)	ng	ng	ng	air	ng	ng	ng	ng	ng	ng	ng	ng	ng	2
Nimmermark (2011)	4 - 16	yes	yes	ng	ng	olfactometer	ng	ng	D + R	ng	yes	yes	yes	
Nishida (1975)	20	ng	ng	air	Fresh air	mask	ng	1 lpm	D	ng	ng	no	yes	
Nishida (1979)	8 - 11	ng	ng	air	carbon	mask	yes	2 lpm	D	A + D	yes	no	yes	19
Nordin (1997)	16	yes	yes	air	air	dynamic olfactometer	yes	100 lpm	D	yes	yes	yes	yes	39

Table 6.2 – Methods Summary of Reviewed Articles, cont.

Source (Year)	Panel			Presentation Apparatus					Presentation Method					Note #
	Size	Selection Criteria	Calibration	Vapor Modality	Diluent	Presentation Mode	Analytic Measure	Flow Rate	Threshold Type	Conc. Series	Trials	Forced - Choice	Conc. Interval	
Novikov (1957)	12	yes	ng	air	ng	ng	ng	ng	MP	ng	yes	ng	yes	36
Odoshashvili (1962)	12	yes	ng	air	carbon filtered	cylinder	yes	15 lpm	MP	nd	yes	yes	yes	36
Olsson (2010)	500	yes	no	air	Diethyl Phthalate	squeeze bottles	no	static	D	A	yes	yes	yes	12
Ozturk (1976)	12-26	yes	ng	water vapor	Distilled water	aerosol bottle	yes	static	D	A	ng	no	yes	
Pangborn (1964)	5	yes	ng	air	pure air	hood	no	ng	D	R	yes	yes	yes	
Parker (1913)	2	ng	ng	air	air	jar	no	static	R	ng	yes	yes	yes	37
Patterson (1993)	40	yes	ng	air	mineral oil	squeeze bottles	yes	static	D	A + D	yes	yes	no	
Piggott (1975)	10	yes	ng	water	water	bottle	no	static	ng	R	yes	no	yes	20
Pliska (1965)	nd	nd	nd	air	nd	nose port	nd	nd	nd	ng	A	nd	nd	
Plotnikova (1957)	nd	nd	nd	air	nd	nd	nd	nd	MP	nd	nd	nd	nd	46
Pogosyan (1965)	18	ng	ng	air	ng	ng	ng	ng	MP	ng	ng	ng	ng	
Polednik (2008)	22	yes	ng	air	air	Room	yes	ng	D	ng	yes	yes	yes	
Polgar (1975)	6	yes	ng	air	pure air	cup	ng	3 lpm	R	A	ng	yes	yes	

**Table 6.2 – Methods Summary of Reviewed Articles, cont.**

	Panel			Presentation Apparatus					Presentation Method					
Source (Year)	Size	Selection Criteria	Calibration	Vapor Modality	Diluent	Presentation Mode	Analytic Measure	Flow Rate	Threshold Type	Conc. Series	Trials	Forced - Choice	Conc. Interval	Note #
Poostchi (1986)	7 to 10	ng	ng	ng	ng	olfactometer	yes	01 lpm	D R	ng	yes	yes	yes	
Popov (1970)	ng	ng	ng	air	ng	ng	ng	ng	MP	ng	ng	ng	ng	
Pozzani (1968)	8 - 9	no	ng	air	air	test room	yes	static	D	R	yes	no	yes	3
Prusakov (1976)	7	nd	nd	air	nd	nd	nd	nd	MP	nd	nd	nd	nd	46
Punter (1983)	26 - 44	nd	nd	air	nitrogen	port	yes	5 lpm	D	A	yes	yes	yes	
Randebroek (1971)	5	ng	ng	air	ng	sniff port	yes	ng	ng	ng	1-5	ng	ng	1
Ripp (1968)	16	yes	ng	air	pure air	cylinder	yes	15 lpm	MP	ng	yes	yes	yes	36
Rylova (1953)	ng	ng	ng	air	ng	ng	ng	ng	ng	ng	ng	ng	ng	1
Sadilova (1968)	17	yes	ng	air	carbon filtered	cylinder	yes	15 lpm	MP	ng	yes	yes	yes	36
Saifudinov (1966)	22	yes	ng	air	carbon filtered	cylinder	yes	15 lpm	MP	ng	yes	yes	yes	36
Sanders (1970)	4	yes	yes	air	Clean air	mask	no	12 lpm	D	A	yes	no	yes	37
Savenhed (1985)	ng	ng	ng	air	ng	G- olfactometry	yes	ng	D	ng	ng	ng	ng	
Scherberger (1958)	3	no	no	air	air	glass	yes	07 or 31 lpm	R	ng	yes	no	ng	1, 37

Table 6.2 – Methods Summary of Reviewed Articles, cont.

	Panel			Presentation Apparatus					Presentation Method					
Source (Year)	Size	Selection Criteria	Calibration	Vapor Modality	Diluent	Presentation Mode	Analytic Measure	Flow Rate	Threshold Type	Conc. Series	Trials	Forced - Choice	Conc. Interval	Note #
Scherberger (1960)	ng	ng	ng	ng	ng	ng	ng	ng	D	ng	ng	ng	ng	
Schmidt (2010)	ng	ng	ng	ng	ng	ng	ng	ng	ng	ng	ng	ng	ng	16
Schneider (1955)	53	yes	ng	air	Odor free air	test room	no	static	D	R + D + A	yes	no	ng	20
Schneider (1966)	8	yes	ng	air	nitrogen	nares piece	yes	0.6-4.8 lpm	R	A	yes	no	ng	1
Schulman (2011)	ng	ng	ng	ng	ng	ng	ng	ng	ng	ng	ng	ng	ng	61
Selyuzhitskii (1976)	ng	ng	ng	air	ng	ng	ng	ng	ng	ng	ng	ng	ng	2
Sgibnev (1968)	nd	nd	nd	air	nd	nd	nd	nd	MP	nd	nd	nd	nd	46
Shalamberidze (1967)	14-15	ng	ng	air	ng	ng	ng	ng	MP	ng	ng	ng	ng	
Shusterman (1997a)	30	yes	no	air	air	nasal canula	no	5 lpm	I (Irritation)	A	no	no	no	49
Sinkuvene (1970)	ng	ng	ng	air	ng	ng	ng	ng	MP	ng	ng	ng	ng	
Slavgorodskiy (1968)	27	ng	ng	air	ng	ng	ng	ng	MP	ng	yes	ng	yes	
Slotnick (1984)	10	ng	ng	ng	air	olfactometer	ng	ng	ng	ng	ng	ng	ng	
Smeets (2002)	26	yes	yes	air	propylene glycol, mineral oil	bottle	Y	static	D and I	A + D	yes	yes	yes	47

**Table 6.2 – Methods Summary of Reviewed Articles, cont.**

	Panel			Presentation Apparatus					Presentation Method					
Source (Year)	Size	Selection Criteria	Calibration	Vapor Modality	Diluent	Presentation Mode	Analytic Measure	Flow Rate	Threshold Type	Conc. Series	Trials	Forced - Choice	Conc. Interval	Note #
Smeets (2007)	24	yes	yes	air	water	bottle	yes	static & Dynamic	D and I	A	yes	yes	yes	49
Smith (1969)	7	ng	ng	air	nitrogen	nose piece	yes	static	R	D	ng	yes	yes	20
Solomin (1961)	nd	nd	nd	air	nd	nd	nd	nd	MP	nd	nd	nd	nd	46
Solomin (1964)	14	yes	ng	air	carbon	cylinder	yes	15 lpm	MP	ng	yes	yes	yes	36
Stalker (1963)	23	yes	no	air	pure air	mask	yes	15-20 lpm	D	A	yes	no	yes	
Steinmetz (1969)	5	yes	ng	air	Odor free air	hood	no	yes	D	R	yes	yes	ng	20, 29
Stephens (1971)	ng	ng	ng	air	pure air	sniff port	yes	yes	ng	ng	ng	yes	ng	1, 29
Stevens (1993)	24	yes	no	air	water	squeeze bottles	yes	static	D	A	yes	yes	yes	63
Stevens (1988)	3	no	yes	ng	water, mineral oil	squeeze bottles	yes	ng	D	A	ng	yes	yes	
Stewart (1974)	9	yes	ng	air	Room air	test room	yes	static	D	R	yes	no	yes	38
Stone (1965)	9	yes	ng	air	Charcoal filtered	hood	yes	yes	D	R	yes	yes	yes	20, 29
Stone (1967b)	6	yes	ng	air	Odor free air	hood	no	yes	D	R	yes	yes	yes	20, 29
Stone (1962)	48	yes	ng	air	Charcoal filtered	hood	yes	yes	D	R	yes	yes	yes	20, 29

Table 6.2 – Methods Summary of Reviewed Articles, cont.

Source (Year)	Panel			Presentation Apparatus					Presentation Method					Note #
	Size	Selection Criteria	Calibration	Vapor Modality	Diluent	Presentation Mode	Analytic Measure	Flow Rate	Threshold Type	Conc. Series	Trials	Forced - Choice	Conc. Interval	
Stone (19623b)	54	no	ng	air	pure air	hood	yes	yes	D	R	yes	yes	yes	20, 29
Stone (1963a)	6	no	ng	air	Charcoal filtered	hood	no	yes	D	R	yes	yes	yes	20, 29
Stone (1967a)	9	yes	ng	air	Charcoal filtered	hood	no	yes	D + R	R	yes	yes	yes	20, 29
Stone (1972)	3 - 5	yes	no	air	air	nose port	yes	yes	D	R	ng	yes	yes	20, 29
Strube (2012)	10	yes	yes	air	air and water	GC-olfactometry	yes	ng	D	ng	ng	ng	ng	
Styazhkin (1973)	17	yes	ng	air	pure air	ng	ng	ng	MP	ng	ng	yes	yes	
Tabakova (1969)	23	yes	ng	air	ng	ng	ng	ng	MP	ng	ng	ng	yes	
Takhiroff (1957)	nd	nd	nd	air	nd	nd	nd	nd	MP	nd	nd	nd	nd	45
Takhirov (1969)	ng	yes	ng	air	carbon filtered	cylinder	yes	15 lpm	MP	ng	yes	yes	yes	36
Tamman (1928)	3-7	ng	ng	air	air	nd	no	yes	D	A	ng	no	yes	29, 37
Tarkhova (1965)	20	yes	ng	air	air	cylinder	yes	15 lpm	MP	ng	yes	yes	yes	36
Tepikina (1968)	24	yes	ng	air	carbon filtered	cylinder	yes	15 lpm	MP	ng	yes	yes	yes	36
Teranishi (1974)	ng	ng	ng	water	water	ng	ng	static	ng	ng	ng	ng	ng	1

**Table 6.2 – Methods Summary of Reviewed Articles, cont.**

Source (Year)	Panel			Presentation Apparatus					Presentation Method				Note #	
	Size	Selection Criteria	Calibration	Vapor Modality	Diluent	Presentation Mode	Analytic Measure	Flow Rate	Threshold Type	Conc. Series	Trials	Forced - Choice		Conc. Interval
Thiele (1979)	3, 15, 150	nd	nd	air	Activated carbon	port	nd	nd	D	nd	nd	yes	yes	
Thriel (2006a)	144	yes	yes	air	water or Mineral Oil	bottle	yes	static	D	A	yes	yes	yes	51
Tkach (1965)	16	yes	ng	air	carbon filtered	cylinder	yes	15 lpm	MP	ng	yes	yes	yes	36
Tkachev (1963)	17	nd	nd	air	nd	nd	nd	nd	MP	nd	nd	nd	nd	46
Tkachev (1969)	ng	ng	ng	air	ng	ng	ng	ng	ng	ng	ng	ng	ng	1
Tkachev (1970)	21	ng	ng	air	ng	ng	ng	ng	ng	ng	ng	ng	ng	
Torkelson (1977)	10	no	no	ng	ng	ng	ng	ng	R	A	ng	ng	yes	
Tsukatani (2003)	31	ng	ng	ng	several	ng	ng	ng	D	A	ng	ng	ng	
Turk (1973)	ng	no	no	air	ambient air	vent	yes	Varied	R	D	yes	no	yes	20
Ubaidullaev (1978)	nd	nd	nd	air	nd	nd	nd	nd	MP	nd	nd	nd	nd	46
Ubaidullaev (1961)	nd	nd	nd	air	nd	nd	nd	nd	MP	nd	nd	nd	nd	46
Ubaidullaev (1966)	25	ng	ng	air	air	ng	yes	ng	MP	ng	ng	ng	ng	
Ueno (2009)	6, 12, 51	yes	yes	ng	air	olfactometer	yes	ng	D	D + A	ng	Triangle Bag Method	yes	

Table 6.2 – Methods Summary of Reviewed Articles, cont.

Source (Year)	Panel			Presentation Apparatus					Presentation Method					Note #
	Size	Selection Criteria	Calibration	Vapor Modality	Diluent	Presentation Mode	Analytic Measure	Flow Rate	Threshold Type	Conc. Series	Trials	Forced - Choice	Conc. Interval	
Ventura (1997)	5	ng	ng	air	water, air	GC / Sniffer	yes	71 cm/s	R	A	ng	ng	yes	
Vermeulen (2006)	2	ng	ng	air	air	GC-olfactometry	yes	20ml/min	D	A	ng	ng	yes	
Viswanathan (1983)	17	no	no	ng	ng	bag and olfactometry	ng	static and dynamic	D	A	ng	yes	yes	
Walker (1996)	ng	ng	ng	ng	ng	ng	ng	ng	ng	ng	ng	ng	ng	
Walker (2003)	7 & 5	yes	yes	air	air	olfactometer	yes	43 lpm	D	A	Y	no	yes	
Weeks (1960)	12	ng	ng	ng	air	Fair Wells Osmoscope	ng	ng	R	ng	ng	no	ng	
Whisman (1978)	6	yes	ng	air	air	test room	yes	yes	D	A	yes	no	yes	29
Wilby (1964)	3-4	yes	ng	air	air	10-inch square port	no	2830 lpm	D	A	yes	no	yes	37
Wilby (1969)	35	no	ng	air	air	10-inch square port	yes	2830 lpm	R	R	yes	no	yes	42
Williams (1977)	10	yes	nd	air	Clean air	nose port	no	0.5 lpm	D	A	ng	yes	yes	
Winneke (1979)	31	nd	nd	air	nd	hood	yes	yes	D	A	nd	no	yes	29
Wise (2007)	20	yes	yes	nitrogen	air	olfactometer	yes	30 lpm	R	A	yes	yes	yes	
Witheridge (1939)	2	yes	ng	air	Clean air	test room	yes	static	D	ng	ng	no	yes	37



Table 6.2 – Methods Summary of Reviewed Articles, cont.

Source (Year)	Panel			Presentation Apparatus					Presentation Method				Note #	
	Size	Selection Criteria	Calibration	Vapor Modality	Diluent	Presentation Mode	Analytic Measure	Flow Rate	Threshold Type	Conc. Series	Trials	Forced - Choice		Conc. Interval
Yang (2008)	3	ng	ng	ng	hexane	GC-olfactometry	yes	2 ml/min	D	ng	ng	no	yes	
Young (1966)	81	yes	ng	air	Filtered air	mask	yes	57 lpm	D	A	yes	no	yes	
Yuldashev (1965)	20	ng	ng	air	ng	ng	yes	ng	MP	ng	ng	ng	ng	
Zarzo (2012)	ng	ng	ng	ng	ng	ng	ng	ng	D	ng	ng	ng	ng	
Zibireva (1967)	ng	ng	ng	air	pure air	cylinder	yes	15 lpm	MP	ng	yes	yes	yes	36
Ziemer (2000)	10	ng	ng	air	air	Devlin olfactometer	yes	ng	D	A	yes	no	ng	

## **Notes to Table 6.2**

- 1 A project note about an experimental paper presenting threshold values.
- 2 Abstract with insufficient information.
- 3 Adaptation effects were avoided with a 45-min interval between concentrations.
- 4 Although a random presentation was used in this study, adaptation effects were avoided by presenting stimuli with 30 minute intervals between concentrations.
- 5 Approximate thresholds determined and no threshold methodology is given.
- 6 Article only provided the range of all measurements and the values detected 90% of the time.
- 7 Article investigated an additive effect of odorants.
- 8 Article focused upon validating olfactometer(s).
- 9 Article investigated whether subjects detected CO<sub>2</sub> in the nose or the mouth first.
- 10 Article investigates the odor detection, discrimination and chemesthetic properties.
- 11 Article investigating nasal irritation sensitivity variation in humans.
- 12 Article investigating odor threshold differences between males, females, osmics and anosmics.
- 13 Article contains good descriptions for the compounds found in orange peel vapor.
- 14 Article investigating the compounds and their organoleptic intensity scales.
- 15 Article mentions new olfactometer. Flow rate difficult to determine.
- 16 Article on good odor measurement methods/studies and the vapor delivery device 8 (VDD8).
- 17 Article refers to a minimal perceptible concentration based on an intensity scale.
- 18 Article refers to a previously published articles for the details of the odor testing. Results are for brief, 2 minute duration, exposures only.
- 19 Ascending/descending patterns with consideration of other factors of the experimental design.
- 20 Concentration series are presented with insufficient time for de-adaptation of the olfactory receptors.
- 21 Concentration series not given, however the 1-hr waiting period used would eliminate adaptation effects.
- 22 Concentrations are presented in ascending, descending, and random order.
- 23 Detection threshold was a 50% response.
- 24 Different subjects were tested at different concentrations to eliminate adaptation effects.
- 25 Evaluation of the repeatability of odor threshold data; determining the precision of odor threshold identification methods. air-dilution olfactometer had good precision (4.2%).
- 26 Experimental purpose was to avoid inhibition.
- 27 Eye irritation and pungency was measured.
- 28 German article. A tenfold concentration step size was used.
- 29 Flow rate difficult to determine.
- 30 Panel was 50% anosmic.
- 31 Investigation of how the detection threshold might change when compounds are presented in mixtures.
- 32 Investigation of olfactory properties of chemicals under hyperbaric atmospheres.
- 33 Investigation of the properties affecting odor thresholds in hydroalcoholic solutions (like wine).
- 34 Investigation to identify and quantify the odorants from apple.
- 35 Investigation to identify the odorants and thresholds from linden tree honey from Romania.
- 36 The MP is the minimum perceptible concentration of the most sensitive subject.
- 37 Number of subjects was insufficient to represent the range of olfactory sensitivity.

- 38 Only one concentration per day was tested to avoid adaptation effects.
- 39 Only the detection threshold for the Controls (without Alzheimers disease) were quoted.
- 40 Panelists completed four scaling tasks in 30 min with 10-sec waiting period between sniffs.
- 41 Participant count is the lowest number of subjects per compound.
- 42 Random presentation order to determine recognition threshold.
- 43 Reported values are for 100 percent recognition.
- 44 Results displayed on small graph in log ppb units; conversion errors may have resulted during conversion.
- 45 Russian article minimal perceptible value was determined from English summary.
- 46 Russian article was categorized based on translation of key words and review of tables presenting minimum perceptible values.
- 47 Study focus was testing olfactory fatigue between exposed and non-exposed workers.
- 48 Study investigated the odor threshold differences between smokers and non-smokers.
- 49 Study of the odor and chemesthesis (pungency and eye irritation).
- 50 Study of the odorant extracts of Lovage using GC-O.
- 51 Study on possible odorants for inert gas and investigated differences in age, sex, and smoking.
- 52 Study to compare the odor detection thresholds for smokers and non-smokers.
- 53 Study to determine the odor recognition thresholds of several organics.
- 54 Study to identify the odor detection thresholds of common food odorants.
- 55 Study to identify the odor thresholds of chemicals in drinking water.
- 56 The study presents air values based on transformed data from water values and a descending series without adequate de-adaptation time.
- 57 Variable presentation was used with intervals between sniffs to reduce adaptation effects.
- 58 Threshold was calculated from the intensity slope at the intercept.
- 59 Thresholds were conducted as training for a field program. Threshold measurement recorded to document panel calibration.
- 60 Up-down technique used is less likely to cause olfactory fatigue than a descending or random pattern.
- 61 U.S. EPA Report on odor detection of methyl tert-butyl ether in water based upon on previously published data.
- 62 Purpose of the experiment was to prove hypothesis on the effects of humidity and temperature on odor thresholds.
- 63 Investigation of the relationship between odor detection thresholds and age.
- 64 Study of glutaraldehyde odor detection threshold, eye and throat sensation threshold, and response over time.
- 65 Only dilution to threshold values presented.
- 66 Study evaluated subjects ability to recognize the odorant, not to determine a threshold.
- 67 Study to evaluate the effect of the molecule length (carbon atoms) on the odor detection threshold.
- 68 Study comparing age-related loss of detection threshold, intensity, pleasantness and repeated exposure effects.
- 69 Using the triangle odor bag method, 12 years of data on 223 compounds was summarized and trends were examined.
- 70 Study of individual's odor detection and hedonic tone from animal production facility odorants.
- 71 An evaluation of the methodology and data analysis to identify the appropriate study size and trials.

### **Table 6.3 – Reported Odor Thresholds from All Sources**

All published odor threshold values for the 295 chemicals with occupational exposure values.

The table provides the following information:

- Chemical Name
- Source (Last name of first author) and publication date
- Type of odor threshold values reported as either detection (d) or recognition (r)
- All threshold values from the Gemert compendium in both mg/m<sup>3</sup> and ppm.

**Note:** *Conversion of units from mg/m<sup>3</sup> to ppm was based on the molecular weight of the compound and the known volume of a perfect gas or vapor at standard temperature and pressure (STP).*

**Table 6.3 – Odor Threshold Values**

*Bold = Lowest Value Reported*

#	Chemical Name	Source	Type of Threshold	Odor Thresholds	
				mg/m <sup>3</sup>	ppm
1	Acetaldehyde	Zwaardemaker 1914	d	0.7	0.39
		Backman 1917	r	0.062 - 0.075	0.034 - 0.042
		Katz & Talbert 1930	d	0.12	0.067
		Balavoine 1943		10	6
		Pliska & Janicek 1965		1,800	1,000
		Gofmekler 1967, 1968	d	0.012	0.0067
		Leonardos et al 1969	r	0.38	0.21
		Hartung et al 1971		0.005	0.0028
		Takhirova 1974		0.49	0.27
		Teranishi et al 1974		0.041	0.023
		Anon. 1980	d	0.0027	<b>0.0015</b>
		Anon. 1980	r	0.027	0.015
		Naus 1982	d	1	0.555
		Naus 1982	r	10	6
		Nagy 1991	d	0.09	0.05
Nagata 2003	d	0.0027	0.0015		
2	Acetic Acid	Passy 1893b, 1893c	d	5 - 10	2.0 - 4.1
		Grijns 1906		49 - 76	20 - 31
		Backman 1917	r	4.8 - 5.0	2.0 - 2.0
		Grijns 1919		2	0.81
		Mitsumoto 1926	r	0.074 - 0.57	0.030 - 0.23
		Hesse 1926	r	0.6	0.24
		Henning 1927	d	3.6	1.5
		Morimura 1934	r	1.82 - 1.91	0.74 - 0.78
		Jung 1936	d	0.025	0.01
		Jung 1936	r	0.05	0.02
		Balavoine 1943, 1948		300 - 500	122 - 204
		Stone 1963c	d	3.9	1.6
		Stone & Bosley 1965	d	4.2	1.7
		Endo et al 1967		6.5	2.65
		Takhirova 1969, 1974		0.6	0.24
		Leonardos et al 1969	r	2.5	1
		Homans et al 1978	d	0.37	0.15
		Naus 1982	d	0.5	0.20
		Naus 1982	r	25	10
		Punter 1983	d	0.09	0.037
		Homans 1984		0.93	0.38
		Walker et al 1990		5	2.04
		Nagy 1991	d	0.37	0.15
		Blank & Schieberle 1993		0.03 - 0.09	0.012 - 0.037
		Walker et al 1996		0.25 - 2.5	0.1 - 1.0
		Cometto - Muniz et al 1998a	d	0.025	0.01
		Cometto - Muniz 1999	d	0.025	0.01

Table 6.3 – Odor Threshold Values, cont.

*Bold = Lowest Value Reported*

#	Chemical Name	Source	Type of Threshold	Odor Thresholds	
				mg/m <sup>3</sup>	ppm
2	Acetic Acid cont.	Nagata 2003	d	0.015	0.006
		Van Thriel et al 2006	d	1.45	0.59
		Wise et al 2007	d	0.017 - 0.020	0.0069 - 0.0081
		Miyazawa et al 2009a	d	0.017 - 0.020	0.0069 - 0.0081
		Miyazawa et al 2009b	d	0.001	<b>0.0004</b>
		Cain et al 2010	d	0.15	0.06
		Cometto - Muniz & Abraham 2010b	d	0.013	0.0053
3	Acetic Anhydride	Takhirova 1969		0.49	<b>0.12</b>
		Hellman & Small 1973a,b, 1974	d	<0.6	<0.14
		Hellman & Small 1973a,b, 1974	r	1.5	0.36
4	Acetone	Zwaardemaker 1914, 1927	d	4 - 7	1.7 - 2.9
		Backman 1917	r	4.1 - 4.3	1.7 - 1.8
		Van Anrooij 1931	d	1.1	0.46
		Jung 1936	d	78	33
		Jung 1936	r	78	33
		Scherberger et al 1958	r	1,900	800
		Stuiver 1958	d	5.8	2.4
		Feldman 1960		1.1	0.46
		Naus 1962	d	4	1.7
		Pogosyan 1965		1.1	0.46
		Tkach 1965		1.1	0.46
		May 1966	d	770	324
		May 1966	r	1,660	699
		Kittel 1968		11 - 240	4.6 - 101
		Leonardos et al 1969	r	240	101
		Kittel & Wendelstein 1971	d	75	32
		Kittel & Wendelstein 1971	r	121	51
		Hartung et al 1971		2.3	0.97
		Dravnieks & Laffort 1972		240	10
		Artho & Koch 1973		1,000 - 10,000	421 - 4,208
		Hellman & Small 1973a,b, 1974	d	48	20
		Hellman & Small 1973a,b, 1974	r	78	33
		Dravnieks 1974	d	1,550	653
		Takhirova 1974		1.15	0.48
		Makeicheva 1978		0.94	<b>0.4</b>
		Anon. 1980	d	72	30
Anon. 1980	r	264	111		
Naus 1982	d	1	0.42		
Naus 1982	r	20	8		

**Table 6.3 – Odor Threshold Values, cont.**

*Bold = Lowest Value Reported*

#	Chemical Name	Source	Type of Threshold	Odor Thresholds	
				mg/m <sup>3</sup>	ppm
4	Acetone cont.	Punter 1983	d	8.6	3.6
		Nagy 1991	d	40	17
		Cometto - Muniz & Cain 1993	d	27,900	11,745
		Cometto - Muniz 1993	d	27,900	11,745
		Dalton et al 1997a	d	199 - 204	84 - 86
		Dalton et al 1997b	d	626 - 936	263 - 394
		Wysocki et al 1997	d	97 - 2,026	41 - 853
		Dalton et al 2000	d	59	25
		Nagata 2003	d	101	43
Cometto - Muniz & Abraham 2009a	d	2	0.84		
5	Acetonitrile	Pozzani et al 1959		<67	<40
		Dravnieks & Laffort 1972		285	170
		Dravnieks 1974	d	1,950	1,161
		Nagata 2003	d	22	<b>13</b>
6	Acetophenone	Imasheva 1963		0.01	0.002
		Tkach 1965		0.01	0.002
		Korneev 1965		0.01	0.002
		Gavaudan & Poussel 1966		0.23	0.047
		Hellman & Small 1973a,b, 1974		1.5	0.305
		Hellman & Small 1973a,b, 1974	r	2.9	0.59
		Savenhed et al 1985	d	0.01 - 0.04	0.002 - 0.008
		Randebroek 1986		0.0012	<b>0.00024</b>
7	Acetylene	Deadman & Prigg 1959	d	240	<b>226</b>
		Babin et al 1965		1,300 - 2,750	1,222 - 2,584
		Nagy 1991	d	510	479
8	Acrolein	Katz & Talbert 1930	d	4.1	1.8
		Plotnikova 1957		0.8	0.35
		Buchberg et al 1961		0.2 - 0.7	0.087 - 0.31
		Leonardos et al 1969	r	0.48	0.21
		Sinkuvene 1970		0.07	0.031
		Knuth 1973		0.14	0.061
		Cormack et al 1974		0.23	0.1
		Teranishi et al 1974		0.05	0.022
		Anon. 1980	d	0.069	0.03
		Anon. 1980	r	0.32	0.14
		Nagata 2003	d	0.0083	<b>0.0036</b>
9	Acrylic Acid	Hellman & Small 1974	d	0.27	<b>0.092</b>
		Hellman & Small 1974	r	3	1
		Piringer & Granzer 1984		2	0.679
		Van Thriel et al 2006	d	1.5	0.51

**Table 6.3 – Odor Threshold Values, cont.***Bold = Lowest Value Reported*

#	Chemical Name	Source	Type of Threshold	Odor Thresholds	
				mg/m <sup>3</sup>	ppm
10	Acrylonitrile	Stalker 1963	d	3.4	<b>1.6</b>
		Leonardos et al 1969	r	47	22
		Nagata 2003	d	19	8.8
11	Allyl Alcohol	Katz & Talbert 1930	d	3.3	1.4
		Jones 1955c	d	83	35
		Dunlap et al 1958		1.9	0.8
		Pliska & Janicek 1965		48	20
		Dravnieks 1974	d	5	2.1
		Dravnieks & Laffort 1972		1.2	<b>0.51</b>
		12	Allyl Chloride	Toxicity Data Sheet 1958a	
Torkelson et al 1959		3 - 9		0.958 - 2.875	
Leonardos et al 1969	r	1.5		<b>0.48</b>	
13	Allyl Isothiocyanate	Allison & Katz 1919		8	1.97
		Katz & Talbert 1930		0.61	0.15
		Stone et al 1967a	d	0.19	0.05
		Stone & Pryor 1967b	d	0.037 - 0.24	<b>0.0091</b> - 0.0592
14	Ammonia	Valentin 1848, 1850		21	30
		Grijns 1906		21.6 - 42	31.0 - 60.3
		Fieldner et al 1921		37	53
		Smolczyk & Cobler 1930		0.71 - 7.1	1.02 - 10.2
		Geier 1936	d	1.25	1.79
		Geier 1936	r	2.5	3.6
		Carpenter et al 1948		0.7	1
		Smyth 1956	r	≤0.7	≤1.00
		Patty 1962a		<3.5	<5.0
		Saifutdinov 1966		0.50 - 0.55	0.72 - 0.79
		Endo et al 1967		37	53
		Leonardos et al 1969	r	33	47
		Hamanabe et al 1969		0.03	<b>0.043</b>
		Stephens 1971		2.7	3.9
		Nishida et al 1975	d	1.8 - 37.8	2.6 - 54.3
		Hill & Barth 1976		21	30
		Schoedder 1977		5.0 - 7.6	7.2 - 10.9
		Logtenberg 1978	d	5.2	7.5
		Nishida et al 1979	d	11.6	16.7
		Anon. 1980	d	0.1	0.14
		Anon. 1980	r	0.4	0.57
		Naus 1982	d	1.5	2.15
		Naus 1982	r	35	50
		Nagy 1991	d	3.7	5.31
		Nagata 2003	d	1.1	1.58
		Van Thriel et al 2006	d	0.04	0.057
		Smeets et al 2007	d	1.8	2.58
		Smeets et al 2007	d	1.8	2.58



**Table 6.3 – Odor Threshold Values, cont.**

*Bold = Lowest Value Reported*

#	Chemical Name	Source	Type of Threshold	Odor Thresholds	
				mg/m <sup>3</sup>	ppm
15	n - Amyl Acetate	Grijns 1919		0.9	0.17
		Allison & Katz 1919		39	7.3
		Jones 1955c	d	1.6	0.3
		Gofmekler 1960		0.6	0.11
		Pliska & Janicek 1960		31	5.8
		Guadagni 1966		0.05	0.0094
		Davis 1973	d	0.04	0.0075
		Hendriks 1979	d	0.27	0.051
		Slotnick 1981		1.3	0.24
		Laing 1982	d	0.95	0.178
		Punter 1983	d	0.27 - 0.28	0.051 - 0.053
		Cristoph 1983	r	0.045 - 0.06	0.00845 - 0.0113
		Walker et al 1990		6.9	1.3
		Cometto - Muniz & Cain 1991		6.3	1.18
		Cometto - Muniz 1993	d	6.3	1.18
		Walker et al 1996		0.53 - 5.3	0.09954 - 0.9954
		Hoshika et al 1997	r	41	7.7
		Ziemer et al 2000	d	0.049	0.0092
		Walker et al 2003		0.038 - 0.89	<b>0.007</b> - 0.167
		Komthong et al 2006		10.7 - 230	2.0 - 43
Olsson & Laska 2010	d	2.2 - 2.7	0.414 - 0.508		
16	Aniline	Tempelaar 1913	d	0.97	0.25
		Huijer 1924	d	0.046	<b>0.012</b>
		Zwaardemaker 1927	d	0.046	0.012
		Backman 1917	r	5.0 - 5.8	1.3 - 1.5
		Geier 1936	d	1.2 - 1.5	0.32 - 0.39
		Geier 1936		2.0 - 2.5	0.53 - 0.66
		Jacobson et al 1958	d	38	10
		Tkachev 1963		0.37	0.097
		Leonardos et al 1969	r	3.8	1
		Ozturk 1976	d	2.21	0.58
		Naus 1982	d	2	0.53
		Naus 1982	r	20	5.3
		17	Arsine	Patty 1962b	
18	Benzaldehyde	Backman 1917	r	0.33 - 0.50	0.05 - 0.076
		Rocen 1920	r	1.7	0.260
		Ohma 1922	d	0.44	0.067
		Katz & Talbert 1930		0.18	0.027
		Jones 1955c	r	4.1	0.626
		Pliska & Janicek 1965		13	2.0
		Knuth 1973		0.27	0.041
		Laing 1975	d	4.3	0.657
		Nishida et al 1979	d	3,400	783

Table 6.3 – Odor Threshold Values, cont.

Bold = Lowest Value Reported

#	Chemical Name	Source	Type of Threshold	Odor Thresholds	
				mg/m <sup>3</sup>	ppm
18	Benzaldehyde cont.	Randebrook 1986		0.014	0.0021
		Stevens & Cain 1987a	d	0.43 - 43	0.0657 - 6.57
		Khiari et al 1992	d	<0.01	<b>&lt;0.0015</b>
		Von Ranson & Belitz 1992b	d	0.61	0.093
		Von Ranson & Belitz 1992b	r	2.1	0.32
		McGee et al 1995	d	0.1 - 1	0.015 - 0.15
		Yang et al 2008		0.085	0.013
19	Benzene	Backman 1917	r	6.6 - 6.9	2.1 - 2.2
		Backman 1918		5 - 5.3	1.7
		Zwaardemaker 1927		5 - 5.3	1.7
		Grijns 1919		420	131
		Zwaardemaker 1927		420	131
		Schley 1934	d	8.8	2.8
		Schley 1934	r	12	3.8
		Jones 1954	r	480 - 510	150 - 160
		Jones 1955c	d	180	56
		Novikov 1957		4.9	1.5
		Deadman & Prigg 1959	d	9	2.8
		Gusev 1965		2.8 - 4	0.88 - 1.3
		Naus 1962	d	6	1.9
		May 1966	d	180	56
		May 1966	r	310	97
		Elfimova 1966		2.5	0.78
		Schutte & Zubek 1967	r	310	97
		Leonardos et al 1969	r	15	4.7
		Alibaev 1970		2.9	0.91
		Dravnieks & O'Donnell 1971		38	12
		Koster 1971	d	37	12
		Dravnieks & Laffort 1972		32.5	10.2
		Laffort & Dravineks 1973		14.5	4.5
		Artho & Koch 1973		100 - 1,000	31.3 - 313
		Dravnieks 1974	d	380	119
		Naus 1982	d	1.5	<b>0.47</b>
		Naus 1982	r	16	5
		Punter 1983	d	108	34
Nagata 2003	d	8.6	2.69		
20	Benzoyl Chloride	Schley 1934	d	0.012 - 0.024	<b>0.0021</b> - 0.0042
		Schley 1934	r	0.012 - 0.036	0.0021 - 0.0063
21	Benzyl Acetate	Appell 1969		0.001	<b>0.00016</b>
		Koster 1971		85 - 135	14 - 22
22	Benzyl Chloride	Katz & Talbert 1930	d	0.21	<b>0.041</b>
		Leonardos et al 1969	r	0.24	0.046

**Table 6.3 – Odor Threshold Values, cont.**

*Bold = Lowest Value Reported*

#	Chemical Name	Source	Type of Threshold	Odor Thresholds	
				mg/m <sup>3</sup>	ppm
23	Biphenyl	Solomin 1961		0.06	0.0095
		Nagy 1991	d	0.0033	<b>0.00052</b>
24	Boron Trifluoride	Torkelson et al 1961		4.2	1.5
25	Bromine	Valentin 1848, 1850		3	0.46
		Henning 1924	d	0.2	0.031
		Rupp & Henschler 1967	d	<0.065	<b>&lt;0.0099</b>
		Rupp & Henschler 1967	r	>6.5	>0.99
		Leonardos et al 1969	r	0.3	0.046
		Randebrook 1986		0.9	0.14
26	Bromoform	Passy 1893a	d	2 - 5	<b>0.19</b> - 0.48
		Backman 1917	r	2.2 - 2.5	0.21 - 0.24
		Grijns 1919		150	15
		Rocen 1920	r	30	2.9
27	1,3 - Butadiene	Mullins 1955	r	169	76
		Deadman & Prigg 1959	d	2.1	0.95
		Ripp 1968		4	1.8
		Laffort & Dravnieks 1973		5.8	2.6
		Hellman & Small 1974	d	1	0.45
		Hellman & Small 1974	r	2.4	1.1
		Jeltes 1975		0.22	<b>0.099</b>
		Nagata 2003	d	0.51	0.23
28	Butane, all isomers	Patty & Yant 1929		12,000	5,048
		Mullins 1955	r	6,160	2,591
		Mullins 1955	r	1,370	576
		Schneider et al 1966		8,700	3,660
		Laffort & Dravnieks 1973		3,000	1,262
		Artho & Koch 1973		1 - 10	<b>0.421</b> - 4.21
		Nagata 2003	d	2,880	1,212
29	Butenes, all isomers	Katz & Talbert 1930		2.1	0.915
		Katz & Talbert 1930		4.8	2.09
		Katz & Talbert 1930		3	1.31
		Mullins 1955	r	39.2	17
		Mullins 1955	r	2,700	1,177
		Mullins 1955	r	28.5	12
		Mullins 1955	r	4,880	2126
		Krasovitskaya & Malyarova 1968		15.4	6.71
		Knuth 1973		1.2	0.523
		Anon 1980	d	15	6.5
		Anon 1980	r	46	20
		Nagata 2003	d	0.83	<b>0.362</b>
		Nagata 2003	d	23	10

**Table 6.3 – Odor Threshold Values, cont.**

*Bold = Lowest Value Reported*

#	Chemical Name	Source	Type of Threshold	Odor Thresholds	
				mg/m <sup>3</sup>	ppm
30	2 - Butoxyethanol	Hellman & Small 1973a,b; 1974	d	0.5	0.1
		Hellman & Small 1973a,b; 1974	r	1.7	0.35
		Nagy 1991		0.4	<b>0.08</b>
		Nagata 2003		1.3	0.27
31	2 - Butoxyethyl Acetate	Hellman & Small 1973a,b; 1974	d	0.7	<b>0.107</b>
		Hellman & Small 1973a,b; 1974	r	1.3	0.198
		Nagy 1991	d	6.5	0.99
32	n - Butyl Acetate	Backman 1917	r	1.3 - 1.7	0.27 - 0.36
		Jung 1936	d	0.044	0.0093
		Jung 1936	r	0.044 - 0.13	0.0093 - 0.027
		Scherberger et al 1958	r	96	20
		Gofmekler 1960		0.6	0.13
		Pliska & Janicek 1960		190	40
		Naus 1962	d	0.7	0.147
		May 1966	d	35	7.4
		May 1966	r	55	12
		Koster 1971	d	480 - 1,750	101 - 368
		Dravnieks & Laffort 1972		0.04	0.008
		Dravnieks 1974	d	3	0.63
		Hellman & Small 1974	d	0.03	0.0063
		Hellman & Small 1974	r	0.18	0.038
		Anon. 1980	d	0.32	0.067
		Anon. 1980	r	2.4	0.505
		Cristoph 1983	r	0.46 - 0.55	0.097 - 0.116
		Scharfenberger 1990		4	0.84
		Cometto - Muniz & Cain 1991, 1993	d	11.5	2.4
		Cometto - Muniz 1993	d	11.5	2.4
		Nagy 1991	d	1	0.21
		Nagy 1991	d	0.521	0.11
		Patterson et al 1993	d	7.7	1.6
		Ziemer et al 2000	d	0.061	0.013
		Cometto - Muniz et al 2002		0.00062	<b>0.00013</b>
		Cometto - Muniz et al 2003		0.009	0.0019
		Nagata 2003	d	0.077	0.016
		Cometto - Muniz et al 2004	d	0.015	0.003
		Komthong et al 2006		165 - 1,570	35 - 330
		Cometto - Muniz et al 2008	d	0.02	0.004
		Cain & Schmidt 2009	d	0.01	0.002

**Table 6.3 – Odor Threshold Values, cont.**

*Bold = Lowest Value Reported*

#	Chemical Name	Source	Type of Threshold	Odor Thresholds	
				mg/m <sup>3</sup>	ppm
33	sec - Butyl Acetate	Cometto - Muniz & Cain 1993	d	22.6	4.76
		Cometto - Muniz 1993	d	22.6	4.76
		Nagata 2003	d	0.012	<b>0.0025</b>
34	tert - Butyl Acetate	Cometto - Muniz & Cain 1993	d	6.2	1.31
		Cometto - Muniz 1993	d	6.2	1.31
		Nagata 2003	d	0.34	0.072
		Cain & Schmidt 2009	d	0.038	<b>0.008</b>
35	Butyl Acrylate	Anon. 1969		0.53	0.1
		Gemert 1973	d	0.005 - 0.01	0.00096 - 0.0019
		Anon. 1980	d	0.0015	<b>0.00029</b>
		Anon. 1980		0.014	0.0027
		Piringer & Granzer		0.01	0.0019
		Nagata 2003	d	0.0029	0.00055
36	n - Butyl Alcohol	Passy 1892c	d	1	0.33
		Backman 1917	r	0.35 - 0.6	0.12 - 0.20
		Zwaardemaker 1927		1	0.33
		Jung 1936	d	0.158 - 0.316	0.052 - 0.10
		Jung 1936	r	0.474 - 0.632	0.16 - 0.21
		Gavaudan et al 1948		0.15	0.049
		Mullins 1955	r	37.2	12
		Jones 1955a	r	3.1	1
		Jones 1955b	r	110 - 285	36 - 94
		Jones 1955c	d	42	14
		Scherberger et al 1958	r	45	15
		Janicek et al 1960		20	6.6
		Naus 1962	d	4	1.3
		Pliska & Janicek 1965		3,000	990
		Gavaudan & Poussel 1966		1.1	0.36
		May 1966	d	33	11
		May 1966	r	48	16
		Dravnieks & Krotoszynski 1968		1.35	0.45
		Khachatryan & Baikov 1969		1.2 - 2	0.7 - 4
		Cain 1969	r	60	20
		Corbit & Engen 1971		13 - 20	4.3 - 6.6
		Dravnieks & Laffort 1972		10	3.3
		Baikov & Khachatryan 1973		1.2	0.396
		Laffort & Dravnieks 1973		0.9	0.3
		Hellman & Small 1974	d	0.9	0.3
		Hellman & Small 1974	r	3	1
Moskowitz et al 1974		186	61		
Jones et al 1975		<132	<44		
Piggott & Harper 1975		4 - 1,000	1.3 - 330		

Table 6.3 – Odor Threshold Values, cont.

*Bold = Lowest Value Reported*

#	Chemical Name	Source	Type of Threshold	Odor Thresholds	
				mg/m <sup>3</sup>	ppm
36	n - Butyl Alcohol cont.	Dravnieks 1976	d	0.36 - 10.2	0.12 - 3.4
		Williams et al 1977	d	0.63 - 1.14	0.21 - 0.38
		Amoore & Buttery 1978	d	2.3	0.76
		Homans et al 1978	d	13.94	4.6
		Jones et al 1978	d	42 - 105	14 - 35
		Laing et al 1978	r	10.5	3.5
		Laing 1982	d	3	1
		Cain et al 1983	d	<4.2	<1.4
		Cain et al 1983	d	<4.2	<1.4
		Cristoph 1983	r	0.7 - 0.9	0.23 - 0.30
		Laing 1983		6	1.98
		Jensen & Flyger 1983		0.10 - 2.4	0.033 - 0.79
		Punter 1983	d	2.6 - 5.3	0.86 - 1.7
		Viswanathan et al 1983		1.26 - 2.4	0.42 - 0.79
		Homans 1984		21.5	7.09
		Murphy & Cain 1985	d	0.39 - 4.26	0.13 - 1.41
		Roos et al 1985	d	0.101 - 0.136	0.033 - 0.45
		Roos et al 1985	d	0.77	0.25
		Don 1986	d	0.77	0.25
		Ahlstrom et al 1986	d	0.136 - 0.224	0.045 - 0.074
		Dravnieks et al 1986	d	0.51 - 4.05	0.168 - 1.34
		Hartigh 1986	d	0.01 - 0.292	<b>0.0033</b> - 0.096
		MacLeod et al 1986		0.69	0.23
		Poostchi et al 1986	d	0.99 - 1.85	0.33 - 0.61
		Poostchi et al 1986	r	3.72 - 4.02	1.23 - 1.33
		Cain et al 1988		1.4	0.46
		Dollnick et al 1988		0.384	0.13
		Stevens et al 1988	d	0.36 - 3.3	0.12 - 1.09
		De Wijk 1989		4.43	1.46
		Hermans 1989		0.15 - 0.214	0.049 - 0.071
		Cometto - Muniz & Cain 1990	d	5.4	1.78
		Cometto - Muniz 1993	d	5.4	1.78
		Scharfenberger 1990		0.5	0.16
		Cain & Gent 1991	d	3 - 9	1 - 3
		Laska & Hudson 1991	d	0.79	0.26
		Lea & Ford 1991		2	0.66
		Nagy 1991	d	3.1	1.02
		Nagy 1991	d	0.591	0.19
		Cometto - Muniz & Cain 1993	d	162	53
		Patterson et al 1993	d	5.4	1.78
Stevens & Dadarwala 1993	d	0.48 - 38.4	0.16 - 13		
Dalton et al 1997a	d	0.61 - 5.5	0.20 - 1.8		

**Table 6.3 – Odor Threshold Values, cont.**

*Bold = Lowest Value Reported*

#	Chemical Name	Source	Type of Threshold	Odor Thresholds	
				mg/m <sup>3</sup>	ppm
36	<b>n - Butyl Alcohol cont.</b>	Dalton et al 1997b	d	8.2 - 15.8	2.7 - 5.2
		Harreveld & Heeres 1997		0.058 - 0.53	0.019 - 0.17
		Wysocki et al 1997	d	0.48 - 9.6	0.16 - 3.2
		Cometto - Muniz et al 1999	d	1.7 - 3.8	0.56 - 1.25
		Molhave et al 2000	d	11	3.63
		Ziemer et al 2000	d	0.15	0.049
		Feddes et al 2001	d	0.17	0.056
		Mannebeck & Mannebeck 2002	d	0.105 - 0.1739	0.035 - 0.057
		Smeets & Dalton 2002	d	42 - 54	14 - 18
		Nagata 2003	d	0.11	0.036
		Cometto - Muniz et al 2004	d	0.97	0.32
		Maxeiner & Mannebeck 2004	d	0.1323 - 0.1957	0.044 - 0.065
		Cometto - Muniz & Abraham 2008	d	0.024	0.008
		Polednik et al 2008	d	0.2 - 0.4	0.0660.13
		Maxeiner 2006	d	0.1071 - 0.1251	0.035 - 0.041
		Maxeiner 2007	d	0.111 - 0.130	0.037 - 0.043
		Ueno et al 2009		0.051	0.017
		Ueno et al 2009		0.26	0.086
		Ueno et al 2009		0.16	0.053
		Ueno et al 2009		0.42	0.14
Cain et al 2010		0.48	0.16		
Nimmermark 2011		0.078 - 1.4	0.026 - 0.46		
37	<b>sec - Butyl Alcohol</b>	Jung 1936	d	7.4	2.4
		Jung 1936	r	14.4	4.8
		Jones 1955c	r	80	26
		Laffort & Dravnieks 1973		9	3
		Hellman & Small 1974	d	0.4	0.12
		Hellman & Small 1974	r	1.2	0.41
		Bedborough & Trott 1979	d	3.3	1.1
		Punter 1983	d	59.1	19.5
		Punter 1983	d	41.8	13.8
		Cometto - Muniz & Cain 1993	d	285	94
		Cometto - Muniz 1993	d	285	94
		Ziemer et al 2000	d	0.13	<b>0.043</b>
		Nagata 2003	d	0.66	0.218
38	<b>tert - Butyl Alcohol</b>	Passy 1892c	d	10 - 20	3.3 - 6.6
		Backman 1917	r	36 - 40	11.875 - 13.195
		Jones 1955c	r	750	247
		Dravnieks & Laffort 1972		71	23
		Dravnieks 1974	d	2,900	957
		Nagy 1991	d	42	14

Table 6.3 – Odor Threshold Values, cont.

Bold = Lowest Value Reported

#	Chemical Name	Source	Type of Threshold	Odor Thresholds	
				mg/m <sup>3</sup>	ppm
38	<b>tert - Butyl Alcohol cont.</b>	Cometto - Muniz & Cain 1993	d	1,827	603
		Cometto - Muniz 1993	d	1,827	603
		Ziemer et al 2000	d	24.2	7.98
		Nagata 2003	d	14	4.62
39	<b>n - Butylamine</b>	Scherberger et al 1960		<0.36	<0.12
		Sutton 1962a		<3	<1
		Hellman & Small 1973a,b, 1974	d	0.24	<b>0.08</b>
		Hellman & Small 1973a,b, 1974	r	0.72	0.24
		Laing et al 1978	r	41.7	13.9
		Nagata 2003	d	0.51	0.17
40	<b>n - Butyl Lactate</b>	Ziemer et al 2000	d	0.000000029	0.00000000485
41	<b>Butyl Mercaptan</b>	Allison & Katz 1919	d	18	4.9
		Katz & Talbert 1930	d	0.0037	0.001
		Deadman & Prigg 1959	d	0.0015	0.00041
		Blinova 1965		0.007 - 0.04	0.0019 - 0.0011
		Kniebes et al 1969		0.003	0.00081
		Wilby 1969	r	0.0027	0.00073
		Patte 1978	d	0.003	0.00081
		Patte & Punter 1979	d	0.003	0.00081
	Nagata 2003	d	0.00001	<b>0.0000027</b>	
42	<b>p - tert Butyl Toluene</b>	Hine et al 1954	r	<30.5	<5.03
43	<b>Butyraldehyde</b>	Backman 1917	r	0.013 - 0.014	0.0044 - 0.0047
		Pliska & Janicek 1965		15	5.09
		Hellman & Small 1973a,b, 1974	d	<0.013	<0.0044
		Hellman & Small 1973a,b, 1974	r	0.027	0.0092
		Teranshi et al 1974		0.042	0.014
		Anon 1980	d	0.00084	<b>0.0003</b>
		Anon 1980	r	0.011	0.0037
		Hall & Andersson 1983	d	0.2	0.068
		Cristoph 1983	r	0.18 - 0.21	0.061 - 0.071
		Cometto - Muniz et al 1998a	d	8.8	2.98
		Cometto - Muniz 1999	d	8.8	2.98
		Nagata 2003	d	0.0019	0.0006
		Cometto - Muniz & Abraham 2010a	d	0.0013	0.0004
		Laska & Ringh 2010	d	0.1	0.034
44	<b>Camphor, synthetic</b>	Passy 1892a, 1892b	d	5	0.8
		Zwaardemaker 1914, 1927	d	0.016 - 2	<b>0.0026</b> - 0.32
		Backman 1917	r	0.76 - 0.88	0.12 - 0.14
		Ohma 1922	d	0.06	0.0096



**Table 6.3 – Odor Threshold Values, cont.**

*Bold = Lowest Value Reported*

#	Chemical Name	Source	Type of Threshold	Odor Thresholds	
				mg/m <sup>3</sup>	ppm
44	<b>Camphor, synthetic cont.</b>	Hofmann & Kohlrausch 1925	r	2 - 33	0.32 - 5.3
		Mitsumoto 1926	r	4.4 - 45.0	0.71 - 7.2
		Tamman & Oelsen 1928	d	6 - 13	0.97 - 2.1
		Morimura 1934	r	1.16 - 32.5	0.19 - 5.2
		Gundlach & Kenway 1939	d	0.49	0.079
		Kleinschmidt 1983	r	3.35	0.54
		De Wijk 1989		2.84	0.46
45	<b>Caprolactam</b>	Krichevskaya 1968		0.3	0.065
46	<b>Carbon Dioxide</b>	Lotsch et al 1997		540,000 - 1,080,000	300,068 - 600,136
		Shusterman & Balmes 1997a, 1997b		486,000	270,000
		Melzner et al 2011	d	95,400	53,000
		Melzner et al 2011	d	81,000	45,000
		Melzner et al 2011	d	75,600	42,000
		Melzner et al 2011	d	70,200	<b>39,000</b>
47	<b>Carbon Disulfide</b>	Deadman & Prigg 1959	d	0.07	0.022
		Hildenskiold 1959		0.05	<b>0.016</b>
		Frantikova 1962		1.3	0.42
		Baikov 1963		0.08 - 0.5	0.026 - 0.16
		Leonardos et al 1969	r	0.65	0.21
		Naus 1982	d	0.1	0.032
		Naus 1982	r	1	0.32
		Kleinschmidt 1983	r	98.9	32
		Moriguchi et al 1983	d	0.11	0.04
		Don 1986	d	0.18	0.06
		Nagy 1991	d	3.9	1.25
		Nagy 1991	d	1.269	0.41
		Nagata 2003	d	0.65	0.21
48	<b>Carbon Tetrachloride</b>	Allison & Katz 1919		4,533	720
		Davis 1934		500	79
		Lehmann & Schmidt - Kehl 1936		900	143
		May 1966	d	1,260	200
		May 1966	r	1,600	254
		Leonardos et al 1969	r	135 - 630	21 - 100
		Belkov 1969		11.5 - 58	1.8 - 9
		Nikiforov 1970		10.58	<b>1.68</b>
		Dravnieks & Laffort 1972		280	45
		Dravnieks 1974	d	3700	588
		Punter 1983	d	884	140
		Nagata 2003	d	29	4.6
49	<b>Carbonyl Sulfide</b>	Polgar et al 1975		0.25	0.102
		Nagata 2003	d	0.14	<b>0.057</b>

Table 6.3 – Odor Threshold Values, cont.

Bold = Lowest Value Reported

#	Chemical Name	Source	Type of Threshold	Odor Thresholds	
				mg/m <sup>3</sup>	ppm
50	Chlorine	Fieldner et al 1921		10	3.4
		Prentiss 1937		10	3.4
		Smolczyk & Cobler 1930		1.43 - 14.3	0.49 - 4.9
		Takhiroff 1957		0.8	0.28
		Beck 1959		0.15 - 0.3	0.05 - 0.10
		Styazhkin 1963		0.7	0.24
		Rupp & Henschler 1967	d	0.06 - 0.15	0.021 - 0.052
		Rupp & Henschler 1967	r	0.3	0.1
		Leonardos et al 1969	r	0.6	0.21
		Kramer 1976		3.2 - 7.8	1.10 - 2.69
		Dixon & Ikels 1977	d	0.23	0.08
		Naus 1982		3	1.03
		Naus 1982		10	3.45
		Randebrook 1986		0.18	0.062
Nagata 2003		0.14	0.048		
51	Chlorine Dioxide	Vincent et al 1946		42	15
52	Chloroacetophenone	Katz & Talbert 1930		0.10 - 0.70	<b>0.016</b> - 0.111
		Prentiss 1937		0.2	0.032
53	Chlorobenzene	Backman 1917	r	7.5 - 8.1	1.6 - 1.8
		Mateson 1955		21.6	4.7
		Tarkhova 1965		0.4	<b>0.087</b>
		Leonardos et al 1969	r	0.97	0.21
		Smith & Hochstettler 1969	r	3	0.65
		Punter 1983	d	5.9	1.3
		Don 1986	d	1	0.217
		Nagy 1991	d	4.5	0.98
		Cometto - Muniz 1993	d	59.3	13
		Cometto - Muniz & Cain 1994	d	59.3	13
54	Chlorodifluoromethane	Braker & Mossman 1980		708,000	200,192
55	Chloroform	Passy 1893a	d	30	6.1
		Tempelaar 1913	d	3,000	614
		Backman 1917	r	14.1 - 15.1	2.9 - 3.1
		Allison & Katz 1919		3,300	676
		Grijns 1919		2,350	481
		Rocen 1920	d	730	150
		Rocen 1920	r	2,500	512
		Mitsumoto 1926	r	353.8 - 589.0	72.7 - 121
		Schley 1934	d	42	8.6
		Schley 1934	r	56	11
		Morimura 1934	r	480 - 622	99 - 128
		Lehmann & Schmidt - Kehl 1936		1,000 - 1,500	205 - 307
		Scherberger et al 1958	r	6,900	1,413

**Table 6.3 – Odor Threshold Values, cont.**

*Bold = Lowest Value Reported*

#	Chemical Name	Source	Type of Threshold	Odor Thresholds			
				mg/m <sup>3</sup>	ppm		
55	Chloroform cont.	Janicek et al 1960		3,700	758		
		Naus 1962	d	3	0.6		
		Dravnieks & Laffort 1972		150	31		
		Dravnieks 1974	d	1,350	276		
		Naus 1982	d	0.5	<b>0.1</b>		
		Naus 1982	r	20	4.1		
		Punter 1983	d	650	133		
		Nagata 2003	d	19	3.9		
56	Chloropicrin	Prentiss 1937		7.3	1.09		
57	$\beta$ - Chloroprene	Mnatsakanyan 1962		0.4 - 2.0	<b>0.11</b> - 0.55		
58	Chlorotoluene	Blackman 1917	r	0.95 - 1.4	<b>0.18</b> - 0.27		
59	Citral	Passy 1892a, 1892b	d	0.1 - 0.5	0.016 - 0.08		
		Tempelaar 1913	d	0.062 - 0.1	0.010 - 0.016		
		Zwaardemaker 1927	d	0.062 - 0.1	0.010 - 0.016		
		Backman 1917	r	0.06 - 0.09	0.0097 - 0.014		
		Ohma 1922	d	0.13	0.0209		
		Schneider & Wolf 1955		0.027	0.0043		
		Schneider et al 1958		0.12	0.0193		
		Apell 1969		0.0005	0.00008		
		Koster 1971	d	0.17 - 0.19	0.027 - 0.031		
		Etzweiler et al 1980		0.02	0.032		
		Randebroek 1986		0.00015	<b>0.000024</b>		
		60	Cresol, all isomers	<b>o - cresol</b>			
				Backman 1917	r	0.004	0.0009
Stuiver 1958	d			0.0004	<b>0.00009</b>		
Kendall et al 1968	r			0.0028	0.00063		
Anon. 1980	d			0.0017	0.00038		
Anon. 1980	r			0.027	0.0061		
Moriguchi et al 1983	d			0.02	0.00452		
Schieberle et al 1988				0.0007 - 0.0027	0.00016 - 0.00061		
Nagata 2003	d			0.0012	0.00027		
Strube et al 2012				0.0078	0.0018		
<b>m - cresol</b>							
Backman 1917	r			0.0007 - 0.0009	0.00016 - 0.00020		
Stuiver 1958	d			0.0004	0.00009		
Nader 1958	d			0.00022 - 0.035	<b>0.000050</b> - 0.0079		
Anon. 1980	d			0.00057	0.00013		
Anon. 1980	r			0.011	0.0025		
Nagata 2003	d			0.00044	0.0001		
<b>p - cresol</b>							
Backman 1917	r			0.03 - 0.04	0.0068 - 0.0090		
Baldus 1936	d			0.0125	0.0028		
Baldus 1936	r			0.015	0.0034		

Table 6.3 – Odor Threshold Values, cont.

Bold = Lowest Value Reported

#	Chemical Name	Source	Type of Threshold	Odor Thresholds	
				mg/m <sup>3</sup>	ppm
60	Cresol, all isomers cont.	Stuiver 1958	d	0.00005	0.000011
		Leonardos et al 1969	r	0.0044	0.00099
		Punter 1975, 1979	d	0.024	0.0054
		Anon. 1980	d	0.00018	0.000041
		Anon. 1980	r	0.0084	0.0019
		Schieberle et al 1988		0.0003 - 0.001	<b>0.00007</b> - 0.00023
		Schieberle & Grosch 1988		0.0003 - 0.001	0.00007 - 0.00023
		Blank et al 1989		0.0003 - 0.001	0.00007 - 0.00023
		Blank 1990		0.0003 - 0.001	0.00007 - 0.00023
		Nagata 2003	d	0.00024	0.000054
61	Crotonaldehyde	Katz & Talbert 1930	d	0.18 - 0.57	0.063 - 0.20
		Teranishi et al 1974		0.42	0.147
		Hall & Andersson 1983	d	1.7	0.59
		Nagata 2003	d	0.067	<b>0.02</b>
62	Cumene	Solomin 1964		0.06	0.012
		Elfimova 1966		0.025	0.0051
		Koster 1971	d	0.25	0.051
		Turk 1973	r	4.8 - 6.4	0.98 - 1.3
		Hellman & Small 1974	d	0.04	0.008
		Hellman & Small 1974	r	0.23	0.047
		Anon. 1980	d	0.074	0.025
		Anon. 1980	r	0.54	0.11
		Punter 1983	d	0.65	0.132
		Bahmuller 1983		0.017 - 1.19	0.035 - 0.242
		Nagy 1991	d	0.6	0.12
		Cometto - Muniz et al 1998b		5.3	1.08
		Cometto - Muniz 1999	d	5.3	1.08
		Nagata 2003	d	0.041	<b>0.008</b>
63	Cumene Hydroperoxide	Solomin 1964		0.03	0.0048
64	Cyanogen	Braker & Mossman 1980		>533	>500
65	Cyanogen Chloride	Prentiss 1937		2.5	0.994
66	Cyclohexane	Schley 1934	d	39	11
		Schley 1934	r	120	35
		Jones 1955c	d	900	261
		Alibaev 1970		1.8	<b>0.52</b>
		Stone et al 1972	d	35.6	10.3
		Dravnieks & Laffort 1972		315	91.5
		Laffort & Dravnieks 1973		165	48
		Dravnieks & Laffort 1972	d	2,700	784
		Nagata 2003	d	8.5	2.47
		67	Cyclohexanol	Dobrinskiy 1964	
Punter 1983	d			0.64	0.156
Van Thriel et al 2006	d			2.01	0.491

**Table 6.3 – Odor Threshold Values, cont.**

*Bold = Lowest Value Reported*

#	Chemical Name	Source	Type of Threshold	Odor Thresholds	
				mg/m <sup>3</sup>	ppm
68	Cyclohexanone	Dobrinsky 1964		0.21	<b>0.052</b>
		Stone et al 1967	d	1.15	0.29
		Koster 1971	d	790 - 880	197 - 219
		Stone et al 1972	d	1.6	0.4
		Hellman & Small 1973a,b, 1974	d	0.48	0.12
		Hellman & Small 1973a,b, 1974	r	0.48	0.12
		Davis 1973	d	2	0.5
		Laing 1975	d	40	10
		Laing 1983		1.0 - 2.4	0.249 - 0.598
		Laska & Hudson 1991	d	0.88 - 1.2	0.219 - 0.299
		Ziemer et al 2000	d	1.1	0.27
Van Thriel et al 2006	d	5.27	1.31		
69	Cyclohexene	Deadman & Prigg 1959	d	0.6	0.18
70	Cyclohexylamine	Van Thriel et al 2006	d	9.83	2.42
71	Cyclopentadiene	Deadman & Prigg 1959	d	5	1.8
72	Decaborane	Krackow 1953		0.3	0.06
73	1 - Decene	Koszinowski & Piringer 1983		37	6.45
74	Diacetone Alcohol	Hellman & Small 1974	d	1.3	<b>0.27</b>
		Hellman & Small 1974	r	5.2	1.1
		Nagy 1991	d	60	13
		Nagy 1991	d	37.418	7.88
75	Diacetyl	Backman 1917	r	0.003 - 0.006	0.00085 - 0.00170
		Van Anrooij 1931	d	0.0025	0.0007
		Apell 1969		0.0026	0.00074
		Artho & Koch 1973		0.00001	0.0000028
		Punter 1975	d	0.000007	<b>0.0000019</b>
		Punter 1979	d	0.000007	0.0000019
		Hall & Andersson 1983	d	0.005	0.0014
		Bahnmueller 1983		0.0007 - 0.087	0.00020 - 0.247
		Randebroek 1986		10.2	2.9
		Blank 1990		0.015 - 0.030	0.0043 - 0.0085
		Blank et al 1992		0.01 - 0.02	0.0028 - 0.0057
Nagata 2003	d	0.00018	0.000051		
76	Diallylamine	Hine et al 1960		8	2
77	Diborane	Krackow 1953		2 - 4	1.8 - 3.5
78	2,3 Dibromo-1-Chloro-propane	Torkelson & Rowe 1981		0.1 - 0.3	0.010 - 0.031
79	Dibutylamine	Hellman & Small 1973a,b, 1974	d	0.42	<b>0.079</b>
		Hellman & Small 1973a,b, 1974	r	1.4	0.265
		Laing et al 1978	r	2.76	0.522

Table 6.3 – Odor Threshold Values, cont.

Bold = Lowest Value Reported

#	Chemical Name	Source	Type of Threshold	Odor Thresholds	
				mg/m <sup>3</sup>	ppm
79	<b>Dibutylamine cont.</b>	Bahnmueller 1984		0.44 - 4.069	0.083 - 0.77
80	<b>Dibutyl Phthalate</b>	Menshikova 1972		0.26	0.023
81	<b>Dichloroacetic Acid</b>	Backman 1917	r	0.232	0.044
82	<b>Dichlorobenzene, o - isomer</b>	Backman 1917	r	0.12	<b>0.02</b>
		Hollingsworth et al 1958		<300	<50
		Punter 1983	d	4.2	0.699
83	<b>Dichlorobenzene, p - isomer</b>	Hollingsworth et al 1956		<90	<15
		Punter 1983	d	0.73	<b>0.121</b>
84	<b>Dichlorodifluoromethane</b>	Braker & Mossman 1980		988,000	199,790
85	<b>1,1 - Dichloroethane</b>	Rylova 1953		200	<b>49</b>
		Janicek et al 1960		5,500	1359
		Irish 1963		2,000 - 4,000	494 - 988
86	<b>1,2 Dichloroethylene, all isomers</b>	Lehmann & Schmidt - Kehl 1936		1,100	277
87	<b>2,4 - Dichlorophenol</b>	Punter 1983	d	0.00027	0.000041
		Strube et al 2012		0.0068	0.00102
88	<b>1,3 - Dichloropropene</b>	Torkelson & Oyen 1977		<4.5	<0.99
89	<b>Dicyclopentadiene</b>	Kinthead et al 1971b		0.016	0.003
		Hellman & Small 1974	d	0.06	0.011
		Hellman & Small 1974	r	0.11	0.02
		Ventura et al 1997		0.001	<b>0.00019</b>
90	<b>Diethanolamine</b>	England et al 1978	r	1.2	0.279
91	<b>Diethylamine</b>	Geier 1936	d	0.01 - 0.1	<b>0.0033</b> - 0.033
		Geier 1936	r	2.25 - 5	0.75 - 1.67
		Kosiborod 1968		0.084	0.028
		Hellman & Small 1973a	d	0.42	0.14
		Hellman & Small 1973a	r	1.5	0.5
		Hellman & Small 1974	d	0.06	0.02
		Hellman & Small 1974	r	0.18	0.06
		Cormack et al 1974		0.09	0.03
		Laing et al 1978	r	42.9	14.3
		Tkachev 1978		0.044 - 0.558	0.015 - 0.187
		Anon. 1980	d	0.09	0.03
		Anon. 1980	r	0.9	0.3
		Laing 1982	d	4	1.3
		Nagata 2003	d	0.14	0.0468
92	<b>2 - Diethylaminoethanol</b>	Hellman & Small 1973a,b, 1974	d	0.05	<b>0.01</b>
		Hellman & Small 1973a,b, 1974	r	0.19	0.04
		England et al 1978	r	1.2	0.25

**Table 6.3 – Odor Threshold Values, cont.**

*Bold = Lowest Value Reported*

#	Chemical Name	Source	Type of Threshold	Odor Thresholds	
				mg/m <sup>3</sup>	ppm
93	Diethylbenzenes, mixed isomers	Nagata 2003	d	0.052	0.0095
		Nagata 2003	d	0.39	0.071
		Nagata 2003	d	0.0021	<b>0.00038</b>
94	Diethyl Ketone	Backman 1917	r	3.8	1.1
		May 1966	d	33	9.4
		May 1966	r	49	14
		Dravnieks 1974	d	3	<b>0.85</b>
95	Diethyl Phthalate	Wunsche et al 1995	d	0.33 - 3.3	0.036 - 0.363
96	Diisobutyl Ketone	Hellman & Small 1973a,b, 1974	d	<0.6	<b>&lt;0.103</b>
		Hellman & Small 1973a,b, 1974	r	1.8	0.309
		Nagy 1991	d	9.3	1.6
97	Diisopropylamine	Hellman & Small 1974	d	0.56	<b>0.14</b>
		Hellman & Small 1974	r	1.6	0.39
		England et al 1978	r	17.4	4.2
98	N,N - Dimethylacetamide	Leonardos et al 1969	r	170	48
99	Dimethylamine	Geier 1936	d	0.65 - 1.0	0.35 - 0.54
		Geier 1936	r	2.2 - 3.0	1.2 - 1.6
		Taylor & Bodurtha 1960		1.1	0.6
		Leonardos et al 1969	r	0.085	0.046
		Stephens 1971		0.16	0.087
		Prusakov et al 1976		0.01 - 0.03	0.005 - 0.016
		Tkachev 1978		0.03	0.016
		Anon. 1980	d	0.0014	<b>0.00076</b>
		Anon. 1980	r	0.023	0.012
		Nagata 2003	d	0.059	0.032
Van Thriel et al 2006	d	7.75	4.2		
100	Dimethylaniline	Backman 1917	r	0.8 - 1.0	0.16 - 0.20
		Geier 1936	d	0.005 - 0.1	<b>0.001</b> - 0.02
		Geier 1936	r	0.05 - 0.25	0.010 - 0.050
		Deadman & Prigg 1959	d	0.012	0.0024
101	Dimethyl Disulfide	Wilby 1969	r	0.029	0.0075
		Lindvall 1970	d	0.003 - 0.014	0.00078 - 0.00363
		Selyuzhitskii 1972		3.5	0.908
		Bedborough & Trott 1979	d	0.046	0.012
		Anon 1980	d	0.0011 - 0.0020	<b>0.00029</b> - 0.00052
		Anon 1980	r	0.011 - 0.017	0.00286 - 0.0044
		Moriguchi et al 1983	d	0.007	0.0018
		Ahlstrom et al 1986	d	0.050 - 0.078	0.0130 - 0.0202
		Nagy 1991	d	0.066	0.017
		Gijs et al 2000		0.82	0.213

Table 6.3 – Odor Threshold Values, cont.

Bold = Lowest Value Reported

#	Chemical Name	Source	Type of Threshold	Odor Thresholds	
				mg/m <sup>3</sup>	ppm
101	Dimethyl Disulfide cont.	Greenman et al 2004		5.6	1.45
		Nagata 2003	d	0.0084	0.0022
102	Dimethyl Ether	Nagy 1991	d	430	228
		Nagy 1991	d	303.967	<b>161</b>
103	Dimethyl Formamide	Odoshashvili 1962		0.14	<b>0.047</b>
		Leonardos et al 1969	r	300	100
104	1,1 - Dimethylhydrazine	Jacobson et al 1955	d	15 - 35	6.1 - 14
		Rumsey & Cesta 1970		<0.75	<b>&lt;0.31</b>
105	Dimethyl Sulfide	Katz & Talbert 1930		0.0094	0.0037
		Nevers & Oister 1965		0.0035	0.0014
		Guadagni 1966		0.003	0.0012
		Leonardos et al 1969	r	0.0025	0.001
		Wilby 1969	r	0.0063	0.0025
		Lindvall 1970	d	0.002 - 0.03	0.00079 - 0.012
		Laffort 1968b		0.014	0.0055
		Laffort & Dravnieks 1973		0.014	0.0055
		Hamanabe et al 1969		0.025	0.0098
		Selyuzhitskii 1972		0.75	0.295
		Ifeadi 1972		0.65	0.256
		Cormack et al 1974		0.0075	0.003
		Nishida et al 1975	d	0.0025 - 0.065	0.00098 - 0.026
		Nishida et al 1979	d	0.16	0.063
		Anon 1980	d	0.0003	<b>0.00012</b>
		Anon 1980	r	0.0058	0.0023
		Moschandreas & Jones 1983	d	0.027	0.011
		Moschandreas & Jones 1983	r	0.049	0.019
		Randebrock 1986		20.6	8.11
		Nagy 1991	d	0.051	0.020
Nagata 2003	d	0.0075	0.003		
Glindemann et al 2006	d	0.001	0.00039		
106	4,6 - Dinitro - o - cresol	Kurtschatowa & Dawidkowa 1970		0.004 - 0.021	<b>0.00049</b> - 0.00259
107	1,4 - Dioxane	Wirth & Klimmer 1937	d	10	2.8
		May 1966	d	620	172
		May 1966	r	1000	278
		Koster 1968a, 1971	d	45 - 9,400	12 - 2609
		Dravnieks & Laffort 1972		30.6	8.5
		Hellman & Small 1973a,b, 1974	d	2.9	<b>0.8</b>
		Hellman & Small 1973a,b, 1974	r	6.5	1.8
		Dravnieks 1974	d	270	75
Nagy 1991	d	46	12.8		



**Table 6.3 – Odor Threshold Values, cont.**

*Bold = Lowest Value Reported*

#	Chemical Name	Source	Type of Threshold	Odor Thresholds	
				mg/m <sup>3</sup>	ppm
108	1,3 - Dioxolane	Hellman & Small 1974	d	51	<b>16.8</b>
		Hellman & Small 1974	r	192	63.4
109	Diphenylamine	Backman 1917	r	0.15 - 0.17	<b>0.022</b> - 0.025
		Nagy 1991	d	1.3	0.188
110	Dodecyl Mercaptan	Kendall et al 1968	r	0.0008	0.000097
		Patte 1978	d	0.0000009	<b>0.0000011</b>
		Patte & Punter 1979	d	0.0000009	0.0000011
111	Epichlorohydrin	Toxicity Data Sheet 1959		38 - 46	10.04 - 12.15
		Fomin 1966		0.3	<b>0.08</b>
112	Ethane	Mullins 1955	r	899,000	730973
		Laffort & Dravnieks 1973		25,000	<b>20328</b>
113	Ethanolamine	Weeks et al 1960	d	6.5	<b>2.6</b>
		Weeks et al 1960	r	60	24
114	2 - Ethoxyethanol	May 1966	d	90	24
		May 1966	r	180	49
		Hellman & Small 1973a,b, 1974	d	1.1	<b>0.3</b>
		Hellman & Small 1973a,b, 1974	r	2	0.54
		Nagy 1991	d	11.6	3.15
		Nagata 2003	d	2.1	0.57
115	2-(2-Ethoxyethoxy) - ethanol	Hellman & Small 1973a,b, 1974	d	<1.2	<b>&lt;0.219</b>
		Hellman & Small 1973a,b, 1974	r	6	1.09
116	2 - Ethoxyethyl Acetate	Hellman & Small 1973a,b, 1974	d	0.3	0.06
		Hellman & Small 1973a,b, 1974	r	0.7	0.13
		Nagy 1991	d	0.48	0.089
		Nagata 2003	d	0.26	<b>0.048</b>
117	Ethyl Acetate	Backman 1917	r	15 - 17.5	4.2 - 4.9
		Allison & Katz 1919		686	190
		Jung 1936	d	3.6	1
		Jung 1936	r	3.6 - 5.4	1.0 - 1.5
		Jones 1955c	d	155	43
		Clausen et al 1955	d	4.8	1.33
		Gofmekler 1960		0.6	0.17
		Janicek et al 1960		1120	311
		May 1966	d	180	50
		May 1966	r	270	75
		Laffort & Dravnieks 1973		27	7.5
		Hellman & Small 1973a, 1974	d	23	6.4
		Hellman & Small 1973a, 1974	r	48	13.3

**Table 6.3 – Odor Threshold Values, cont.***Bold = Lowest Value Reported*

#	Chemical Name	Source	Type of Threshold	Odor Thresholds	
				mg/m <sup>3</sup>	ppm
117	Ethyl Acetate cont.	Anon. 1980	d	0.9	0.25
		Anon. 1980	r	5	1.39
		Bahnmuller 1983		3.7 - 25	1.027 - 6.9
		Cristoph 1983	r	4.6 - 5.0	1.3 - 1.4
		Randebrock 1986		0.34	0.09
		Scharfenberger 1990		141	39
		Cometto - Muniz & Cain 1991	d	623	173
		Cometto - Muniz 1993	d	623	173
		Nagy 1991	d	28	8
		Ziemer et al 2000	d	4.6	1.28
		Nagata 2003	d	3.1	0.86
		Higuchi & Masuda 2004	d	2.0 - 3.0	0.555 - 0.833
		Komthong et al 2006		1,030	286
		Van Thriel et al 2006	d	5.36	1.49
		Cometto - Muniz et al 2008	d	0.88	0.24
		Ueno et al 2009		1.3	0.36
Ueno et al 2009		6.1	1.69		
Ueno et al 2009		4.7	1.30		
Ueno et al 2009		4.3	1.19		
118	Ethyl Acrylate	Leonardos et al 1969	r	0.0019	0.00046
		Hellman & Small 1973a, 1974	d	0.001	0.00024
		Hellman & Small 1973a, 1974	r	0.0015	0.00037
		Anon. 1980	d	0.00082	0.0002
		Anon. 1980	r	0.0053	0.0013
		Piringer & Granzer 1984		0.001	0.00024
		Nagy 1991	d	0.013	0.0032
		Nagata 2003	d	0.0011	0.00026
		Van Thriel et al 2006	d	0.000027	0.0000066
119	Ethyl Alcohol	Passy 1892c	d	250	133
		Parker & Stabler 1913	r	17	9
		Backman 1917	r	175 - 200	93 - 106
		Grijns 1919		2,600	1380
		Zwaardemaker 1927		2,600	1380
		Henning 1924	d	183	97
		Jung 1936	d	7.8	4.1
		Jung 1936	r	11.7 - 14	6.2 - 7.4
		Balavoine 1943		10,000	5,307
		Mullins 1955	r	9,230	4,898
		Scherberger et al 1958	r	665	353
		Janicek et al 1960		884	469
		Naus 1962	d	2	1.1
		Pliska & Janicek 1965		76,000	40,334
Ubaidullaev 1966b		7.1	3.77		

**Table 6.3 – Odor Threshold Values, cont.**

*Bold = Lowest Value Reported*

#	Chemical Name	Source	Type of Threshold	Odor Thresholds	
				mg/m <sup>3</sup>	ppm
119	Ethyl Alcohol cont.	Guadagni 1966		100	53
		May 1966	d	93	49
		May 1966	r	190	101
		Leonardos et al 1969	r	19	10
		Dravnieks & Laffort 1972		640	340
		Dravnieks 1974	d	1,350	716
		Nishida et al 1979	d	302	159
		Anon. 1980	d	0.64	0.34
		Anon. 1980	r	11.6	6.2
		Naus 1982	d	2	1.06
		Naus 1982	r	20	10.61
		Cristoph 1983	r	8.7 - 9.2	4.6 - 4.9
		Cometto - Muniz & Cain 1990		154	82
		Cometto - Muniz 1993	d	154	82
		Scharfenberger 1990		988	524
		Nagy 1991	d	36	19
Nagata 2003	d	0.99	0.525		
Cain et al 2005	d	0.17	<b>0.09</b>		
Cometto - Muniz & Abraham 2008	d	0.62	0.329		
120	Ethylamine	Tkachev 1969		0.05	<b>0.027</b>
		Hellman & Small 1974	d	0.5	0.27
		Hellman & Small 1974	r	1.5	0.81
		Laing et al 1978	r	6.5	3.5
		Nagata 2003	d	0.083	0.045
121	Ethyl Amyl Ketone	Toxicity Data Sheet 1958b		31	5.9
122	Ethyl Benzene	Ivanov 1964		2 - 2.6	0.46 - 0.60
		Koster 1971	d	0.4	0.092
		Nagy 1991	d	1.9	0.44
		Khiari et al 0992		<0.01	<b>&lt;0.002</b>
		Cometto - Muniz 1993	d	78.3	18
		Cometto - Muniz & Cain 1994	d	78.3	18
		Nagata 2003	d	0.73	0.17
		Cometto - Muniz & Abraham 2009b	d	0.026	0.006
123	Ethyl Bromide	Backman 1917		12.1 - 16	<b>2.7</b> - 3.6
124	Ethyl Chloride	Backman 1917	r	10 - 12	<b>3.8</b> - 4.5
		Nagy 1991	d	>1000	>379
125	Ethylene	Mullins 1955	r	1,180	1,029
		Deadman & Prigg 1959	d	125	109
		Krasovitskaya & Malyarova 1968		20	<b>17</b>
		Laffort & Dravnieks 1973		1,100	959

Table 6.3 – Odor Threshold Values, cont.

Bold = Lowest Value Reported

#	Chemical Name	Source	Type of Threshold	Odor Thresholds	
				mg/m <sup>3</sup>	ppm
125	Ethylene cont.	Hellman & Small 1974	d	310	270
		Hellman & Small 1974	r	480	418
126	Ethylene Chlorohydrin	Semenova et al 1980		1.2	0.36
127	Ethylenediamine	Hellman & Small 1974	d	3.2	1.3
128	Ethylene Dibromide	Olmstead 1972	r	<77	<10
129	Ethylene Dichloride	McCawley 1942		1,200 - 4,000	297 - 988
		Jones 1955c	d	1,500	371
		Borisova 1957		17.5 - 23.2	<b>4.3</b> - 5.7
		Scherberger et al 1958	r	820	203
		Irish 1963		200	49
		May 1966	d	450	111
		May 1966	r	750	185
		Dravnieks & O'Donnell 1971		190	47
		Hellman & Small 1974	d	25	6
		Hellman & Small 1974	r	165	41
		Kleinschmidt 1983	r	350	86
		130	Ethylene Glycol	Nagy 1991	d
131	Ethyleneimine	Carpenter et al 1948		3.6	2
		Berzins 1967		1.25 - 3.5	<b>0.71</b> - 1.99
132	Ethylene Oxide	Jacobson et al 1956	d	1,260	690
		Yuldashev 1965		1.5	<b>0.82</b>
		Hellman & Small 1974	d	470	257
		Hellman & Small 1974	r	900	493
133	Ethyl Ether	Passy 1892a,b,d	d	0.5 - 4	<b>0.165</b> - 1.32
		Allison & Katz 1919		5,833	1,924
		Grijns 1919		<50	<16.49
		Henning 1924	d	0.75	0.25
		Zwaardemaker 1927		1	0.33
		Jung 1936	d	35	12
		Jung 1936	r	35	12
		Scherberger et al 1958	r	210	69
		Flemming & Johnstone 1977	r	4.8	1.58
Nagy 1991	d	0.95	0.31		
134	Ethyl Formate	Backman 1917	r	54 - 61	18 - 20
		Nagata 2003	d	8.1	<b>2.67</b>
		Van Thriel et al 2006	d	90.9	30
135	Ethylidene Norbornene	Kinthead et al 1971a		0.035 - 0.07	<b>0.007</b> - 0.014
		Hellman & Small 1974	d	0.1	0.02
		Hellman & Small 1974	r	0.4	0.08
136	Ethyl Mercaptan	Allison & Katz 1919		46	18
		Katz & Talbert 1930	d	0.00066 - 0.0076	0.00026 - 0.0030
		Thomas et al 1943		0.005	0.002
		Stuiver 1958	d	0.0001	0.0004

**Table 6.3 – Odor Threshold Values, cont.**

*Bold = Lowest Value Reported*

#	Chemical Name	Source	Type of Threshold	Odor Thresholds	
				mg/m <sup>3</sup>	ppm
136	<b>Ethyl Mercaptan cont.</b>	Sales 1958		0.0025 - 0.0045	0.00098 - 0.0018
		Blinova 1965		0.006 - 0.03	0.002 - 0.01
		Endo et al 1967		0.00065	0.00026
		Leonardos et al 1969	r	0.0025	0.0098
		Wilby 1969	r	0.001	0.0004
		Blanchard 1976		0.016	0.0063
		Selyuzhitskii et al 1978		0.000095	0.000037
		Whisman et al 1978	d	0.00025 - 0.0005	0.000098 - 0.00020
		Bedborough & Trott 1979	d	0.00033	0.00013
		Anon 1980	d	0.000043	0.000017
		Anon 1980	r	0.00073	0.00029
		Cristoph 1983	r	0.0008 - 0.0009	0.00031 - 0.00035
		Stevens et al 1987		0.0019 - 0.021	0.00075 - 0.00826
		Stevens & Cain 1987b	d	0.0019 - 0.021	0.00075 - 0.00826
Nagata 2003	d	0.000022	<b>0.0000087</b>		
137	<b>N - Ethylmorpholine</b>	Hellman & Small 1974	d	0.4	<b>0.085</b>
		Hellman & Small 1974	r	1.2	0.25
138	<b>Ethyl Silicate</b>	Smyth & Seaton 1940	d	<720	<85
		Hellman & Small 1974	d	31	<b>3.6</b>
		Hellman & Small 1974	r	43	5
139	<b>Fluorine</b>	Belles et al 1965		0.15 - 0.30	<b>0.097</b> - 0.19
140	<b>Formaldehyde</b>	Backman 1917	r	0.033 - 0.036	<b>0.027</b> - 0.029
		Melekhina 1958		0.07	0.057
		Buchberg et al 1961		1.1 - 2.2	0.90 - 1.8
		Pliska & Janicek 1965		12,000	9,770
		Sgibnev 1968		0.3 - 0.4	0.24 - 0.33
		Leonardos et al 1969	r	1.2	0.98
		Feldman & Bonashevskya 1971		0.073	0.059
		Takhirov 1974		0.065	0.053
		Makeicheva 1978		0.077	0.063
		Anon. 1980	d	0.49	0.4
		Anon. 1980	r	2.3	1.9
		Berglund et al 1984	d	0.06	0.049
		Ahstrom et al 1986b	d	0.06	0.049
		Berglund et al 1987	d	0.14 - 0.21	0.114 - 0.171
		Winneke et al 1988		0.15 - 0.29	0.122 - 0.236
		Nagy 1991	d	2.2	1.8
		Berglund & Nordin 1992	d	0.066 - 0.11	0.054 - 0.09
		Berglund & Esfandabad 1992		0.18	0.15
		Berglund & Esfandabad 1992		0.69	0.56
		Nagata 2003	d	0.6	0.49

Table 6.3 – Odor Threshold Values, cont.

Bold = Lowest Value Reported

#	Chemical Name	Source	Type of Threshold	Odor Thresholds	
				mg/m <sup>3</sup>	ppm
141	Formic Acid	Passy 1893b, 1893c	d	25 - 50	13 - 27
		Zwaardemaker 1914	d	640	340
		Backman 1917	r	21 - 24	11 - 13
		Schley 1934		3.0 - 6.0	1.6 - 3.2
		Guadagni 1966		450	239
		Naus 1982	d	2	1.06
		Naus 1982	r	20	10.63
		Kleinschmidt 1983	r	453	241
		Cometto - Muniz et al 1998a		14.5	7.7
		Cometto - Muniz 1999	d	14.5	7.7
		Van Thriel et al 2006	d	12.4	6.59
Cometto - Muniz & Abraham 2010b	d	0.98	<b>0.52</b>		
142	Furan	Nagata 2003	d	28	10.06
143	Furfural	Ubaidullaev 1961		1	0.254
		Apell 1969		0.008	<b>0.002</b>
		Makeicheva 1978		0.98	0.249
		Bedborough & Trott 1979	d	0.25	0.0636
		Nagy 1991	d	2.8	0.713
144	Furfuryl Alcohol	Jacobson et al 1958	d	32	8
145	Glutaraldehyde	Colwell 1976	r	0.16	0.039
		Cain et al 2007b	d	0.0015	<b>0.00037</b>
146	Halothane	Flemming & Johnstone 1977	r	267	33
147	Heptane, all isomers	Patty & Yant 1929		410	100
		Mullins 1955	r	2,240	547
		Jones 1955c	d	750	183
		May 1966	d	930	227
		May 1966	r	1,350	329
		Dravnieks & Laffort 1972		870	212
		Laffort & Dravnieks 1973		165	40
		Dravnieks 1974	d	3,000	732
		Nagy 1991	d	110	27
		Nagata 2003	d	2.7	0.66
		Nagata 2003	d	1.7	<b>0.41</b>
		Nagata 2003	d	3.4	0.83
		Nagata 2003	d	156	38
		Nagata 2003	d	18	4.39
Nagata 2003	d	3.9	0.95		
148	Hexachlorocyclopentadiene	Treon et al 1955		1.7	0.15
149	1,6 Hexamethylene Diisocyanate	Kimmerle 1971		0.035 - 0.07	<b>0.005</b> - 0.010

**Table 6.3 – Odor Threshold Values, cont.**

*Bold = Lowest Value Reported*

#	Chemical Name	Source	Type of Threshold	Odor Thresholds	
				mg/m <sup>3</sup>	ppm
150	n - Hexane	Patty & Yant 1929		875	248
		Laffort & Dravnieks 1973		230	65
		De Wijk 1989		107	30
		Nagata 2003	d	5.3	1.5
151	Hexane, all isomers, except n - Hexane	Nagata 2003	d	5.8	1.68
		Nagata 2003	d	25	7
		Nagata 2003	d	31	9
		Nagata 2003	d	70	20
		Nagata 2003	d	1.5	<b>0.426</b>
152	1,6 - Hexanediamine	Kulakov 1964		0.0032	0.00067
153	1 - Hexene	Nagata 2003	d	0.48	0.139
154	sec - Hexyl Acetate	Stone et al 1972	d	2.3	0.39
		Hellman & Small 1974	d	<0.4	<b>&lt;0.068</b>
		Hellman & Small 1974	r	1.4	0.237
155	n - Hexyl Alcohol	Backman 1917	r	1.0 - 1.3	0.24 - 0.31
		Mullins 1955	r	9.94	2.38
		Pliska 1962		65	16
		Cain 1969	r	3.5	0.837
		Stone et al 1972	d	1.5	0.359
		Dravnieks & Laffort 1972		0.01	<b>0.0024</b>
		Dravnieks 1974	d	0.3	0.072
		Hellman & Small 1974	d	0.04	0.01
		Hellman & Small 1974	r	0.38	0.091
		Punter 1983	d	1.93	0.46
		Cristoph 1983	r	0.10 - 0.15	0.024 - 0.036
		Cometto - Muniz & Cain 1990	d	4	0.96
		Cometto - Muniz 1993	d	4	0.96
		Ferreira et al 1998		0.74	0.18
		Nagata 2003	d	0.025	0.006
		Komthong et al 2006		12.3	2.9
		Cometto - Muniz & Abraham 2008	d	0.034	0.0081
156	Hexylene Glycol	Nagy 1991	d	19	3.93
157	Hydrazine	Jacobson et al 1955	d	3.9 - 5.2	<b>3.0 - 4.0</b>
		Jacobson et al 1958	d	5.2	4
158	Hydrogen Chloride	Schley 1934		4.5	3.02
		Elfimova 1959		0.1 - 0.2	0.067 - 0.134
		Heyroth 1963	d	1.5 - 7.5	1.01 - 5.03
		Styazhkin 1963		0.2	0.134
		Melekhina 1968	d	0.39	0.262
		Leonardos et al 1969	r	15	10
		Takhirov 1974		0.38	0.255

**Table 6.3 – Odor Threshold Values, cont.**

*Bold = Lowest Value Reported*

#	Chemical Name	Source	Type of Threshold	Odor Thresholds	
				mg/m <sup>3</sup>	ppm
158	Hydrogen Chloride cont.	Naus 1982	d	7	4.69
		Naus 1982	r	15	10
		Van Thriel et al 2006	d	0.09	<b>0.06</b>
159	Hydrogen Cyanide	Sherrard 1928		6	5.43
		Smolczyk & Cobler 1930		<1.1	<1
		Prentiss 1937		1	0.905
		Artho & Koch 1973		0.01 - 0.1	<b>0.009</b> - 0.09
		Braker & Mossman 1980	r	2.2 - 5.6	1.99 - 5.07
160	Hydrogen Fluoride	Sadilova 1968		0.03	0.04
161	Hydrogen Selenide	Dudley & Miller 1941		<1	<0.3
162	Hydrogen Sulfide	Valentin 1848, 1850		2	1.4
		Lehmann 1897		<2	<1.4
		Kulka & Homma 1910		0.2 - 0.3	0.14 - 0.22
		Henderson & Haggard 1922		<0.001	<0.00072
		Henning 1924	d	0.0001	0.00007
		Katz & Talbert 1930	d	0.18	0.13
		Thomas et al 1943		0.035	0.025
		Loginova 1957		0.04	0.029
		Duan - Fen - Djuj 1959		0.012 - 0.03	0.0086 - 0.022
		Sanders & Dechant 1961		0.04 - 0.10	0.029 - 0.072
		Baikov 1963		0.014 - 0.03	0.010 - 0.022
		Young & Adams 1966	d	0.008 - 0.011	0.0057 - 0.0079
		Cederlof et al 1966	d	0.01	0.0072
		Sakuma et al 1967		0.007	0.005
		Endo et al 1967		1.4	1
		Basmadzhieva & Argirova 1968		0.012	0.0086
		Adams et al 1968	d	0.0047 - 0.0090	0.0034 - 0.0065
		Leonardos et al 1969	r	0.00066 - 0.0066	0.00047 - 0.0047
		Pomeroy & Cruse 1969		0.0042 - 0.042	0.003 - 0.030
		Wilby 1969	r	0.0063	0.0045
		Lindvall 1970	d	0.00021 - 0.0016	0.00015 - 0.0017
		Stephens 1971		0.067	0.048
		Randebrook 1971		0.012	0.0086
		Nishida et al 1975	d	0.0014 - 0.055	0.0010 - 0.039
		Winkler 1975	d	0.003	0.0022
		Winkler 1975	r	0.03	0.022
		Hill & Barth 1976		0.0007	0.0005
		Williams et al 1977	d	0.27	0.019
		Logtenberg 1978	d	0.002	0.0014
Nishida et al 1979	d	0.074	0.053		
Winneke et al 1979	d	0.00265	0.0019		
Thiele 1979	d	0.0016	0.0011		



**Table 6.3 – Odor Threshold Values, cont.**

*Bold = Lowest Value Reported*

#	Chemical Name	Source	Type of Threshold	Odor Thresholds	
				mg/m <sup>3</sup>	ppm
162	Hydrogen Sulfide cont.	Bedborough & Trott 1979	d	0.0036	0.0026
		Brunekreef & Harssema 1980		0.0011 - 0.0024	0.00079 - 0.0017
		Anon. 1980	d	0.0007	0.0005
		Anon. 1980	r	0.0078	0.0056
		Thiele et al 1981		0.0013 - 0.0053	0.00093 - 0.0038
		Thiele 1982		0.0028	0.062
		Naus 1982	d	0.1	0.072
		Naus 1982	r	5	3.6
		Jensen & Flyger 1983		0.0038 - 0.0067	0.0027 - 0.0048
		Kobal & Thiele 1983		0.0022	0.0016
		Bahmuller 1983		0.0014 - 0.023	0.001 - 0.017
		Moriguchi et al 1983	d	0.0007	0.0005
		Bahmuller 1984		0.0012 - 0.0073	0.00086 - 0.0052
		Thiele 1984		0.0018	0.0013
		Roos et al 1985	d	0.00085 - 0.00105	0.00061 - 0.00075
		Roos et al 1985	d	0.0004 - 0.00043	0.00029 - 0.00031
		Don 1986	d	0.0004 - 0.00043	0.00029 - 0.00031
		Hoshika et al 1993	d	0.0004 - 0.00043	0.00029 - 0.00031
		Randebroek 1986		0.0096	0.0069
		Heeres et al 1986		0.0004 - 0.0052	0.00029 - 0.0037
		Dollnick et al 1988		0.00166	0.0012
		Winneke et al 1988		0.0015 - 0.0026	0.0011 - 0.0019
		Hermans 1989		0.000056 - 0.001545	<b>0.00004</b> - 0.0011
		Nagy 1991	d	0.0055	0.0039
		Hoshika et al 1993	d	0.0007	0.0005
		Lotsch et al 1997		0.14 - 2.8	0.10 - 2
		Mannebeck & Mannebeck 2002	d	0.000491 - 0.000946	0.00035 - 0.00068
		Nagata 2003	d	0.00057	0.00041
		Greenman et al 2004		0.022	0.0157
		McGinley & McGinley 2004		0.00070 - 0.003	0.0005 - 0.0022
		McGinley & McGinley 2004	r	0.00064 - 0.0013	0.00046 - 0.00093
		McGinley & McGinley 2004	d	0.00057 - 0.00142	0.00041 - 0.0010
McGinley & McGinley 2004	r	0.00071 - 0.0032	0.00051 - 0.0023		
Glindemann et al 2006	d	0.001	0.00072		
Ueno et al 2009		0.00045	0.00032		
Ueno et al 2009		0.0018	0.013		
163	Indene	Deadman & Prigg 1959	d	0.02	0.0042
		Moriguchi et al 1983	d	0.013	<b>0.0027</b>
164	Iodine	Randebroek 1986		10.1	0.973
165	Iodoform	Passy 1893a	d	0.06 - 0.7	0.0037 - 0.043
		Berthelot 1901		0.0003 - 0.03	<b>0.000019</b> - 0.0019

Table 6.3 – Odor Threshold Values, cont.

Bold = Lowest Value Reported

#	Chemical Name	Source	Type of Threshold	Odor Thresholds			
				mg/m <sup>3</sup>	ppm		
165	Iodoform cont.	Backman 1917	r	0.095	0.0059		
		Allison & Katz 1919		18	1.12		
		Zwaardemaker 1927		0.03	0.0019		
		Cancho et al 2001		<0.14	<0.0087		
166	Isoamyl Acetate	Hermanides 1909	r	0.09	0.017		
		Zwaardemaker 1914	d	0.09	0.017		
		Backman 1917	r	0.18 - 0.29	0.034 - 0.054		
		Katz & Talbert 1930	d	0.018	0.0034		
		Jung 1936	d	0.20	0.038		
		Jung 1936	r	0.2	0.038		
		Kerka & Humphreys 1956		0.2	0.038		
		Pliska & Janicek 1965		5	0.94		
		Appell 1969		0.004	<b>0.00075</b>		
		Nishida et al 1979	d	1,100	209		
		Punter 1983	d	0.070 - 0.084	0.013 - 0.016		
		Cristoph 1983	r	0.015 - 0.02	0.0028 - 0.0038		
		Don 1986	d	0.075	0.014		
		Lea & Ford 1991		0.5	0.094		
		Laska & Hudson 1991	d	0.13 - 0.14	0.024 - 0.026		
		Hoshika et al 1997	r	8	1.5		
		Langridge 2004		0.2289	0.043		
		Langridge 2004		0.0107	0.002		
		Atanasova et al 2005	d	0.018 - 0.919	0.0034 - 0.173		
		Atanasova et al 2005	r	0.067 - 0.918	0.013 - 0.172		
Komthong et al 2006		1,950	366				
167	Isoamyl Alcohol	Passy 1892c	d	0.1	0.028		
		Backman 1917	r	0.26	0.072		
		Jung 1936	d	0.08	0.022		
		Jung 1936	r	0.16	0.044		
		Bahmuller 1983		0.019 - 0.547	0.0053 - 0.1517		
		Bahmuller 1984		0.030 - 0.16	0.0083 - 0.0444		
		Dollnick et al 1988		0.116	0.032		
		Guth 1997		0.125	0.035		
		Guth 1997		6.3	1.75		
		Ferreira et al 1998		2.8	0.777		
		Nagata 2003	d	0.0061	<b>0.00169</b>		
		168	Isobutyl Acetate	Backman 1917	r	1.9 - 2.1	0.40 - 0.44
				May 1966	d	17	3.6
May 1966	r			34	7.2		
Hellman & Small 1974	d			1.7	0.36		
Hellman & Small 1974	r			2.4	0.51		
Cristoph 1983	r			0.42 - 0.52	0.088 - 0.109		
Nagata 2003	d			0.038	<b>0.008</b>		
Komthong et al 2006				21.1 - 612	4.4 - 129		

**Table 6.3 – Odor Threshold Values, cont.**

*Bold = Lowest Value Reported*

#	Chemical Name	Source	Type of Threshold	Odor Thresholds	
				mg/m <sup>3</sup>	ppm
169	Isobutyl Alcohol	Passy 1892c	d	1	0.33
		Zwaardemaker 1914	d	500	165
		Backman 1917	r	0.2 - 0.4	0.066 - 0.13
		Jones 1955c	r	31	10
		May 1966	d	120	40
		May 1966	r	160	53
		Laffort & Dravnieks 1973		3	1
		Hellman & Small 1973a,b, 1974	d	2	0.66
		Hellman & Small 1973a,b, 1974	r	5.4	1.8
		Anon. 1980	d	0.036	0.012
		Anon. 1980	r	0.66	0.218
		Punter 1983	d	3.8 - 8.1	1.25 - 2.67
		Cristoph 1983	r	0.7 - 1.0	0.23 - 0.33
		Nagy 1991	d	2.64	0.87
		Nagy 1991	d	1.73	0.57
		Guth 1997		0.64	0.21
		Guth 1997		200	66
Nagata 2003	d	0.033	<b>0.011</b>		
170	Isobutyraldehyde	Hellman & Small 1973a,b, 1974	d	0.14	0.0475
		Hellman & Small 1973a,b, 1974	r	0.41	0.139
		Amoore 1977	d	0.015	0.0051
		Hendriks 304	d	0.022	0.0075
		Nagata 2003	d	0.001	<b>0.00034</b>
171	Isooctyl Alcohol	Tsulaya et al 1972		0.26	0.049
		Nagata 2003	d	0.049	<b>0.0092</b>
172	Isophorone	Hellman & Small 1974	d	1.1	0.19
		Hellman & Small 1974	r	3	0.53
		Ziemer et al 2000	d	0.0017	<b>0.0003</b>
173	Isoprene	Artho & Koch 1973		1 - 10	0.359 - 3.59
		Nagata 2003	d	0.13	<b>0.047</b>
174	Isopropyl Acetate	Backman 1917	r	27 - 33	6.5 - 7.9
		Jung 1936	d	1.9	0.45
		Jung 1936	r	1.9 - 2.9	0.45 - 0.69
		May 1966	d	140	34
		May 1966	r	170	41
		Hellman & Small 1974	d	2.1	0.5
		Hellman & Small 1974	r	3.8	0.91
		Scharfenberger 1990		68	16
		Nagy 1991	d	9.4	2.25
Nagata 2003	d	0.67	<b>0.16</b>		

Table 6.3 – Odor Threshold Values, cont.

Bold = Lowest Value Reported

#	Chemical Name	Source	Type of Threshold	Odor Thresholds	
				mg/m <sup>3</sup>	ppm
175	Isopropyl Alcohol	Passy 1892c	d	40	16
		Backman 1917	r	18 - 24	7.3 - 9.8
		Jung 1936	d	3.9 - 32.4	1.6 - 13.2
		Jung 1936	r	7.8 - 31.2	3.2 - 12.7
		Scherberger et al 1958	r	500	203
		Cheesman & Kirkby 1959	d	43 - 290	17 - 118
		May 1966	d	90	37
		May 1966	r	120	49
		Gorlova 1970		2.5	<b>1</b>
		Koster 1968a, 1971	d	64 - 5,400	26 - 2,197
		Dravnieks & Laffort 1972		57.4	23.4
		Dravnieks 1974	d	1,500	610
		Hellman & Small 1974	d	8	3.3
		Hellman & Small 1974	r	18.8	7.6
		Scharfenberger 1990		491	200
		Nagy 1991	d	180	73
		Cometto - Muniz & Cain 1993		1,245	507
		Cometto - Muniz 1993, 1999	d	1,245	507
		Smith & Duffy 1995	d	103	42
		Smith & Duffy 1965	r	228	93
Smeets & Dalton 2002	d	28 - 98	11 - 40		
Nagata 2003	d	65	26		
176	Isopropyl amine	Hellman & Small 1974	d	0.5	0.21
		Hellman & Small 1974	r	1.7	0.7
		Nagata 2003	d	0.06	<b>0.025</b>
177	Isopropyl Ether	Hellman & Small 1974	d	0.07	<b>0.017</b>
		Hellman & Small 1974	r	0.22	0.053
178	d-Limonene	Fuller et al 1964	r	0.058	0.01
		Apell 1969		0.01	<b>0.0018</b>
		Dravnieks et al 1986	d	1.7	0.31
		Nagata 2003	d	0.21	0.04
		Langridge 2004		1.6878	0.3
		Langridge 2004		0.0539	0.01
179	Maleic Anhydride	Grigorieva 1964		1.0 - 1.3	<b>0.25</b> - 0.32
180	Mercaptoethanol	Vermeulen & Collin 2006		0.24	0.075
181	Mesityl Oxide	Toxicity Data Sheet 1957		48	12
		Hellman & Small 1974	d	0.07	<b>0.017</b>
		Hellman & Small 1974	r	0.2	0.05
182	Methacrylic Acid	Piringer & Granzer 1984		10	2.84
		Nagy 1991	d	1.9	<b>0.54</b>
183	Methacrylonitrile	Pozzani et al 1968	d	19	6.9
		Nagata 2003	d	8.1	<b>2.95</b>
184	Methane	Laffort & Gortan 1987		1,900,000	2,896,197

**Table 6.3 – Odor Threshold Values, cont.**

*Bold = Lowest Value Reported*

#	Chemical Name	Source	Type of Threshold	Odor Thresholds	
				mg/m <sup>3</sup>	ppm
185	2 - Methoxyethanol	May 1966	d	190	61
		May 1966	r	280	90
		Hellman & Small 1973a,b, 1974	d	<0.3	<b>&lt;0.096</b>
		Hellman & Small 1973a,b, 1974	r	0.7	0.22
186	2 - Methoxyethyl Acetate	Hellman & Small 1973a,b, 1974	d	1.6	<b>0.33</b>
		Hellman & Small 1973a,b, 1974	r	3.1	0.64
187	1 - Methoxy - 2 - Propanol	Stewart et al 1970	d	37	10
		Nagy 1991	d	121	33
		Nagy 1991	d	30.908	<b>8.39</b>
188	1-Methoxy-2-Propyl Acetate	Nagy 1991	d	0.7	0.13
		Ziemer et al 2000	d	0.016	<b>0.0029</b>
189	Methyl Acetate	Zwaardemaker 1914, 1927	d	2	0.66
		Backman 1917	r	67	22
		Gofmekler 1960		0.5	<b>0.17</b>
		Janicek et al 1960		5250	1733
		Naus 1962	d	0.7	0.231
		May 1966	d	550	182
		May 1966	r	900	297
		Anon. 1980	d	22	7.3
		Anon. 1980	r	63	20.8
		Scharfenberger 1990		579	191
		Cometto - Muniz & Cain 1991		8,628	2,848
		Cometto - Muniz 1993	d	8,628	2,848
190	Methyl Acrylate	Nagata 2003	d	5.1	1.68
		Bezpalkova 1967a, b		0.017	0.0048
		Anon. 1980	d	0.01	<b>0.003</b>
		Anon. 1980	r	0.06	0.020
		Bahnmuller 1984		0.015 - 0.088	0.004 - 0.025
		Priinger & Granzer 1984		0.05	0.0142
		Nagy 1991	d	0.061	0.0173
		Nagata 2003	d	0.012	0.0034
191	Methyl Alcohol	Passy 1892c	d	1,000	764
		Zwaardemaker 1914	d	600	458
		Backman 1917	r	900 - 1,000	687 - 763
		Grijns 1919		2,150	1,643
		Zwaardemaker 1927		2,150	1,643
		Jung 1936	d	23.4 - 54.6	17.9 - 41.7
		Jung 1936	r	54.6 - 62.4	41.7 - 47.7
		Gavaudan et al 1948		150	115
Mullins 1955	r	19,300	14,746		

**Table 6.3 – Odor Threshold Values, cont.**

*Bold = Lowest Value Reported*

#	Chemical Name	Source	Type of Threshold	Odor Thresholds	
				mg/m <sup>3</sup>	ppm
191	Methyl Alcohol cont.	Scherberger et al 1958	r	1,950	1,490
		Chao - Chen - Tzi 1959		4.3	3.3
		Janicek et al 1960		4,000	3,056
		Pliska & Janicek 1965		260,000	198,656
		May 1966	d	7,800	5,960
		May 1966	r	11700	8,940
		Ubaidullaev 1966a		4.5	3.4
		Leonardos et al 1969	r	130	99
		Hellman & Small 1974	d	5.5	4.2
		Hellman & Small 1974	r	69	53
		Anon. 1980	d	74	57
		Anon. 1980	r	260	199
		Naus 1982	d	4	<b>3.05</b>
		Naus 1982	r	10	7.63
		Cometto - Muniz & Cain 1990	d	2,096	1,599
		Cometto - Muniz 1993	d	2,096	1,599
		Scharfenberger 1990		1,975	1,507
Nagata 2003	d	43	33		
192	Methylamine	Leonardos et al 1969	r	0.027	0.021
		Nishida et al 1975	d	0.065	0.0512
		Nishida et al 1979	d	6.1	4.8
		Anon. 1980	d	0.0012	<b>0.0009</b>
		Anon. 1980	r	0.012	0.009
		Hill & Barth 1976		0.027	0.021
		Nagy 1991	d	0.23	0.18
		Nagata 2003	d	0.046	0.036
193	Methyl n - Amyl Ketone	Stone et al 1962	d	0.9	0.19
		Pangborn et al 1964	d	0.82	0.18
		Teranishi et al 1974		0.84	0.18
		Hall & Andersson 1983	d	1.3	0.28
		Nagy 1991	d	1.2	0.26
		Nagy 1991	d	0.398	0.085
		Cometto - Muniz & Cain 1993	d	3.3	0.71
		Cometto - Muniz 1993	d	3.3	0.71
		Cometto - Muniz et al 1999	d	0.29 - 0.65	0.062 - 0.139
		Ziemer et al 2000	d	0.045	0.01
		Nagata 2003	d	0.032	0.007
		Cometto - Muniz et al 2004	d	0.47	0.1
		Cain et al 2008	d	0.062	0.013
		Cometto - Muniz et al 2008	d	0.023	0.0049
		Cometto - Muniz & Abraham 2009a	d	0.023	0.0049
Yang et al 2008		0.0035	<b>0.00075</b>		

**Table 6.3 – Odor Threshold Values, cont.**

*Bold = Lowest Value Reported*

#	Chemical Name	Source	Type of Threshold	Odor Thresholds	
				mg/m <sup>3</sup>	ppm
194	<b>N - Methyl Aniline</b>	Backman 1917	r	6.9 - 8.6	<b>1.6</b> - 2.0
195	<b>2 - Methylbutyl Acetate</b>	Cristoph 1983	r	0.14 - 0.21	<b>0.026</b> - 0.039
196	<b>Methyl tert Butyl Ether</b>	Smith & Duffy 1995	d	0.11	<b>0.03</b>
		Smith & Duffy 1995	r	0.22	0.06
		Prah et al 1994	d	0.63	0.17
		Schulman 2001	d	0.63	0.17
197	<b>Methyl n - Butyl Ketone</b>	Backman 1917	r	0.28 - 0.35	0.068 - 0.085
		Hall & Andersson 1983	d	4.7	1.15
		Nagata 2003	d	0.098	<b>0.024</b>
198	<b>Methyl Chloride</b>	Leonardos et al 1969	r	>21	>10
199	<b>Methyl Chloroform</b>	Scherberger et al 1958	r	1,650	302
		May 1966	d	2,100	385
		May 1966	r	3,900	715
		Kendall et al 1968	r	88	16
		Don 1986	d	5.3	<b>0.97</b>
200	<b>Methyl - 2 - Cyanoacrylate</b>	McGee et al 1968		4.5 - 13.5	<b>0.99</b> - 2.97
201	<b>Methyl Cyclohexane</b>	Nagata 2003	d	0.6	0.149
202	<b>2 - Methyl Cyclohexanone</b>	Van Thriel et al 2006	d	0.83	0.181
203	<b>Methylene Bisphenyl Isocyanate</b>	Woolrich 1982		4	0.39
204	<b>Methylene Chloride</b>	Lehmann & Schmidt - Kehl 1936		1,100	317
		Scherberger et al 1958	r	1,530	440
		May 1966	d	550	158
		May 1966	r	790	227
		Leonardos et al 1969	r	730	210
		Basmadhijewa et al 1970	d	4.1 - 33.2	<b>1.2</b> - 9.6
		Don 1986	d	640	184
		Nagata 2003	d	560	161
205	<b>Methyl Ethyl Ketone</b>	Backman 1917	r	63 - 70	21 - 24
		May 1966	d	80	27
		May 1966	r	163	55
		Leonardos et al 1969	r	29	9.8
		Hartung et al 1971		7	2.4
		Mukhitov & Azimbekov 1971		0.75	0.25
		Dravnieks & Laffort 1972		22	7.5
		Artho & Koch 1973		100 - 1,000	34 - 339
		Dravnieks 1974	d	250	85
		Hellman & Small 1974	d	5.8	2
		Hellman & Small 1974	r	16	5.4
		Anon. 1980	d	8.4	2.8
		Anon. 1980	r	29	9.8
		Hall & Andersson 1983	d	61	21
		Doty et al 1988	d	16.5 - 23.9	5.6 - 8.1

Table 6.3 – Odor Threshold Values, cont.

Bold = Lowest Value Reported

#	Chemical Name	Source	Type of Threshold	Odor Thresholds	
				mg/m <sup>3</sup>	ppm
205	Methyl Ethyl Ketone cont.	Scharfenberger 1990		126	43
		Doty 1991	d	2.9 - 51.6	0.983 - 17.5
		Nagy 1991	d	4.8	1.63
		Ziemer et al 2000	d	0.21	<b>0.07</b>
		Nagata 2003	d	1.3	0.44
206	Methyl Formate	Backman 1917	r	165 - 180	<b>67</b> - 73
		May 1966	d	5,000	2,035
		May 1966	r	6,900	2,809
		Nagata 2003	d	325	132
207	Methyl Hydrazine	Jacobson et al 1955		1.9 - 5.7	<b>1</b> - 3
208	Methyl Isoamyl Ketone	Hellman & Small 1974	d	0.06	0.013
		Hellman & Small 1974	r	0.23	0.049
		Nagy 1911	d	0.63	0.135
		Nagata 2003	d	0.0099	<b>0.0021</b>
209	Methyl Isobutyl Ketone	Backman 1917	r	0.6 - 0.8	0.15 - 0.2
		Middleton 1956	r	1.9	0.46
		May 1966	d	32	7.8
		May 1966	r	64	16
		Stone et al 1967b, 1972	d	0.97 - 9.7	0.24 - 2.4
		Steinmetz et al 1969	d	1.21	0.3
		Leonardos et al 1969	r	1.9	0.46
		Hellman & Small 1974	d	0.4	0.1
		Hellman & Small 1974	r	1.1	0.27
		Anon. 1980	d	0.7	0.17
		Anon. 1980	r	2.8	0.68
		Dravnieks et al 1986	d	0.14	<b>0.03</b>
		Nagy 1911	d	6.3	1.54
		Dalton et al 2000	d	41	10.00
		Ziemer et al 2000	d	1.1	0.27
		Nagata 2003	d	0.7	0.17
210	Methyl Isocyanate	Kimmerle & Eben 1964		5	2.14
211	Methyl Isopropyl Ketone	Backman 1917	r	15 - 17	4.3 - 4.8
		Nagata 2003	d	1.8	<b>0.51</b>
212	Methyl Mercaptan	Katz & Talbert 1930	d	0.081	0.041
		Bozza & Colombo 1949		1	0.51
		Freudenberg & Reichert 1955		0.0005	0.00025
		Guadagni 1966		0.0002	0.0001
		Endo et al 1967		1.1	0.56
		Leonardos et al 1969	r	0.0042	0.0021
		Wilby 1969	r	0.002	0.001
		Hamanabe et al 1969		0.0002	0.0001
		Sanders et al 1970		0.0019	0.00097
Selyuzhitskii 1972		0.0005	0.00025		



**Table 6.3 – Odor Threshold Values, cont.**

*Bold = Lowest Value Reported*

#	Chemical Name	Source	Type of Threshold	Odor Thresholds	
				mg/m <sup>3</sup>	ppm
212	Methyl Mercaptan cont.	Artho & Koch 1973		0.000000000001	<b>0.0000000000051</b>
		Blanchard 1976		0.003	0.0015
		Williams et al 1977	d	0.0000003	0.0000002
		Nishida et al 1979	d	0.038	0.019
		Bedborough & Trott 1979	d	0.00016	0.000081
		Anon. 1980	d	0.00024	0.00012
		Anon. 1980	r	0.0013	0.00066
		Nagy 1991	d	0.0024	0.0012
		Nagata 2003	d	0.00014	0.000071
		Greenman et al 2004		0.00048	0.00024
		Clindemann et al 2006	d	0.001	0.00051
213	Methyl Methacrylate	Filatova 1962		0.2	0.049
		Leonardos et al 1969	r	0.85	0.21
		Hellman & Small 1973a,b, 1974	d	0.2	0.049
		Hellman & Small 1973a,b, 1974	r	1.4	0.34
		Holland 1974		0.057	<b>0.014</b>
		Anon. 1980	d	0.62	0.15
		Anon. 1980	r	1.9	0.46
		Piringer & Granzer 1984		0.7	0.17
		Nagy 1991	d	2.7	0.66
		Nagata 2003	d	0.86	0.21
214	2 - Methyl-naphthalene	Moriguchi et al 1983	d	0.004	0.00069
215	Methyl Parathion	Akhmedov 1968		0.0125	0.0012
216	4 - Methyl - 2 - Pentanol	Hellman & Small 1974	d	1.4	<b>0.335</b>
		Hellman & Small 1974	r	2.2	0.526
217	Methyl Propyl Ketone	Backman 1917	r	11 - 15	3.1 - 4.3
		May 1966	d	27	7.7
		May 1966	r	48	13.6
		Hall & Andersson 1983	d	22	6.24
		Laska & Hudson 1991	d	6.7 - 8.3	1.9 - 2.4
		Cometto - Muniz & Cain 1993	d	30.1	8.54
		Cometto - Muniz 1993	d	30.1	8.54
		Patterson et al 1993	d	9.1	2.58
		Nagata 2003	d	0.098	<b>0.028</b>
		Komthong et al 2006		230	65
		Cometto - Muniz et al 2008	d	0.35	0.099
		Cometto - Muniz & Abraham 2009a	d	0.35	0.099
218	1 - Methyl - 2 - Pyrrolidone	Nagy 1991	d	41	10
		Nagy 1991	d	17.113	<b>4.22</b>

Table 6.3 – Odor Threshold Values, cont.

Bold = Lowest Value Reported

#	Chemical Name	Source	Type of Threshold	Odor Thresholds	
				mg/m <sup>3</sup>	ppm
219	Methyl Styrene	Wolf et al 1956		48 - 240	9.9 - 49.7
		Minaev 1966		0.1	<b>0.02</b>
		Hellman & Small 1974	d	0.25	0.05
		Hellman & Small 1974	r	0.75	0.16
		Nagy 1991	d	2.2	0.46
220	Methyl Vinyl Ketone	Martirosyan 1970		0.5	0.174
221	Monochloroacetic Acid	Backman 1917	r	0.6	0.155
		Smith & Hochstetler 1969	r	0.05	<b>0.013</b>
222	Morpholine	Hellman & Small 1973a,b, 1974	d	0.04	<b>0.01</b>
		Hellman & Small 1973a,b, 1974	r	0.25	0.07
223	Naphthalene	Backman 1917	r	0.05 - 0.055	0.0095 - 0.0105
		Mitsumoto 1926	r	4.0 - 4.4	0.76 - 0.84
		Hesse 1928	r	0.3	0.057
		Morimura 1934	r	3.37 - 5.34	0.64 - 1.02
		Robbins 1951		<1.6	<0.31
		Punter 1983	d	0.2	0.038
		Moriguchi et al 1983	d	0.007	<b>0.0013</b>
		Savenhed et al 1985	d	0.01 - 0.04	0.0019 - 0.0076
	Nagy 1991	d	0.45	0.086	
224	1 - Naphthylamine	Backman 1917		0.14 - 0.29	<b>0.024</b> - 0.050
225	2 - Naphthylamine	Backman 1917	r	1.4 - 1.9	<b>0.24</b> - 0.32
226	Nickel Carbonyl	Armit 1907		3.5	<b>0.5</b>
		Kincaid 1956		7 - 21	1.0 - 3.0
227	Nicotine	Walker et al 1996		0.066	0.0099
228	Nitric Acid	Melekhina 1968	d	0.7	0.27
229	Nitrobenzene	Hermanides 1909	r	0.0412	0.0082
		Zwaardemaker 1914, 1927	d	0.04 - 0.041	0.0079 - 0.0081
		Backman 1917	r	0.34 - 0.70	0.068 - 0.14
		Allison & Katz 1919		146	29
		Henning 1924	d	0.0065	0.0013
		Katz & Talbert 1930	d	9.6	1.9
		Van Anrooij 1931	d	0.019	0.0038
		Janicek et al 1960		19	3.78
		Andrcescheva 1964		0.0182	0.0036
		Gavaudan & Poussel 1966		0.15	0.03
		Leonardos et al 1969	r	0.024	0.0048
		Randebroek 1971		0.002	<b>0.0004</b>
		Ozturk 1976	d	0.363	0.072
		Naus 1982	d	0.2	0.04
		Naus 1982	r	20	3.97
	Randebroek 1986		0.0053	0.0011	

**Table 6.3 – Odor Threshold Values, cont.**

*Bold = Lowest Value Reported*

#	Chemical Name	Source	Type of Threshold	Odor Thresholds	
				mg/m <sup>3</sup>	ppm
230	Nitrogen Dioxide	Beck 1959		0.2 - 1.0	0.1 - 0.5
		Henschler et al 1960	d	0.2 - 1.0	0.1 - 0.5
		Shalamberidze 1967		0.23	0.12
		Rumsey & Cesta 1970		<1	<0.5
		Knuth 1973		0.11	<b>0.058</b>
		Prusakov et al 1976		0.2 - 0.26	0.11 - 0.14
		Braker & Mossman 1980		<9.4	<5
		Nagata 2003		0.23	0.12
231	Nitromethane	Nagy 1991	d	124	50
232	1 - Nitropropane	Dravnieks 1974	d	510	140
		Dravnieks & Laffort 1972		28.2	<b>7.7</b>
233	2 - Nitropropane	Treon & Dutra 1952		297 - 1,050	82 - 288
		Hine et al 1978	r	580	159
		Crawford et al 1984	d	18	<b>4.94</b>
234	N - Nitrosodimethyl Amine	Prusakov et al 1976		0.024 - 0.04	<b>0.0079</b> - 0.013
235	Nonane	Mullins 1955	r	108	21
		Laffort & Dravnieks 1973		60	11
		Nagata 2003	d	12	<b>2.3</b>
236	Octane, all isomers	Jones 1955c	d	550	118
		May 1966	d	710	152
		May 1966	r	1,100	235
		Laffort & Dravnieks 1973		71	15
		Nagy 1991	d	61.8	13
		Nagy 1991	d	90.102	19
		Nagata 2003	d	8	1.71
		Nagata 2003	d	3.1	<b>0.66</b>
237	1 - Octanol	Backman 1917	r	0.005 - 0.008	0.0009 - 0.0015
		Rcoen 1920	r	0.005	<b>0.0009</b>
		Gavaudan et al 1948		0.02	0.0038
		Mullins 1955	r	5.44	1.02
		Pliska & Janicek 1960		0.14	0.026
		Pliska 1962		9	1.69
		Stone et al 1967	d	0.05	0.009
		Cain 1969	r	0.5	0.09
		Punter 1983	d	0.73	0.137
		Cristoph 1983	r	0.03 - 0.05	0.0056 - 0.009
		Cometto - Muniz & Cain 1990	d	0.037	0.0069
		Cometto - Muniz 1993	d	0.037	0.0069
		Nagata 2003	d	0.014	0.0026
		Cometto - Muniz & Abraham 2008	d	0.023	0.0043
		Yang et al 2008		0.022	0.0041

Table 6.3 – Odor Threshold Values, cont.

Bold = Lowest Value Reported

#	Chemical Name	Source	Type of Threshold	Odor Thresholds	
				mg/m <sup>3</sup>	ppm
238	1 - Octene	Dravnieks & Laffort 1972		0.33	0.07
		Dravnieks 1974	d	5	1.09
		Kosinowski & Piringer 1983		37	8.06
		Cometto - Muniz 1993	d	945	206
		Cometto - Muniz & Cain 1994	d	945	206
		Nagata 2003	d	0.0046	<b>0.001</b>
239	Oxygen Difluoride	Lester & Adams 1965		0.22	0.0996
240	Ozone	Witheridge & Yaglou 1939		0.02 - 0.03	0.010 - 0.015
		Wilska 1951		<0.2	<0.10
		Beck 1959		<0.04	<0.02
		Henschler et al 1960		<0.04	<0.020
		Buchberg et al 1961		0.07 - 0.5	0.036 - 0.25
		Eglite 1968		0.015	0.0076
		Nagata 2003	d	0.0064	<b>0.0033</b>
		Cain et al 2007a	d	0.014	0.0071
241	Pentaborane	Krackow 1963		2.5	0.97
242	Pentane, all isomers	Patty & Yant 1929		1450	491
		Mullins 1955	r	3,090	1,147
		Laffort & Dravnieks 1973		350	119
		Nagata 2003	d	4.1	1.39
		Nagata 2003	d	3.8	<b>1.29</b>
243	2,4 - Pentanedione	Hellman & Small 1974	d	0.04	<b>0.0098</b>
		Hellman & Small 1974	r	0.08	0.0195
244	Pentanol, all isomers	Backman 1917	r	1.0 - 1.2	0.28 - 0.33
		Allison & Katz 1919		225	62
		Jung 1936	d	0.4 - 0.81	0.11 - 0.22
		Jung 1936	r	1.62	0.45
		Janicek et al 1960		11	3.05
		Naus 1962	d	4	1.11
		Pliska & Janicek 1965		1,100	305
		Gavaudan & Poussel 1966		0.4	0.11
		May 1966	d	35	9.71
		May 1966	r	80	22.00
		Stone et al 1972	d	1.2	0.33
		Baikov et al 1973		0.1	0.028
		Hellman & Small 1974	d	0.8	0.22
		Hellman & Small 1974	r	1.1	0.31
		Naus 1982	d	4	1.11
		Naus 1982	r	30	8.32
		Punter 1983	d	2	0.55
		Cristoph 1983	r	1.0 - 1.1	0.28 - 0.30
		Cometto - Muniz & Cain 1990	d	5	1.39
		Cometto - Muniz 1993	d	5	1.39

**Table 6.3 – Odor Threshold Values, cont.**

*Bold = Lowest Value Reported*

#	Chemical Name	Source	Type of Threshold	Odor Thresholds	
				mg/m <sup>3</sup>	ppm
244	Pentanol, all isomers cont.	Lindell 1991	d	1.3	0.36
		Ziemer et al 2000	d	0.02	<b>0.0055</b>
		Nagata 2003	d	0.36	0.10
		Yang et al 2008		0.153	0.04
		Nagata 2003	d	1	0.28
		Passy 1892c	d	20 - 40	5.5 - 11
		Backman 1917	r	2.0 - 3.0	0.55 - 0.83
		Nagata 2003	d	0.32	0.089
		Backman 1917	r	1.4 - 1.7	0.39 - 0.47
		Hellman & Small 1973a,b, 1974	d	0.14	0.039
		Hellman & Small 1973a,b, 1974	r	0.83	0.23
		Cristoph 1983	r	0.9 - 1.0	0.25 - 0.28
		Komthong et al 2006		329	91
		Nagata 2003	d	1	0.28
245	Perchloroethylene	Carpenter 1937		<340	<50
		May 1966	d	320	47
		May 1966	r	480	71
		Leonardos et al 1969	r	32	5
		Anon. 1980	d	12	2
		Anon. 1980	r	55	8
		Torkelson & Rowe 1981		340	50
		Don 1986	d	8.1 - 8.3	1.19 - 1.22
		Hoshika et al 1993	d	8.1 - 8.3	1.19 - 1.22
		Hoshika et al 1993	d	12	2
		Nagata 2003	d	5.2	<b>0.767</b>
246	Perchloryl Fluoride	Braker & Mossman 1980		42	14.58
247	Phenol	Grijns 1906		2.2 - 6.8	0.57 - 1.8
		Zwaardemaker 1914, 1927	d	4	1
		Backman 1917	r	0.13 - 0.26	0.034 - 0.068
		Henning 1924	d	1.2	0.31
		Mukhitov 1962, 1963		0.022	0.0057
		Itskovich & Vinogradova 1962		3	0.78
		Pogosyan 1965		0.022	0.0057
		Komeev 1965		0.0172	<b>0.0045</b>
		Makhinya 1966		0.022	0.0057
		Basmdzhieva & Argirova 1968		0.021	0.0055
		Leonardos et al 1969	r	0.18	0.047
		Takhirov 1974		0.022	0.0057
		Punter 1975, 1979	d	0.8	0.21
Makeicheva 1978		0.027	0.007		

Table 6.3 – Odor Threshold Values, cont.

Bold = Lowest Value Reported

#	Chemical Name	Source	Type of Threshold	Odor Thresholds	
				mg/m <sup>3</sup>	ppm
247	Phenol cont.	Anon. 1980	d	0.046	0.012
		Anon. 1980	r	0.22	0.057
		Naus 1982	d	0.2	0.2
		Naus 1982	r	20	20
		Punter 1983	d	0.23	0.06
		Moriguchi et al 1983	d	0.046	0.012
		Kohler & Homans 1980		5.8 - 7.5	1.51 - 1.95
		Homans 1984		5.8 - 7.5	1.51 - 1.95
		Hoshika et al 1993	d	0.039	0.010
		Don 1986	d	0.039	0.010
		Nagy 1991	d	0.5	0.130
		Hoshika et al 1993	d	0.046	0.012
		Nagata 2003	d	0.021	0.0055
248	Phenyl Mercaptan	Katz & Talbert 1930	d	0.0012	0.00027
		Stuiver 1958	d	0.00014	<b>0.00003</b>
249	Phosgene	Fieldner et al 1921		23	5.7
		Suchier 1929		4	1
		Schley 1934	d	0.5	<b>0.12</b>
		Schley 1934	r	0.5 - 1	0.12 - 0.25
		Prentiss 1937		4.4	1.09
		Patty 1962c		2	0.49
		Leonardos et al 1969	r	4	1
250	Phosphine	Valentin 1848		1.4	1
		Valentin 1850		0.13	0.094
		Singh et al 1967	d	7	5
		Berck 1968	r	<2	<1.4
		Leonardos et al 1969	r	0.03	0.022
		Dumas & Bond 1974	d	>280	>201
		Fluck 1976	r	0.014 - 2.8	<b>0.010</b> - 2.014
251	Phthalic Anhydride	Slavgorodskiy 1968		0.32	0.053
252	Picolines, all isomers	Hellman & Small 1974	d	0.05	0.0131
		Hellman & Small 1974	r	0.09	0.0236
		Moriguchi et al 1983	d	0.01	<b>0.0026</b>
253	Piperdine	Geier 1936	d	0.5	<b>0.14</b>
		Geier 1936	r	2	0.57
		Nawakowski 1980		<7	<2
254	Propane	Patty & Yant 1929		36,000	19,964
		Laffort & Dravnieks 1973		22,000	12,200
		Nagata 2003		2,700	<b>1,497</b>
255	Propionaldehyde	Backman 1917	r	0.02	0.0084
		Pliska & Janicek 1965		240	101
		Hartung et al 1971		1.7	0.716
		Knuth 1973		0.026	0.011

**Table 6.3 – Odor Threshold Values, cont.**

*Bold = Lowest Value Reported*

#	Chemical Name	Source	Type of Threshold	Odor Thresholds	
				mg/m <sup>3</sup>	ppm
255	Propionaldehyde cont.	Hellman & Small 1974	d	0.02	0.008
		Hellman & Small 1974	r	0.1	0.042
		Teranishi et al 1974		0.02	0.008
		Bedborough & Trott 1979	d	0.014	0.006
		Anon 1980	d	0.0036	0.0015
		Anon 1980	r	0.036	0.015
		Hall & Andersson 1983	d	0.69	0.29
		Cristoph 1983	d	0.33 - 0.40	0.139 - 0.168
		Nagy 1991	d	0.21	0.088
		Nagata 2003	d	0.0024	<b>0.001</b>
		Cometto - Muniz & Abraham 2010a	d	0.0048	0.002
256	Propionic Acid	Passy 1893b, 1893c	d	0.05	0.017
		Backman 1917	r	0.5	0.17
		Grijns 1919		0.6	0.2
		Mitsumoto 1926	r	1.7 - 2.55	0.56 - 0.84
		Hesse 1926	r	4.6	1.5
		Morimura 1934	r	1.77 - 2.38	0.58 - 0.79
		Stone 1963a, 1963c	d	0.39 - 0.68	0.13 - 0.22
		Stone & Bosley 1965	d	0.89	0.29
		Goldenberg 1967	d	0.003	<b>0.00099</b>
		Hellman & Small 1974	d	0.08	0.026
		Hellman & Small 1974	r	0.1	0.033
		Anon. 1980	d	0.0051	0.0017
		Anon. 1980	r	0.025	0.0083
		Punter 1983	d	0.44 - 0.58	0.15 - 0.19
		Dollnick et al 1988		0.147	0.049
		Walker et al 1990		14.1	4.65
		Nagy 1991	d	1.2	0.4
		Walker et al 1996		0.3 - 3	0.099 - 0.99
		Nagata 2003	d	0.017	0.0056
		Van Thriel et al 2006	d	1	0.33
257	n - Propyl Acetate	Backman 1917	r	12	2.9
		Jung 1936	d	0.35	0.084
		Jung 1936	r	0.35 - 0.62	0.084 - 0.15
		May 1966	d	70	17
		May 1966	r	110	26
		Hellman & Small 1974	d	0.2	<b>0.048</b>
		Hellman & Small 1974	r	0.6	0.14
		Cometto - Muniz & Cain 1991	d	104	25
		Cometto - Muniz & Cain 1991	d	104	25
		Nagata 2003	d	1	0.239
		Komthong et al 2006		363	87

Table 6.3 – Odor Threshold Values, cont.

Bold = Lowest Value Reported

#	Chemical Name	Source	Type of Threshold	Odor Thresholds	
				mg/m <sup>3</sup>	ppm
258	Propyl Alcohol	Passy 1892c	d	5 - 10	2.0 - 4.1
		Backman 1917	r	3 - 5	1.2 - 2.0
		Jung 1936	d	0.8 - 8	0.33 - 3.3
		Jung 1936	r	8 - 24	3.3 - 9.8
		Jones 1955c	d	140	57
		Janicek et al 1960		540	220
		Pliska & Janicek 1965		25,000	10,172
		Guadagni 1966		9	3.7
		May 1966	d	80	33
		May 1966	r	150	61
		Khachatryan et al 1968		1.25	0.51
		Cain 1969	r	660	269
		Corbit & Engen 1971		46 - 51	19 - 21
		Stone et al 1972	d	2.8	1.1
		Dravnieks & Laffort 1972		32.3	13
		Dravnieks 1974	d	100	41
		Hellman & Small 1974	d	<0.075	<b>&lt;0.031</b>
		Hellman & Small 1974	r	0.2	0.081
		Laing 1975	d	100	41
		Naus 1982	d	2	0.81
Naus 1982	r	20	8		
Punter 1983	d	5.9	2.4		
Cristoph 1983	r	2.9 - 3.2	1.18 - 1.3		
Cometto - Muniz & Cain 1990, 1993	d	27.5 - 35	11 - 14		
Cometto - Muniz 1993	d	27.5 - 35	11 - 14		
Scharfenberger 1990		16	6.5		
Nagata 2003	d	0.24	0.098		
259	Propylene	Krasovitskaya & Malyarova 1968		17.3	<b>10.1</b>
		Laffort & Dravnieks 1973		170	99
		Hellman & Small 1974	d	38	22.1
		Hellman & Small 1974	r	115	67
		Nagata 2003	d	22	13
260	Propylene Dichloride	Hellman & Small 1974	d	1.2	<b>0.26</b>
		Hellman & Small 1974	r	2.4	0.52
		Nagy 1991	d	40	8.66
261	Propylene Glycol	Nagy 1991	d	16	5.14
262	Propylene Glycol Dinitrate	Stewart et al 1974	d	1.6	0.236
263	Propylene Oxide	Jacobson et al 1956	d	473	199
		Hellman & Small 1974	d	24	<b>10</b>
		Hellman & Small 1974	r	84	35



Table 6.3 – Odor Threshold Values, cont.

Bold = Lowest Value Reported

#	Chemical Name	Source	Type of Threshold	Odor Thresholds	
				mg/m <sup>3</sup>	ppm
264	Pyridine	Hermanides 1909	r	0.16	0.05
		Zwaardemaker 1914, 1927	d	0.04	0.012
		Backman 1917	r	0.2	0.062
		Allison & Katz 1919		32	9.9
		Katz & Talbert 1930	d	0.74	0.23
		Van Anrooij 1931	d	0.078	0.024
		Geier 1936	d	0.09	0.029
		Geier 1936	r	0.095	0.029
		Jones 1955c	d	40	12
		Sales 1958		0.42	0.13
		Janicek et al 1960		4.6	1.4
		Sutton 1962b		<3.2	<0.99
		Kristesashvili 1965		0.21	0.065
		Leonardos et al 1969	r	0.067	0.021
		Dravnieks & Laffort 1972		0.33	0.1
		Laffort & Dravnieks 1973		0.74	0.23
		Dravnieks 1974	d	6	1.9
		Amoore & Buttery 1978	d	2.4	0.74
		Laing et al 1978	r	2.4	0.74
		Hangartner 1981		0.08 - 2.9	0.025 - 0.90
		Naus 1982	d	1	0.31
		Naus 1982	r	10	3.1
		Moriguchi et al 1983	d	0.023	<b>0.007</b>
		Bahnmuller 1983		0.132 - 1.21	0.041 - 0.374
		Ahlstrom et al 1986a	d	0.124 - 0.146	0.038 - 0.045
		Amoore 1986a,b	d	2.1	0.65
		Don 1986	d	0.12	0.04
		Hartigh 1986	d	0.15 - 0.29	0.046 - 0.090
		MacLeod et al 1986		0.054	0.017
		Cain et al 1987	d	0.34	0.11
		Steven et al 1988	d	0.13 - 1.2	0.040 - 0.371
		Cometto - Muniz & Cain 1990		4.1	1.27
		Cometto - Muniz 1993	d	4.1	1.27
		Cain & Gent 1991	d	0.32	0.1
		Laska & Hudson 1991	d	0.039	0.012
		Nagy 1991	d	1.5	0.46
		Berglund & Esfandabad 1992		0.31	0.1
		Berglund & Esfandabad 1992		2.5	0.77
		Nordin et al 1997		0.34	0.11
		Nagata 2003	d	0.2	0.062
Cain et al 2010	d	0.32	0.1		

Table 6.3 – Odor Threshold Values, cont.

Bold = Lowest Value Reported

#	Chemical Name	Source	Type of Threshold	Odor Thresholds	
				mg/m <sup>3</sup>	ppm
265	Quinoline	Geier 1936	d	0.03	<b>0.0057</b>
		Geier 1936	r	0.05 - 0.1	0.0095 - 0.019
		Gundlach & Kenway 1939	d	28	5.3
266	Quinone	Backman 1917	r	0.047 - 0.050	<b>0.0106</b> - 0.0113
		Oglesby et al 1947		0.44	0.1
267	Styrene, Monomer	Wolf et al 1956		43 - 258	10 - 61
		Deadman & Prigg 1959	d	0.11	0.026
		Li - Shen 1961		0.02	0.0047
		Stalker 1963	d	0.073	0.017
		Muhlen 1968	r	4.3	1
		Leonardos et al 1969	r	0.2 - 0.4	0.047 - 0.094
		Smith & Hochstettler 1969	r	0.2	0.047
		Dravnieks & Laffort 1972		1.7	0.4
		Hellman & Small 1973a,b, 1974	d	0.22 - 0.64	0.052 - 0.15
		Hellman & Small 1973a,b, 1974	r	0.64	0.15
		Dravnieks 1974	d	8	1.9
		Anon. 1980	d	0.14	0.033
		Anon. 1980	r	0.73	0.17
		Don 1986	d	0.068	0.016
		Hoshika et al 1993	d	0.068	0.016
		Randebrook 1986		0.012	<b>0.0028</b>
		Nagy 1991	d	1.3	0.305
		Hoshika et al 1993	d	0.14	0.033
		Nagata 2003	d	0.15	0.035
		Dalton et al 2007	d	26.4	6.2
268	Sulfur Dioxide	Holmes et al 1915	d	5 - 10	1.9 - 3.8
		Holmes et al 1915	r	10 - 13	3.8 - 5.0
		Smolczyk & Cobler 1930		<4	<1.5
		Thomas et al 1943		1.3 - 1.6	0.50 - 0.61
		Popov et al 1952		4 - 6.5	1.5 - 2.5
		Amdur et al 1953		2.6 - 21	0.99 - 8.0
		Dubrovskaya 1957		2.6 - 3.0	0.99 - 1.1
		Beck 1959		1.3 - 2.6	0.50 - 0.99
		Henschler et al 1960	d	1.3 - 2.6	0.50 - 0.99
		Bushtueva 1960		1.5	0.57
		Bushtueva 1962		1.6 - 2.6	0.61 - 0.99
		Makhinya 1966		0.87 - 0.88	<b>0.33 - 0.34</b>
		Shalamberidze 1967		1.6	0.61
		Leonardos et al 1969	r	1.2	0.46
		Nagata 2003	d	2.3	0.88
269	Sulfur Hexafluoride	Kleinbeck et al 2011		1.434 - 8.307	0.547 - 3.17
		Laffort 1968a		24,000,000	4,017,527

**Table 6.3 – Odor Threshold Values, cont.**

*Bold = Lowest Value Reported*

#	Chemical Name	Source	Type of Threshold	Odor Thresholds	
				mg/m <sup>3</sup>	ppm
270	Sulfuric Acid	Melekhina 1968	d	0.6	0.15
271	1,1,2,2 - Tetrabromoethane	Hollingsworth et al 1963	r	<14	<0.99
272	1,1,2,2 - Tetrachloroethane	Lehmann & Schmidt - Kehl 1936		20	2.9
		Dravnieks & Laffort 1972		1.6	<b>0.233</b>
		Dravnieks 1974	d	50	7.3
273	Tetrahydrofuran	May 1966	d	90	30.5
		May 1966	r	180	61
		Popov 1970		0.27	<b>0.092</b>
		Kendall et al 1968	r	7.3 - 10.2	2.5 - 3.5
		Nagy 1991	d	18	6.1
274	Thioglycolic Acid	Dravnieks et al 1986	d	0.0008	0.00021
275	Toluene	Backman 1917	r	3.5 - 3.6	0.93 - 0.96
		Backman 1918		2	0.53
		Grijns 1919		170	45
		Zwaardemaker 1927		170	45
		Zwaardemaker 1927		2	0.53
		Schley 1934	d	6	1.6
		Schley 1934	r	16	4.2
		Nader 1958	d	0.08 - 1.9	0.021 - 0.50
		Deadman & Prigg 1959	d	5.5	1.5
		Naus 1962	d	2	0.53
		Stalker 1963	d	1	0.27
		Gusev 1965		1.5 - 3.2	0.40 - 0.85
		May 1966	d	140	37
		May 1966	r	260	69
		Leonardos et al 1969	r	8.1 - 17.8	2.1 - 4.7
		Dravnieks & O'Donnell 1971		45	12
		Koster 1971	d	13.7	3.6
		Dravnieks & Laffort 1972		0.53	0.14
		Artho & Koch 1973		100 - 1,000	26.5 - 265
		Hellman & Small 1973a,b, 1974	d	0.6	0.16
		Hellman & Small 1973a,b, 1974	r	7	1.9
		Dravnieks 1974	d	60	16
		Winneke & Kastka 1975		46 - 84	12 - 22
		Anon. 1980	d	3.5	0.93
		Anon. 1980	r	18	4.8
		Naus 1982	d	2	0.53
		Naus 1982	r	20	5.3
		Punter 1983	d	25.4	6.7
		Bahmuller 1983		5.85 - 29.8	1.55 - 7.9

Table 6.3 – Odor Threshold Values, cont.

Bold = Lowest Value Reported

#	Chemical Name	Source	Type of Threshold	Odor Thresholds	
				mg/m <sup>3</sup>	ppm
275	Toluene cont.	Don 1986	d	3.7 - 3.8	0.98 - 1.01
		Hoshika et al 1993	d	3.7 - 3.8	0.98 - 1.01
		Scharfenberger 1990		17	4.51
		Nagy 1991	d	12	3.18
		Cometto - Muniz 1993	d	590	157
		Cometto - Muniz & Cain 1994	d	590	157
		Hoshika et al 1993	d	305	81
		Cometto - Muniz et al 2002		0.4	0.11
		Cometto - Muniz et al 2003		0.098	<b>0.026</b>
		Nagata 2003	d	1.3	0.35
Cometto - Muniz et al 2004	d	0.12 - 0.38	0.032 - 0.10		
Cometto - Muniz & Abraham 2009b	d	0.3	0.08		
276	Toluene 2,4 - & 2,6 - Diisocyanate	Zapp 1957		2.8	0.39
		Henschler et al 1962		0.14 - 0.35	<b>0.020</b> - 0.049
		Chizhikov 1963		0.2	0.028
		Leonardos et al 1969	r	15	2
277	o - Toluidine	Huijjer 1924	d	29	6.6
		Backman 1917	r	4.0 - 5.4	0.91 - 1.23
		Stuiver 1958	d	0.11	<b>0.025</b>
278	m - Toluidine	Huijjer 1924	d	26	5.9
		Backman 1917	r	3.0 - 3.9	0.68 - 0.089
		Stuiver 1958	d	2	<b>0.46</b>
279	p - Toluidine	Huijjer 1924	d	14	3.2
		Backman 1917	r	1.0 - 1.3	0.23 - 0.30
		Stuiver 1958	d	0.12	<b>0.027</b>
280	Trichloroacetic Acid	Backman 1917	r	1.6 - 2.5	<b>0.24</b> - 0.37
281	1,2,4 - Trichlorobenzene	Rowe 1975		22	2.96
282	Trichloroethylene	Lehmann & Schmidt - Kehl 1936		900	167
		Weitbrecht 1957		110	20
		Scherberger et al 1958	r	410	76
		Frantikova 1962		69	13
		Naus 1962	d	3	0.56
		May 1966	d	440	82
		May 1966	r	580	108
		Malyarova 1967		2.5 - 21	<b>0.5</b> - 4
		Leonardos et al 1969	r	115	21
		Torkelson & Rowe 1981		538	100
		Naus 1982	d	3	0.56
		Naus 1982	r	20	3.72
		Don 1986	d	3.9	0.73
Nagata 2003	d	21	3.91		

**Table 6.3 – Odor Threshold Values, cont.**

*Bold = Lowest Value Reported*

#	Chemical Name	Source	Type of Threshold	Odor Thresholds	
				mg/m <sup>3</sup>	ppm
283	Trichlorofluoromethane	Hellman & Small 1974	d	28	<b>5</b>
		Hellman & Small 1974	r	760	135
		Braker & Mossman 1980		1,124,000	200,057
284	Triethanolamine	Nagata 2003	d	>61	>10
285	Triethylamine	Tkachev 1970		0.33	0.08
		Hellman & Small 1974	d	<0.4	<0.10
		Hellman & Small 1974	r	1.1	0.27
		Laing et al 1978	r	11.9	2.9
		Homans et al 1978	d	2.7	0.65
		Dravnieks et al 1986	d	1	0.24
		Nagata 2003	d	0.022	<b>0.0053</b>
		Tempelaar 1913	d	2.1	0.87
286	Trimethylamine	Rotenberg & Mashbits 1967		2	0.83
		Sakuma et al 1967		0.0007	0.00029
		Leonardos et al 1969	r	0.0005	0.00021
		Stephens 1971		0.0014	0.00058
		Amoore 1977		0.0025	0.001
		Bedborough & Trott 1979	d	0.0012	0.0005
		Anon. 1980	d	0.00026	0.00011
		Anon. 1980	r	0.0034	0.0014
		Jensen & Flyger 1983		0.0031 - 0.027	0.00128 - 0.1117
		Langenhove & Schamp 1984		0.002	0.00083
		Homans 1984		4.4	1.82
		Nagy 1991		0.0059	0.0024
		Greenman et al 2004		0.000041 - 0.0011	<b>0.00002</b> - 0.00045
		Nagata 2003		0.000077	0.000032
		Van Thriel et al 2006		0.63	0.26
		287	Trimethyl Benzene, all isomers	Backman 1917	r
Backman 1917	r			0.3 - 0.35	0.061 - 0.071
Backman 1918				0.2	0.041
Deadman & Prigg 1959	d			0.7	0.14
Deadman & Prigg 1959	d			0.2	0.041
Dravnieks & Laffort 1972				0.03	<b>0.006</b>
Knuth 1973				1.2	0.24
Dravnieks 1974	d			12	2.4
Anon. 1980	d			0.14	0.028
Anon. 1980	r			1.1	0.22
Anon. 1980	d			0.18	0.037
Anon. 1980	r			1.4	0.28
Punter 1983	d			10.7	2.2
Nagata 2003	d			0.59	0.12
288	Trimethyl Phosphite	Levin & Gabriel 1973		0.0005	0.000099

Table 6.3 – Odor Threshold Values, cont.

Bold = Lowest Value Reported

#	Chemical Name	Source	Type of Threshold	Odor Thresholds	
				mg/m <sup>3</sup>	ppm
289	Turpentine and selected monoterpenes	Apell 1969		0.016	0.0029
		Cristoph 1983	r	25 - 29	4.49 - 5.20
		Cristoph 1983	r	35 - 38	6.3 - 6.8
		Randerbrock 1986		0.00036	<b>0.00006</b>
		Laska & Hudson 1991	d	0.23 - 0.36	0.041 - 0.065
		Lindell 1991	d	2.1	0.38
		Lindell 1991	d	3.3	0.59
		Lindell 1991	d	8.9	1.60
		Cometto - Muniz et al 1998b	d	105	19
		Cometto - Muniz et al 1998b	d	65	12
		Jagella & Grosch 1998		0.035	0.0063
		Jagella & Grosch 1998		0.018	0.0032
		Cometto - Muniz 1999	d	105	19
		Cometto - Muniz 1999	d	65	12
		Molhave et al 2000	d	23	4.13
		290	n - Valeraldehyde	Buettmer & Schieberle 2001a, 2001b	
Nagata 2003	d			0.18	0.032
Nagata 2003	d			0.1	0.0179
Backman 1917	r			0.009 - 0.01	0.0026 - 0.0028
Teranishi et al 1974				0.072	0.02
Anon. 1980	d			0.0025	0.0007
Anon. 1980	r			0.013	0.0037
Hall & Andersson 1983	d			0.034	0.0097
Cristoph 1983	r			0.14 - 0.15	0.04 - 0.043
Lindell 1991	d			0.092	0.026
Von Ronson & Belitz 1992a	d			0.12	0.034
Von Ronson & Belitz 1992a	r			0.22	0.062
Cometto - Muniz et al 1998a				17.5	4.97
Cometto - Muniz 1999	d			17.5	4.97
Nagata 2003	d			0.0014	<b>0.0004</b>
Laska & Ringh 2010	d			0.85	0.24
291	Vanillin	Passy 1892a,b,d	d	0.00007 - 0.005	0.000011 - 0.000803
		Tempelaar 1913	d	0.00018 - 0.0002	0.000029 - 0.000032
		Zwaardemaker 1927	d	0.00018 - 0.0002	0.000029 - 0.000032
		Backman 1917	r	0.0015 - 0.002	0.000241 - 0.000321
		Baldus 1936	d	0.000001	0.00000016
		Baldus 1936	r	0.000004	0.000000643
		Appell 1969		0.000001	<b>0.00000016</b>
		Randerbrock 1971		0.000006	0.000000964

**Table 6.3 – Odor Threshold Values, cont.**

*Bold = Lowest Value Reported*

#	Chemical Name	Source	Type of Threshold	Odor Thresholds	
				mg/m <sup>3</sup>	ppm
291	Vanillin cont.	Herrmann & Abel El Salam 1980a,b		0.08 - 0.12	0.0129 - 0.0193
		Kleinschmidt 1983	r	0.578	0.0929
		Randebrook 1986		0.000033	0.00000530
		Blank et al 1989, 1992		0.0006 - 0.0012	0.000096 - 0.000193
292	Vinyl Acetate	Gofmekler 1960		1	0.28
		Deese & Joyner 1969	r	≤1.4	≤0.40
		Hellman & Small 1973a,b, 1974	d	0.4	<b>0.12</b>
		Hellman & Small 1973a,b, 1974	r	1.4	0.4
293	Vinyl Chloride	Hori et al 1972		520 - 910	203 - 356
294	Vinylidene Chloride	Rylova 1953		200	<b>50</b>
		Janicek et al 1960		5500	1387
		Irish 1962		2,000 - 4,000	504 - 1,009
295	Xylene (o -, m -, p -, isomers)	Backman 1917	r	1.0 - 1.2	0.23 - 0.28
		Backman 1917	r	1.1 - 1.3	0.25 - 0.30
		Backman 1917	r	1.4 - 1.5	0.32 - 0.35
		Backman 1918		0.8	0.18
		Stuiver 1958	d	2.1	0.48
		Stuiver 1958	d	0.35	0.081
		Stuiver 1958	d	0.6	0.14
		Naus 1962	d	1	0.23
		Gusev 1965		0.6 - 1.9	0.14 - 0.44
		May 1966	d	100	23
		May 1966	r	1,370	316
		Koster 1965, 1968a,b, 1971	d	0.6 - 86	0.16 - 20
		Koster 1968a, 1971	d	11	2.5
		Koster 1968a, 1971	d	8	1.8
		Leonardos et al 1969	r	2	0.46
		Dravnieks & O'Donnell 1971		1.3	0.3
		Knuth 1973		0.8	0.18
		Anon. 1980	d	0.77	0.18
		Anon. 1980	r	3.1	0.71
		Anon. 1980	d	0.52	0.12
		Anon. 1980	r	2.4	0.55
		Anon. 1980	d	0.52	0.12
Anon. 1980	r	2.2	0.51		
Punter 1983	d	23.6	5.4		
Punter 1983	d	1.5 - 4.9	0.35 - 1.1		
Punter 1983	d	9.1	2.1		
Don 1986	d	0.52 - 0.54	0.120 - 0.124		

**Table 6.3 – Odor Threshold Values, cont.**

*Bold = Lowest Value Reported*

#	Chemical Name	Source	Type of Threshold	Odor Thresholds	
				mg/m <sup>3</sup>	ppm
295	Xylene (o -, m -, p -, isomers) cont.	Hoshika et al 1993	d	0.52 - 0.54	0.120 - 0.124
		Hoshika et al 1993	d	0.052	<b>0.012</b>
		Nagata 2003	d	1.6	0.37
		Nagata 2003	d	0.18	0.041
		Nagata 2003	d	0.25	0.058



## Table 6.4 – Odor Character List

The Table contains the following information:

- Odor Character
- Chemical Name

*Note: Odor character is affected by odor concentration.*

**Table 6.4 – Odor Character List**

Odor Character	Chemical(s)
Acetone	Diethyl Ketone; 2-Methylcyclohexanone
Acid	Acetic Anhydride; Cyanogen Chloride; Phenol
Acrid	Maleic Anhydride
Airplane glue	Methyl Acrylate
Alcohol	n-Butyl Alcohol; sec-Butyl Alcohol; tert-Butyl Alcohol; 1,4-Dioxane; Ethyl Acrylate, Ethyl Alcohol; Ethyl Silicate; Furfuryl Alcohol; Isobutyl Alcohol; 2-Methoxyethanol; Methyl Alcohol; Propyl Alcohol
Alliaceous	Bromine; Iodine
Almond	Acetophenone; Benzaldehyde; Chlorobenzene; Cyanogen; Furfural; Hydrogen Cyanide; Nitrobenzene
Amine	Dibutylamine; Diethanolamine; Diethylamine; 2-Diethylaminoethanol; Diisopropylamine; Isopropylamine; n-Methyl-2-Pyrrolidone; Morpholine; o-Toluidine; p-Toluidine; Triethylamine
Ammonia; Ammonical	n-Butylamine; Cyclohexylamine; Diethanolamine; Diethylaminoethanol; n,n-Dimethylacetamide; Dimethylamine; Ethanolamine; Ethyl Acrylate; Ethylamine; Ethylenediamine; Ethyleneimine; n-Ethylmorpholine; Hydrazine; Isopropylamine; 1-Methoxy-2-Propanol; Methyl Hydrazine; Triethanolamine
Anesthetic	Ethyl Ether; Methyl tert-Butyl Ether
Anise	Benzyl Acetate
Aromatic	Benzene; 1,3-Butadiene; Carbon Tetrachloride; Chlorotoluene; 1,1-Dichloroethane; Ethyl Formate; Isoprene; Propylene; o-Toluidine; 1,2,4-Trichlorobenzene; Trimethyl Benzene
Asparagus	Dimethyl Disulfide; Dimethyl Sulfide
Banana	n-Amyl Acetate; n-Butyl Acetate; sec-Hexyl Acetate; Isoamyl Acetate
Bitter	Isoamyl Alcohol
Bitter almond	Benzaldehyde
Bleach	Chlorine; Nitrogen Dioxide
Bread	Furfural
Burnt	Pyridine, Toluene
Burnt plastic	Cresol
Butter-like; buttery	Biphenyl; Diacetyl
Camphor; camphorous	Camphor; Cyclohexanol; o-Dichlorobenzene; p-Dichlorobenzene; 1,1,2,2-Tetrabromoethane; Turpentine
Caramel	Vanillin
Chemical	Iodoform

**Table 6.4 – Odor Character List, cont.**

<b>Odor Character</b>	<b>Chemical(s)</b>
<b>Chlorine</b>	Chlorine Dioxide; Chloropicrin
<b>Chloroform</b>	Bromoform; 1,1-Dichloroethane; Epichlorohydrin; Halothane; Vinylidene Chloride
<b>Choking</b>	Phthalic Anhydride
<b>Citrus</b>	Citral; d-Limonene
<b>Creosote</b>	Cresol; Naphthalene; Phenol
<b>Decayed</b>	n-Valeraldehyde
<b>Disagreeable</b>	Diallylamine; Dimethyl Sulfide; Propylene Glycol Dinitrate
<b>Dry cleaner</b>	Carbon Tetrachloride
<b>Empyreumatic</b>	Aniline; Benzene; Cresol; Naphthalene; Phenol; o-Toluidine; m-Toluidine; p-Toluidine; Xylene
<b>Ester</b>	2-Butoxyethanol, 2-Ethoxyethyl Acetate; Ethyl Acrylate; Isobutyl Acetate; 2-Methoxyethyl Acetate; n-Propyl Acetate
<b>Ether; Etherish; Etherous; Ethereal</b>	Acetone; Acetonitrile; n-Amyl Acetate; Benzyl Acetate; Carbon Tetrachloride; Chlorodifluoromethane; Chloroform; Dichlorodifluoromethane; Dimethyl Ether; Ethyl Acetate; Ethyl Bromide; Ethylene Chlorohydrin; Ethyl Ether; Furfuryl Alcohol; Iodoform; 1-Methoxy-2-propanol; Methyl Chloride; Methyl Chloroform; Methyl Formate; Perchloroethylene; Tetrahydrofuran; Trichloroethylene
<b>Faint</b>	n-Nitrosodimethylamine
<b>Fingernail polish</b>	Ethyl Acetate; Methyl Propyl Ketone
<b>Fingernail polish remover</b>	Diethyl Ketone
<b>Fir needles</b>	Turpentine
<b>Fish sauce</b>	2,4-Dichlorophenol
<b>Fishy</b>	Diethylamine; Diisopropylamine; Dimethyl Formamide; 1,1-Dimethylhydrazine; Methylamine; Morpholine; Triethylamine; Trimethylamine
<b>Floral; Flowery</b>	Citral; Diphenylamine
<b>Fresh</b>	Acetaldehyde; Isoamyl Acetate
<b>Fruity</b>	Acetaldehyde; Acetone; Benzaldehyde; 2-Butoxyethyl Acetate; sec-Butyl Acetate; 2-Chloroacetophenone; Cyclopentadiene; Ethyl Acetate; sec-Hexyl Acetate; Isopropyl Acetate; Methyl Acetate; 2-Nitropropane; Propionaldehyde
<b>Fusel</b>	Isobutyl Alcohol
<b>Garlic</b>	Acetylene; Acrylonitrile; Arsine; Dimethyl Disulfide; Hydrogen Selenide; Methyl Mercaptan; Phosphine
<b>Gasoline</b>	p-tert-Butyl Toluene; Heptane; Hexane, Nonane; Octane
<b>Gassy</b>	Acetylene; Propylene
<b>Grassy</b>	Ethylene; n-Hexyl Alcohol
<b>Green</b>	Acetaldehyde
<b>Hay like</b>	Phosgene
<b>Highly corrosive</b>	Hydrogen Fluoride
<b>Ink</b>	Phenol
<b>Irritating</b>	Allyl Isothiocyanate; Ammonia; Bromine; Cresol; Cumene Hydroperoxide; 2,3-Dibromo-1-chloropropane; Hydrogen Chloride; Hydrogen Fluoride

**Table 6.4 – Odor Character List, cont.**

<b>Odor Character</b>	<b>Chemical(s)</b>
<b>Latex</b>	Isobutyl Alcohol
<b>Leather-like</b>	2,4-Dichlorophenol
<b>Lemon</b>	Citral; d-Limonene
<b>Malty</b>	n-Butyl Alcohol; sec-Butyl Alcohol; Isoamyl Alcohol
<b>Medicinal</b>	n-Butyl Alcohol; Carbon Disulfide; 2,4-Dichlorophenol; Isobutyl Acetate; Phenol
<b>Metallic</b>	Sulfur Dioxide
<b>Mild</b>	tert-Butyl Acetate; n- Butyl Lactate; Caprolactam; 2-(2-Ethoxyethoxy)ethanol; Hexylene Glycol; Triethanolamine
<b>Mothballs</b>	p-Dichlorobenzene; Naphthalene
<b>Mushroom</b>	Methyl n-amyl Ketone
<b>Mustard</b>	allyl alcohol
<b>Musty</b>	2-Butoxyethanol; Diethylamine; 2-Ethoxyethanol; Isobutyl Alcohol
<b>Natural gas</b>	Butane; Propane
<b>Nauseating</b>	Pyridine
<b>Oil</b>	Octane
<b>Oily</b>	Aniline; Dimethylaniline; Ethyl Benzene
<b>Olefinic</b>	Ethylene Oxide
<b>Onion</b>	acrylonitrile
<b>Oranges</b>	acetophenone
<b>Paint</b>	Methyl n-Butyl Ketone
<b>Pears</b>	Benzyl Acetate; sec-Hexyl Acetate
<b>Peculiar</b>	Oxygen Difluoride; Quinoline
<b>Penetrating</b>	1-Octanol
<b>Pepper</b>	Piperdine
<b>Peppermint</b>	Diisobutyl Ketone
<b>Petroleum</b>	Butenes; 1-Hexene; Methylcyclohexane
<b>Phenol; phenolic</b>	Cresol; 2,4-Dichlorophenol
<b>Pine</b>	Cyclopentadiene; Turpentine
<b>Plastic</b>	Acrylic Acid; Benzyl Acetate; n-Butyl Acrylate; Ethyl Acrylate; n-Hexyl Alcohol; d-Limonene; Methyl Acrylate; Methyl Metacrylate
<b>Pleasant</b>	Biphenyl; 1-Decene; Diacetyl; 1,2-Dichloroethylene; 2-(2-Ethoxyethoxy)ethanol; Isooctyl Alcohol
<b>Pungent</b>	Acetaldehyde; Acetic Acid; Acetophenone; Acrolein; Allyl Chloride; Ammonia; Aniline; Benzyl Chloride; Boron Trifluoride; Butyraldehyde; Crotonaldehyde; Cyclohexane; Decaborane; Ethyl Chloride; Fluorine; Formaldehyde; Hexachlorocyclopentadiene; Isobutyraldehyde; Methacrylic Acid; Methyl Parathion; Methyl Vinyl Ketone; Nitrobenzene; Ozone; Pentaborane; Pyridine; Quinone; 1,1,2,2-Tetrabromoethane; Trimethylamine; Trimethyl Phosphite
<b>Putrid</b>	Dimethyl Disulfide; Dimethyl Sulfide; Phenyl Mercaptan
<b>Rancid</b>	Acrylic Acid; n-Butyl Acrylate; Isoamyl Alcohol; n-Valeraldehyde
<b>Repulsive</b>	Diborane

**Table 6.4 – Odor Character List, cont.**

<b>Odor Character</b>	<b>Chemical(s)</b>
<b>River water</b>	Acetophenone
<b>Rosiny</b>	Turpentine
<b>Rotten cabbage</b>	Ethyl Mercaptan; Methyl Mercaptan
<b>Rotten eggs</b>	Hydrogen Sulfide
<b>Rotten fish</b>	Diethanolamine; Dimethylamine
<b>Rubber</b>	1,3-Butadiene; Chloroprene; Isoamyl Alcohol; Isobutyl Alcohol; d-Limonene
<b>Rubbing alcohol</b>	Isopropyl Alcohol
<b>Sharp</b>	Bromine; Chlorine; Cumene; Cumene Hydroperoxide; Cyclohexanone; Dicyclopentadiene; 2-Diethylaminoethanol; Ethyl Acrylate; Ethyl Amyl Ketone; Formic Acid; Hydrogen Chloride; Iodine; Isophorone; Isopropyl Alcohol; Methyl Acrylate; Methyl Ethyl Ketone; Methyl Isoamyl Ketone, Methyl Isobutyl Ketone; Methyl Methacrylate; Styrene; Vinyl Acetate
<b>Shoe polish</b>	Chlorobenzene; Nitrobenzene
<b>Sickening</b>	n-Valeraldehyde
<b>Skunk</b>	Butyl Mercaptan; Dodecyl Mercaptan
<b>Smoky</b>	Cresol
<b>Solvent</b>	Benzene; Ethyl Amyl Ketone; Ethyl Benzene; 1,1,2,2-Tetrachloroethane; Trichloroethylene
<b>Sooty</b>	Nickel Carbonyl
<b>Sour</b>	Acetic Anhydride; n-Butylamine; Methyl Alcohol; Propionic Acid; Toluene; Vinyl Acetate
<b>Strong</b>	Furan; Oxygen Difluoride; Picolines
<b>Suffocating</b>	Acetaldehyde; Bromoform; Chlorine; Chloroform; Nitric Acid
<b>Sulfide</b>	Carbon Disulfide
<b>Sweet</b>	Acetone; Acetophenone; Acrylic Acid; Benzene; Bromoform; 2-Butoxyethanol; n-Butyl Acetate; n-Butyl Acrylate; n-Butyl Alcohol; sec-Butyl Alcohol; tert-Butyl Alcohol; Carbon Tetrachloride; Chloroform; Cyclohexanone; Cyclohexene; Diacetone Alcohol; Dicyclopentadiene; 1,4-Dioxane; 2-Ethoxyethanol; 2-Ethoxyethyl Acetate; Ethyl Acetate; Ethyl Acrylate; Ethylene Dibromide; Ethylene Dichloride; Ethylene Oxide; Ethyl Silicate; Furfuryl Alcohol; Hexylene Glycol; Isoamyl Alcohol; Isobutyl Acetate; Isobutyl Alcohol; Isopropyl Ether; Mesityl Oxide; 2-Methoxyethanol; 2-Methoxyethyl Acetate; Methyl Alcohol; Methyl n-amyyl Ketone; Methyl n-Butyl Ketone; Methyl Chloride; Methyl Chloroform; Methylene Chloride; Methyl Ethyl Ketone; Methyl Isoamyl Ketone; Methyl Isobutyl Ketone; Pentane; Perchloryl Fluoride; n-Propyl Acetate; Propyl Alcohol; Propylene Dichloride; Propylene Oxide; Styrene; Vanillin; Vinyl Chloride; Xylene
<b>Tar</b>	Naphthalene
<b>Terpene-like; terpeny</b>	Cyclopentadiene; d-Limonene
<b>Thunder storm</b>	Ozone
<b>Turpentine</b>	Ethylidene Norbornene; Turpentine
<b>Unpleasant</b>	Carbonyl Sulfide; Picolines; Thioglycolic Acid
<b>Vanilla</b>	Vanillin
<b>Vegetable</b>	Carbon Disulfide
<b>Vinegar</b>	acetic acid
<b>Vinous</b>	Ethyl Alcohol

## Table 6.5 – Synonyms of Chemical Names

The table contains the following information:

- Synonyms in alphabetical order
- Chemical Name

**Table 6.5 – Synonyms**

Synonym Name	Chemical Name
Acetanhydride	Acetic Anhydride
Acetic Acid Benzyl Ester	Benzyl Acetate
Acetic Acid Dimethylamine	Dimethylamine
Acetic Aldehyde	Acetaldehyde
Acetic Oxide	Acetic Anhydride
Acetoacetone	2,4-Pentanedione
Acetylene Tetrachloride	1,1,2,2-Tetrachloroethane
Acetyl Oxide	Acetic Anhydride
Acroleic Acid	Acrylic Acid
Acrylic Acid n-butyl Ester	n-Butyl Acrylate
Acrylic Acid Ethyl Ester	Ethyl Acetate
Adronal	Cyclohexanol
Aethyl Chloride	Ethyl Chloride
Aldehyde Crotonique	Crotonaldehyde
Allyl Mustard Oil	Allyl Isothiocyanate
1-Aminobutane	n-Butylamine
Aminocyclohexane	Cyclohexylamine
2-Aminoethanol	Ethanolamine
1-Amino-2-Methylbenzene	o-Toluidine
1-Amino-3-Methylbenzene	m-Toluidine
1-Amino-4-Methylbenzene	p-Toluidine
1-Aminonaphthalene	1-Naphthylamine
2-Aminonaphthalene	2-Naphthylamine
2-Aminopropane	Isopropylamine
2-Aminotoluene	o-Toluidine
4-Aminotoluene	p-Toluidine
3-Aminotoluene	m-Toluidine
n-Amyl Alcohol	Pentanol, all isomers
Amyl Ethyl Ketone	Ethyl Amyl Ketone
1-Benzazine	Quinoline
Benzene Chloride	Chlorobenzene

Synonym Name	Chemical Name
Benzenethiol	Phenyl Mercaptan
Benzinofonn	Carbon Tetrachloride
1,4-Benzoquinone	Quinone
Biethylene	1,3-Butadine
2-Bromo-2-Chloro-1,1,1-Trifluoroethane	Halothane
Butadien	1,3-Butadiene
Butanal	Butyraldehyde
2,3-Butanedione	Diacetyl
1-Butanethiol	Butyl Mercaptan
n-Butanol	n-Butyl Alcohol
2-Butanol	sec-Butyl Alcohol
tert-Butanol	tert-Butyl Alcohol
Butanone	Methyl Ethyl Ketone
2-Butenal	Crotonaldehyde
1-Buten-3-one	Methyl Vinyl Ketone
n-Butyl-1-Butanamine	Dibutylamine
Butyl Cellosolve	2-Butoxyethanol
Butyl Cellosolve Acetate	2-Butoxyethanol Acetate
1-Butylene	Butenes
Butyl 2-Hydroxypropanoate	Butyl Lactate
1-tert-Butyl-2-Methylbenzene	p-tert-Butyl Toluene
Butyl-2-Propenoate	n-Butyl Acrylate
Butyric Alcohol	n-Butyl Alcohol
Carbolic Acid	Phenol
Carbon Bisulfide	Carbon Disulfide
Carbonic Chloride	Phosgene
Carbon Nitride	Cyanogen
Carbonyl Chloride	Phosgene
Cellosolve	2-Ethoxyethanol
Cellosolve Acetate	2-Ethoxyethyl Acetate

**Table 6.5 – Synonyms, cont.**

Synonym Name	Chemical Name
Chlorene	Ethyl Chloride
Chlorine Cyanide	Cyanogen Chloride
Chloformyl Chloride	Phosgene
3-Chloroallyl Chloride	1,3-Dichloropropene
Chloroben	Dichlorobenzene, o-isomer
Chlorobenzol	Chlorobenzene
2-Chloro-1,3-butadiene	__-Chloroprene
4-Chlorocarbonyl Polystyrene	Benzoyl Chloride
Chlorocyanogen	Cyanogen Chloride
Chloroethane	Ethyl Chloride
2-Chloroethanol	Ethylene Chlorohydrin
Chloroethene	Vinyl Chloride
Chloroethylene	Vinyl Chloride
1-Chloro-2,3-Epoxypropane	Epichlorohydrin
Chloromethane	Methyl Chloride
1-Chloro-2-Methylbenzene	Chlorotoluene, o-isomer
3-Chloro-1-Propene	Allyl Chloride
3-Chloropropylene	Allyl Chloride
__-Chlorotoluene	Benzyl Chloride
Cinnamene	Styrene, monomer
Cresylic Acid	Cresol, all isomers
Cyano Acrylic Acid Methyl Ester	Methyl 2-Cyanoacrylate
Cyanomethane	Acetonitrile
2,5-Cyclohexadiene Dioxide	Quinone
Cyclohexyl Alcohol	Cyclohexanol
DCPD	Dicyclopentadiene
DEAE	2-Diethylaminoethanol
1,2-Diaminoethane	Ethylenediamine
1,2-Dibromoethane	Ethylene Dibromide
1,2-Dichloroethane	Ethylene Dichloride
Dichloromethane	Methylene Chloride
1,2-Dichloropropane	Propylene Dichloride
1,3-Dichloropropylene	1,3-Dichloropropene
Dicyan	Cyanogen
N-Diethylethanolamine	2-Diethylaminoethanol
Diethyl- 2-hydroxyethyl amine	2-Diethylaminoethanol
Diethylene Oxide	Tetrahydrofuran

Synonym Name	Chemical Name
1,4-Diethylene Dioxide	Dioxane
Diethyl Ether	Ethyl Ether
Diethylolamine	Diethanolamine
Difluorochloromethane	Chlorodifluoromethane
2,2-Dihydroxy Diethylamine	Diethanolamine
2,4-Diisocyanato-1-Methylbenzene	Toluene-2,4-Diisocyanate
Dimethyl Acetate	N,N-Dimethylaniline
Dimethyl Benzene	Xylene
1,3-Dimethylbutyl Acetate	sec-Hexyl Acetate
Dimethyl Diketone	Diacetyl
Dimethyl Glyoxal	Diacetyl
Dimethylformaldehyde	Acetone
Dimethylnitromethane	2-Nitropropane
Dimethyl Nitrosamine	N-Nitrosodimethylamine
o,o-Dimethyl, o,p-Nitrophenyl Phosphorothioate	Methyl Parathion
3,7-Dimethyl-2,6-Octadienal	Citral
Dimethyl Oxide	Dimethyl Ether
Diphenyl	Biphenyl
Diphenylmethane 4,4-Diisocyanate	Methylene Bisphenyl Isocyanate
Di-2-Propenylamine	Diallylamine
Divinyl	1,3-Butadiene
DMA	Dimethylamine
DMN	N-Nitrosodimethylamine
DMNA	N-Nitrosodimethylamine
DMF	Dimethyl Formamide
1-Dodecanethiol	Dodecyl Mercaptan
EAK	Ethyl Amyl Ketone
EGBE	2-Butoxyethanol
EGBEA	2-Butoxyethyl Acetate
EgMEA	2-Methoxyethyl Acetate
1,4-Epoxybutane	Tetrahydrofuran
1,2-Epoxy-3-Chloropropane	Epichlorohydrin
2,3-Epoxypropyl Chloride	Epichlorohydrin
1,2-Epoxyethane	Ethylene Oxide
Erythrene	1,3-Butadiene

**Table 6.5 – Synonyms, cont.**

Synonym Name	Chemical Name
Ethanal	Acetaldehyde
1,2-Ethanediamine	Ethylenediamine
1,2-Ethanediol	Ethylene Glycol
Ethanenitrile	Acetonitrile
Ethanethiol	Ethyl Mercaptan
Ethanoic Acid	Acetic Acid
Ethanol	Ethyl Alcohol
Ethene	Ethylene
Ethenyl Acetate	Vinyl Acetate
Ethenyl Benzene	Styrene
Ethenyl Cyanide	Acrylonitrile
Ethoxyethane	Ethyl Ether
Ethyl Acetone	Methyl Propyl Ketone
Ethylene Bromide	Ethylene Dibromide
Ethylene Carboxylic Acid	Acrylic Acid
Ethylene Chloride	Ethylene Dichloride
Ethylene Glycol Methylene Ether	1,3-Dioxolane
Ethylene Glycol Monobutyl Ether	2-Butoxyethanol
Ethylene Glycol Monoethyl Ether	2-Ethoxyethanol
Ethylene Glycol Monoethyl Ether Acetate	2-Ethoxyethyl Acetate
Ethylene Glycol Methyl Ether	2-Methoxyethanol
Ethylene Glycol Monomethyl Acetate	2-Methoxyethyl Acetate
Ethylethylene	Butenes, all isomers
Ethyl Fonic Acid	Propionic Acid
Ethyl Glycol	2-Ethoxyethanol
Ethyl Glycol Acetate	2-Ethoxyethyl Acetate
5-Ethylidenebicyclo[2.2.1]-2-Heptene	Ethylidenenorbornene
Ethylidene Chloride	1,1-Dichloroethane
Ethylmethyl Carbinol	sec-Butyl Alcohol
Ethyl Nitrile	Acetonitrile
Ethylolamine Glycol	Ethanolamine
Ethyl 2-Propenoate	Ethyl Acrylate
Ethylene	Acetylene

Synonym Name	Chemical Name
Formic Acid Ethyl Ester	Ethyl Formate
Formic Nitrate	Hydrogen Cyanide
Freon 12	Dichlorodifluoromethane
Freon 22	Chlorodifluoromethane
Freon 30	Methylene Chloride
Freon 40	Methyl Chloride
2,5-Furandione	Maleic Anhydride
2-Furanmethanol	Furfuryl Alcohol
Glacial Acrylic Acid	Acrylic Acid
Glycinol	Ethanolamine
2-Heptanone	Methyl n-Amyl Ketone
Hexahydroaniline	Cyclohexylamine
Hexahydroazine	Piperidine
Hexahydrobenzene	Cyclohexane
Hexahydrophenol	Cyclohexanol
Hexahydrotoluene	2-Methylcyclohexane
Hexalin	Cyclohexanol
Hexamethylene	Cyclohexane
Hexanaphthalene	Cyclohexane
Hexanaphthylene	Cyclohexene
1-Hexanol	Hexyl Alcohol
2-Hexanone	Methyl n-Butyl Ketone
Hexone	Methyl Isobutyl Ketone
Hydroxy Benzene	Phenol
Hydrocyclohexane	Cyclohexanol
1-Hydroxy-2,4-Dichlorobenzene	2,4-Dichlorophenol
2-Hydroxyethyl Chloride	Ethylene Dichloride
4-Hydroxy-4-Methyl-2-Pentanone	Diacetone Alcohol
2,2-Iminodiethanol	Diethanolamine
1,3-Isobenzofurandione	Phthalic Anhydride
Isobutanol	Isobutyl Alcohol
Isobutyl Carbinol	Isoamyl Alcohol
Isocyanic Acid-Methyl Ester	Methyl Isocyanate
Isonitropropane	2-Nitropropane
Isopentyl Acetate	Isoamyl Acetate
Isopentyl Alcohol	Isoamyl Alcohol

**Table 6.5 – Synonyms, cont.**

Synonym Name	Chemical Name
Isopropanol	Isopropyl Alcohol
Isopropenylbenzene	Methyl Styrene
Isopropenyl Cyanide	Methacrylonitrile
4-Isopropenyl-1-Methyl-1-Cyclohexene	d-Limonene
2-Isopropoxypropane	Isopropyl Ether
Isopropyl Benzene	Cumene
Isopropyl Benzene Hydroperoxide	Cumene Hydroperoxide
Isopropylidene Acetone	Mesityl Oxide
Ketohexamethylene	Cyclohexanone
MBK	Methyl n-Butyl Ketone
MEK	Methyl Ethyl Ketone
Mercaptoacetic Acid	Thioglycolic Acid
Mesitylene	Trimethyl Benzene
Methanal	Formaldehyde
Methanamine	Methylamine
Methanethiol	Methyl Mercaptan
Methanol	Methyl Alcohol
Methanone	Acetaldehyde
Methoxymethane	Dimethyl Ether
Methylacetic Acid	Propionic Acid
Methyl Acetone	Methyl Ether Ketone
_ -Methylacrolein	Crotonaldehyde
2-Methylaniline	o-Toluidine
3-Methylaniline	m-Toluidine
4-Methylaniline	p-Toluidine
Methyl-2-Butanone	Methyl Isopropyl Ketone
2-Methylbenzenamine	o-Toluidine
3-Methylbenzenamine	m-Toluidine
4-Methylbenzenamine	p-Toluidine
Methylbenzene	Toluene
3-Methyl-1-butanol	Isoamyl Alcohol
3-Methyl-1-butanol Acetate	Isoamyl Acetate
3-Methyl-2-butanone	Methyl Isopropyl Ketone
Methyl Cellosolve	2-Methoxyethanol
Methyl Cyanide	Acetonitrile
Methyl Dithiomethane	Dimethyl Disulfide

Synonym Name	Chemical Name
Methyl Glycol	2-Methoxyethanol
2-Methyl-1-Heptanol	Isooctyl Alcohol
5-Methyl-3-Heptanone	Ethyl Amyl Ketone
5-Methyl-3-Hexanone	Methyl Isoamyl Ketone
Methyl Isoamyl Acetate	sec-Hexyl Acetate
N-Methylmethanamine	Dimethylamine
Methyl 2-Methyl-2-Propenoate	Methyl Methacrylate
N-Methyl-N-Nitrosomethanamine	N-Nitrosodimethylamine
Methyloxidrane	Propylene Oxide
2-Methyl-2,4-Pentanediol	Hexylene Glycol
4-Methyl-2-Pentanone	Methyl Isobutyl Ketone
4-methyl-3-Pentene-2-One	Mesityl Oxide
4-Methylpentyl-2-Acetate	sec-Hexyl Acetate
Methylphenol	Cresol, all isomers
4-Methyl-1,3-Phenylenediisocyanate	Toluene-2,4- or 2,6-Diisocyanate
Methyl 2-Propenoate	Methyl Acrylate
2-Methylpropenoic Acid	Methacrylic Acid
1-Methyl Propanol	sec-Butyl Alcohol
2-Methyl-1-Propanol	Isobutyl Alcohol
2-Methyl-2-Propanol	tert-Butyl Alcohol
2-Methyl-2-Propenitrile	Methacrylonitrile
2-Methylpropyl Acetate	Isobutyl Acetate
2-Methylpyridine	Picolines
MIBK	Methyl Isobutyl Ketone
MIC	Methyl Isocyanate
MIPK	Methyl Isopropyl Ketone
Monochlorobenzene	Chlorobenzene
Monoethyl Ether of Ethylene Glycol	2-Ethoxyethanol
Necatorina	Carbon Tetrachloride
Nitro	Methyl Parathion
Nitrox	Methyl Parathion
1-NP	1-Nitropropane
2-NP	2-Nitropropane
3-Octanone	Ethyl Amyl Ketone
Orthodichloro Benzene	Dichlorobenzene, o-isomer



**Table 6.5 – Synonyms, cont.**

Synonym Name	Chemical Name
Oxacyclopentadiene	Furan
Oxirane	Ethylene Oxide
2-Oxobutane	Methyl Ethyl Ketone
Oxocyclohexane	Cyclohexanone
Oxybenzene	Phenol
Oxybismethane	Dimethyl Ether
PAN	Phthalic Anhydride
Parazene	Dichlorobenzene, p-isomer
Paracide	Dichlorobenzene, p-isomer
Parton M	Methyl Parathion
Pentanal	n-Valeraldehyde
1,5-Pentanedial	Glutaraldehyde
1-Pentanol	Amyl Alcohol
2-Pentanone	Methyl Propyl Ketone
3-Pentanone	Diethyl Ketone
Pentyl Acetate	n-Amyl Acetate
Perchlorocyclopentadiene	Hexachloro-1.3-Cyclopentadiene
Perchloromethane	Carbon Tetrachloride
PGDN	Propylene Glycol Dinitrate
Phenacyl Chloride	2-Chloroacetophenone
Phenyl Benzene	Biphenyl
N-Phenylbenzene Amine	Diphenylamine
Phenyl Chloride	Chlorobenzene
Phenyl Ethane	Ethyl Benzene
Phenyl Ethylene	Styrene, Monomer
Phenyl Hydroxide	Phenol
Phenylic Acid	Phenol
Phenylmethane	Toluene
Phenylmethyl Acetate	Benzyl Acetate
2-Phenyl-1-Propane	Methyl Styrene
Phosphorothiol	Methyl Parathion
Propanal	Propionaldehyde
2-Propanamine	Isopropylamine
1,2-Propanediol	Propylene Glycol
1,2-Propanediol Dinitrate	Propylene Glycol Dinitrate
Propane Nitrile	Acrylonitrile
Propane Oxide	Propylene Oxide

Synonym Name	Chemical Name
Propanoic Acid	Propionic Acid
1-Propanol	Propyl Alcohol
2-Propanol	Isopropyl Alcohol
2-Propanone	Acetone
2-Propenal	Acrolein
Propene	Propylene
Propene Acid	Acrylic Acid
Propenenitrile	Acrylonitrile
Propene Oxide	Propylene Oxide
Propenoic Acid	Acrylic Acid
2-Propenoic Acid	Acrylic Acid
2-Propenoic Acid Butyl Ester	n-Butyl Acrylate
2-Propen-1-ol	Allyl Alcohol
1-Propenol-3	Allyl Alcohol
2-Propenyl Isothiocyanate	Allyl Isothiocyanate
Propional	Propionaldehyde
2-Propylamine	Isopropylamine
Propylene Chloride	N,N-Dimethylacetamide
Santochlor	Dichlorobenzene, p-isomer
Silicic Acid Tetraethyl Ester	Ethyl Silicate
TCE	Trichloroethylene
Termitkiln	Dichlorobenzene, o-isomer
sym-Tetrachlorethane	1,1,2,2-Tetrachlorethane
Tetrachloroethene	Perchloroethylene
Tetrachloroethylene	Perchloroethylene
Tetrachloromethane	Carbon Tetrachloride
3a,4,7,7a-Tetrahydro-4,7-Methanoindene	Dicyclopentadiene
Tetramethylene Oxide	Tetrahydrofuran
THF	Tetrahydrofuran
Thiophenol	Phenyl Mercaptan
TMA	Trimethylamine
Toluol	Toluene
Tribromoethane	Bromoform
1,1,1-Trichloroethane	Methyl Chloroform
Triiodomethane	Iodoform
1,7-Trimethylbicyclo[2.2.1]-2-Heptanone	Camphor

**Table 6.5 – Synonyms, cont.**

Synonym Name	Chemical Name
3,5,5-Trimethyl-2-Cyclohexenone	Isophorone
Trioxychlorofluoride	Perchloryl Fluoride
Vinyl Benzene	Styrene, monomer
Vinyl Carbinol	Allyl Alcohol
Vinyl Cyanide	Acrylonitrile
Vinyl Formic Acid	Acrylic Acid
Wofatos	Methyl Parathion

**Table 6.6 – Chemical Abstract Numbers & Chemical Names**

The table contains the following information:

- Chemical Abstract Number (CAS) in numerical order
- Chemical Name

**Table 6.6 – Chemical Abstract Numbers**

CAS #	Chemical Name	CAS #	Chemical Name
50-00-0	Formaldehyde	74-90-8	Hydrogen Cyanide
54-11-5	Nicotine	74-93-1	Methyl Mercaptan
56-23-5	Carbon Tetrachloride	74-96-4	Ethyl Bromide
57-06-7	Allyl Isothiocyanate	74-98-6	Propane
57-14-7	1,1-Dimethylhydrazine	75-00-3	Ethyl Chloride
57-55-6	Propylene Glycol	75-01-4	Vinyl Chloride
60-24-2	Mercaptoethanol	75-04-7	Ethylamine
60-29-7	Ethyl Ether	75-05-8	Acetonitrile
60-34-4	Methyl Hydrazine	75-07-0	Acetaldehyde
62-53-3	Aniline	75-08-1	Ethyl Mercaptan
62-75-9	N-Nitrosodimethylamine	75-09-2	Methylene Chloride
64-17-5	Ethyl Alcohol	75-15-0	Carbon Disulfide
64-18-6	Formic Acid	75-18-3	Dimethyl Sulfide
64-19-7	Acetic Acid	75-21-8	Ethylene Oxide
67-56-1	Methyl Alcohol	75-25-2	Bromoform
67-63-0	Isopropyl Alcohol	75-28-5	Butane, all isomers
67-64-1	Acetone	75-31-0	Isopropylamine
67-66-3	Chloroform	75-34-3	1,1-Dichloroethane
68-11-1	Thioglycolic Acid	75-35-4	Vinylidene chloride
68-12-2	Dimethyl Formamide	75-44-5	Phosgene
71-23-8	Propyl Alcohol	75-45-6	Chlorodifluoromethane
71-36-3	n-Butyl Alcohol	75-47-8	Iodoform
71-41-0	Pentanol, all isomers	75-50-3	Trimethylamine
71-43-2	Benzene	75-52-5	Nitromethane
71-55-6	Methyl Chloroform	75-56-9	Propylene Oxide
74-82-8	Methane	75-65-0	tert-Butyl Alcohol
74-84-0	Ethane	75-69-4	Trichlorofluoromethane
74-85-1	Ethylene	75-71-8	Dichlorodifluoromethane
74-86-2	Acetylene	75-83-2	Hexane, all isomers except n-hexane
74-87-3	Methyl Chloride	75-85-4	Pentanol, all isomers
74-89-5	Methylamine	76-03-9	Trichloroacetic acid

**Table 6.6 – Chemical Abstract Numbers, cont.**

CAS #	Chemical Name	CAS #	Chemical Name
76-06-2	Chloropicrin	93-82-8	Cumene
76-22-2	Camphor, synthetic	95-13-6	Indene
77-47-4	Hexachlorocyclopentadiene	95-47-6	Xylene (o-, m-, p- isomers)
77-73-6	Dicyclopentadiene	95-48-7	Cresol, all isomers
78-10-4	Ethyl Silicate	95-49-8	Chlorotoluene, o-isomer
78-59-1	Isophorone	95-50-1	Dichlorobenzene, o- isomer
78-78-4	Pentane, all isomers	95-53-4	o-Toluidine
78-79-5	Isoprene	95-63-6	Trimethyl Benzene, all isomers
78-83-1	Isobutyl Alcohol	96-12-8	2,3-Dibromo-1-Chloropropane
78-84-2	Isobutyraldehyde	96-14-0	Hexane, all isomers except n-hexane
78-87-5	Propylene Dichloride	96-22-0	Diethyl Ketone
78-92-2	sec-Butyl Alcohol	96-33-3	Methyl Acrylate
78-93-3	Methyl Ethyl Ketone	96-37-7	Hexane, all isomers except n-hexane
78-94-4	Methyl Vinyl Ketone	98-00-0	Furfuryl Alcohol
79-01-6	Trichloroethylene	98-01-1	Furfural
79-09-4	Propionic Acid	98-51-1	p-tert-Butyl toluene
79-10-7	Acrylic Acid	98-82-8	Cumene
79-11-8	Monochloroacetic Acid	98-83-9	Methyl Styrene
79-20-9	Methyl Acetate	98-86-2	Acetophenone
79-27-6	1,1,2,2-Tetrabromoethane	98-88-4	Benzoyl Chloride
79-29-8	Hexane, all isomers except n-hexane	98-95-3	Nitrobenzene
79-34-5	1,1,2,2-Tetrachloroethane	100-37-8	2-Diethylaminoethanol
79-41-4	Methacrylic acid	100-41-4	Ethyl Benzene
79-43-6	Dichloroacetic Acid	100-42-5	Styrene, monomer
79-46-9	2-Nitropropane	100-44-7	Benzyl Chloride
80-15-9	Cumene Hydroperoxide	100-52-7	Benzaldehyde
80-56-8	Turpentine & selected monoterpenes	100-61-8	N-Methyl Aniline
80-62-6	Methyl Methacrylate	100-74-3	N-Ethylmorpholine
84-66-2	Diethyl Phthalate	101-68-8	Methylene Bisphenyl Isocyanate
84-74-2	Dibutyl Phthalate	102-71-6	Triethanolamine
85-44-9	Phthalic Anhydride	105-05-5	Diethylbenzenes, mixed isomers
91-08-7	Toluene Diisocyanate	105-46-4	sec-Butyl Acetate
91-20-3	Naphthalene	105-60-2	Caprolactam
91-22-5	Quinoline	106-42-3	Xylene (o-, m-, p- isomers)
91-57-6	2-Methylnaphthalene	106-44-5	Cresol, all isomers
91-59-8	2-Naphthylamine	106-46-7	Dichlorobenzene, p-isomer
92-52-4	Biphenyl	106-49-0	p-Toluidine

**Table 6.6 – Chemical Abstract Numbers, cont.**

CAS #	Chemical Name
106-51-4	Quinone
106-89-8	Epichlorohydrin
106-93-4	Ethylene Dibromide
106-97-8	Butane, all isomers
106-98-9	Butenes, all isomers
106-99-0	1,3-Butadiene
107-01-7	Butenes, all isomers
107-02-8	Acrolein
107-05-1	Allyl Chloride
107-06-2	Ethylene Dichloride
107-07-3	Ethylene Chlorohydrin
107-13-1	Acrylonitrile
107-15-3	Ethylenediamine
107-18-6	Allyl Alcohol
107-21-1	Ethylene Glycol
107-31-3	Methyl Formate
107-41-5	Hexylene glycol
107-83-5	Hexane, all isomers except n-hexane
107-87-9	Methyl Propyl Ketone
107-98-2	1-Methoxy-2-Propanol
108-03-2	1-Nitropropane
108-05-4	Vinyl Acetate
108-08-7	Heptane, all isomers
108-10-1	Methyl Isobutyl Ketone
108-11-2	4-Methyl-2-Pentanol
108-18-9	Diisopropylamine
108-20-3	Isopropyl Ether
108-21-4	Isopropyl Acetate
108-24-7	Acetic Anhydride
108-31-6	Maleic Anhydride
108-38-3	Xylene (o-, m-, p- isomers)
108-39-4	Cresol, all isomers
108-44-1	m-Toluidine
108-65-6	1-Methoxy-2-Propyl Acetate
108-67-8	Trimethyl Benzene, all isomers
108-83-8	Diisobutyl Ketone
108-84-9	sec-Hexyl Acetate

CAS #	Chemical Name
108-87-2	Methylcyclohexane
108-88-3	Toluene
108-89-4	Picolines
108-90-7	Chlorobenzene
108-91-8	Cyclohexylamine
108-93-0	Cyclohexanol
108-94-1	Cyclohexanone
108-95-2	Phenol
108-98-5	Phenyl Mercaptan
108-99-6	Picolines
109-06-8	Picolines
109-60-4	n-Propyl Acetate
109-66-0	Pentane, all isomers
109-73-9	n-Butylamine
109-79-5	Butyl Mercaptan
109-86-4	2-Methoxyethanol
109-89-7	Diethylamine
109-94-4	Ethyl Formate
109-99-9	Tetrahydrofuran
110-00-9	Furan
110-12-3	Methyl Isoamyl Ketone
110-19-0	Isobutyl Acetate
110-43-0	Methyl n-amyl Ketone
110-49-6	2-Methoxyethyl Acetate
110-54-3	n-Hexane
110-62-3	n-Valeraldehyde
110-80-5	2-Ethoxyethanol
110-82-7	Cyclohexane
110-83-8	Cyclohexene
110-86-1	Pyridine
110-89-4	Piperdine
110-91-8	Morpholine
111-15-9	2-Ethoxyethyl Acetate
111-27-3	n-Hexyl Alcohol
111-30-8	Glutaraldehyde
111-42-2	Diethanolamine
111-65-9	Octane, all isomers

**Table 6.6 – Chemical Abstract Numbers, cont.**

CAS #	Chemical Name
111-66-0	1-Octene
111-76-2	2-Butoxyethanol
111-84-2	Nonane
111-87-5	1-Octanol
111-90-0	2-(2-Ethoxyethoxy)ethanol
111-92-2	Dibutylamine
112-07-2	2-Butoxyethyl Acetate
112-55-0	Dodecyl Mercaptan
115-07-1	Propylene
115-10-6	Dimethyl Ether
115-11-7	Butenes, all isomers
120-82-1	1,2,4-Trichlorobenzene
120-83-2	2,4-Dichlorophenol
121-33-5	Vanillin
121-44-8	Triethylamine
121-45-9	Trimethyl Phosphite
121-69-7	Dimethylaniline
122-39-4	Diphenylamine
123-38-6	Propionaldehyde
123-42-2	Diacetone Alcohol
123-51-3	Isoamyl Alcohol
123-54-6	2,4-Pentanedione
123-72-8	Butyraldehyde
123-73-9	Crotonaldehyde
123-86-4	n-Butyl Acetate
123-91-1	1,4-Dioxane
123-92-2	Isoamyl Acetate
124-02-7	Diallylamine
124-09-4	1,6-Hexanediamine
124-38-9	Carbon Dioxide
124-40-3	Dimethylamine
126-98-7	Methacrylonitrile
126-99-8	b-Chloroprene
127-18-4	Perchloroethylene
127-19-5	N,N-Dimethylacetamide
127-91-3	Turpentine & selected monoterpenes
134-32-7	1-Naphthylamine & selected monoterpenes

CAS #	Chemical Name
135-01-3	Diethylbenzenes, mixed isomers
137-05-3	Methyl 2-Cyanoacrylate
137-32-6	Pentanol, all isomers
138-22-7	n-Butyl lactate
138-86-3	d-Limonene
140-11-4	Benzyl Acetate
140-88-5	Ethyl Acrylate
141-32-2	n-Butyl Acrylate
141-43-5	Ethanolamine
141-78-6	Ethyl Acetate
141-79-7	Mesityl Oxide
141-93-5	Diethylbenzenes, mixed isomers
142-82-5	Heptane, all isomers
151-56-4	Ethyleneimine
151-67-7	Halothane
156-59-2	1,2-Dichloroethylene, all isomers
156-60-5	1,2-Dichloroethylene, all isomers
298-00-0	Methyl Parathion
302-01-2	Hydrazine
431-03-8	Diacetyl
460-19-5	Cyanogen
463-58-1	Carbonyl Sulfide
463-82-1	Pentane, all isomers
506-77-4	Cyanogen Chloride
526-73-8	Trimethyl Benzene, all isomers
532-27-4	2-Chloroacetophenone
534-52-1	4,6-Dinitro-o-Cresol
540-59-0	1,2-Dichloroethylene, all isomers
540-84-1	Octane, all isomers
540-88-5	tert-Butyl Acetate
541-85-5	Ethyl Amyl Ketone
542-75-6	1,3-Dichloropropene
542-92-7	Cyclopentadiene
563-80-4	Methyl Isopropyl Ketone
565-59-3	Heptane, all isomers
583-60-8	2-Methylcyclohexanone
584-84-9	Toluene 2,4- or 2,6-Diisocyanate

**Table 6.6 – Chemical Abstract Numbers, cont.**

CAS #	Chemical Name
589-34-4	Heptane, all isomers
590-18-1	Butenes, all isomers
590-35-2	Heptane, all isomers
591-76-4	Heptane, all isomers
591-78-6	Methyl n-Butyl Ketone
592-41-6	1-Hexene
624-41-9	2-Methyl Butyl Acetate
624-64-6	Butenes, all isomers
624-83-9	Methyl Isocyanate
624-92-0	Dimethyl Disulfide
628-63-7	n-Amyl Acetate
646-06-0	1,3-Dioxolane
822-06-0	1,6-Diisocyanatohexane
872-05-9	1-Decene
872-50-4	n-Methyl-2-Pyrrolidone
1319-77-3	Cresol, all isomers
1330-20-7	Xylene (o-,m-, p- isomers)
1634-04-4	Methyl tert-Butyl Ether
2551-62-4	Sulfur Hexafluoride
4170-30-3	Crotonaldehyde
5392-40-5	Citral
6032-29-7	Pentanol, all isomers
6423-43-4	Propylene Glycol Dinitrate
7446-09-5	Sulfur Dioxide
7553-56-2	Iodine
7616-94-6	Perchloryl Fluoride
7637-07-2	Boron Trifluoride
7647-01-0	Hydrogen Chloride
7664-39-3	Hydrogen Fluoride
7664-41-7	Ammonia
7664-93-9	Sulfuric Acid
7697-37-2	Nitric Acid
7726-95-6	Bromine
7782-41-4	Fluorine
7782-50-5	Chlorine
7783-06-4	Hydrogen Sulfide
7783-07-5	Hydrogen Selenide

CAS #	Chemical Name
7783-41-7	Oxygen Difluoride
7784-42-1	Arsine
7785-26-4	Turpentine & selected monoterpenes
7785-70-8	Turpentine & selected monoterpenes
7803-51-2	Phosphine
8006-64-2	Turpentine & selected monoterpenes
8014-95-7	Sulfuric Acid
10028-15-6	Ozone
10049-04-4	Chlorine Dioxide
10102-44-0	Nitrogen Dioxide
13463-39-3	Nickel Carbonyl
13466-78-9	Turpentine & selected monoterpenes
16219-75-3	Ethylidene Norbornene
17702-41-9	Decaborane
19287-45-7	Diborane
19624-22-7	Pentaborane
25167-67-3	Butenes, all isomers
25340-17-4	Diethylbenzenes, mixed isomers
25551-13-7	Trimethyl Benzene, all isomers
26952-21-6	Isooctyl Alcohol
60435-70-3	Isooctyl Alcohol
86290-81-5	Octane, all isomers

## References

1. **Adams, D.F., F.A. Young, and R.A. Luhr:** Evaluation of an Odor Perception Threshold Test Facility. *Tappi*. 51(3):62A–67A (1968).
2. **Ahlström, R., B. Berglund, U. Berglund, T. Lindvall and A. Wernberg:** Impaired Odor Perception Tank Cleaners. *Scand. J. Work Environ. Health* 12:574–81 (1986).
3. **Ahlström, R., B. Berglund, U. Berglund and T. Lindvall:** Formaldehyde Odor and its Interaction with the Air of a Sick Building. *Environ. Inter.* 12:289–95 (1986).
4. **Akhmedov, B.K.:** Methapos as an Air Pollutant. *Hyg. Sanit. USSR* 33:11–16 [English translation of Russian article.] (1968).
5. **Alibaev, T.S.:** Hygienic Standards for Cyclohexane and Its Mixture with Benzene in Air. *Hyg. Sanit. USSR* 35:22–28 [English translation of Russian article] (1970).
6. **Allison, V.C. and S.H. Katz:** An Investigation of Stenches and Odors for Industrial Purposes. *J. Ind. Eng. Chem.* 11:336–38 (1919).
7. **Amdur, M.O., W.W. Melvin, and P. Drinker:** Effects of Inhalation of Sulphur Dioxide by Man. *Lancet* 265:758–59 (1953).
8. **Amoore, J.E.:** Specific Anosmia and the Concept of Primary Odors. *Chem. Senses Flavor* 2:267–81 (1977).
9. **Amoore, J.E. and R.G. Buttery:** Partition Coefficients and Comparative Olfactometry. *Chem. Senses Flavor* 3:57–71 (1978).
10. **Amoore, J.E.:** The Chemistry and Physiology of Odor Sensitivity. *J. Am. Water Works Assoc.* 78:70–76 (1986).
11. **Amoore, J.E.:** Effects of Chemical Exposure on Olfaction in Humans. In *Toxicology of the Nasal Passages* edited by C.S Barrow, Washington: Hemisphere Publishing Corporation, 1986. pp. 155–190..
12. **Andreescheva, N.G.:** Substantiation of the Maximum Permissible Concentration of Nitrobenzene in the Atmospheric Air. *Hyg. Sanit. USSR* 29:5–10 [English translation of Russian article] (1964).
13. **Andreescheva, N.G.:** The Effects of Certain Aromatic Carbohydrates in the Air. *Hyg. Sanit. USSR* 33:12–16 [English translation of Russian article] (1968).
14. **Anon.:** Industrial Hygiene Department, E.K. Co., June 6. In *Documentation of the Threshold Limit Values*. 3rd edition. 1971, Cincinnati, OH: ACGIH®, 1969. p. 411.
15. **Anon.:** *Reports of Studies on the Measurement of the Offensive Odors (from 1972–1980) in Japan*. Tokyo: Japan Environment Agency, 1980.



16. **Anrooij, A. Van:** "Desodorisatie Langs Photochemischen Weg." Thesis, Utrecht, 1931.
17. **Appell, L.:** Physical Foundations in Perfumery VIII. The Minimum Perceptible. *Am. Perfum. Cosmet.* 84:45–50 (1969).
18. **Armit, H.W.:** The Toxicology of Nickel Carbonyl, *J. Hyg.* 7:525–51 (1907).
19. **Artho, A. and R. Koch:** *Caractérisation Olfactive des Composés de la Fumée de Cigarettes. A. du Tabac. Sect. 1-11 Paris-Seita* 37–43 (1973).
20. **Atanasova, B.T. Thomas-Danguin, D. Langlois, S. Nicklaus, C. Chabanet and P. Etievant:** Perception of Wine Fruity and Woody Notes: Influence of Perithreshold Odorants. *Food Qual. Prefer.* 16:504–10 (2005).
21. **Babin, I.N., I.I. Strizhevskii, and E.I. Kordysh:** Use of the Odorimeter VNIIT for Comparative Tests of the Intensity of the Odor of Carbide and Electrolysis Acetylene. *Khim. Tekhnol. Topl. Prod. Ego Pererab.* 152–55 (1965). [In Chem. Abstr. 66:31815m (1967)].
22. **Backman, E.L.:** Experimentella Undersökningar Öfver Luktsinnets Fysiologi. Upsala Laekarefoeren. *Foerh.* 22:319–470 (1917).
23. **Backman, E.L.:** The Olfactologie of the Methylbenzol Series. *Onderzoekingen Gedaan in Het Fysiologisch Laboratorium der Utrechise Hogeschool S5* 18:349–64 (1918).
24. **Bahn Müller, H.:** Comparison of Individual Odor Thresholds of Various Compounds – Results of a Ring Test Performed by BASF AG, Hoechst AG, Bayer Ag, and E. Merck, In: *Proc. 6th World Congress on Air Quality, Paris, May 1983*, 2:417–22, (1983).
25. **Bahn Müller, H.:** Olfatometric von Dibutylamin, Acrylsäuremethylester, Isoamyalkohol und Eines Spritzverdünners für Autolacke – Ergebnisse Eines VDI-Ringvergleichs, *Staub* 44:352–58 (1984).
26. **Baikov, B.K.:** Experimental Data for Substantiating the Maximum Permissible Concentration of Carbon Disulfide in Combination with Hydrogen Disulfide in the Atmospheric Air. *Gig. Sanit.* 28:3–8 [English summary of Russian article] (1963).
27. **Baikov, B.K., M.K. Khachatryan, E.V. Bordodina, N.G. Feldman, and A.M. Tambovtseva:** Data for Hygienic Standardization of Amyl Alcohol in the Atmosphere. *Gig. Sanit.* 38:10–14 [English summary of Russian article] (1973.).
28. **Baikov, B.K. and M.Kh. Khachatryan:** Hygienic Assessment of the Reflex Action on a Body of Small Concentrations of Butyl Alcohol in the Atmosphere. [In Russian], *Gig. Sanit.* 38(12):7–11 (1973).

29. **Balavoine, P.:** L'olfaction au Service de L'appréciation des Denrées Alimentaires et des Boissons. *Mitt. Geb. Lebensmittelunters. Hyg.* 34:68–74 (1943).
30. **Balavoine, P.:** Observatiojns sur les Qualités Olfactives et Gustatives des Ali-ments, *Mitt. Geb. Lebensmittelunters. Hyg.* 39:342–50 (1948).
31. **Baldus, C.:** "Untersuchung Uber Geruchsswellen." Thesis, Wurzburg, 1936.
32. **Basmadshijewa, K., M. Argirowa, and E. Dawidkswa:** Hygienische Begründung des MIK-Wertes von Methylendichlorid in der Atmosphärischen Luft. *Wiss. Z. Humboldt-Univ. Berlin, Math. Naturwiss. Reihe 19:*469–70. [English summary of German article] (1970).
33. **Basmadzhieva, K. and M. Argirova:** Maximum Permissible Concentration of Hydrogen Sulfide and Phenol during their Simultaneous Presence in the Atmosphere. *Chem. Abstr.* 69:69522p (1968).
34. **Baydar, A., M. Petrzilka and M.-P Schott:** Olfactory Thresholds for Androstene and Galaxolide: sensitivity, insensitivity and specific anosmia. *Chem. Senses* 18(6):661–68 (1993).
35. **Beck, H.:** Experimentelle Ermittlung von Geruchsschwellen Ciniger Wichtiger Reizasc (Chlor, Schwefeldioxyd, Oson, Nitrose) und Erscheinungen bei Einwirkung Geringer Konzentrationen auf den Menschen, *Thesis*, Wurzburg, 1959.
36. **Bedborough, D.R. and P.E. Trott:** *The Sensory Measurement of Odours by Dynamic Dilution*. Report No.: LR 299(AP), Stevenage: Warren Spring Laboratory, 1979.
37. **Belkov, A.N.:** Action of Small Concentrations of Carbon Tetrachloride on the Human Body, *Tr., Tsent. Inst. Usoversh. Vrachei* 135:90–96 (1969) [from *Chem. Abstr.* 74:2340n (1971)].
38. **Belles, F. (ed.):** *Fluorine Handbook*. Cleveland, OH: National Aerospace and Space Administration, Lewis Research Center, Cited by P. M. Ricca, Exposure Criteria for Fluorine Rocket Propellants. *Arch. Environ. Health* 12:399–407 (1966) (1965).
39. **Berck, B.:** Sorption of Phosphine by Cereal Products, *J. Agric. Food Chem.* 16:419–25 (1968).
40. **Berglund, B., U. Berglund, I. Johansson, and T. Lindvall:** Formaldehyde – Absolute Odor Threshold and Perceived Odor Intensity. In *Proc. Indoor Air Pollution*, Stockholm, August, 1984. pp. 89–96.
41. **Berglund, B., U. Berglund, L. Hogman, I. Johansson, and T. Lindvall:** The Psychophysics of Formaldehyde Odor, In *Indoor Air*. Seifer, B., H. Esdorn, M. Fisher, H. Ruden, and J. Wegner (eds.). (1987). Institute for Water, Soil and Air Hygiene, Berlin, 2:655–60 [cited by B. Berglund and S. Nordin, Detectability and Perceived

Intensity for Formaldehyde in Smoker and Non-smokers. *Chem. Senses* 17:291–306 (1992)].

42. **Berglund, B. and S. Nordin:** Detectability and Perceived Intensity for Formaldehyde in Smoker and Non-smokers. *Chem. Senses* 17:291–306 (1992).
43. **Berglund, B. and H.S. Esfandabad:** Humans as Discerner of Odor and Irritation. *Abstracts 10th Congress of the European Chemoreception Research Organisation (ECRO)*, August 23–28, Munich, 1992.
44. **Berthelot, M.:** Observations sur les Procèdes Propres a Determiner les Limites de la Sensibilite Olfactive. *Ann. Chim. Phys.* 22:460–64 [French] (1901).
45. **Berzins, A.:** Determination of the Threshold of Olfactory Perception of Ethylenimine. *Aktual. Chem. Abstr.* 72:35503e (1970).
46. **Bezpakova, L.E.:** Experimental Data for Determining the Hygienic Standards of Methylacrylate in the Atmosphere. [Russian], *Gig. Sanit.* 10:3–7 (1967).
47. **Bezpakova, L.E.:** The Threshold of Reflex Effects of Methylacrylate Vapours upon Human Organism. [Czech.] *Cesk. Hyg.* 12:577–82 (1967).
48. **Bilings, C.E. and L.C. Jonas:** Odor Thresholds in Air as Compared to Threshold Limit Values, *Am. Ind. Hyg. Assoc. J.* 42:479–80 (1981).
49. **Blanchard, J.:** Relazioni tra l'odore di un Gas ed il Suo Tenore in Mercaptani. [Italian], *Gas (Rome)* 26:287–309 (1976).
50. **Blank, I., K.-H Fischer, and W. Grosch:** Intensive Neutral Odourants of Linden Honey, Differences from Honeys of Other Botanical Origin. *Z. Lebensm. Unters. Forsch.* 189:426–33 (1989).
51. **Blank, I.:** Thesis. TU Munchen [cited by M. Rychlik, P. Schieberle and W. Grosch, Compilation of Odor Thresholds, Odor Qualities and Retention Indices of Key Food Odorants, *Deutsche Forschungsanstalt für Lebensmittelchemie and Institut für Lebensmittelchemie der Technischen Universität München*, 1998, pp 63.]
52. **Blank, I., A. Sen and W. Grosch:** Potent Odorants of the Roasted Powder and Brew of Arabica Coffee. *Z. Lebensm. Unters. Forsch.* 195:239–45 (1992).
53. **Blank, I. and P. Schieberle:** Analysis of the Seasoning-like Flavour Substances of a Commercial Lovage Extract (*Levisticum officinale* Koch). *Flav. Fragr. J.* 8:191–95 (1993).
54. **Blinova, A.E.:** Industrial Standards for Substances Emitting Strong Odors. *Hyg. Sanit. USSR* 30:1–22 [English translation of Russian article] (1965).
55. **Bocca, E. and M.N. Battiston:** Odour Perception and Environment Conditions. *Acta Oto-Laryngol.* 57:391–400 (1964).

56. **Bokowa, A. and J.A. Beukes:** How Reliable are Odour Assessments? *Water Science & Technology* 66(10):2049–55 (2012).
57. **Borisova, M.K.:** Experimental Data for Determination of the Maximum Allowable Concentration of Dichloroethane in the Atmosphere. *Gig. Sanit.* 22:13–19, [English summary of Russian article] (1957).
58. **Bozza, G. and P. Colombo:** Eliminazione Degli Odori di una Fabbrica di Cellulosa al Solfato. *Chem. Ind. (Milan)* 31:233–41 [Italian] (1949).
59. **Braker, W. and A.L. Mossman:** *Matheson Gas Data Book*. 6th edition. Basking Ridge, NJ: Matheson Gas, Lyndhurst. 1980.
60. **Brunekreef, B. and H. Harssema:** Viscose Odors in Ambient Air. *Water, Air, Soil Pollut.* 13:439–46 (1980).
61. **Buchberg, H., M.H. Jones, H.G. Lindh, and K.W. Wilson:** *Air Pollution Studies with Simulated Atmospheres* (Report no. 61-44). Los Angeles, CA: California University, 1961.
62. **Buettner, A. and P. Schieberle:** Evaluation of Key Aroma Compounds in Hand Squeezed Grapefruit Juice (Citrus paradise Macfayden) by Quantitation and Flavor Reconstitution Experiments. *J. Agric. Food Chem.* 49:1358–63 (2001).
63. **Buettner, A. and P. Schieberle:** Evaluation of Aroma Differences between Hand Squeezed Juices from Valencia Late and Navel Oranges by Quantitation and Flavor Reconstitution Experiments. *J. Agric. Food Chem.* 49:2387–94 (2001).
64. **Buhr, K.:** Untersuchungen zu Struktur Geruchsbeziehungen von Lactonen, Ph.D. Thesis, Bergishen Universitat including corrections by J.C. Leffingwell, *Chirality and Odour Perception: The p-mentane lactones – Part 1 & Part 2*, 2006.
65. **Bushtueva, K.A.:** Threshold Reflex Effect of SO<sub>2</sub> and Sulfuric acid Aerosol Simultaneously Present in the Air. In *Limits of Allowable Concentrations of Atmospheric Pollutants* edited by V.A. Ryazanov, Book 4:72–80. Washington, D.C.: U.S. Department of Commerce, Office of Technical Services, 1960.
66. **Bushtueva, K.A.:** New Studies of the Effect of Sulfur Dioxide and of Sulfuric Acid Aerosol on Reflex Activity of Man. In *Limits of Allowable Concentrations of Atmospheric Pollutants*. Ryazanov, V.A. (ed.). Book 5:86–92, 1962.
67. **Buttery, R.C., R.M. Seifert, D.G. Guadagni, and L.C. Ling:** Characterization of Some Volatile Constituents of Bell Peppers. *J. Agric. Food Chem.* 17:1322–27 (1969).
68. **Cain, W.S.:** Odor Intensity: Differences in the Exponent of the Psychophysical Function. *Percept. Psychophys.* 6(6A):349–54 (1969).
69. **Cain, W.S.:** Differential Sensitivity for Smell: Noise at the Nose. *Science* 195:796–98 (1977).

70. **Cain, W.S.: Odor Identification by Males and Females:** Prediction Versus Performance. *Chem. Sens.* 7:129–42 (1982).
71. **Cain, W.S., J. Gent, F.A. Catalanotto and R.B. Goodspeed:** Clinical Evaluation of Olfaction, *Am. J. Otolaryngol.* 4:252–56 (1983).
72. **Cain, W.S., B.P. Leadere, L. Cannon, T. Tosun and H. Ismail:** Odorization of Inert Gas for Occupational Safety Psychophysical Considerations. *Am. Ind. Hyg. Assoc. J.* 48:47–55 (1987).
73. **Cain, W.S., J.F. Gent, R.B. Goodspeed and G. Leonard:** Evaluation of Olfactory Dysfunction in the Connecticut Chemosensory Clinical Research Center. *Laryngoscope* 98:83–88 (1988).
74. **Cain, W.S.:** Testing Olfaction in a Clinical Setting. *Ear, Nose and Throat J.* 68(4):316, 322–28 (1989).
75. **Cain, W.S. and J.F. Gent:** Olfactory Sensitivity: Reliability, Generality and Association with Aging. *J. Exp. Psychol. Hum. Percept. Perform.* 17:382–91 (1991).
76. **Cain, W.S., R.A. De Wijk, A.A. Jalowayski, G. Pila Caminha, and R. Schmidt:** Odor and Chemesthesis from Brief Exposures to TXIB. *Indoor Air* 15:445–57 (2005).
77. **Cain, W.S., R. Schmidt, and P.Wolkoff:** Olfactory Detection of Ozone and D-limonene: Reactants in Indoor Spaces. *Indoor Air* 17:337–47 (2007).
78. **Cain, W.S., R. Schmidt, and A.A. Jalowayski:** Odor and Chemesthesis from Exposures to Glutaraldehyde Vapor. *Int. Arch. Occup. Environ. Health* 80:721–31 (2007).
79. **Cain, W.S., R.A. De Wijk, S. Nordin, and M. Nordin:** Independence of Odor Quality and Absolute Sensitivity in a Study of Aging. *Chem. Percept.* 1:24–33 (2008).
80. **Cain, W.S. and R. Schmidt:** Can We Trust Odor Databases? Example of i- and n-butyl acetate. *Atmos. Environ.* 43:2591–601 (2009).
81. **Cain, W.S., M.S. Dourson, M.J. Kohrman-Vincent, and B.C. Allen:** Human Chemosensory Perception of Methyl Isothiocyanate: Chemesthesis and Odor. *Reg. Toxicol. Pharmacol.* 58:173–80 (2010).
82. **Cancho, B., C. Fabrellas, A. Diaz, F. Ventura, and T. Galceran:** Determination of the Odor Threshold Concentrations of Iodinated Trihalomethanes in Drinking Water. *J. Agric. Food Chem.* 49:1881–84 (2001).
83. **Carpenter, C.P.:** The Chronic Toxicity of Tetrachloroethylene. *J. Ind. Hyg. Toxicol.* 19:323–36 (1937).

84. **Carpenter, CP., H.F. Smyth, Jr., and C.B. Shaffer:** The Acute Toxicity of Ethylene Imine to Small Animals. *J. Ind. Hyg. Toxicol.* 30:2–6 (1948).
85. **Catana, I, S. Negoias, A. Maniu, M. Porojan, and M. Cosgarea:** A Modified Version of “Sniffin Sticks” Odor Identification Test: The Romanian Cultural Adaptation. *Clujul Medical* 85(2):218–23 (2012).
86. **Cederlof, R., M.L. Edfors, L. Friberg, and T. Lindvall:** On the Determination of Odor Thresholds in Air Pollution Control. An Experimental Field Study on Flue Gases from Sulfate Cellulose Plants. *J. Air Pollut. Control Assoc.* 16:92–94 (1966).
87. **Cerny, C. and W. Grosch:** Precursors of Ethyldimethylpyrazones Isomers and 2,3-dimethyl-5-methylpyrazine Formed in Roasted Beef. *Z. Lebensm. Unters. Forsch.* 198:210–14 (1994).
88. **Chao-Chen-Tzi:** Data for Determining the Standard Maximum Permissible Concentration of Methanol Vapours in the Atmospheric Air. *Gig. Sanit.* 24:7–12 [English summary of Russian article] (1959).
89. **Cheesman, G.H. and H.M. Kirkby:** An Air Dilution Olfactometer Suitable for Group Threshold Measurements. *Q.J. Exp. Psychol.* 11:115–23 (1959).
90. **Chizhikov, V.A.:** Data for Substantiating the Maximum Permissible Concentration of Toluilendiisocyanate in the Atmospheric Air. *Gig. Sanit.* 28:8–15 [English translation of Russian article] (1963).
91. **Clausen, D.F., C.E. Felt, L.F. Borchardt, and A.A. Reinsch:** Quantitative Odor Measurement. *Mod. Packag.* 29:149–56, 212, 215, 218, 220 (1955).
92. **Colwell, C.E.:** Human Sensory Irritation Threshold of Glutaraldehyde Vapor. In *Documentation of the Threshold Limit Values*, 4th edition. Cincinnati, OH: ACGIH®, 1980. pp. 204–205.
93. **Cometto-Muñiz, J.E. and W.S. Cain:** Thresholds for Odor and Nasal Pungency. *Physiol. Behav.* 48:719–25 (1990).
94. **Cometto-Muñiz, J.E. and W.S. Cain:** Nasal Pungency, Odor, and Eye Irritation Thresholds for Homologous Acetates. *Pharmacol. Biochem. Behav.* 39:983–89 (1991).
95. **Cometto-Muñiz, J.E.:** “Personal Communication”. New Haven, CT: The John B. Pierce Laboratory, 1993.
96. **Cometto-Muñiz, J.E. and W.S. Cain:** Efficacy of Volatile Organic Compounds in Evoking Nasal Pungency and Odor. *Arch. Environ, Health* 48:309–14 (1993).
97. **Cometto-Muñiz, J.E. and W.S. Cain:** Sensory Reactions of Nasal Pungency and Odor to Volatile Organic Compounds: the Alkybenzenes. *Am. Ind. Hyg. Assoc. J.* 55:811–17 (1994).

98. **Cometto-Muñiz, J.E., W.S. Cain and M.H. Abraham:** Nasal Pungency and Odor of Homologous Aldehydes and Carboxylic Acids. *Exp. Brain Res.* 118:180–88 (1998).
99. **Cometto-Muñiz, J.E., W.S. Cain, M.H. Abraham and R. Kumarsingh:** Trigeminal and Olfactory Chemosensory Impact of Selected Terpenes. *Pharmacol. Biochem. Behav.* 60:765–70 (1998).
100. **Cometto-Muñiz, J.E., W.S. Cain, M.H. Abraham, J.M.R. Gola:** Chemosensory Detectability of 1-butanol and 2-heptanone Singly and in Binary Mixtures, *Physiol. Behav.* 67:269–76 (1999).
101. **Cometto-Muñiz, J.E.:** “Personal Communication”. San Diego, CA: Chemosensory Perception Laboratory, Dept. of Surgery (Otolaryngology). 1999.
102. **Cometto-Muñiz, J.E., W.S. Cain, M.H. Abraham and J.M.R. Gola:** Psychometric Functions for the Olfactory and Trigominal Detectability of Butyl Acetate and Toluene. *J. Appl. Toxicol.* 22:25–30 (2002).
103. **Cometto-Muñiz, J.E., W.S. Cain and M.H. Abraham:** Dose Addition of Individual Odorants in the Odor Detection of Binary Mixtures. *Behav. Brain Res.* 138:95–105 (2003).
104. **Cometto-Muñiz, J.E., W.S. Cain and M.H. Abraham:** Detection of Single and Mixed VOC’s by Smell and by Sensory Irritation. *Indoor Air* 14(8):108–17 (2004).
105. **Cometto-Muñiz, J.E., W.S. Cain and M.H. Abraham:** Odor Detection of Single Chemicals and Binary Mixtures. *Behav. Brain Res.* 156:115–23 (2005).
106. **Cometto-Muñiz, J.E. and M.H. Abraham:** Human Olfactory Detection of Homologous n-Alcohols Measured via Concentration Response Functions. *Pharmacol. Biochem. Behav.* 89:279–91 (2008).
107. **Cometto-Muñiz, J.E., W.S. Cain, M.H. Abraham and J. Gil-Lostes:** Concentration Detection Functions for the Odor of Homologous n-Acetate Esters. *Physiol. Behav.* 95:658–67 (2008).
108. **Cometto-Muñiz, J.E. and M.H. Abraham:** Olfactory Psychometric Functions for Homologous 2-Ketones. *Behav. Brain Res.* 201:207–15 (2009).
109. **Cometto-Muñiz, J.E. and M.H. Abraham:** Olfactory Detectability of Homologous n-Alkylbenzenes as Reflected by Concentration Detection Functions in Humans. *Neuroscience* 161:236–48 (2009).
110. **Cometto-Muñiz, J.E. and M.H. Abraham:** Odor Detection by Humans of Lineal Aliphatic Aldehydes and Helional as Gauged by Dose Response Functions. *Chem. Senses* 35:289–99 (2010).

- 111. Cometto-Muñiz, J.E. and M.H. Abraham:** Structure Activity Relationships on the Odor Detectability of Homogous Carboxylic Acids by Humans. *Exp. Brain Res.* 207:75–84 (2010).
- 112. Corbit, T.E. and T. Engen:** Facilitation of Olfactory Detection. *Percept. Psychophys.* 10:433–36 (1971).
- 113. Cormack, D., T.A. Dorling, and B.W.J. Lynch:** Comparison of Techniques for Organoleptic Odour-intensity Assessment. *Chem. Ind. (London)* 16:857–61 (1974).
- 114. Crawford, G.N., R.P. Garrison and D.R. McFee:** Odor Threshold Determination for 2-nitro-propane. *Am. Ind. Hyg. Assoc. J.* 45(2):B7–B8 (1984).
- 115. Cristoph, N.:** Die Anwendung der Gaschromatografischen Sniffing Technik zur Gestimmung von Geruschsschwellen und Aromaweten. *Thesis, TU Munchen*, 1983.
- 116. Czerny, M., et al.:** Re-investigation on Odour Thresholds of Key Food Aroma Compounds and Development of an Aroma Language Based on Odor Qualities of Defined Aqueous Odorant Solutions. *Eur. Food Res. Technol.* 228:265–73 (2008).
- 117. Czerny, M., R. Brueckner, E. Kirchhoff, R. Schmitt, and A. Buettner:** The Influence of Molecular Structure on Odor Qualities and Odor Detection Thresholds of Volatile Alkylated Phenols. *Chem. Senses* 36:539–53 (2011).
- 118. Dalton, P., C.J. Wysocki, M.J. Brody and H. J. Lawley:** Perceived Odor, Irritation and Health Symptoms Following Short-term Exposure to Acetone. *Am. J. Ind. Med.* 31:558–69 (1997).
- 119. Dalton, P., C.J. Wysocki, M.J. Brody and H. J. Lawley:** The Influence of Cognitive Bias on the Perceived Odor, Irritation and Health Symptoms from Chemical Exposure. *Int. Arch. Occup. Environ. Health* 69:407–17 (1997b).
- 120. Dalton, P., D.D. Dilks and M.I. Banton:** Evaluation of Odor and Sensory Irritation Thresholds for Methyl Isobutyl Ketone in Humans. *Am. Ind. Hyg. Assoc. J.* 61:340–50, (2000)
- 121. Dalton, P., B. Cowart, D. Dilks, M. Gould, P.S.J. Lees, A. Stefaniak and E. Emmett:** Olfactory Function in Workers Exposed to Styrene in the Reinforced Plastics Industry. *Am. J. Ind. Med.* 44:1–11 (2003), [converted data taken from: P. Dalton, et.al, Evaluation of Long-term Occupational Exposure to Styrene Vapor on Olfactory Function. *Chem. Senses* 32:739–47 (2007)].
- 122. Dalton, P., et al.:** Evaluation of Long-term Occupational Exposure to Styrene Vapor on Olfactory Function. *Chem. Senses* 32:739–47 (2007).
- 123. Davis, P.A.:** Carbon Tetrachloride as an Industrial Hazard. *J. Am. Med. Assoc.* 103:962–66 (1934).



124. **Davis, R.G.:** Olfactory Psychophysical Parameters in Man, Rat, Dog and Pigeon. *J. Compo Physiol. Psychol.* 85:221–32 (1973).
125. **Deadman, K.A. and J.A. Prigg:** *The Odour and Odorization of Gas.* (Research Communication GC59), London: The Gas Council, 1959.
126. **Deese, D.E. and R.E. Joyner:** Vinyl Acetate: A Study of Chronic Human Exposure. *J. Am. Ind. Hyg. Assoc.* 30:449–57 (1969).
127. **Dixon, G.A. and K.G. Ikels:** *Olfactory Threshold of Chlorine in Oxygen.* (AD Rep., AD-A 046015). Springfield, VA: National Technical Information Service, Dept. of Commerce, 1977.
128. **Dobrinskii, A.A.:** Establishing Hygienic Norms for Some Intermediates in Caprolactam Production in the Air. *Hyg. Sanit.* 29:8–13 [English translation of Russian article] (1964).
129. **Doll, W. and K. Boumot:** Über den Geruch Optischer Antipoden. *Pharmazie* 4:224–227, (1949).
130. **Dollnick, H.W.O., V. Thiele, and F. Drawert:** Olfatometrie von Schwefelwasserstoff, n-Butanol, Isoamyalkohol, Propionsäure und Dibutylamin. *Staub.* 48:325–31 (1988).
131. **Don, J.A.:** Odour Measurement and Control. Paper presented at 'Odour Control in Industry', London, March 18 (to be published in 'Filtration and Separation'), (1986).
132. **Doty, P., R.L., P. Shaman, S.L. Appelbaum, R. Giberson, L. Sikorski, and L. Rosenberg:** Smell Identification Ability: Changes with Age. *Science* 226:1441–43 (1984).
133. **Doty, R.L., D.A. Deems, R.E. Fyre, R. Perlberg and A. Shapiro:** Olfactory Sensitivity, Nasal Resistance, and Autonomic Function in Patients with Multiple Chemical Sensitivities. *Arch. Otolaryngol. Head Neck Sur.* 114:1422–27 (1988).
134. **Doty, R.L.:** "Unpublished" (cited by R.L. Doty, Psychophysical Measurement of Odor Perception in Hymans. In *The Human Sense of Smell.* Laing, D.G., R.L. Doty, W. Breipohl (eds.). Berlin: Springer-Verlag, 1991. pp. 95–134.
135. **Dravnieks, A.:** Development of a Dynamic Forced-choice Triangle Olfactometer and of a Sample Storage System for the Odor Unit Content Determination of Rendering Plant Exhausts. *IITRI Project C8217-2, Final Report*, Chicago, IL: IITRI Research Institute, 1973.
136. **Dravnieks, A.:** A Building-Block Model for the Characterization of Odorant Molecules and Their Odors. *Ann. NY Acad. Sci.* 237:144–63 (1974).

- 137. Dravnieks, A.:** Development of a Dynamic Forced Choice Triangle Method for the Measurement of Emission and Ambient Odors. *IITRI Project C 8217*, Final Report. Chicago, IL: IIT Research Institute, 1976.
- 138. Dravnieks, A. and B.K. Krotoszynski:** Collection and Processing of Airborne Chemical Information II. *J. Gas Chromatogr.* 6:144–49 (1968).
- 139. Dravnieks, A. and A. O'Donnell:** Principles and Some Techniques of High-Resolution Headspace Analysis. *J. Agric. Food Chem.* 19:1049–56 (1971).
- 140. Dravnieks, A. and P. Laffort:** Physio-Chemical Basis of Quantitative and Qualitative Odor Discrimination in Humans. In *Olfaction and Taste IV*. Schneider, D. (ed.). Proc. Fourth Internat. Symp., August 1971, Stuttgart: Wissenschaftliche Verlagsgesellschaft MBH, 1972, pp. 142–48.
- 141. Dravnieks, A., W. Schmidtsdorff and M. Meilgaard:** Odor Thresholds by Forced Choice Dynamic Triangle Olfactometry: Reproducibility and Methods of Calculation. *J. Air Pollut. Control Assoc.* 36:900–05 (1986).
- 142. Duan-Fen-Djuy:** Data for Determining the Maximum Permissible Concentration of Hydrogen Sulfide in the Atmospheric Air. *Gig. Sanit.* 24:12–17, [English summary of Russian article] 1959.
- 143. Dubrovskaya, F.I.:** Hygienic Evaluation of Pollution of Atmospheric Air of a Large City with Sulfur Dioxide Gas. In *Limits of Allowable Concentrations of Atmospheric Pollutants*, edited by V. A. Ryazanov, Book 3, Washington, D.C.: Dept. of Commerce, 1957. pp. 37–51.
- 144. Dubrovskaya, F.I.:** One-Time Maximum Concentration of Caproic Acid in Air. *Hyg. Sanit. USSR* 34:331–35 [English translation of Russian article] (1969).
- 145. Dubrovskaya, F.I., et al.:** Atmospheric Air Pollution with Discharges from Synthetic Fatty Acids and Alcohols Producing Industries and Their Effect on the Health of the Population. *Gig. Sanit.* 26:3–8 [English summary of Russian article] (1961).
- 146. Dubrovskaya, F.I. and M. Khachatryan:** Sanitary Evaluation of Valeric Acid as an Atmospheric Pollutant. In *American Institute of Crop Ecology (AICE) Survey of U.S.S.R. Air Pollution Literature*, edited by M. Y. Nuttonson. 19:98–103, [English translation of Russian article] (1973).
- 147. Dudley, H.C. and J.W. Miller:** Toxicology of Selenium VI. Effects of Subacute Exposure to Hydrogen Selenide. *J. Ind. Hyg. Toxicol.* 23:470–77 (1941).
- 148. Dumas, T. and E.J. Bond:** Separation of Phosphine from Odour Producing Impurities. *J. Stored Prod. Res.* 10:67–68 (1974).
- 149. Dunlap, M.K., J.K. Kodama, J.S. Wellington, H.H. Anderson, and C.H. Hine:** The Toxicity of Allyl Alcohol. *AMA Arch. Ind. Health* 18:303–11 (1958).

150. **Eglite, M.E.:** A Contribution to the Hygienic Assessment of Atmospheric Ozone. *Hyg. Sanit. USSR* 33:18–23 [English translation of Russian article] (1968).
151. **Elfimova, E.V.:** Data for Determination of the Maximum Permissible Concentration of Hydrogen Chloride Aerosols in the Atmospheric Air (in Russian). *Gig. Sanit.* 24(1):13–20 (1959).
152. **Elfimova, E.V.:** Biological Activity of Low Atmospheric Concentrations of Isopropylbenzene and Benzene. *Chem. Abstr.* 66:93947g (1967).
153. **Endo, R., T. Kohgo, and T. Oyaki:** Research on Odor Nuisance in Hokkaido (part 2). Chemical Analysis of Odors. In *Proceedings of the 8th Annual Meeting of the Japan Society of Air Pollution, Symposium (I): Hazard Nuisance Pollution Problems Due to Odors* [English Translation foAPTIC by SCITRAN]. 1967.
154. **England, A., D.G. Laing, and H. Panhuber:** *Odorous Pollutants*. CSIRO Division of Food Research, Report of Research 1977–1978. 1978. p. 23.
155. **Etzweiler, F., N. Neuner-Jehle and E. Senn:** Einsatz der Quantitativen Head-space-Analyse zur Untersuchung des Behaltens von Riechstoffen in Seifen un Detergentien. *Seife-Ole-Fette-Wachse* 106:419–27 (1980).
156. **Eykman, H. J.:** “Klinische Olfactometrie”. Thesis, Utrecht, [German], 1927.
157. **Fassett, D.W.:** Esters. In *Industrial Hygiene and Toxicology. Volume II. Toxicology*. Patty, F.A. (ed.). New York: Wiley Interscience, 1962. pp.1847-1934.
158. **Feddes, J.J.R., G. Qu, C.A. Quelette and J.J. Leonard:** Development of an Eight Panelist Single Port, Forced Choice, Dynamic Dilution Olfactometer. *J. of Canadian Biosystems Engineering* 43:6.1–6.5 (2001).
159. **Feldman, Y.G.:** Data for Determining the Maximum Permissible Concentration of Acetone in the Atmosphere. *Gig. Sanit.* 25:3–10 [English summary of Russian article] (1960).
160. **Feldman, Y.G.:** The Experimental Determination of the Maximum Permissible One-Time Concentration of Diketene in the Atmosphere. *Hyg. Sanit. USSR* 32:9–14 [English translation of Russian article] (1967).
161. **Feldman, Y.G. and T.I. Bonashevskaya:** On the Effect of Low Concentrations of Formaldehyde. *Hyg. Sanit. USSR* 36:174–80, [English translation of Russian article] (1971).
162. **Ferreira, V., M. Ardanuy, R. Lopex, and J.F. Cacho:** Relationship between Flavor Dilution Values and Odor Unit Values in Hydroalcoholic Solutions: Role of Volatility and a Practical Rule for its Estimation. *J. Agric. Food Chem.* 46:4341–46 (1998).
163. **Fieldner, A.C., S.H. Katz, and S.P. Kinney:** *Gas Masks for Gases Met in Fighting Fires*. Technical Paper no. 248. U.S. Bureau of Mines, 1921.

- 164. Filatova, V.I.:** Data for Substantiating the Maximum Permissible Concentration of Methylmethacrylate in the Atmospheric Air. *Gig. Sanit.* 27:3–8 [English summary of Russian article] (1962).
- 165. Fischer, A., W. Grab, and P. Schieberle:** Characterisation of the Moset Odour-active Compounds in a Peel Oil Extract from Ponitanak Oranges (*Citrus nobilis* var. *Lour. microcarpa* Hassk.). *Eur. Food Res. Technol.* 227:735–44 (2008).
- 166. Flemming, D.C. and R.E. Johnstone:** Recognition Thresholds for Diethyl Ether and Halothane. *Anesthesiol.* 46:68–69 (1977).
- 167. Fluck, E.:** The Odor Threshold of Phosphine. *J. Air Pollut. Control Assoc.* 26:795 (1976).
- 168. Flury, F.:** Uber Kampfgasvergiftungen IX. Lokal Reizende Arsenverbindungen. *Z. Gesamte Exp. Med.* 13:523–78 [German] (1921).
- 169. Fomin, A.P.:** Biological Effect of Epichlorohydrine and Its Hygienic Significance as an Atmospheric Pollutant. *Hyg. Sanit. USSR* 31:357–63, [English translation of Russian article] (1966).
- 170. Frantikova, D.:** Determination of the Olfactory Threshold to Industrial Poisons. *Act. Nerv. Super.* 4:184–85 [Czech.] (1962).
- 171. Freudenberg, K. and M. Reichert:** Sulphate Pulp Mill Odors. *Tappi.* 38:165A–66A (1955).
- 172. Fuller, G.H., R. Steltenkamp and G.A. Tisserand:** The Gas Chromatograph with Human Sensor: Perfumer Model. *Ann. N.Y. Acad. Sci.* 116:711–24 (1964).
- 173. Gavaudan, P., H. Poussel, G. Brebion, and M.P. Schutzenberger:** L'étude des Conditions Thermodynamiques de l'excitation Olfactive et les Theories de l'olfaction. *C.R. Hebd. Seances Acad. Sci.* 226:1995–96 [French] (1948).
- 174. Gavaudan, P. and H. Poussel:** Conditions Physio-chimiques du Declenchement des Sensations Olfactives. *C.R. Congr. Natl. Soc. Savantes Sect. Sci.* 90:607–13 [French] (1966).
- 175. Geier, F.:** "Beitrag zur Bestimmung von Geruchsschwellen". Thesis, Wurzburg, [German], 1936.
- 176. Gemert, L.J. Van:** Unpublished data. Zeist, The Netherlands: Central Institute for Nutrition and Food Research TNO, 1973. [cited in Gemert, 1982.]
- 177. Gijs, L., G. Piraprez, P. Perpetè, E. Spinnler, and S. Collin:** Retention of Sulfur Flavours by Food Matrix and Determination of Sensorial Data Independent of the Medium Composition. *Food Chem.* 69:319–30 (2000).

178. **Glindemann, D., J. Novak, and J. Witherspoon:** Dimethyl Sulfoxide (DMSO) Waste Residues and Municipal Waste Odor by Dimethyl Sulfide (DMS): the North-Cast WPCP Plant of Philadelphia. *Environ. Sci. Technol.* 40:202–07 (2006).
179. **Gofmekler, V. A.:** Data to Substantiate the Maximum Permissible Concentration of Acetates in the Atmosphere. *Gig. Sanit.* 25:9–15 [English summary of Russian article] (1960).
180. **Gofmekler, V. A.:** Experimental Observation of the Reflex Effect of Acetaldehyde upon Human Organism. *Cesk. Hyg.* 12:369–75, [English summary Czech. article] (1967).
181. **Gofmekler, V.A.:** Effect of Acetaldehyde on Reflexes. *Hyg. Sanit.* 33(4-6):109–10 (1968).
182. **Goldenberg, D.M.:** *Geruchswahrnehmung und Schwellen von Duftgemischen Beim Menschen.* Hals-, Nasen- und Ohrenheilkunde, Zwanglose Schriftenreihe, Heft 19, Leipzig: Barth, 1967.
183. **Gorlova, O.E.:** Hygienic Evaluation of Isopropyl Alcohol as an Air Pollutant. *Hyg. Sanit. USSR* 35:178–84 [English translation of Russian article] (1970).
184. **Green, D.M. and J.A. Sweets:** *Signal Detection Theory and Psychophysics.* New York: John Wiley & Sons, 1966.
185. **Greenman, J., et al.:** Study on the Organoleptic Intensity Scale for Measureing Malodor. *J. Dent. Res.* 83:81–85 (2004).
186. **Grigorieva, K.V.:** Studies on Pollution of Atmospheric Air with Maleic Anhydride. *Hyg. Sanit. USSR* 29:7–11 [English translation of Russian article] (1964).
187. **Grijns, G.:** Messungen der Riechscharfe bei European und Javanen. *Arch. Anat. Physiol., Physiol. Abt.* 30:509–17 (1906).
188. **Grijns, G.:** Y-a-t-il une Relation Entre le Pouvoir Absorbant a l'egard de la Chaleur Rayonnante et le Pouvoir Odorant des Substances. *Arch. Neerl. Physioi.* 3:377–90 [French] (1919).
189. **Guadagni, D.G.:** Private Communication. (1966), [cited by R. Teranishi, High Resolution Gas Chromatography in Aroma Research. *Am. Perfum. Cosmet.* 82:43–44,46,48,51 (1967)].
190. **Gundlach, R.H. and G. Kenway:** A Method for the Determination of Olfactory Thresholds in Humans. *J. Exp. Psychol.* 24:192-201, (1939).
191. **Gusev, I.S.:** Reflective Effects of Microconcentrations of Benzene, Toluene, Xylene, and Their Comparative Assessment. *Hyg. Sanit. USSR* 30:331–36 [Russian], (1965).

192. **Guth, H.:** Objectivation of White Wine Aromas. Thesis, TU Munchen, (1997), [cited by W. Grosch, Evaluation of the Key Odorants of Foods by Dilution Experiments, Aroma Models and Omission, *Chem. Senses* 26:533–45 (2001)].
193. **Hall, G. and J. Andersson:** Volatile Fat Oxidation Products 1. Determination of Odour Thresholds and Odour Intensity Functions by Dynamic Olfactometry. *Lebensm. Wiss. Technol.* 16:354–61 (1983).
194. **Hamanabe, Y., S. Harima, T. Miura, T. Domon and T. Namiki:** The Sensitive Measurement of Bad Odor in Using of an Odorless Chamber. *J. Jpn. Soc. Air Pollut.* 4(1):115. [Japanese, English Abstract], (1969).
195. **Hangartner, M.:** Geruchsprobleme bei der Abfallbeseitigung, Haustechnik-Bauphysik-Unwelttechnik-Gesund. *Ing.* 102:117–20 (1981).
196. **Harrevelde, A.Ph. Van and P. Heeres:** The Validation of the Draft European CEN Standard for Dynamic Olfactometry by an Interlaboratory Comparison on Butanol. *Gefahrstoffe-Reinhaltung der Luft* 57:393–98 (1997).
197. **Hartigh, P. Den:** The Development of Odour Determination Methods in the Rijnmond Area. *VDI-Berichte* 56:277–87 (1986).
198. **Hartung, L.D., E.G. Hammond, and J.R. Miner:** Identification of Carbonyl Compounds in a Swine-Building Atmosphere. *Livest. Waste Manage. Pollut. Abatement, Proc. Int. Symp.* 105–06 (1971).
199. **Heeres, P., H. Harssema, and C.J. Smitshoek:** The Influence of the Dilution Air on the Threshold Value of H<sub>2</sub>S. *Staub* 46:426–428, (1986).
200. **Hellman, T.M. and F.H. Small:** Characterization of Petrochemical Odors. *Chem. Eng. Progr.* 69:75–77 (1973).
201. **Hellman, T.M. and F.H. Small:** *The Characterization of Odor from the Petrochemical Industry Using Sensory Methods*. MS. No 4988 [unpublished]. New York: American Institute of Chemical Engineers, 1973.
202. **Hellman, T.M. and F.H. Small:** Characterization of the Odor Properties of 101 Petrochemicals Using Sensory Methods. *J. Air Pollut. Contr. Assoc.* 24:979–82, (1974).
203. **Henderson, Y. and H.W. Haggard:** The Elimination of Industrial Organic Odors. *J. Ind. Eng. Chem.* 14:548–51 (1922).
204. **Hendriks, A.P.J.:** *Zelfadaptatie en Kruisadaptatie met enige Primaire en Niet-Primarie Geurstoffen*. [Unpublished]. Utrecht, The Netherlands: University of Utrecht, Psychology Laboratory, 1979.
205. **Henning, H.:** *Der Geruch (Zweite Ausgabe)*. Leipzig: Barth, [German reviewed at TRC]. 1924.

206. **Henning, H.:** Psychologische Studien am Geruchssinn. In *Handbuch der Biologischen Arbeitsmethoden VI*, edited by E. Abderhalden. Berlin/Wien: Urban and Schwarzenberg. A:741–836 [German reviewed at TRC] 1927.
207. **Henschler, D., A. Stier, H. Beck, and W. Neumann:** Olfactory Threshold of Some Important Irritant Gases and Manifestation in Man by Low Concentrations. *Arch. Gewerbepathol. Gewerbehyg.* 17:547–70 [German] (1960).
208. **Henschler, D., W. Assmann, and K.O. Meyer:** Zur Toxikologie der Toluylendiisocyanate. *Arch. Toxicol.* 19:364–87 [German reviewed at TRC], (1962).
209. **Hermanides, J.:** "Over de Constanten der in de Olfactometri Gebruikelijke Negen Standaardgeuren". Thesis. Utrecht, The Netherlands, University of Utrecht, 1909.
210. **Hermans, L.T.M.:** *Ringonderzock Olfactometers*. Publikatiereeks Lucht 80, Ministerie van Volkshuisvesting, Ruimteijke Ordening en Milieubeheer, Leidschendam, 1989.
211. **Herrmann, J. and I. Abd El Salam:** Die Sensorische Bestimmung der Verteilungskoeffizienten und Dampfdrucke von Aromastoffen mit Starken Positiven Bzw. Negativen Abweichungen vom Raoultischen Gesetz in Wasser-Ol-Gemengen. *Lebensm. Will. Technol.* 13:123–27 (1980).
212. **Herrmann, J. and I. Abd El Salam:** Neue Methoden der Be- und Auswertung Sensorischer Eigenschaften von Lebensmitteln und der Berchnung Ihrer Veränderungen. 13. Mitt. Die Ermittlung der Geruchsintensität Reiner Aromastoffe Soowie Deren Losungen als Methode sur Bestimmung Inrer Dampfdrucke, *Nahrung.* 24:389–97 (1980).
213. **Hesse, W.:** Bestimmung von Geruchsschwellen in Absoluten Werten. *Z. Hals-, Nasen-, Ohrenheilk.* 16:359–73 [German review at TRC] (1926).
214. **Hesse, W.:** Ein Einfacher Apparat zur Bestimmung von Geruchsschwellen in Absoluten Werten. *Z. Hals-, Nasen-, Ohrenheilk.* 19:348–52, [German reviewed at TRC] (1928).
215. **Heyroth, F. F.:** Halogens. In *Industrial Hygiene and Toxicology, Volume 11 Toxicology*. Patty, F.A. (ed.). New York/London: Interscience, 1963, pp. 849–851.
216. **Higuchi, T. and J. Masuda:** Interlaboratory Comparison of Olfactometry in Japan. *Water Sci. Technol.* 50(4):147–52 (2004).
217. **Hildenskiold, R.S.:** Maximum Permissible Concentration of Carbon Bisulphide in the Atmospheric Air of Residential Districts. *Gig. Sanit.* 24:3–8 [English summary of Russian article] (1959).
218. **Hill, D.T. and C.L. Barth:** Quantitative Prediction of Odor Intensity. *Transactions of the ASAE* 939–44 (1976).

- 219. Hine, C.H., H. Ungar, H.H. Anderson, J.K. Kodama, J.K. Critchlow, and N.W. Jacobson:** Toxicological Studies on p-Tertiary-Butyltoluene. *AMA Arch. Ind. Hyg. Occup. Med.* 9:227–44 (1954),
- 220. Hine, C.H., J.K. Kodama, R.J. Guzman, and G.S. Loquvam:** The Toxicity of Allylamines. *Arch. Environ. Health* 1:343–52 (1960).
- 221. Hine, C.H., R.J. Guzman, M.K. Dunlap, R. Lima, and G.S. Loquvam:** Studies on the Toxicity of Glycidaldehyde. *Arch. Environ. Health* 2:23–30 (1961).
- 222. Hine, C.H., A. Pasi, and B.G. Stephens:** Fatalities Following Exposure to 2-nitropropane. *J. Occup. Med.* 20:333–37 (1978).
- 223. Hofmann, F.B. and A. Kohlrausch:** Bestimmung von Geruchsschwellen. *Biochem., Z.* 156:287–94 [German] (1925).
- 224. Holland, J.:** Memorandum to S. C. Kelton Jr., Rohm & Haas Co. (1974). [Cited in *Documentation of the Threshold Limit Values*, 3rd edition. Cincinnati, OH: ACGIH®, 1971, 4th printing, 1977. p. 411].
- 225. Hollingsworth, R.L., V.K. Rowe, F. Oyen, H.R. Hoyle, and H.C. Spencer:** Toxicity of Paradichlorobenzene. *AMA Arch. Ind. Health* 14:138–47 (1956).
- 226. Hollingsworth, R.L., V.K. Rowe, F. Oyen, T.R. Torkelson, and E.M. Adams:** Toxicity of o-dichlorobenzene. *AMA Arch. Ind. Health* 17:180–87 (1958).
- 227. Hollingsworth, R.L., V.K. Rowe and F. Oyen:** Toxicity of Acetylene Tetrabromide Determined on Experimental Animals. *Am. Ind. Hyg. Assoc. J.* 24:28–35 (1963).
- 228. Holmes, J.A., E.C. Franklin, and R.A. Gould:** *Report of Selby Smelter Commission.* (Department of the Interior, Bureau of Mines, Bulletin 98.) Washington, D.C.: U.S. Dept. of Commerce, 1915.
- 229. Homans, J.W., H. Kohler, E. Paul, and D. Bardtke:** Bestimmung von Geruchsschwellenwerten mit dem Olfaktometer T04 in Kombination mit dem Flammenionisationsdetektor (FID). *Staub.* 38:217–21 [German] 1978.
- 230. Homans, W.J.:** Source Emission Odour Measurement by Combining an Olfactometer and a F.I.D. In *Pro. Int. Symposium Characterization and Control of Odoriferous Pollutants in Process Industries*, Louvain-La-Neuve, April 25–27, 1984. pp. 203–12.
- 231. Hori, M., Y. Kobayashi, and Y. Ota:** Vinyl Chloride Monomer Odor Concentration. *Plast. Ind. News* 18:164–68 (1972).
- 232. Hoshika, Y., T. Imamura, G. Muto, L.J. Van Gemert, J.A. Don and J.I. Walpot:** International Comparison of Odor Threshold Values of Several Odorants in Japan and in The Netherlands. *Environ. Res.* 61:78–83 (1993).



- 233. Hoshika, Y., M. Nishikitani, K. Yokoyama and S. Araki:** A Study on the Determining of Odor Recognition Threshold Values of n- and iso-amyl(pentyl) acetates using Tedlar Bag Static Methods. *Anal. Sci.* 13:505–08 (1997).
- 234. Huijjer, H.:** “De Olfactologie van Aniline en Homolegen.” Thesis, Utrecht, 1924.
- 235. Ifeadi, N.G.:** “Quantitative Measurement and Sensory Evaluation of Dairy Waste Odor.” Thesis. Columbus, OH: Ohio State University, 1972.
- 236. Imasheva, N.B.:** The Substantiation of the Maximum Permissible Concentrations of Acetophenon in the Atmospheric Air. *Gig. Sanit.* 28:3-8, [Russian], (1963).
- 237. Irish, D.D.: Halogenated Hydrocarbons:** I. Aliphatic. In *Industrial Hygiene and Toxicology, Volume 11. Toxicology*. Patty, F.A. (ed.). New York/London: Interscience, 1963, pp. 1305–1307.
- 238. Itskovich, A.A. and V.A. Vinogradova:** Norms for Phenol in the Atmosphere. *Okhrana Prirody Sibiri I Dal'nego Vostoka Sb.* 1:139–45 (1962). [from *Chem. Abstr.* 61:6256b, (1964)].
- 239. Ivanov, S.V.:** Materials on Toxicology and Hygienic Rating of Ethylbenzol Content in the Atmosphere of Industrial Premises. *Gig. Tr. Prof. Zabol.* 8:9–14 [English summary of Russian article] (1964).
- 240. Jacobson, K.H., J.H. Clem, H.J. Wheelwright, W.E. Rinehart, and N. Mayes:** The Acute Toxicity of the Vapors of Some Methylated Hydrazine Derivatives. *AMA Arch. Ind. Health* 12:609–16 (1955).
- 241. Jacobson, K.H., E.B. Hackley, and L. Feinsilver:** The Toxicity of Inhaled Ethylene Oxide and Propylene Oxide Vapors. *AMA Arch. Ind. Health* 13:237–44 (1956).
- 242. Jacobson, K.H., W.E. Rinehart, H.J. Wheelwright, M.A. Ross, J.L. Papin, R.C. Daly, E.A. Greene, and W.A. Groff:** The Toxicology of an Aniline-Furfuryl-Alcohol-Hydrazine Vapor Mixture. *Am. Ind. Hyg. Assoc. J.* 19:91–100 (1958).
- 243. Jagella, T. and W. Grosch:** In Preparation. [cited by: M. Rychlik, P. Schieberle and W. Grosch, *Compilation of Odor Thresholds, Odor Qualities and Retention Indices of Key Food Odorants*, Deutsche Forschungsanstalt für Lebensmittelchemie and Institut für Lebensmittelchemie der Technischen Universität München, 1998. p. 63].
- 244. Janicek, G., V. Pliska, and J. Kubátová:** Olfactometric Estimation of the Threshold of Perception of Odorous Substances by a Flow Olfactometer. *Cesk Hyg.* 5:441–47 [Czech.]. (1960).
- 245. Jeltens, R.:** Measurements of Organic Substances in the Atmosphere in the Netherlands in Relation to Their Possible Detrimental Effect. In *Verunreinigung der Luft*. Hambraeus, G. and I. Jung (eds.). (IVA-Rapport 78.) Stockholm, Sweden, 1975. pp. 21–39.

246. **Jensen, F.P. and H. Flyger:** Application of Reference Odours for Comparison of Sensory Measurements of Odours. In: *Proc. 6th World Congress on Air Quality*, Paris, May 1983, 2:385–92 (1983).
247. **Jones, B.P., H.R. Moskowitz, N. Butters, and G. Glosser:** Psychophysical Scaling of Olfactory, Visual, and Auditory Stimuli by Alcoholic Korsakoff Patients. *Neuropsychol.* 13:387–93 (1975).
248. **Jones, B.P., N. Butters, H.R. Moskowitz, and K. Montgomery:** Olfactory and Gustatory Capacities of Alcoholic Korsakoff Patients. *Neuropsychol.* 16:323–37 (1978).
249. **Jones, F.N.:** An Olfactometer Permitting Stimulus Specification in Molar Terms. *Am. J. Psychol.* 67:147–51 (1954).
250. **Jones, F.N.:** The Reliability of Olfactory Thresholds Obtained by Sniffing. *Am. J. Psychol.* 68:289–90 (1955).
251. **Jones, F.N.:** A Comparison of the Methods of Olfactory Stimulation; Blasting vs. Sniffing. *Am. J. Psychol.* 68:486–88 (1955).
252. **Jones, F.N.:** Olfactory Absolute Thresholds and Their Implications for the Nature of the Receptor Process. *J. Psychol.* 40:223–27 (1955).
253. **Jung, J.:** “Untersuchung Uber Geruchsschwellen.” Thesis, Wurzburg, 1936.
254. **Kaloyanova, F., A. Bainova, G. Dimov, and M. Mukhtarova:** Experimental Determination of the Maximum Permissible Concentration of Dipterex in the Atmosphere. *Hyg. Sanit. USSR* 32:325–30 [English translation of Russian article] (1967).
255. **Kaloyanova, E, G. Ivanova, G. Dimov, and M. Mukhtarova:** Determination of the Maximum Permissible Concentration of Bi-58 (Rogor, Phosphamide) in Air. *Hyg. Sanit. USSR* 33:303–07 [English translation of Russian article] (1968).
256. **Katz, S.H. and E.J. Talbert:** Intensities of Odors and Irritating Effects of Warning Agents for Inflammable and Poisonous Gases. U.S. Bureau of Mines, Technical Report no. 480, Washington, D.C.: U.S. Dept. of Commerce, 1930.
257. **Kay, R.E., J.T. Eichner, and D.E. Gelvin:** Quantitative Studies on the Olfactory Potentials of *Lucilia Sericata*. *Am. J. Physiol.* 213:1–10 (1967).
258. **Kazakova, M.N.:** Unreferenced citation in *The Effects of Certain Aromatic Carbohydrates in the Air*, by N.G. Andreescheva. *Hyg. Sanit. USSR* 33:12–16, (1968).
259. **Kendall, D.A., G. Leonardus, and E.R. Rubin:** *Parameters Affecting the Determination of Odor Thresholds*. (Report C-68988), Cambridge, MA: Arthur D. Little, Inc. 1968.

260. **Kerka, W.F. and C.M. Humphreys:** Temperature and Humidity Effect on Odor Perception. *Heat. Piping Air Cond.* 28:129–36 (1956).
261. **Khachatryan, M., M.I. Gusev, and O.E. Gorlova:** The Effect of Weak Olfactory Stimuli on the Latent Period of Motor Reaction to Light in Man. *Hyg. Sanit. USSR* 33:274–76 [English translation of Russian article] (1968).
262. **Khachatryan, M.K. and B.K. Baikov:** On the Possibility of Detecting the Effect of Reflexes of Low Butanol Concentrations by the Inculcation of Conditioned Inhibition. *Hyg. Sanit. USSR* 34:251–54 [English translation of Russian article] (1969).
263. **Khiari, D., L. Brenner, G.A. Burlingame, and I.H. Suffet:** Sensory Gas Chromatography for Evaluation of Taste and Odor Events in Drinking Water. *Water Sci. Technol.* 25:97–104 (1992).
264. **Khikmatullaeva, S.S:** Maximum Permissible Concentration of Thiophene in the Atmosphere. *Hyg. Sanit. USSR* 32:319–23 [English translation of Russian article] (1967).
265. **Kimmerle, G.:** Persönliche Mitteilung (1971), Institut für Toxikologie der Farbenfabrikanten Bayer A. G., Wuppertal-Elberfeld. [taken from: D. Henschler (Herausgeber), Gesundheitschädliche Arbeitsstoffe (2, Lieferung.) Weinheim: Verlag Chemie, 1973.
266. **Kimmerle, G. and A. Eben:** Zur Toxizität von Methylisocyanat und Dessen Quantitativer Bestimmung in der Luft. *Arch. Toxicol.* 20:235–41 [German, reviewed at TRC] (1964).
267. **Kincaid, J.E, E.L. Stanley, C.H. Beckworth, and EW. Sunderman:** Nickel Poisoning. *Am. J. Clin. Pathol.* 26:107–19 (1956).
268. **Kinkead, E. R., U. C. Pozzani, D. L. Geary, and C. P. Carpenter:** The Mammalian Toxicity of Ethylidenenorbornene (5-Ethylidenebicyclo (2,2,1) Hept-2-ene). *Toxicol. Appl. Pharmacol.* 20:250–59 (1971).
269. **Kinkead, E.R., U.C. Pozzani, D.L. Geary, and C.P. Carpenter:** The Mammalian Toxicity of Dicyclopentadiene. *Toxicol. Appl. Pharmacol.* 20:552–61 (1971).
270. **Kittel, G.:** Die Moderne Olfaktometrie und Odorimetrie. *Z. Laryngol., Rhinol., Otol. Ihre Grenzgeb.* 12:893–903 [German] (1968).
271. **Kittel, G. and P.G.J. Wendelstein:** Zur Differenzierung der Olfaktiven Wahrnehmungs- und Erkennungsschwelle. *Arch. Klin. Exp. Ohren- Nasen- Kehkopfheilkd. (Berlin)* 199:683–87 (1971).
272. **Kleinbeck, S., M. Schäper, S.A. Juran, E. Kiesswetter, M. Blaszkewicz, K. Golka, A. Zimmerman, T. Brüning, and C. van Thriel:** Odor Thresholds and Breathing Changes of Human Volunteers as Consequences of Sulphur Dioxide

- Exposure Considering Individual Factors. *Safety and Health at Work* 2:355–64 (2011).
- 273. Kleinschmidt, E.-G.:** Untersuchungen zum Zusammenhang Zwischen Riechschwelle des Menschen für Einige Substanzen und Deren Chemischer Struktur. *Wiss. Z. Wilhelm-Pieck-Univ. Rostock, Naturwiss. Reihe* 32(7):54–58 (1983).
- 274. Kniebes, D.V., J.A. Chisholm, and R.C. Stubbs:** Olfactory Response for Natural Gas Odorants. Presented at the American Gas Association, Houston, TX, (1968), *Odorizzanti del Gas Naturale: Fattori de Sensibilita' Olfattiva. Gas (Rome)* 19:211–17 (1969).
- 275. Knuth, H.W.:** "Personal Communication" (1973). Institute für Chemische Technologie, TU Clausthal, [cited in Gemert and Nettenbreijer, (1977)].
- 276. Kobal, G. and V. Thiele:** Comparison of Objective and Subjective Methods in Olfactometric Measurements: Example H<sub>2</sub>S. In: *Proc. 6th World Congress on Air Quality*, Paris, May 1983, 2:435–41 (1983).
- 277. Kohler, H. and W.J. Homans:** Kombination von Olfactometer und Flammenionisationsdetektor zur Bestimmung von Geruchsschweilwerten – Einsatz der mcBmethodik in der Praxis. *Staub.* 40:331–35 (1980).
- 278. Komthong, P., S. Hayakawa, T. Katoh, N. Igura, and M. Shimoda:** Determination of Potent Odorants in Apple by Headspace Gas Dilution Analysis. *Lebensm. Wiss. Technol.* 39:472–78 (2006).
- 279. Korneev, Y.E.:** Effect of the Combined Presence of Low Concentrations of Phenol and Acetophenone in the Urban Atmosphere. *Hyg. Sanit. USSR* 30:336–45 [English translation of Russian article] (1965).
- 280. Kosiborod, N.R.:** Hygienic Significance of Low Diethylamine Concentrations in the Atmosphere. *Hyg. Sanit. USSR* 33:31–35 [English translation of Russian article] (1968).
- 281. Köster, E.P.:** Olfactory Sensitivity and the Menstrual Cycle. *Inter. Rhinol.* 3:57–64 (1965).
- 282. Köster, E.P.:** Recovery of Olfactory Sensitivity after Adaptation. In: *Theories of Odor and Odor Measurement* edited by N.N. Tanyolac, Robert College, Istanbul, 1968. pp. 307–336.
- 283. Köster, E.P.:** Olfactory Sensitivity and Ovulatory Cycle Duration. *Olfactologia* 1:43–51 (1968).
- 284. Köster, E.P.:** "Adaptation and Cross-adaptation in Olfaction". Thesis, Utrecht. 1971.
- 285. Koszinowski, J. and O. Piringer:** Die Bedeutung von Oxidationsprodukten Un-gesättigter Kohlenwasserstoffe für die Sensorischen Eigenschaften von Lebens-

mittelverpackungen, 1. Mitteilung: Sauerstoffabkommlinge der 1-Alkene. *Dtsch. Lebensm. Rundsch.* 79:179–83 (1983).

- 286. Krackow, E.H.:** Toxicity and Health Hazards of Boron Hydrides. *AMA Arch. Ind. Hyg. Occup. Med.* 8:335–39 (1953).
- 287. Kramer, C.B.:** "Written Communication" (1974). In *Criteria for a Recommended Standard. Occupational Exposure to Chlorine*, (HEW/NIOSH Publication No. 76-170), Cincinnati, OH: NIOSH, 1976. p. 34.
- 288. Krasovitskaya, M.L. and L.K. Malyarova:** Small Concentrations of Hydrocarbons in the Air of Naphthachemical Plants. *Chem. Abstr.* 65:14324d (1966).
- 289. Krasovitskaya, M.L. and L.K. Malyarova:** Low Unsaturated Ethylene Hydrocarbon Concentrations in the Air Surrounding Operating Crude Oil Complexes (Combines). In *U.S.S.R. Literature on Air Pollution and Related Occupational Diseases*, edited by B. S. Levine. Vol. 16. Springfield, VA: National Technical Information Service, Dept. of Commerce, 1968.
- 290. Krichevskaya, I.M.:** Biological Effect of Caprolactam and Its Sanitary-Hygienic Assessment as an Atmospheric Pollutant. *Hyg. Sanit. USSR* 33:24–31 [English translation of Russian article] (1968).
- 291. Kristesashvili, T.S.:** The Maximum Permissible Concentration of Pyridine in the Air. *Hyg. Sanit. USSR* 30:173–77 [English translation of Russian article] (1965).
- 292. Kulakov, A.E.:** The Effect of Low Concentrations of Hexamethylene Diamine on the Human Body. *Hyg. Sanit. USSR* 29:7–14 [English translation of Russian article] (1964).
- 293. Kulka, W. and E. Homma:** Beitrage zur Kenntnis der Laboratoriumsluft und Deren Schadlichen Beimengungen. *Z. Anal. Chem.* 50:1–11 [German] (1910).
- 294. Kurtschatowa, G. and E. Dawidkowa:** Hygienische Begrundung des Kurzzeit-MIK-Wertes von 4,6,-Dinitroo-Kresol in der Atmospharischen Luft. *Wiss. Z. Humboldt Univ. Berlin, Math.-Naturwiss. Reihe* 19:467–68 (1970).
- 295. Laffort, P.:** Some New Data on the Physico-Chemical Determinants of the Relative Effectiveness of Odorants. In *Theories of Odor and Odor Measurement*, edited by N. N. Tanyoac. Bebek, Istanbul: Robert College, 1968, pp. 247–270.
- 296. Laffort, P.:** Interactions Quantitatives dans un Melange d'odeurs: Niveau Liminaire. *Olfactologia* 1:95–104 [French] (1968).
- 297. Laffort, P. and A. Dravnieks:** An Approach to a PhysicoChemical Model of Olfactory Stimulation in Vertebrates by Single Compounds. *J. Theor. Biol.* 38:335–45 (1973).
- 298. Laffort, P.:** "Personal Communication" (1974). Centre National de la Recherche Scientifique College de France, Paris, [cited in Gemert and Nettenbreijer 1977].

- 299. Laffort, P. and C. Gortan:** Olfactory Properties of Some Gases in Hyperbaric Atmosphere. *Chem. Senses*. 12:139–42 (1987).
- 300. Laing, D.G.:** A Comparative Study of the Olfactory Sensitivity of Humans and Rats. *Chem. Senses Flavor* 1:257–69 (1975).
- 301. Laing, D.G., H. Panhuber, and R.1. Baxter:** Olfactory Properties of Amines and n-Butanol. *Chem. Senses Flavor* 3:149–66 (1978).
- 302. Laing, D.G.:** Characterisation of Human Behaviour during Odour Perception. *Perception* 11:221–30 (1982).
- 303. Laing, D.G.:** Natural Sniffing Gives Optimum Odour Perception for Humans. *Perception*. 12:99–117 (1983).
- 304. Langenhove, H. Van, and N. Schamp:** Chemical and Olfactometric Measurement of Odours. In: *Proc. Int. Symposium Characterization and Control of Odoriferous Pollutants in Process Industries*, Louvain-La-Neuve, April 25-27, 1984, pp. 23–36.
- 305. Laska, M. and R. Hudson:** A comparison of the Detection Thresholds of Odour Mixtures and Their Components. *Chem. Senses* 16:651–62 (1991).
- 306. Laska, M. and A. Ringh:** How Big is the Gap between Olfactory Detection and Recognition of Aliphatic Aldehyde? *Attention, Perception & Psychophysics* 73:860–72 (2010).
- 307. Lea, A.G.H. and G.D. Ford:** Instrumentelle und Sensorische Charakterisierung Handelsüblicher Apfelsäfte. *Flussiges Obst*. 58:121–25 (1991).
- 308. Lehmann, K.B.:** Die Bestimmung Minimaler Schwefelwasserstoffmengen in der Luft. *Arch. Hyg.* 30:262–66 (1897).
- 309. Lehmann, K.B. and L. Schmidt-Kehl:** Die 13 Wichtigsten Chlorkohlwasserstoffe der Fettreihe vom Standpunkt der Gewerbehygiene. *Ach. Hyg. Bakteriol.* 116:131–268 [German] (1936).
- 310. Leonardos G., D. Kendall, and N. Barnard:** Odor Threshold Determinations of 53 Odorant Chemicals. *J. Air Pollut. Control Assoc.* 19:91–95 (1969).
- 311. Lester, D. and W.R. Adams:** The Inhalational Toxicity of Oxygen Difluoride. *Am. Ind. Hyg. Assoc. J.* 26:562–67 (1965).
- 312. Levin, L. and K.L. Gabriel:** The Vapor Toxicity of Trimethyl Phosphite. *Am. Ind. Hyg. Assoc. J.* 34:286–91 (1973).
- 313. Lindell, H.:** A Study of Odour and Taste Originating from Food Packaging Board Analysed by Chromatographic Techniques and Sensory Evaluation. *Acta Academiæ Aboensis, Ser. B, Mathematica et Physica*, 51, no. 4, (Dissertation, Abo Akademi), (1991).

314. **Lindvall, T.:** On Sensory Evaluation of Odorous Air Pollutant Intensities. *Nord. Hyg. Tidskr.* 51(2):1–181 (1970).
315. **Li-Shen:** Data for Substantiating the Maximum Permissible Concentration of Styrol in Atmospheric Air. *Gig. Sanit.* 26:11–17 [English summary of Russian article] (1961).
316. **Loginova, R.A.:** Basic Principles for the Determination of Limits of Allowable Concentrations of Hydrogen Sulfide in Atmospheric Air. In *Limits of Allowable Concentrations of Atmospheric Pollutants*, edited by V. A. Ryazanov. Book 3. Washington, D.C.: U.S. Dept. of Commerce, 1957. pp. 52-68.
317. **Logtenberg, M.T.:** Onderzoek Naar de Nauwkeurigheid van Geurconcentratie-metingen met de TNO Snuifkar. Division for Physical and Chemical Technology TNO, Apeldoorn. Unpublished Report (1978). [Cited in Gemert 1982].
318. **Lötsch, J., S. Nordin, T. Hummel, C. Murphy and G. Kobal:** Chronobiology of Nasal Chemosensitivity: Do Odor or Trigeminal Pain Thresholds Follow a Circadian Rhythm? *Chem. Senses* 22:593–98 (1997).
319. **Macleod, P., J.M. Fauconnier, Ph. Guillet and M.F. Thal:** Development of an Olfactometer According to the French Odour Measurement Standard AFNOR. *VDI-Berichte* 56:271–75 (1986).
320. **Makeicheva, N.A.:** Data for Hygienic Standardization of a Mixture of Furfural, Formaldehyde, Phenol and Acetone in the Atmosphere. *Gig. Sanit.* 9:314–17 [English summary of Russian article] (1978).
321. **Makhinya, A.P.:** Hygienic Assessment of Atmospheric Air Polluted by Sulfur Dioxide Phenol. *Hyg. Sanit., USSR* 31:103–05 [English translation of Russian article] (1966).
322. **Malyarova, L.K.:** Biological Action and Hygienic Significance of Trichloroethylene as an Atmospheric Pollutant. In: A.P. Shitskova: *Mater. Nauch.-Prakt. Knof. Molodykh Gig. Sanit. Vrachei* 11:56–69 (1967), [from *Chem. Abstr.* 72:24265c, (1970)].
323. **Mannebeck, D. and H. Mannebeck:** Qualität und Vergleichbarkeit Olfactometrischer Messungen. *Gefahrstoffe Reinh. Luft.* 62:135–40 (2002).
324. **Marin, A.B., T.E. Acree, and J. Barnard:** Variation in Odor Detection Thresholds Determined by Charm Analysis. *Chem. Senses* 13(3):435–44 (1988).
325. **Martirosyan, A.S.:** Maximum Permissible Concentration of Methyl Vinyl Ketone in the Air of a Working Area. *Tr. Klin. Otd. Nauch.-Issled. Inst. Gig. Profzabol.* 1:113–15 (1970). [From *Chem. Abstr.* 77:156013g, (1972)].
326. **Mateson, J.F.:** Olfactometry; Its Techniques and Apparatus. *J. Air Pollut. Control Assoc.* 5:167–70 (1955).

- 327. Maxeiner, B. and D. Mannebeck:** Ringversuch Olfactometric 2003. *Gefahrstoffe Reinh. Luft* 64:118–23 (2004).
- 328. Maxeiner, B.:** Olfactometric Interlaboratory Comparison Test 2005. *Proceedings of the Water Environment Federation “WEF/AWWA Odors and Air Emissions 2006”*, pp. 688-699.
- 329. Maxeiner, B.:** Ringversuch Olfactometric 2007. Ringversuch zur Dynamischen Olfactometric Nach DIN EN 13725:2003, *VDI-Berichte*, nr. 1995, 31–45 (2007).
- 330. May, J.:** Odor Thresholds of Solvents for Assessment of Solvent Odors in the Air. *Staub*. 26:385–89 [Eng. translation of German article] (1966).
- 331. McCawley, E.L.:** The Comparative Toxicity of 2-chloro-2-butene, 1, 2, 3-trichlorobutane and 1,2-dichloroethane. *Univ. Calif. Publ. Pharmacol.* 2:89–98 (1942).
- 332. McGee, T., C. Yeager and J.M. Decazes:** Ethnic and Cultural Influences on Odour Perception. In: *Proc. 13th Int. Congress of Flavours, Fragrances and Essential Oils* edited by K.H.C. Baer, Istanbul, Turkey, 15-19 October 1995, 3:351–67 (1995).
- 333. McGee, W.A., F.L. Oglesby, R.L. Raleigh, and D.W. Fassett:** The Determination of a Sensory Response to Alkyl 2-Cyanoacrylate Vapor in Air. *Am. Ind. Hyg. Assoc. J.* 29:558–61 (1968).
- 334. McGinley, M.A. and C.M. McGinley:** Comparison of Field Olfactometers in a Controlled Chamber Using Hydrogen Sulfide as the Test Odorant, *Water Sci. Technol.* 50(4):75–82 (2004).
- 335. Melekhina, V.P.:** The Maximum Permissible Concentration of Formaldehyde in the Atmospheric Air. *Gig. Sanit.* 23:10–14 [English summary of article] (1958).
- 336. Melekhina, V.P.:** The Problem of Combined Action of Three Mineral Acids. In *USSR Literature on Air Pollution and Related Occupational Diseases*, Vol. 16. Levine, B.S. (ed.). Washington, D.C.: U.S. Dept. of Commerce. 16:76–81 (1968).
- 337. Melzner, J., T. Bitter, O. Guntinas-Lichius, R. Gottschall, M. Walther and H. Gudziol:** Comparison of the Orthonasal and Retronasal Detection Thresholds for Carbon Dioxide in Humans. *Chem. Senses* 36:435–41 (2011).
- 338. Menshikova, T.A.:** Effect of Small Concentrations of Dibutyl Phthalate on the Organism. *Tr. Nauch. Kong., Nauch. Issled. Inst. Gig. Vod. Transp.* 2:106–10 [from *Chem. Abstr.* 81:21988r, (1972)].
- 339. Middleton, F.M.:** “Personal Communication.” Cited by W.H. Stahl, *Compilation of Odor and Taste Threshold Values Data, ASTM Data Series DS48*. Philadelphia, PA: American Society for Testing and Materials, 1956.
- 340. Minaev, A.A.:** Determination of the Maximum Permissible Concentration of *V*-Methylstyrene Vapours in the Atmosphere. *Hyg. Sanit., USSR* 31:157–61. [English translation of Russian article] (1966).



341. **Minaev, A.A.:** Investigation to Substantiate the Hygienic Standards for Alpha-methylstyrol Vapours in the Atmosphere. (Russian), *Gig. Sanit.* 31(2):3–6 (1966).
342. **Miryakubova, D.A.:** Basis for the Maximum Single Permissible Concentration of Hexachloran in the Atmosphere. *Med. Zh. Uzb.* 9:29–31 (1970). [From Chem. Abstr. 74:90813x, (1971)].
343. **Mitsumoto, T.:** Olfaktometrische Untersuchungen. *Z. Sinnesphysiol.* 57:114–65 317 (1926).
344. **Miyazawa, T., M. Gallagher, G. Petri and P.M. Wise:** Odor Detection of Mixtures of Homologous Carboxylic Acids and Coffee Aroma Compounds by Humans. *J. Agric. Food Chem.* 57(21):9895–901 (2009).
345. **Miyazawa, T., M. Gallagher, G. Petri and P.M. Wise:** Methodological Factors in Odor Detection by Humans. *Chem. Percept.* 2:195–202, (2009).
346. **Mnatsakanyan, A.V.:** Basic Experimental Information for the Determination of the Limit of Allowable Chloroprene Concentration in Atmospheric Air. In *Limits of Allowable Concentrations of Atmospheric Pollutants*. Book 5. Ryazanov, V.A. (ed.). Washington, D.C.: U.S. Dept. of Commerce, 1962, pp. 79–85.
347. **Molhave, L., et al.:** The Eye Irritation and Odor Potencies of Four Terpenes which are Major Constituents of the Emissions of VOCs from Nordic Soft Woods. *Indoor Air* 10:315–318, (2000).
348. **Moriguchi, Y., Y. Kojima, and R. Orii:** The Report of Investigative Methods of Olfactive Materials and their Sources in the Environment of an Ironworks. In: *Proc. VIth World Congress on Air Quality*, Paris, 1983. 2:401–408 (1983).
349. **Morimura, S.:** Untersuchung uber den Geruchssinn. *Tohoku J. Exp. Med.* 22:417–48 [German] (1934).
350. **Moschandreas, D.J. and D.R. Jones:** Odor Detection, Recognition and Fatigue. In: *Proc. VIth World Congress on Air Quality*, Paris, May 1983 2:401–08 (1983).
351. **Moskowitz, H.R. and A. Dravnieks:** Odor Intensity and Pleasantness of Butanol. *J. Exp. Psychol.* 103:216–23 (1974).
352. **Moulton, D.G. and D.A. Marshall:** Odor Detection in Dog and Man. In *Abstracts of the First Congress of the European Chemoreception Research Organization*. Paris, 1974.
353. **Moulton, D.G.:** Olfactory Performance and Receptor Processes in the Dog. In *Food Intake and Chemical Senses*. Katsuki, Y., M. Sato, S.F. Takagi, and Y. Oomura (eds.). Japan Scientific Societies Press, 1977, pp. 73–78.
354. **Mühlen, T. Z.:** Messungen von Styrol-Emissionen und Immissionen Mit Hilfe derr Gaschromatographie. *Zentralbl. Arbeitsmed. Arbeitsschutz* 18:41–43. [German reviewed at TRC], (1968).

- 355. Mukhamedova, S.Yu.:** Maximum Permissible Concentration of Butiphos in Air. *Hyg. Sanit. USSR* 33:328–34 [English translation of Russian article] (1968).
- 356. Mukhitov, B.M.:** Experimental Data for Substantiating Maximum Permissible Concentration of Phenol in the Atmospheric Air. *Gig. Sanit.* 27:16–24 [Russian] (1962).
- 357. Mukhitov, B.:** Influence of Small Concentrations of Phenol on Humans and Animals and its Hygienic Evaluation. *Predel'no Dopustimyye Kontsentratsii Atm. Zagryazn.* 7:76–98 (1963). [From: *Chem. Abstr.* 61:6256d, (1964)].
- 358. Mukhitov, B.M. and A.A. Azimbekov:** Hygienic Evaluation of Methyl Ethyl Ketone as an Atmospheric Pollutant. In: A.P. Filin (Ed.), *Vop. Gig. Tr. Profzabol., Mater. Nauch. Konf.* 232–34 (1971). [From *Chem. Abstr.* 81:10876v, (1974)].
- 359. Mullins, L.J.:** Olfaction. *Ann. N.Y. Acad. Sci.* 62:247–76 (1955).
- 360. Murphy, C. and W.S. Cain:** Odor Identification: the Blind are Better. *Physiol Behav.* 37:177–80 (1985).
- 361. Nader, J.S.:** An Odor Evaluation Apparatus for Field and Laboratory Use. *Am. Ind. Hyg. Assoc. J.* 19:1–7 (1958).
- 362. Nagata, Y.:** Measurement of Odor Threshold by Triangle Odor Bag Method. In *Odor Measurement Review*. Japan: Office of Odor, Noise and Vibration Environmental Management Bureau, Ministry of the Environment, Government of Japan, 2003. pp. 118–127.
- 363. Nagy, G.Z.:** The Odor Impact Model. *J. Air Waste Manage. Assoc.* 41:1360–62 (1991).
- 364. Naus, A.:** *Olphatoric Properties of Industrial Matters*. Prague: Charles University, 1975.
- 365. Naus, A.:** Olfactory Thresholds of Some Industrial Substances (Czech.), *Prac. Lek.* 34:217–18 (1982).
- 366. Naves, Y.R.:** Etudes sur les Matieres Vegetales Volatiles XLVI. Sur le Dedoublement de la d,l-a-ionone. *Helv. Chim. Acta* 30:769–74 [French] (1947).
- 367. Naves, Y.R. and M. Delepine:** Sur les iso-a-irones Actives et Racemique et Leurs Derives. *C.R. Hebd. Seances Acad. Sci.* 237:1167–68 [French] (1953).
- 368. Nawakowski, A.C.:** “Unpublished data (1980).” Environmental Health and Safety. Kalamazoo, MI: The Upjohn Company. [Cited in *Workplace Environmental Exposure Level Guide: Piperidine. Am. Ind. Hyg. Assoc. J.* 43:B91–B92 (1982)].
- 369. Neuhaus, W.:** Wahrnehmungsschwelle und Erkennungsschwelle Beim Riechen des Hundes in Vergleich zu den Riechwahrnehmungen des Menschen. *Z. Vgl. Physiol.* 39:624–33 [German reviewed at TRC] (1957).

- 370. Nevers, A.D. and W.H. Oister:** Problems in the Critical Comparison of Odor Intensities. *Proc. Oper. Sect. Am. Gas. Assoc.* 65:126–29 (1965).
- 371. Nikiforov, B.:** Basis for the Single Maximum Permissible Level of Carbon Tetrachloride in the Atmosphere. *Khig. Zdraveopazvane* 13(4):365–70 (1970), [from *Chem. Abstr.* 74:57059h (1971)].
- 372. Nimmermark, S.:** Influence of Odour Concentration and Individual Odour Thresholds on the Hedonic Tone of Odour from Animal Production. *Biosystems Engineering* 108:211–19 (2011).
- 373. Nishida, K., T. Honda, and K. Bota:** New Apparatus on the Measurement of Odor Intensity. *68th Ann. Meeting Air Pollut. Control Assoc.*, June 15–20, Boston, MA, Paper 75-55.3, 1975.
- 374. Nishida, K., M. Yamakawa, and T. Honda:** Experimental Investigations on the Combined Actions of Components Mixed in Odourous Gas. *Mem. Fac. Eng., Kyoto Univ.* 41(4):552–65 (1979).
- 375. Nordin, S., O. Almkvist, B. Berglund and L.-O. Wahlund:** Olfactory Dysfunction for Pyridine and Dementia Progression in Alzheimer Disease. *Arch. Neurol.* 54:993–98, (1997).
- 376. Novikov, Y.V.:** The Determination of Limits of Allowable Concentrations of Benzene in Atmospheric Air. In *Limits of Allowable Concentrations of Atmospheric Pollutants*. Ryazanov, V.A. (ed.). Translated by B. S. Levine. Book 3. Washington, D.C.: United States Department of Commerce, 1957, pp. 69–87.
- 377. Nyström, A.E.:** Health Hazards in the Chloroprene Rubber Industry and Their Prevention. *Acta Med. Scand., Suppl.* 219:1–125, (1948).
- 378. Odoshashvili, D.G.:** Data for Substantiating the Level of Maximum Permissible Concentration of Dimethylformamide in the Atmospheric Air. *Gig. Sanit.* 27:3–7 [English summary of Russian article] (1962).
- 379. Oglesby, F.L., J.H. Sterner, and B. Anderson:** Quinone Vapors and Their Harmful Effects II. Plant Exposures Associated with Eye Injuries. *J. Ind. Hyg. Toxicol.* 29:74–84 (1947).
- 380. Ohma, S.:** La Classification des Odeurs Aromatiques en Sousclasses. *Arch. Neerl. Physiol.* 6:567–90 [French] (1922).
- 381. Olmstead, E.V.:** Dibromoethane. In *Encyclopedia of Occupational Health and Safety* 1. Geneva: International Labour Office, 1972. pp. 384–385.
- 382. Olsson, P. and M. Laska:** Human Male Superiority in Olfactory Sensitivity to the Sperm Attractant Odorant Bourgeonal. *Chem. Senses* 35:427–32 (2010).

- 383. Öztürk, S.:** "A Study of Selected Environmental Influences on the Organoleptic Properties of Drinking Water." Thesis, Drexel University, *Dissertation Abstracts International B*, 37:1859–60 (1976).
- 384. Pangborn, R.M., H.W. Berg, E.B. Roessler, and A.D. Webb:** Influence of Methodology on Olfactory Response. *Percept. Mot. Skills* 18:91–103 (1964).
- 385. Parker, G.H. and E.M. Stabler:** On Certain Distinctions between Taste and Smell. *Am. J. Physiol.* 32:230–40 (1913).
- 386. Passy, J.:** Sur Quelques Minimums Perceptibles d-Odeurs. *C.R. Hebd. Seances Acad. Sci.* 114:786–88 [French] (1892).
- 387. Passy, J.:** Sur la Perception des Odeurs. *C.R. Soc. Bioi.* 44:239–43 [French] (1892).
- 388. Passy, J.:** Les Proprieties Odorantes des Alcools de la Serie Grasse. *C.R. Hebd. Seances Acad. Sci.* 114:1140–43 [French] (1892).
- 389. Passy, JH.,** Sur Les Minimums Perceptibles de Quelques Odeurs. *C.R. Hebd. Seances Acad. Sci.* 114:306–08 [French] (1892).
- 390. Passy, J.:** Pouvoir' Odorant du Chloroforme, du Bromoforme et de L'iodoforme. *C. R. Hebd. Seances Acad. Sci.* 116:769–70 [French] (1893).
- 391. Passy, J.:** Forme Periodique du Pourvoir Odorant Dans la Serie Grasse. *C.R. Hebd. Seances Acad. Sci.* 116:1007–10 [French] (1893).
- 392. Passy, J.:** L'odeur dans la Serie Grasse. *C.R. Soc. Bioi.* 45:479–81 [French] (1893).
- 393. Passy, J.:** Revenue Generale sur les Sensations Olfactives. *Annee Psychol.* 2:363–410 [French] (1895).
- 394. Patte, F.:** "Le Codage Moleculaire de la Stimulation Olfactive-Clarifications et Mises au Point." Thesis, Lyon, 1978.
- 395. Patte, F. and P.H. Punter:** Experimental Assessment of Human Olfactory Thresholds in Air for Some Thiols and Alkanes. *Chem. Senses Flavour* 4:351–54 (1979).
- 396. Patterson, M.Q., J.C. Stevens, W.S. Cain and J.E. Cometto-Muñiz:** Detection Thresholds for an Olfactory Mixture and its Three Constituent Compounds. *Chem. Senses* 18:723–34 (1993).
- 397. Patty, F.A. and W.P. Yant:** *Odor Intensity and Symptons Produced by Commercial Propane, Butane, Pentane, Hexane, and Heptane Vapor.* U.S. Bureau of Mines, R.I. 1929. p. 2979.
- 398. Patty, F.A.:** Alkaline Materials. In *Industrial Hygiene and Toxicology Volume II. Toxicology.* Patty, F.A. (ed.). New York/London: Interscience, 1962, pp. 859–869.

- 399. Patty, F.A.:** Arsenic, Phosphorus, Selenium, Sulfur and Tellurium. In *Industrial Hygiene and Toxicology Volume II. Toxicology*. Patty, F.A. (ed.). New York/London: Interscience, 1962, pp. 871–910.
- 400. Patty, F.A.:** Inorganic Compounds of Oxygen, Nitrogen and Carbon. In *Industrial Hygiene and Toxicology Volume II. Toxicology*. Patty, F.A. (ed.). New York/London: Interscience, 1962, pp. 911–940.
- 401. Piggott, J. R. and R. Harper:** Ratio Scales and Category Scales of Odour Intensity. *Chem. Senses Flavor* 1:307–16 (1975).
- 402. Piringer, O. and R. Granzer:** Geruchsschwellenkonzentrationen Einiger Ester der Acryl- und Methacrylsäure in Luft und Prüfebensmitteln. Tätigkeitsbericht, Fraunhofer Institut für Lebensmitteltechnologie und Verpackung, 1984. pp. 49–55.
- 403. Pliska, V. and G. Janicek:** Olfaktometrische Studien 1. Die Definition der Wahrnehmungsschwelle - Konzentration Mittels Volumengraden. *Sb. Vys. Sk. Chem.-Technol. Praxe, Oddil Fak. Potravin. Technol.* 4(2):115–27 [German; reviewed at TRC] (1960).
- 404. Pliska, V.:** Olfaktometrische Studien IV. Fehleranalyse 01-faktometrischer Messungen. *Sb. Vys. Sk. Chem.-Technol. Praxe, Potravin* 6(1):37–46 (1962).
- 405. Pliska, V. and G. Janicek:** Die Veränderungen der Wahrnehmungsschwellen-Konzentration der Riechstoffe in Einiger Homologischen Serien. *Arch. Int. Pharmacodyn.* 156:211–16 [German reviewed at TRC] (1965).
- 406. Plotnikova, M.M.:** Data on Hygienic Evaluation of Acrolein as a Pollution of the Atmosphere. *Gig. Sanit.* 22:10–15 [Russian] (1957).
- 407. Pogosyan, U.G.:** The Effect on Man of the Combined Action of Small Concentrations of Acetone and Phenol in the Atmosphere. *Hyg. Sanit. USSR* 30:1–9 [English translation of Russian article] (1965).
- 408. Polednik, B. M. Skwarczynski, M. Dudzinska and W. Krawczyk:** Odor Detection Threshold of n-Butanol in Indoor Air. *Proc. ECOpole* 2(1):85–90 (2008).
- 409. Polgar, L.G., R.A. Duffee, and L.J. Updyke:** Odor Characteristics of Mixtures of Sulfur Compounds Emitted from the Viscose Process. Paper 75-55.2 presented at the 68th Annual Meeting Air Pollut. Control Assoc., Boston, Mass., June 15-20, (1975).
- 410. Pomeroy, R.D. and H. Cruse:** Hydrogen Sulfide Odor Threshold. *J. Am. Water Works Assoc.* 61:677 (1969).
- 411. Poostchi, E., A.W. Gnypp and C.C. St. Pierre:** Comparison of Models Used for the Determination of Odor Thresholds. *Atmos. Environ.* 20:2459–64 (1986).
- 412. Popov, I.N., E.F. Cherkasov, and O.L. Trakhtman:** Determination of Sulfur Dioxide Odor Threshold Concentration. *Gig. Sanit.* 17:16–20 [Russian] (1952).

- 413. Popov, V.A.:** Hygienic Evaluation of Tetrahydrofuran as an Atmospheric Pollutant. *Hyg. Sanit. USSR* 35:178–82 [English translation of Russian article], (1970).
- 414. Pozomiek, E.J., et al.:** Research in the Olfactory Detection of Chemical Agents. *Microchem. J.* 16:136–44 (1971).
- 415. Pozzani, U.C., C.P. Carpenter, P.E. Palm, E.S. Weil, and J.H. Nair:** An Investigation of Mammalian Toxicity of Acetonitrile. *J. Occup. Med.* 1: 634–42 (1959).
- 416. Pozzani, U.C., E.R. Kinkead, and J.M. King:** The Mammalian Toxicity of Methacrylonitrile. *Am. Ind. Hyg. Assoc. J.* 29: 202–10 (1968).
- 417. Prah, J.D., D. Ashley, D. House, K.L. Cohen and T. Gerrity:** Sensory, Symptomatic, Inflammatory, and Ocular Responses to and the Metabolism of Methyl Tertiary Butyl Ether in a Controlled Human Exposure Experiment. *Inhal. Toxicol.* 6:521–38 (1994).
- 418. Prentiss, A.M.:** *Chemicals in War*. New York: McGraw-Hill Book Company, Inc., 1937. p. 262.
- 419. Prusakov, V.M., K.K. Dushutin, N.N. Ladygina, and E.A. Verzhbitskaya:** A Combined Action of the Products of the Interaction of Dimethylamine and Nitrogen Dioxide on the Olfactory Analyzer of Man. *Gig. Sanit.* 41:14–18 [Russian] (1976).
- 420. Punter, P.H.:** “Unpublished report.” (1975) Onderzoek naar de Reukgevoeligheid en Mogelijke Mengeffecten van de 6 Belangrijkste Componenten van Varkensmest. Psychological Laboratory, University of Utrecht. [cited in Gemert and Nettenbreijer 1977].
- 421. Punter, P.H.:** “Personal Communication.” Utrecht, The Netherlands: Psychological Laboratory, University of Utrecht. [Cited in Gemert and Nettenbreijer 1977].
- 422. Punter, P.H.:** Measurement of Human Olfactory Thresholds for Several Groups of Structurally Related Compounds. *Chem. Senses* 7:215–35 (1983).
- 423. Randebroek, E.E.:** Molecular Theory of Odor with the Vheliz as Potential Perceptor. In *Gustation and Olfaction*, edited by G. Ohloff and A. F. Thomas. London/New York: Academic Press, 1971. pp. 111–125.
- 424. Randebroek, R.E.:** Geruch und Konstitution. Eine Beweisführung für die Molekulartheorie des Geruches. Teil IV: Olfaktometerversuche zur Molekulartheorie des Geruches, *Parfuem. Kosmt.* 67(1):10–24 (1986).
- 425. Ranson, C. Von and H.-D. Belitz:** Untersuchungen zur StrukturAktivitätsbeziehung bei Geruchsstoffen. 2. Mitteilung: Wahrnehmungs- und Erkennungsschwellenwerte Sowie Geruchsqualitäten Gesättigter und Ungesättigter Aliphatischer Aldehyde. *Z. Lebensm. Unters. Forsch.* 195:515–22 (1992).

- 426. Ranson, C. Von and H.-D Belitz:** Untersuchungen zur Struktur Aktivitätsbeziehung bei Geruchsstoffen. 2. Mitteilung: Wahrnehmungs und Erkennungsschwellenwerte Sowie Geruchsqualitäten Alicyclischer und Aromatischer Aldehyde. *Z. Lebensm. Unters. Forsch.* 195:523–26 (1992).
- 427. Ripp, G.K.:** Hygienic Basis for the Determination of the Allowable Concentration Limit of Divinyl in Atmospheric Air. In *U.S.S.R. Literature on Air Pollution and Related Occupational Diseases*. Translated by B.S. Levine. Volume 17. Washington, D.C.: U.S. Department of Commerce, 1968, pp. 18–32.
- 428. Robbins, M.C.:** Determination of Naphthalene in Air. *AMA Arch. Ind. Hyg. Occup. Med.* 4:85–87 (1951).
- 429. Rocén, E.:** Contribution to the Localisation of "Sweet Smell." *Scand. Arch. Physiol.* 40:129–44 (1920).
- 430. Roos, C., J. Joziassse, J.V. Klarenbeek, P.H. Punter and J. Schaefer:** Standardizatic Olfactometers. *Publikaticreeks Lucht 49*, Ministerie van Volkshyisvesting, Ruimtelijke Ordening on Milieubeheer, Leidschendam, 1985.
- 431. Rotenberg, Y.S. and F.D. Mashbits:** Concerning Toxic Effect of Low Trimethylamine Concentrations. *Gig. Tr. Prof. Zabol.* 11(4):26–30 (1967).
- 432. Rowe, V.K. and M.A. Wolf:** Ketones. In *Industrial Hygiene and Toxicology*. Volume 11. Patty, F.A. (ed.). New York/London: Interscience, 1962, pp. 1719–70.
- 433. Rowe, V.K.:** "Private Communication", (1975). Cited in *Documentation of the Threshold Limit Values*, 3rd edition, 4th printing. Cincinnati, OH: ACGIH®, 1977. pp. 400–401.
- 434. Rumsey, D.W. and R.P. Cesta:** Odor Threshold Levels for UDMH and N02. *Am. Ind. Hyg. Assoc. J.* 31:339–42 (1970).
- 435. Rupp, H. and D. Henschler:** Effects of Low Chlorine and Bromine Concentrations on Man. *Int. Arch. Gewerbepathol. Gewerbehyg.* 23:79–90 [National Translations Center] (1967).
- 436. Ryazanov, V.A., K.A. Bushtueva, and Y.V. Novikov:** Methods for the Experimental Determination of Limits of Allowable Concentrations of Atmospheric Air Pollutants. In *Limits of Allowable Concentrations of Atmospheric Pollutants*. Ryazanov, V.A. (ed.). Translated by B.S. Levine, Book 3. Washington, D.C.: U.S. Department of Commerce, 1957, pp. 96–128.
- 437. Rylova, M.L.:** Toxicity of 1,1-ethene Dichloride. *Farmakol. Toksikol. (Moscow)* 16:47–50 [National Translations Center] (1953).
- 438. Sadilova, M.S.:** Studies in the Standardization of Maximum Allowable Hydrogen Fluoride Concentrations in the Air of Inhibited Areas. In *U.S.S.R. Literature on Air*

*Pollution and Related Occupational Diseases*, translated by B. S. Levine. Volume 17. Washington, D.C.: U.S. Department of Commerce, 1968, pp. 118–128.

- 439. Saifutdinov, M.M.:** Maximum Permissible Concentration of Ammonia in the Atmosphere. *Hyg. Sanit. USSR* 31:171–76 [English translation of Russian article] (1966).
- 440. Sakuma, K., T. Miura, T. Domon, K. Kanazawa, S. Harima, and T. Namiki:** "17. Sensory Measurement of Odors Using an Odorless Chamber". In *Proceedings of the 8th Annual Meeting of the Japan Society of Air Pollution: Symposium (1): Hazard Nuisance Pollution Problems Due to Odors*, [English Translation], 1967.
- 441. Sales, M.:** Odeur et Odorisation des gaz (IGU/36-58). *Congres International de l'industrie du Gaz*, Rome, 1958.
- 442. Sanders, G. and R. Dechant:** Hydrogen Sulphide Odor Threshold Study. Unpublished report of the Air and Industrial Hygiene Laboratory, California, Department of Public Health, (1961). [Cited in A Study of the Variation of Odor Threshold Concentrations Reported in the Literature, by H. F. Droege. Paper No. 67-117 presented at the *Proceedings 60th Annual Meeting Air Pollution Control Association*, Cleveland, OH, 1967.
- 443. Sanders, G.R., R.A. Umbraco, S. Twiss, and P.K. Mueller:** *The Measurement of Malodor in a Community by Dynamic Olfactometry*. (Report no. 86.) Air and Industrial Hygiene Laboratory, Laboratory Services State of California Department of Public Health, Berkeley, 1970.
- 444. Sävenhed, R., H. Borén and A. Grimvall:** Stripping Analysis and Chromatographic Sniffing for the Source Identification of Odorous Compounds in Drinking Water. *J. Chrom.* 328:219–31 (1985).
- 445. Scharfenberger, G.:** Alternative Bestimmung der Geruchsintensitat. *Papier Kunststoff Verarbeiter* 25(10):36:39–40 (1990).
- 446. Scherberger, R.F., G.P. Happ, F.A. Miller, and D.W. Fassett:** A Dynamic Apparatus for Preparing Air-Vapor Mixtures of Known Concentrations. *Am. Ind. Hyg. Assoc. Quart.* 19:494–98 (1958).
- 447. Scherberger, R.F., F.A. Miller and D.W. Fassett:** The Determination of n-butylamine in Air. *Am. Ind. Hyg. Assoc. J.* 21:471–74 (1960).
- 448. Schieberle, P., H. Ehrmeier and W. Grosch:** Aromastoffe aus dem Saurckatalysierten Abbau von Citral. *Z. Lebensm Unters. Forsch.* 187:35–39 (1988).
- 449. Schieberle, P. and W. Grosch:** Quantitative Analysis of Important Volatile Flavour Compounds in Fresh and Stored Lemon oil/Citric acid Emulsions. *Lebens. Wiss. Technol.* 21:158–62 (1988).
- 450. Schley, O.-H.:** "Untersuchung uber Geruchsschwellen." Thesis, Wurzburg, 1934.



- 451. Schmidt, R. and W.S. Cain:** Making Scents: Dynamic Olfactometry for Threshold Measurements. *Chem. Senses* 35:109–20 (2010).
- 452. Schneider, R.A. and S. Wolf:** Olfactory Perception Thresholds for Citral Utilizing a New Type Olfactorium. *J. Appl. Physiol.* 8:337–42 (1955).
- 453. Schneider, R.A., J.P. Costiloe, R.P. Howard and S. Wolf:** Olfactory Perception Thresholds in Hypogonadal Women: Changes Accompanying Administration of Androgen and Estrogen. *J. Clin. Endocrinol. Metab.* 18:379–90 (1958).
- 454. Schneider, R.A., C.E. Schmidt, and J.P. Costiloe:** Relation of Odor Flow Rate and Duration to Stimulus Intensity Needed for Perception. *J. Appl. Physiol.* 21:10–14 (1966).
- 455. Schoedder, F.:** Messen von Geruchsstoffkonzentrationen, Erfassen von Geruch. *Grundl. Landtechnik* 27:73-82 [German] (1977).
- 456. Schulman, A.E.:** *Statistical Analysis of MTBE Odor Detection Thresholds in Drinking Water.* U.S. Environmental Protection Agency, Washington, D.C., 2011.
- 457. Schutte, W. and J.P. Zubek:** Changes in Olfactory and Gustatory Sensitivity After Prolonged Visual Deprivation. *Canadian J. Physiol.* 21:337–45 (1967).
- 458. Schwarz, R.:** Uber die Riechscharfe der Honigbiene. *Z. Vgl. Physiol.* 37:180–10 [German reviewed at TRC] (1955).
- 459. Selyuzhitskii, G.V.:** Experimental Data Used to Determine the Maximum Permissible Concentration of Methyl Mercaptan, Dimethyl Sulfide, and Dimethyl Disulfide in the Air of the Production Area of Paper and Pulp Plants. *Gig. Tr. Prof. Zabol.* 16:46–47 [Russian] (1972).
- 460. Selyuzhitskii, G.V., V.P. Timofeev, A.M. Nikon, and S.I. Ivanov:** Possible Use of Natural Mercaptans for Imparting Odor to Producer Gas. In *Tezisy Dokl. Nauchn. Sess. Khim. Tekhnol. Org. Soedin. Sery Sernistykh Neftei*, 14th edition. Bakhtalze, I.G. (ed.). 1975. pp. 105–106, [Russian] [from: *Chem. Abstr.* 89:101091y, (1978)].
- 461. Semenova, V.N., O.V. Sadovnik, V.N. Fedyanina, E.V. Kreslina and V.A. Kopanev:** Study of the Biological Action of Ethylene Chlorohydrin as a Factor in Air Pollution. [Russian], *Gig. Sanit.* 1:84–85 (1980).
- 462. Sfiras, J. and A. Demeillers:** Application of Gas Chromatography to Olfactometric Research. *Recherches* 16:47–57 (1967).
- 463. Sgibnev, A.K.:** The Action of Low Formaldehyde Fumes Concentrations on the Human Organism. *Gig. Tr. Prof. Zabol.* 12:20–25 [English summary of Russian article] (1968).
- 464. Shalamberidze, O.P.:** Reflex Action of a Mixture of Sulfur Dioxide and Nitrogen Dioxides. *Hyg. Sanit. USSR* 32:10–15 [English translation of Russian article] (1967).

- 465. Sherrard, G.C.:** The Practical Application of Two Qualitative Tests for HCN in Ship Fumigation. *Public Health Rep.* 43:1016–22 (1928).
- 466. Shusterman, D. and J. Balmes:** Measurement of Nasal Irritant Sensoritivity to Pulsed Carbon Dioxide: A Pilot Study. *Arch. Environ. Health* 52:334–40 (1997).
- 467. Shusterman, D. and J. Balmes:** A Comparison of Two Methods for Determining Nasla Irritant Sensitivity. *Am. J. Rhinol.* 11:371–78 (1997).
- 468. Singh, P., T. Ramasivan, and K. Krishnamurthy:** A Simple Phosphine Detector. *Bull. Grain Technol.* 5:24–26 (1967).
- 469. Sinkuvene, D.S.:** Hygienic Evaluation of Acrolein as an Air Pollutant. *Hyg. Sanit. USSR* 35:325–29 [English translation of Russian article] (1970).
- 470. Slavgorodskiy, L.P.:** Biological Effects and Hygienic Evaluation of Air by Phthalic Anhydride. In *USSR Literature on Air Pollution and Related Occupational Diseases*, translated by B.S. Levine. Vol. 17. Washington, D.C.: U.S. Department of Commerce, 1968, pp. 54–59.
- 471. Slotnick, B.M.:** Absolute Olfactory Thresholds in Rats and Humans. Third Annual Meeting Program, The Association of Chemoreception Sciences, Abstract Volume, cited by B.M. Slotnick and F.W. Schoonover, Olfactory Thresholds in Unilaterally Bulbectomized Rats, *Chem. Senses* 9:325–40 (1984).
- 472. Smeets, M. and P. Dalton:** Perceived Odor and Irritation of Isopropanol: a Comparison between Naive Controls and Occupationally Exposed Workers. *Int. Arch. Occup. Environ. Health* 75:541–48 (2002).
- 473. Smeets, M.A.M., et al.:** Odor and Irritation Thresholds for Ammonia: a Comparison Between Static and Dynamic Olfactometry. *Chem. Senses* 32:11–20 (2007).
- 474. Smith, H.O. and A.D. Hochstettler:** Determination of Odor Thresholds in Air Using C14-Labeled Compounds to Monitor Concentrations. *Environ. Sci. Technol.* 3:169–70 (1969).
- 475. Smith, S.L. and L.K. Duffy:** Odor and Health Complaints with Alaskan Pipelines. *Chem. Health Saf.* 2:32–38 (1995).
- 476. Smolczyk, E. and H. Cobler:** Chemischer Nachweis von Atemgiften und Subjektive. Empfindlichkeit. *Gasmasker* 2:27–33 (1930).
- 477. Smyth, H.F. and J. Seaton:** Acute Response of Guinea Pigs and Rats to Inhalation of the Vapors of Tetraethyl Orthosilicate (Ethyl Silicate). *J. Ind. Hyg. Toxicol.* 22:288–96 (1940).
- 478. Smyth, H.F.:** Improved Communication-Hygienic Standards for Daily Inhalation. *Am. Ind. Hyg. Assoc. Q.* 17:129–85 (1956).

- 479. Solomin, G.I.:** Data Substantiating Maximum Permissible Concentration of Dinyll in Atmospheric Air. *Gig. Sanit.* 26:3–8, [English summary of Russian article] (1961).
- 480. Solomin, G.I.:** Experimental Data for Hygienic Substantiation of the Single Permissible Concentrations of Isopropylbenzol and Isopropylbenzene Hydrogenperoxide in Atmospheric Air. *Hyg. Sanit. USSR* 29:1–8 [English translation of Russian article] (1964).
- 481. Spadone, J.-C., W. Matthey-Doret and I. Blank:** Formation of Methyl (methylthio) methyl Disulfide in Broccoli (*Brassica oleracea* (L.) var.), In *Flavour Science: Recent Advances and Trends*. Bredie, W.L.P. and M.A. Petersen (eds.). Elsevier B.V., 2006, pp. 309–314.
- 482. Stang, A.Q. and G. Helmchen:** Enansioselective Syntheses and Gragrance Properties of the Four Stereoisomers of Magnolione® (Magnolia Ketone). *Helv. Chim. Acta* 88:2738–46 (2005).
- 483. Stalker, W.W.:** Defining the Odor Problem in a Community. *Am. Ind. Hyg. Assoc. J.* 24:600–05 (1963).
- 484. Steinmetz, G., G.T. Pryor, and H. Stone:** Effect of Blank Samples on Absolute Odor Threshold Determinations. *Percept. Psychophys.* 6:142–44 (1969).
- 485. Stephens, E.R.:** Identification of Odors from Cattle Feed Lots. *Calif. Agric.* 25:10–11 (1971).
- 486. Stevens, J.C., W.S. Cain, and D.E. Weinstein:** Aging Impairs the Ability to Detect Gas Odor. *Fire Technol.* 23:198–204 (1987).
- 487. Stevens, J.C. and W.S. Cain:** Old-age Deficits in the Sense of Smell as Gauged by Thresholds, Magnitude Matching, and Odor Identification. *Psychol. Aging* 2:36–42 (1987).
- 488. Stevens, J.C. and W.S. Cain:** Detecting Gas Odor in Old Age. *Ann. N.Y. Acad. Sci.* 510:644–46 (1987).
- 489. Stevens, J.C., W.S. Cain and R.J. Burke:** Variability of Olfactory Thresholds. *Chem. Senses* 13:643–53 (1988).
- 490. Stevens, J.C. and A.D. Dadarwala:** Variability of Olfactory Threshold and its Role in Assessment of Aging. *Percept. Psychophys.* 54:296–302 (1993).
- 491. Stewart, R.D., E.D. Baretta, H.C. Dodd, and T.R. Torkelson:** Experimental Human Exposure to Vapor of Propylene Glycol Monomethyl Ether. *Arch. Environ. Health* 20:218–23 (1970).
- 492. Stewart, R.D., et al.:** Experimental Human Exposure to Propylene Glycol Dinitrate. *Toxicol. Appl. Pharmacol.* 30:377–95 (1974).

- 493. Stone, H., C.S. Ough, and R.M. Pangborn:** Determination of Odor Difference Thresholds. *J. Food Sci.* 27:197–202, (1962).
- 494. Stone, H.:** Techniques for Odor Measurement: Olfactometric vs. Sniffing. *J. Food Sci.* 28:719–25 (1963).
- 495. Stone, H.:** Determination of Odor Difference Limens for Three Compounds. *J. Exp. Psychol.* 66:466–73 (1963).
- 496. Stone, H.:** "Some Factors Affecting Olfactory Sensitivity and Odor Intensity." Thesis, University of California, 1963.
- 497. Stone, H. and J.J. Bosley:** Olfactory Discrimination and Weber's Law. *Percept. Mot. Skills* 20:657–65 (1965).
- 498. Stone, H., and G. Pryor:** Some Properties of the Olfactory System of Man. *Percept. Psychophys.* 2:516–18 (1967).
- 499. Stone, H., G.T. Pryor, and J. Colwell:** Olfactory Detection Thresholds in Man under Conditions of Rest and Exercise. *Percept. Psychophys.* 2:167–70 (1967).
- 500. Stone, H., G.T. Pryor, and G. Steinmetz:** A Comparison of Olfactory Adaptation among Seven Odorants and Their Relationship with Several Physiochemical Properties. *Percept. Psychophys.* 12:501–04 (1972).
- 501. Stone, H.:** "Personal Communication". Palo Alto, CA: Tragon Corporation, 1979.
- 502. Strube, A., A. Buettner, and M. Czerny:** Influence of Chemical Structure on Absolute Odour Thresholds and Odour Characteristics of ortho- and para-Halogenated Phenols and Cresols. *Flavour and Fragrance J.* 27:304–12 (2012).
- 503. Stuiver, M.:** "Biophysics of the Sense of Smell". Thesis, Groningen. 1958.
- 504. Styazhkin, V.M.:** Experimental Basis for the Determination of Allowable Concentrations of Chlorine and HCl Gas Simultaneously Present in Atmospheric Air. In *U.S.S.R. Literature on Air Pollution and Related Occupational Diseases*, translated by B.S. Levine. Vol. 8. Washington, D.C.: U.S. Department of Commerce, 1963. pp. 158–164.
- 505. Styazhkin, V.M. and H.K. Khachatryan:** Data for Validating the Maximum Permissible Concentration of Butyric Acid in Atmospheric Air. In *American Institute of Crop Ecology (AICE) Survey of USSR Air Pollution Literature*. Nuttinson, M.Y. (ed.). 19:104–10 (1973).
- 506. Suchier, A.:** Reagenspapier zum Nachweis Kleiner Mengen Phosgen. *Z. Anal. Chem.* 79:183–85 [German reviewed at TRC] (1929).
- 507. Sutton, W.L.:** Aliphatic and Alicyclic Amines. In *Industrial Hygiene and Toxicology Volume II. Toxicology*. Patty, F.A. (ed.). New York/London: Interscience. pp. 2037–2067, 1962.

- 508. Sutton, W.L.:** Heterocyclic and Miscellaneous Nitrogen Compounds. In *Industrial Hygiene and Toxicology Volume II. Toxicology*. Patty, F.A. (ed.). New York/London: Interscience, 1962, pp. 2189–2191.
- 509. Tabakova, S.:** Experimental Data for Hygienic Standards for Dipterex in Air. *Hyg. Sanit. USSR* 34:187–93 [English translation of Russian article] (1969).
- 510. Takhiroff, M.T.:** Maximum Allowable concentration of Chlorine in the Atmosphere Based on Experimental Data. *Gig. Sanit.* 22:13–18 [English summary of Russian article] (1957).
- 511. Takhirov, M.T.:** Hygienic Standards for Acetic and Acetic Anhydride in Air. *Hyg. Sanit. USSR* 34:122–25 [English translation of Russian article] (1969).
- 512. Takhirov, M.T.:** Experimental Study of the Combined Action of Six Atmospheric Pollutants on the Human Organism. *Gig. Sanit.* 39:100–02 [Russian] (1974).
- 513. Tamman, G. and W. Oelsen:** Zur Bestimmung der Dampfdrucke von Riechstoffen. *Z. Anorg. Allg. Chem.* 172:407–13 [German reviewed at TRC] (1928).
- 514. Tarkhova, L.P.:** Materials for Determining the Maximum Permissible Concentration of Chlorobenzol in Atmospheric Air. *Hyg. Sanit. USSR* 30:327–33 [English translation of Russian article] (1965).
- 515. Taylor, E.F. and F.T. Bodurtha:** Control of Dimethylamine Odors. *Ind. Wastes* 5:92–94 (1960).
- 516. Tempelaar, H.C.G.:** "Over den Invloed van Licht op Reukstoffen". Thesis, Utrecht, 1913.
- 517. Tepikina, L.A.:** Biological Effects of Mesidine as an Atmospheric Pollutant. *Hyg. Sanit. USSR* 33:299–302 [English translation of Russian article] (1968).
- 518. Teranishi, R., R.G. Buttery, and D.G. Guadagni:** Odor Quality and Chemical Structure in Fruit and Vegetable Flavors. *Ann. N. Y. Acad. Sci.* 237:209–16 (1974).
- 519. Thiele, V.:** Experimentelle Untersuchungen zur Ermittlung eines Geruchsschwellenwertes für Schwefelwasserstoff. *Staub.* 39:159–60 [German reviewed at TRC] (1979).
- 520. Thiele, V., J. Kastka, G. Winneke and B. Prinz:** Zur Problem der Geruchsmessung – vergleichende Untersuchungen zwischen MIU and LIS. *Staub.* 41:289–95 (1981).
- 521. Thiele, V.:** Olfaktometrie von H<sub>2</sub>S-Ergebnisse des VDI-Ringvergleiches. *Staub* 42:11–15 (1982).
- 522. Thiele, V.:** Olfaktometrie an einer Emissionsquelle – Ergebnisse des VDI-Ringvergleiches. *Staub.* 44:342–51 (1984).

- 523. Thomas, M.D., J.O. Ivie, J.N. Abersold, and R.H. Hendricks:** Automatic Apparatus for Determination of Small Concentrations of Sulfur Dioxide in Air. *Ind. Eng. Chem. Anal. Ed.* 15:287–90 (1943).
- 524. Thrirel, C., et al.:** From Chemosensory Thresholds to Whole Body Exposures Experimental Approaches Evaluating Chemosensory Effects of Chemicals. *Int. Arch. Occup. Environ. Health* 79:308–21 (2006).
- 525. Tkach, N.Z.:** Combined Effect of Acetone and Acetophenone in the Atmosphere. *Hyg. Sanit. USSR* 30:179–85 [English translation of Russian article] (1965).
- 526. Tkachev, P.G.:** Data for Substantiating the Maximum Permissible Concentration of Aniline in the Atmospheric Air. *Gig. Sanit.* 28:3–11 [English summary of Russian article] (1963).
- 527. Tkachev, P.G.:** Monoethylamine in the Atmosphere; Hygienic Significance and Standards. *Hyg. Sanit. USSR* 34:149–53 [English translation of Russian article] (1969).
- 528. Tkachev, P.G.:** Triethylamine in Air: Hygienic Significance and Hygienic Standards. *Hyg. Sanit. USSR* 35:8–13 [English translation of Russian article] (1970).
- 529. Tkachev, P.G.:** Methods for Determining Threshold of Olfactory Sensation for Chemical Substances Having an Odor. *Gig. Sanit.* 3:112–13 [Russian] (1978).
- 530. Torkelson, T.R., M.A. Wolf, F. Oyen, and V.K. Rowe:** Vapor Toxicity of Allyl Chloride as Determined in Laboratory Animals. *Am. Ind. Hyg. Assoc. J.* 20:217–23 (1959).
- 531. Torkelson, T. R., S.E. Sadek, and V.K. Rowe:** The Toxicity of Boron Trifluoride when Inhaled by Laboratory Animals. *Am. Ind. Hyg. Assoc. J.* 22:263–70 (1961).
- 532. Torkelson, T.R., and F. Oyen:** The Toxicity of 1,3-Dichloropropane as Determined by Repeated Exposure of Laboratory Animals, *Am. Ind. Hyg. Assoc. J.* 38:217–23 (1977).
- 533. Torkelson, T.R. and V.K. Rowe:** Halogenated Aliphatic Hydrocarbons Containing Chlorine, Bromine and Iodine. In *Patty's Industrial Hygiene and Toxicology, Volume 2B, Toxicology*, 3rd rev. edition. Clayton, G.D. and F.E. Clayton (eds.). New York: John Wiley & Sons, 1981, pp. 3433–3601.
- 534. Toxicity Data Sheet: Mesityl Oxide,** Shell Chemical Corporation Industrial Hygiene Bulletin, SC, 57-106. New York: Shell Chemical Corporation, 1957.
- 535. Toxicity Data Sheet: Allyl Chloride,** Shell Chemical Corporation Industrial Hygiene Bulletin, SC, 57-80. New York: Shell Chemical Corporation, 1958a.
- 536. Toxicity Data Sheet: Ethyl Amyl Ketone,** Shell Chemical Corporation Industrial Hygiene Bulletin, SC, 57-99. New York: Shell Chemical Corporation, 1958b.

- 537. Toxicity Data Sheet: Epichlorohydrin**, Shell Chemical Corporation Industrial Hygiene Bulletin, SC, 57-86. New York: Shell Chemical Corporation, 1959.
- 538. Treon, J.F. and F.R. Dutra:** Physiological Response of Experimental Animals to the Vapor of 2-Nitropropane. *AMA Arch. Ind. Hyg. Occup. Med.* 5:52-61 (1952).
- 539. Treon, J.F., F.P. Cleveland, and J. Cappel:** The Toxicity of Hexachlorocyclopentadiene. *AMA Arch. Ind. Health* 11:459-72 (1955).
- 540. Tsukatani, T., T. Miwa, M. Furukawa, and R.M. Costanzo:** Detection Thresholds for Phenyl Ethyl Alcohol Using Serial Dilutions in Different Solvents. *Chem. Senses* 28:25-32 (2003).
- 541. Tsulaya, V.R., L.I. Zefirova and E.A. Pereverzeva:** Data for Hygienic Standardization of Izoctanol in the Atmosphere [(Russian), *Gig. Sanit.* 8:16-19 (1972)].
- 542. Turk, A.:** Expressions of Gaseous Concentration and Dilution Ratios. *Atmos. Environ.* 7:967-72 (1973).
- 543. Ubaidullaev, R.:** Data for Substantiating the Maximum Permissible Concentration of Fürförol in Atmospheric Air. *Gig. Sanit.* 26:3-10 [English summary of Russian article] (1961).
- 544. Ubaidullaev, R.:** Effect of Small Concentration of Methanol Vapours on Man and Animals. *Hyg. Sanit., USSR* 31:8-12 [English translation of Russian article] (1966).
- 545. Ubaidullaev, R.:** The Toxicity of Small Concentrations of Hydrolysed Alcohol in the Atmosphere. [Russian], *Gig. Sanit.* 31(5):3-7, (1966).
- 546. Ubaidullaev, R.U. and U.A. Madzhidov:** Investigation Data on Reflex and Resorptive Action of Small Concentrations and Doses of Kilwal. *Gig. Sanit.* 3:11-14 [Russian] (1978).
- 547. Ueno, H., S. Amano, B. Merecka and J. Kosmider:** Difference in the Odor Concentrations Measured by the Triangle Odor Bag Method and Dynamic Olfactometry. *Water Sc. Technol.* 59:1339-42 (2009).
- 548. Valentin, G.:** *Lehrbuch der Physiologie des Menschen*. Braunschweig: Friedrich Bieweg und Sohn. 2.2, 1848. pp. 279-283.
- 549. Valentin, G.:** *Lehrbuch der Physiologie des Menschen*. Braunschweig: Friedrich Bieweg und Sohn, 2.3, 1850. pp. 271-274.
- 550. Ventura, F., J. Romero & J. Pares:** Determination of Dicyclopentadiene and Its Derivatives as Compounds Causing Odors in Groundwater Supplies. *Environ. Sci. Technol.* 31:2368-74 (1997).
- 551. Ventura, F., J. Romero and J. Pares:** Determination of Dicyclopentadiene and its Derivatives as Compounds Causing Odors in Groundwater Supplies. *Environ. Sci. Technol.* 31:2368-74 (2001).

- 552. Vermeulen, C. and S. Collin:** Combinatorial Synthesis and Screening of Novel Odorants Such as Polyfunctional Thiols. *Combinatorial Chemistry & High Throughput Screening*, 9:583–90 (2006).
- 553. Vincent, G.P., J.O. MacMahon, and J.E. Synan:** The Use of Chlorine Dioxide in Water Treatment. *Am. J. Public Health* 36:1035–37 (1946).
- 554. Viswanathan, S., G.P. Mathur, A.W. Gnyp and C.C. St. Pierre:** Application Probability Models to Odor Threshold Determination. *Atmos. Environ.* 17:139–43 (1983).
- 555. Walker, J.C., J.H. Reynolds, D.W. Warren and J.D. Sidman:** Responses of Normal and Anosmic Subjects to Odorants, In *Chemical Senses Vol. 2, Irritation*. Green, B.G., J.R. Masson, and M.R. Kare (eds.). New York: Marcel Dekker, Inc., 1990, 95-121.
- 556. Walker, J.C., M. Kendal-Reed, C.J. Keiger, M. Bencherif and W.L. Silver:** Olfactory and Trigeminal Responses to Nicotine. *Drug Dev. Res.* 38:160–68 (1996).
- 557. Walker, J.C., S.B. Hall, D.B. Walker, M.S. Kendal-Reed, A.F. Hood and X.F. Niu:** Human Odor Detectability: New Methodology Used to Determine Threshold and Variation. *Chem. Senses* 28:817–26 (2003).
- 558. Weeks, M.H.:** Monoethanolamine. In *Transactions of the Symposium on Health Hazards of Military Chemicals*, (Chemical Warfare Laboratories Special Publication 2-10), 1958. pp. 15-23.
- 559. Weeks, M.H., T.O. Downing, N.P. Musselman, T.R. Carson, and W.A. Groff:** The Effects of Continuous Exposure of Animals to Ethanolamine Vapor, *Am. Ind. Hyg. Assoc. J.* 21:374–81 (1960).
- 560. Weitbrecht, U.:** Beurteilung der Trichloroäthylen-Gefährdung im Betrieb. *Zentralbl. Arbeitsmed. Arbeitsschutz.* 7:55–58 [German reviewed at TRC] (1957).
- 561. Whisman, M.L., J.W. Goetzinger, F.O. Cotton, and D.W. Brinkman:** Odorant Evaluation: A Study of Ethanethiol and Tetrahydrothiophene as Warning Agents in Propane. *Environ. Sci. Technol.* 12:1285–88 (1978).
- 562. Widder, S., T. Dittner, C. Sabater-Lüntzel, W. Pickenhagen and J. Vollhardt:** 3-Mercapto-2-alkyl-alkan-1-ole und ihre Anwendungen als Duft- und Aromastoffe. *Europäische Patentschrift* 924198, (2002).
- 563. Wijk, R.A. De:** "Temporal Factors in Human Olfactory Perception". Thesis, Utrecht, 1989.
- 564. Wilby, E.V.:** The Odor Comparator; an Improved Instrument for Quantitative Odor Measurement. *Proc. Oper. Sect. Am. Gas Assoc.* 64:225–31 (1964).
- 565. Wilby, E.V.:** Variation in Recognition Odor Threshold of a Panel. *J. Air Pollut. Control Assoc.* 19:96–100 (1969).



- 566. Williams, A.A., A.G.H. Lea, and C.F. Timberlake:** Measurement of Flavor Quality in Apples, Apple Juices and Fermented Ciders. In *Flavor Quality: Objective Measurement* edited by R.A. Sanlan, ACS Symposium Series no. 51. Washington, D.C.: American Chemical Society, 1977. pp. 71–88.
- 567. Williams, F.D., J.E. Emele, and M.C. Alford:** The Application of the Dynamic Triangle Olfactometer to the Evaluation of Oral Odor. *Chem. Senses Flavour* 2:497–502 (1977).
- 568. Wilska, S.:** Ozone. Its Physiological Effects and Analytical Determination in Laboratory Air. *Acta Chem. Scand.* 5:1359–67 (1951).
- 569. Winkler, K.:** Zur Diskussion Gestellt: Immissionsgrenzwerte zur Verhinderung von Geruchsbelastigungen. *Wasser, Luft Betr.* 19:411 (1975).
- 570. Winneke, G. and J. Kastka:** Wirkung von Geruchsstoffen auf den Menschen. In *Geruchsprobleme bei Tierhaltung und Tierkörperbeseitigung*. Dusseldorf: VDI-Berichte 226, VDIVerlag GmbH, 1975, pp. 11–24.
- 571. Winneke, G., J. Kotalik, H. O. Keldenich, and J. Kastka:** Zur Wahrnehmung von Schwefelwasserstoff unter Laborund Feldbedingungen. *Staub* 39:156–59 [German reviewed at TRC] (1979).
- 572. Winneke, G., H.W. Berresheim, J. Kotalik and A. Kabat:** Vergleichende Olfactometrische Untersuchungen zu Formaldehyd und Schwefelwasserstoff. *Staub* 48:319–23 (1988).
- 573. Wirth, W. and O. Klimmer:** Zur Toxikologie der Organischen Lösungsmittel, 1–4-Dioxan (Diathylendioxyd). *Arch. Gewerbepathol. Gewerbehyg.* 7:192–206 [German reviewed at TRC], (1937).
- 574. Wise, P.M., T. Miyazawa, M. Gallagher and G. Petri:** Human Odor Detection of Homologous Carboxylic Acids and Their Binary Mixtures. *Chem. Senses* 32:475–82 (2007).
- 575. Witheridge, W.N. and C.P. Yaglou:** Ozone in Ventilation – Its Possibilities and Limitations. *Heating, Piping & Air Condo* 2:648–53 (1939).
- 576. Wolf, M.A., V.K. Rowe, D.D. McCollister, R.I. Hollingsworth, and E Oyen:** Toxicological Studies of Certain Alkylated Benzenes and Benzene. *AMA Arch. Ind. Health* 14:387–98 (1956).
- 577. Woolrich, P.E:** Toxicology, Industrial Hygiene and Medical Control of TDI, MDI and PMPPI. *Am. Ind. Hyg. Assoc. J.* 43:89–97 (1982).
- 578. Wünsche, L.F., C. Vuilleumier, U. Keller, M.P. Byfield, I.P. May and M.J. Kearney:** Scent Characterisation: from Human Perception to Electronic Noses. In *Proc. 13th Int. Congress of Flavours, Fragrances and Essential Oils* edited by K.H.C. Baser, Istanbul, Turkey. October 15-19, 1995. 3:295–313, (1995).

- 579. Wysocki, C.J., P. Dalton, M.J. Brody and H.J. Lawley:** Acetone Odor and Irritation Thresholds Obtained from Acetone-Exposed Factory Workers and from Control (occupationally unexposed) Subjects. *Am. Ind. Hyg. Assoc. J.* 58:704–12 (1997).
- 580. Yang, D.S., R.L. Shewfelt, K.S. Lee, and S.J. Kays:** Comparison of Odor Active Compounds from Six Distinctly Different Rice Flavor Types. *J. Agric. Food Chem.* 56:2780–87 (2008).
- 581. Young, F.A. and D.F. Adams:** "Comparison of Olfactory Thresholds Obtained on Trained and Untrained Subjects." *Proc. 74th Annual Conv. Am. Psychol. Assoc.*, 1966. pp. 75–76.
- 582. Yuldashev, T.:** Maximum Permissible Concentration of Ethylene Oxide in the Atmosphere. *Hyg. Sanit. USSR* 30:1–6 [English translation of Russian article] (1965).
- 583. Zapp, J.A.:** Hazards of Isocyanates in Polyurethane Foam Plastic Production. *Arch. Ind. Health* 15:324–30 (1957).
- 584. Zarzo, M.:** Effect of Functional Group and Carbon Length on the Odor Detection Threshold of Aliphatic Compounds. *Sensors* 12:4105–112 (2012).
- 585. Zibireva, I.A.:** Relative Toxicity of Microconcentrations of Two Chlorisocyanates (para- and meta-isomers). *Hyg. Sanit. USSR* 32:3–9 [English translation of Russian article] (1967).
- 586. Ziemer, P.D., J. Woo, and T. Anagnostou:** Study of Odor Qualification of Solvents Used in Coating Compositions. *Journal of Coating Technology* 72:97–102 (2000).
- 587. Zwaardemaker, H.:** Geruch und Geschmack. In *Handbuch der Physiologischen Methodik III*, 1. Tigerstedt, R. Leipzig and S. Hirzel (eds.). [Corrected by Tempelaar (1913)], 1914, pp. 46–108.
- 588. Zwaardemaker, H. and K. Komuro:** Ruiken bij Volledige Vennoeienis, Resp. Adaptatie Voor een Bepaalde Geur. Kon. Ned. Akad. Wetensch. Versl. Gewone Vergad. Afd. *Natuurk.* 30:1189–95 (1921).
- 589. Zwaardemaker, H.:** The Sense of Smell. *Acto-Laryngol* 11:13–15 (1927).